

Electromagnetic separators

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Chartreuse mountains

Ecrins mountains

Vercors mountains

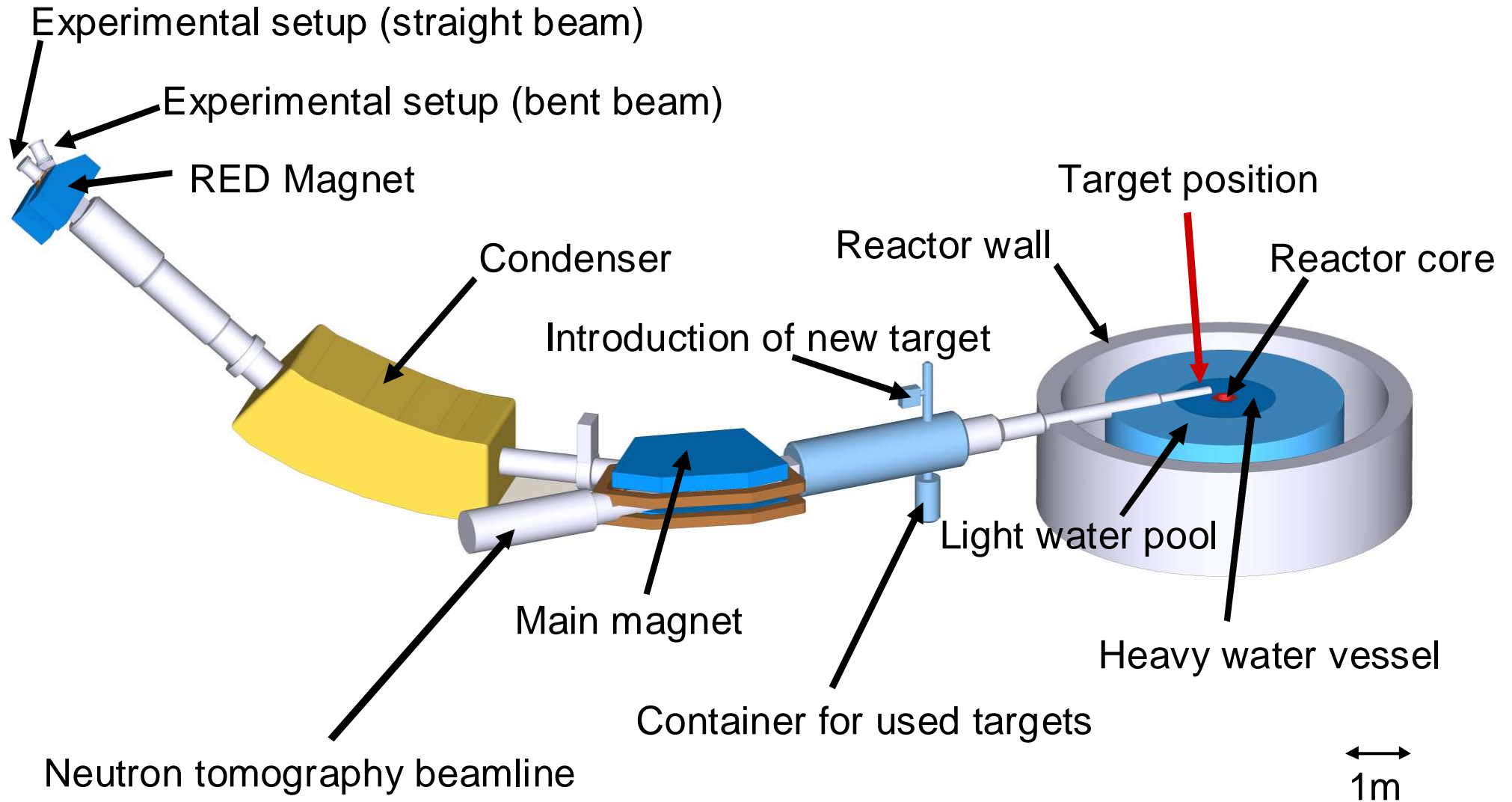
Belledonne mountains

Taillefer mountains



- founded 1967
- today 13 member states
- operates most powerful neutron source of the world:
58 MW high flux reactor, $1.5 \cdot 10^{15}$ n./cm²/s maximum neutron flux
- over 40 instruments, mainly for neutron scattering
- user facility: 2000 scientific visitors from 45 countries per year
- Director General: Richard Wagner
- Nuclear physics instruments: LOHENGRIN, GAMS

LOHENGRIN Setup



Outline

- 1. Definitions and history**
- 2. Basics of ion optics and dispersive elements**
- 3. Static fields**
 - a) deflection spectrometer
 - b) retardation spectrometer
- 4. Dynamic fields/separation**
 - a) Time-of-Flight spectrometer
 - b) Radiofrequency spectrometer
 - c) Traps
- 5. Technical realization (ion sources, etc.)**
- 6. “Real examples” for nuclear physics applications**
 - a) ISOL
 - b) Recoil separators
 - c) Fragment separators
 - d) Spectrometer

Definitions

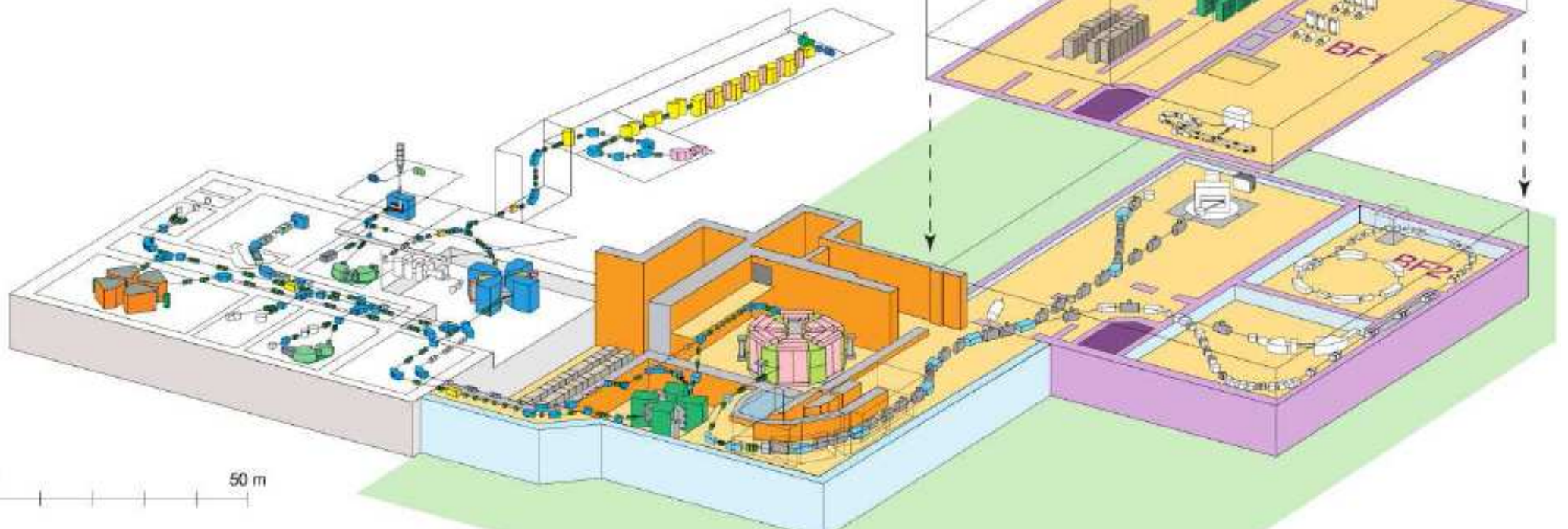
- **spectrometer**: electrical detection
- **spectrograph**: photographic or other non-electrical detection
- **also used**: spectroscope

- **mass / energy / isotope separator**: assures a physical separation of different masses / energies / isotopes

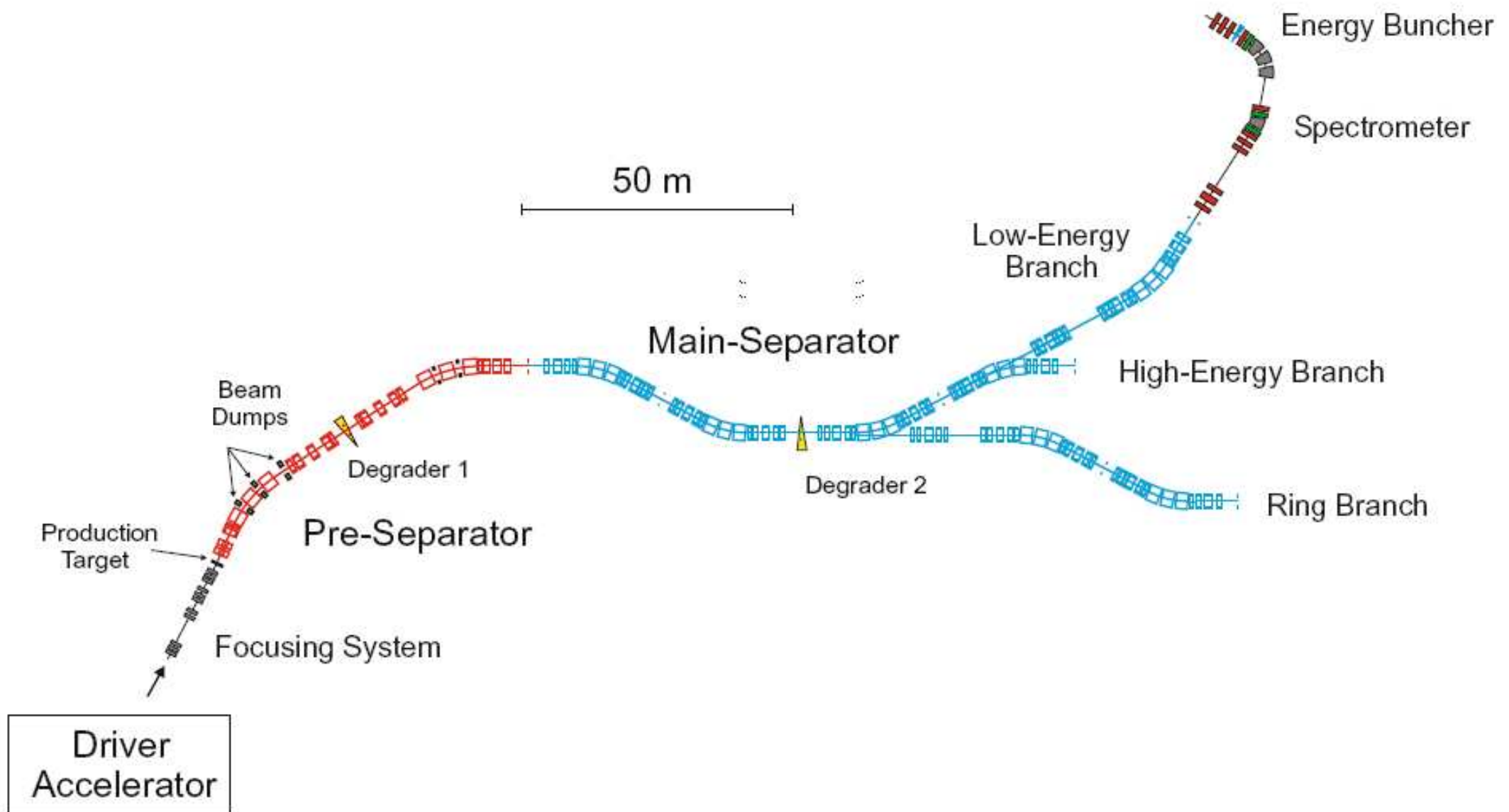
BigRIPS at RIBF Facility, RIKEN, Japan



- $K = 2500$ MeV
- 8300 tons
- 5.36 m extraction radius
- 6 sector magnets
- four main RF cavities



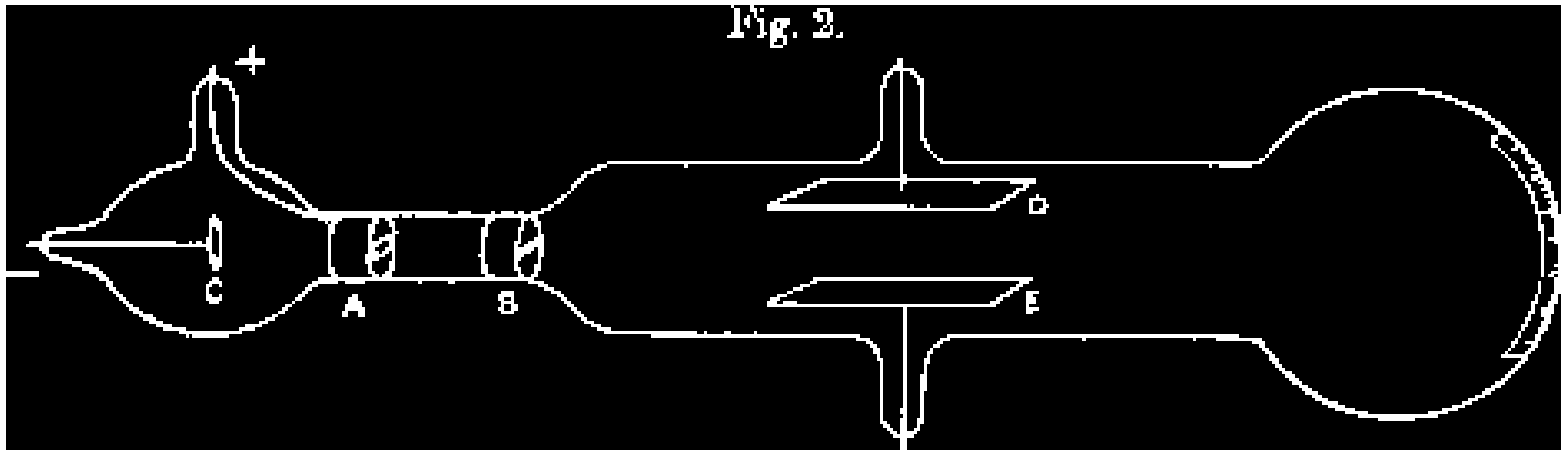
Super-FRS at FAIR, Germany



Importance of electromagnetic spectrometers



Thomson 1897: cathode rays



“Cathode rays”, J.J. Thomson, *Phil. Mag.* 44 (1897) 293.

Noble prize in physics 1906 for discovery of the electron and the determination of its m/q ratio.

Goldstein 1886: *Kanalstrahlen*

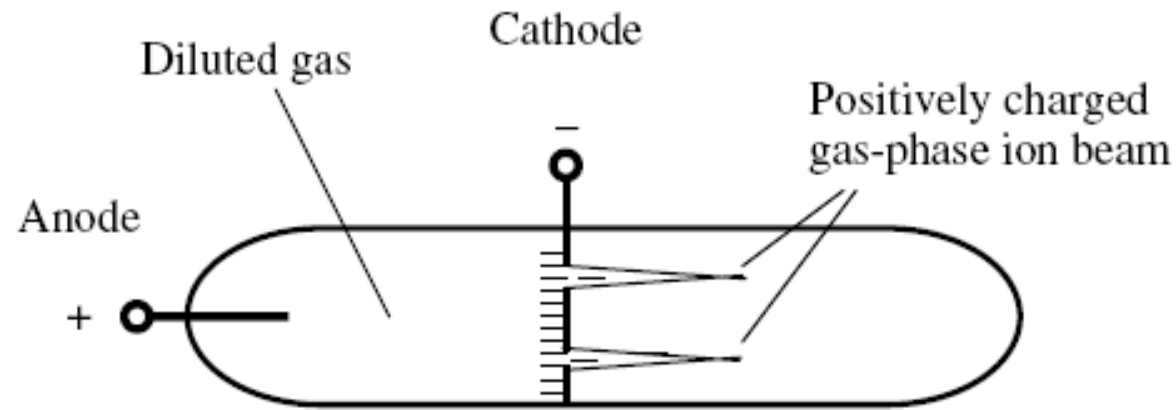
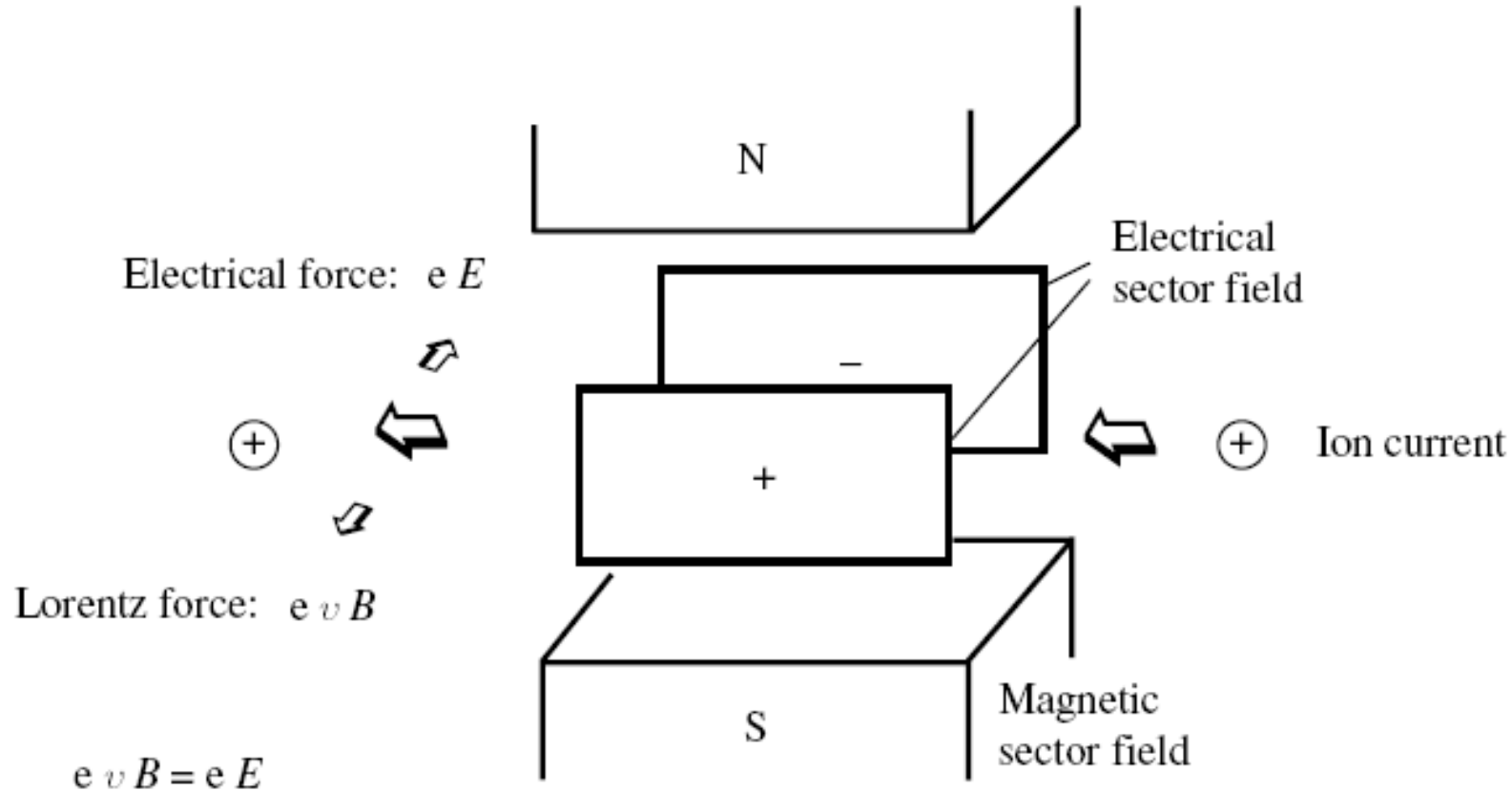


Figure 1.3 Goldstein's glow discharge tube (1886) for generation of positively charged ions. (C. Brunnée, *Int. J. Mass. Spectrom. Ion Proc.* 76, 125 (1987). Reproduced by permission of Elsevier.)

First fluorescent lamp and ion source.

Wien 1902: Wien filter



Electric field perpendicular to magnetic field

Figure 1.4 Schematic of a Wien velocity filter with EB configuration: combination of electric (E) and magnetic (B) field (Wien, 1898). (C. Brunnée, Int. J. Mass. Spectrom. Ion Proc. 76, 125 (1987). Reproduced by permission of Elsevier.)

Wien: Nobel price in physics 1911 for discovery that “Kanalstrahlen” carry positive charge

Thomson 1910: parabola mass spectrograph

Electric field parallel to magnetic field

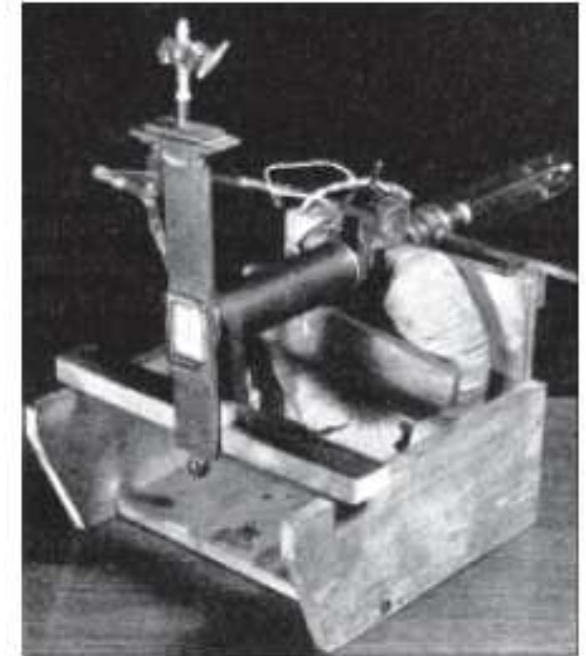
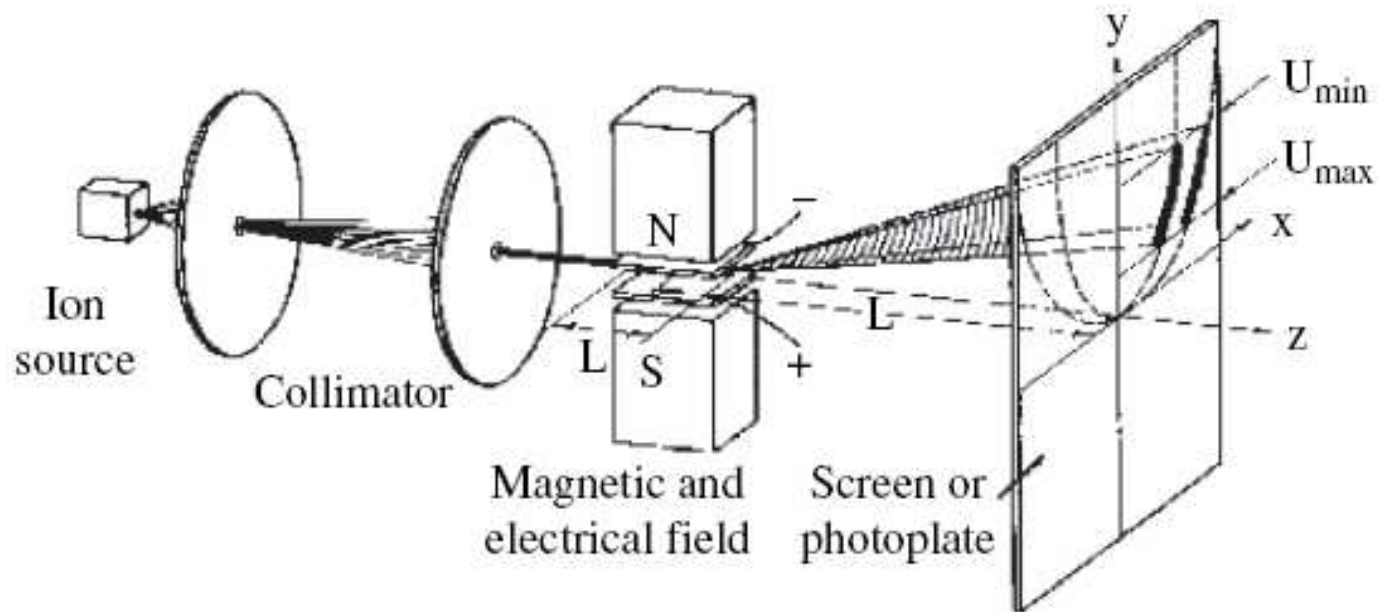


Figure 1.5 Parabola mass spectrograph constructed by J.J. Thomson (1910) with a discharge tube as ion source, a superimposed electrical field and a magnetic field oriented parallel to it for ion separation, and a photoplate for ion detection. (H. Kienitz (ed.), *Massenspektrometrie* (1968), Verlag Chemie, Weinheim. Reproduced by permission of Wiley-VCH.)

Neon consists of two isotopes with mass 20 and 22

Thomson 1913: mass spectrum of neon

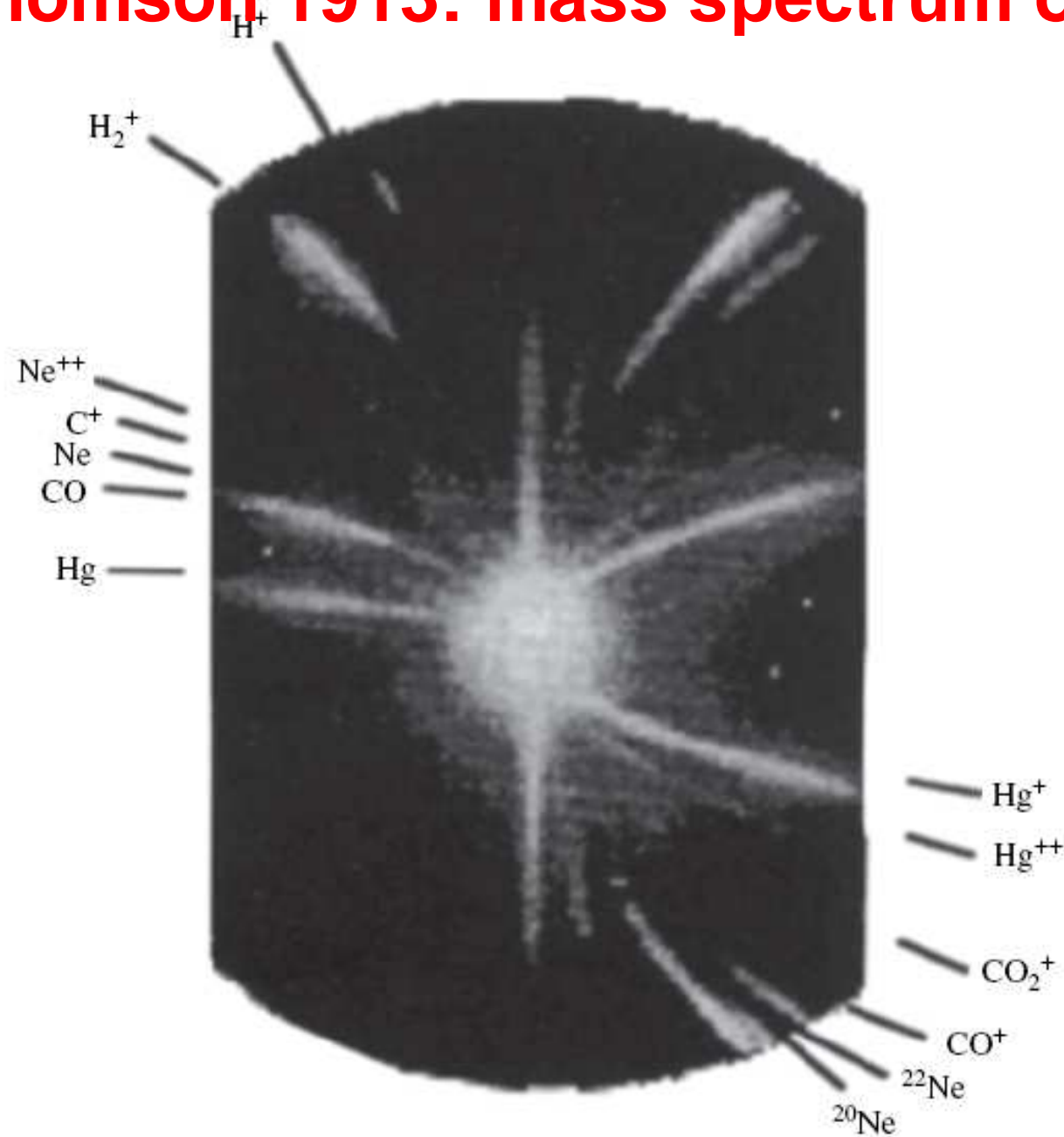
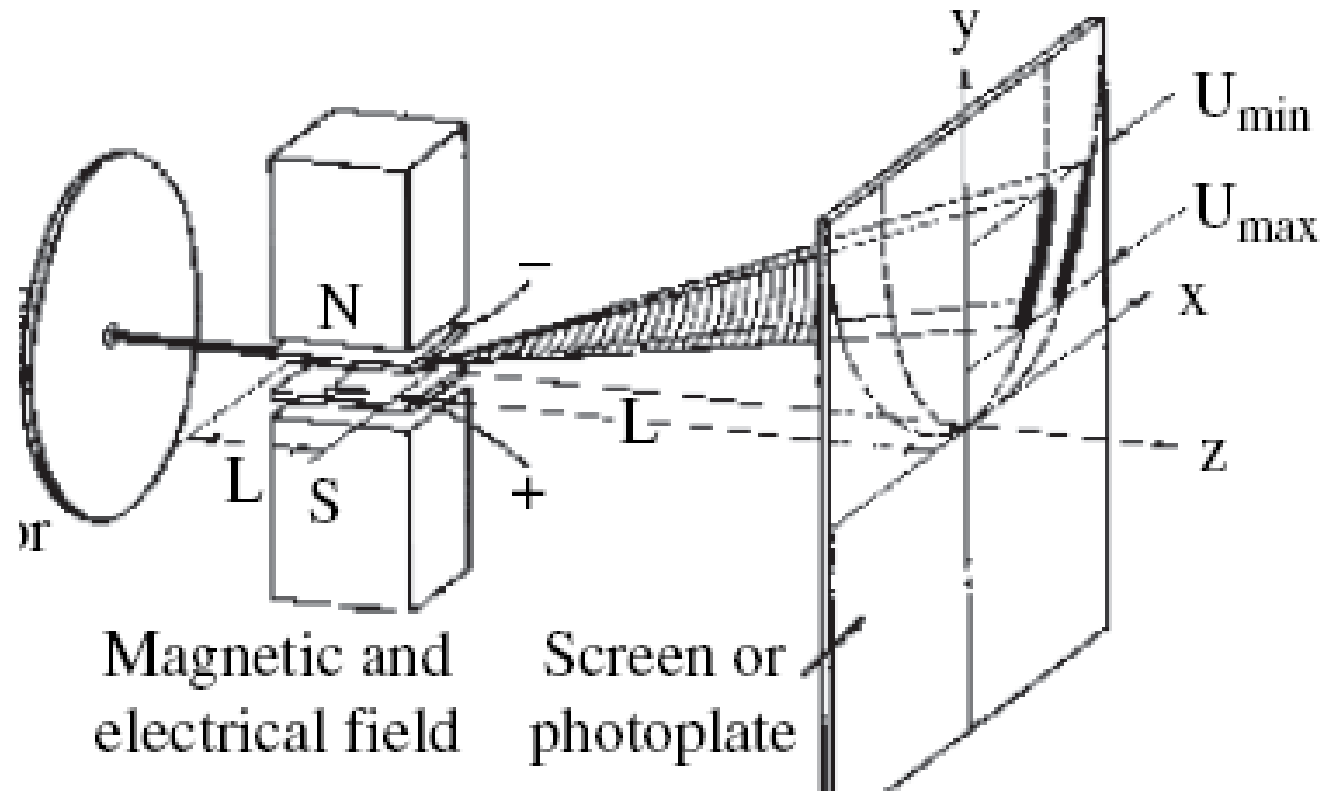


Figure 1.6 Mass spectrum of neon with masses 20 and 22 u measured by J.J. Thomson (1913) using his parabola mass spectrograph is shown in Figure 1.5. (H. Kienitz (ed.), *Massenspektrometrie* (1968), Verlag Chemie, Weinheim. Reproduced by permission of Wiley-VCH.)

Parabola spectrograph



transit time through field:

$$t = L/v$$

vertical displacement:

$$y = \frac{1}{2} U/d \ q/m \ (L/v)^2$$

horizontal displacement:

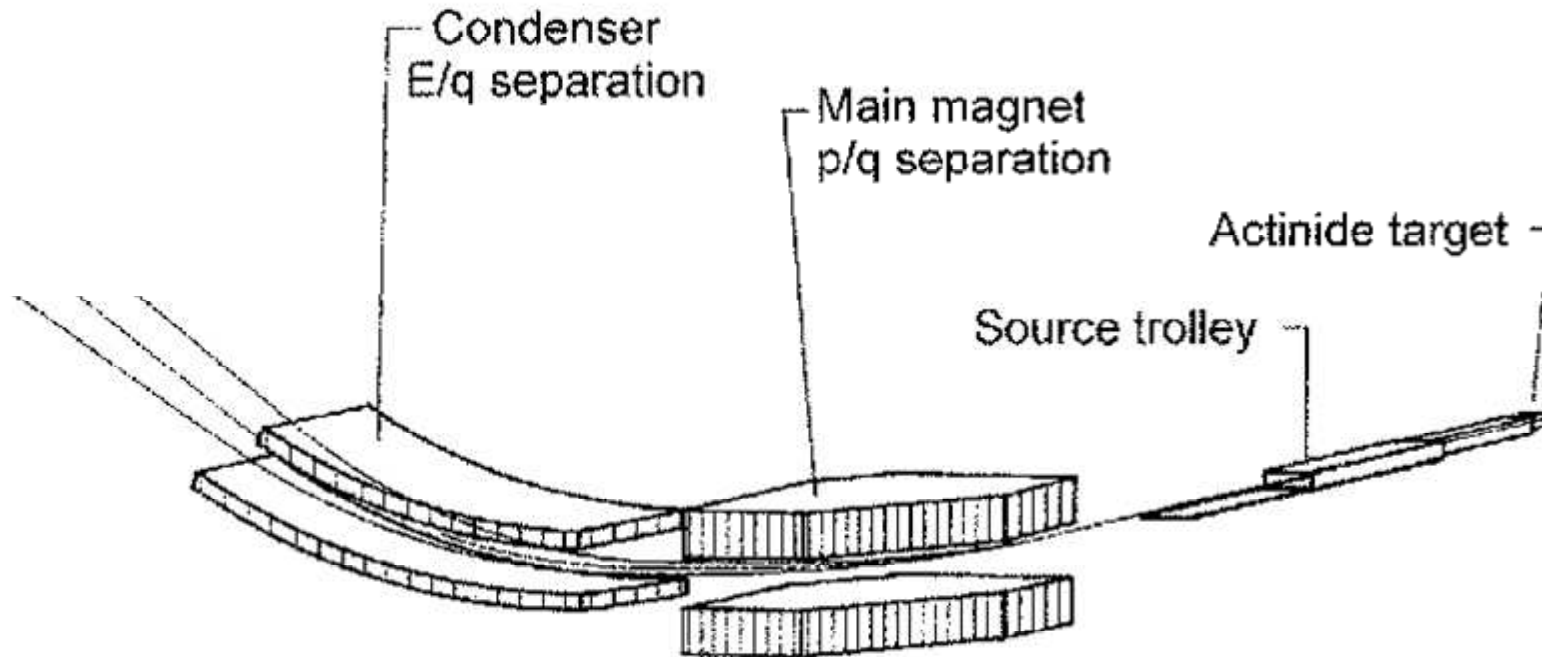
$$x = \frac{1}{2} B \ q/m \ L^2/v$$

$$y = k \ m/q \ x^2;$$

$$k = 2 \ U/(d \ B^2 \ L^2)$$

The LOHENGRIN fission fragment separator

Angular focusing in x and y direction.



$$m v^2 / r_{el} = q E$$

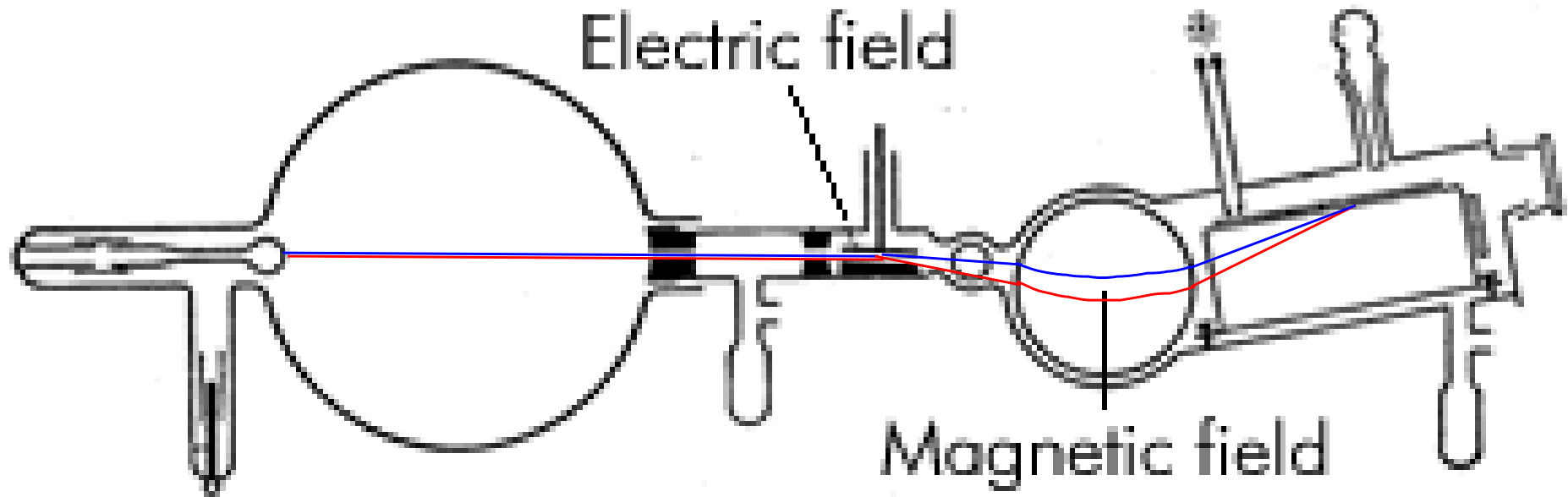
$$E_{kin} / q = E / 2 r_{el}$$

$$m v^2 / r_{magn} = q v B$$

$$m v / q = B r_{magn}$$

Animation!

Aston 1919: velocity focusing spectrograph



Aston's design for the mass spectrograph.

Aston: **velocity focusing** gives factor 10 improvement in mass resolution ($\Delta m/m = 1/130$)

Noble prize in chemistry 1922 for the discovery that elements may have isotopes of different mass (^{20}Ne , ^{21}Ne and ^{22}Ne).

Dempster 1918: 180 degree spectrometer

- electron bombardment ion source for **monoenergetic** ions
- 180 degree magnetic field provides **angular focusing**
- **scan of magnetic field** to measure mass spectra

1920: discovery of isotopes in Mg, Li, K, Ca, Zn

$$q/m = 2 U / (B^2 r^2)$$

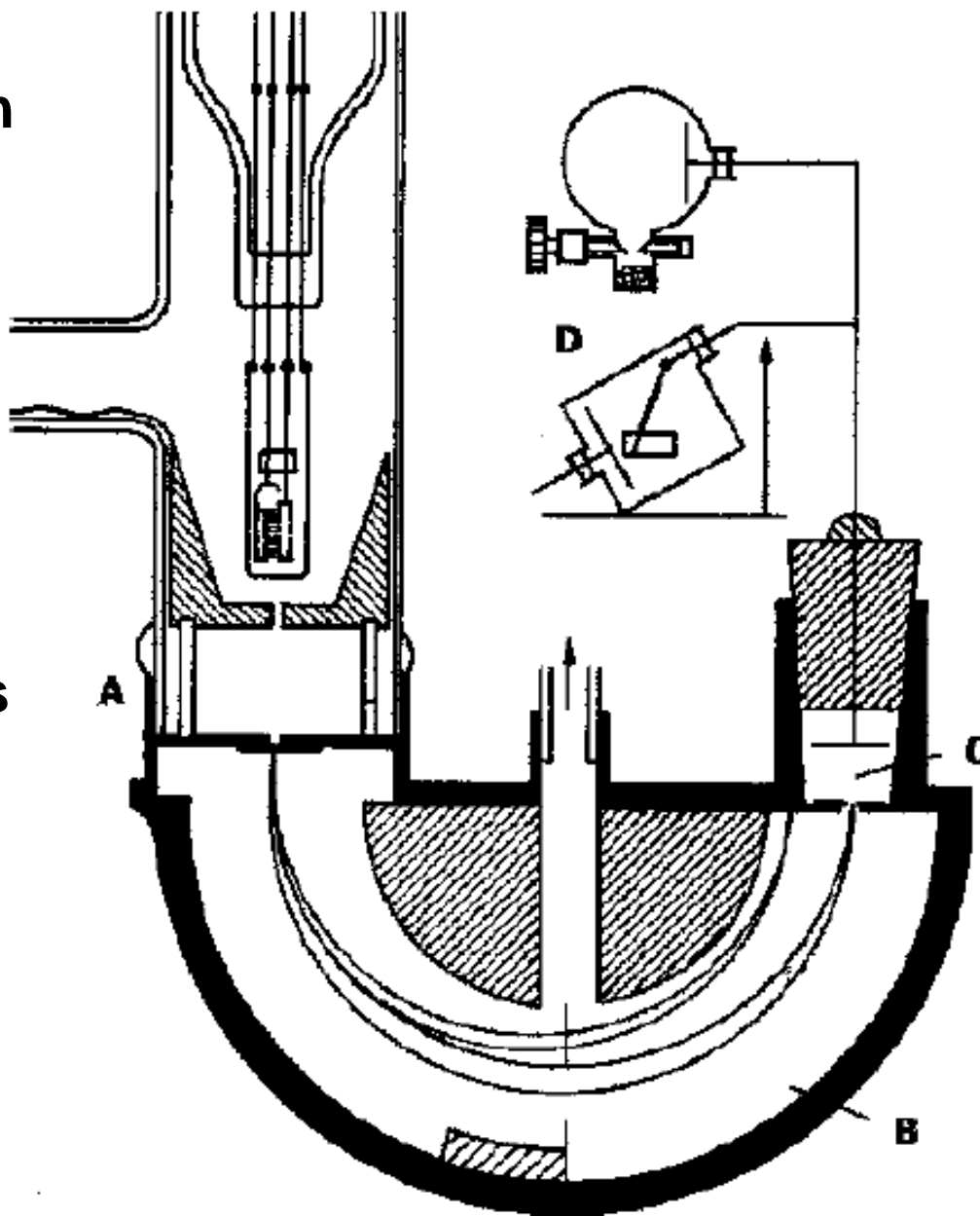
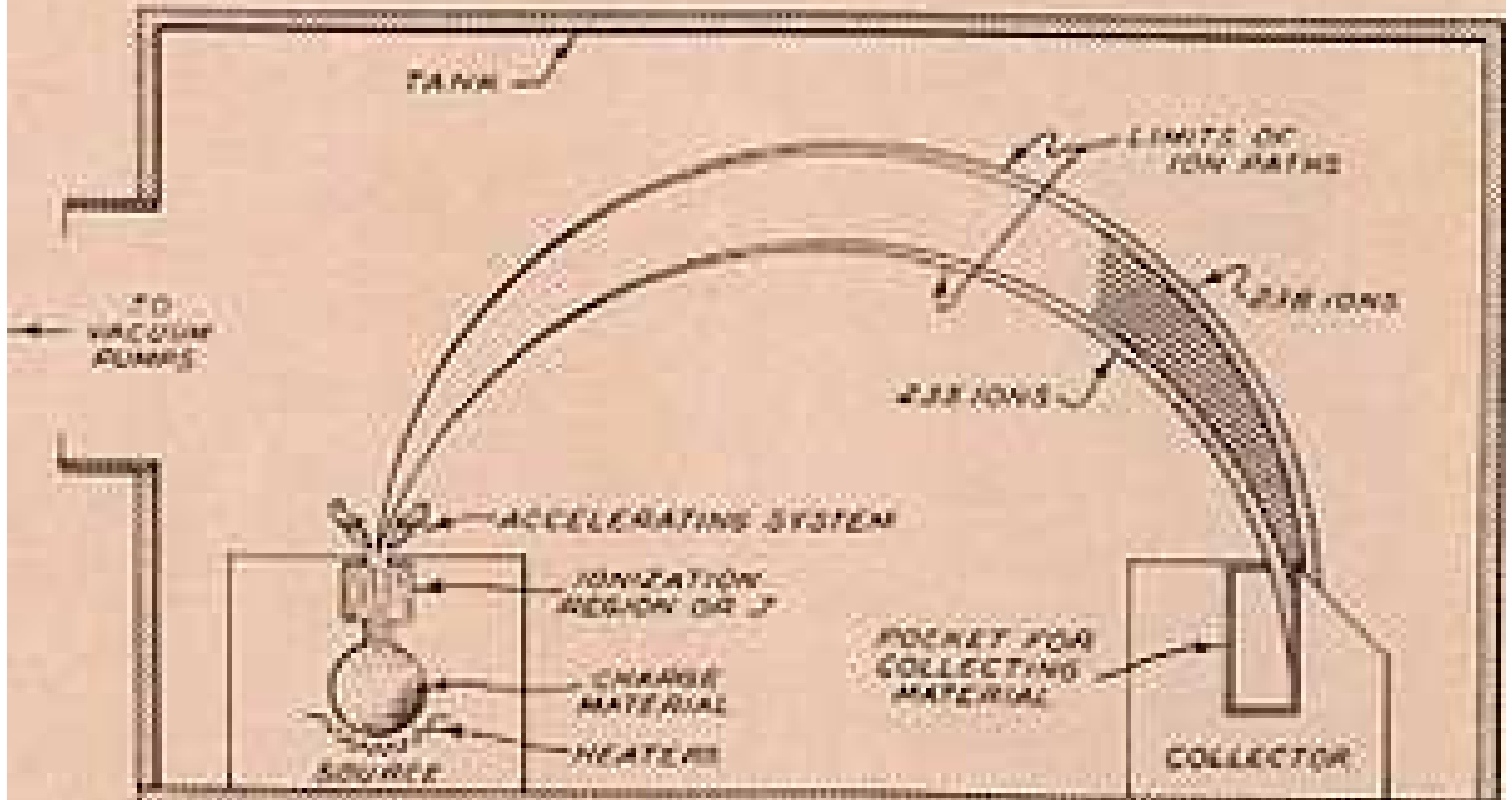


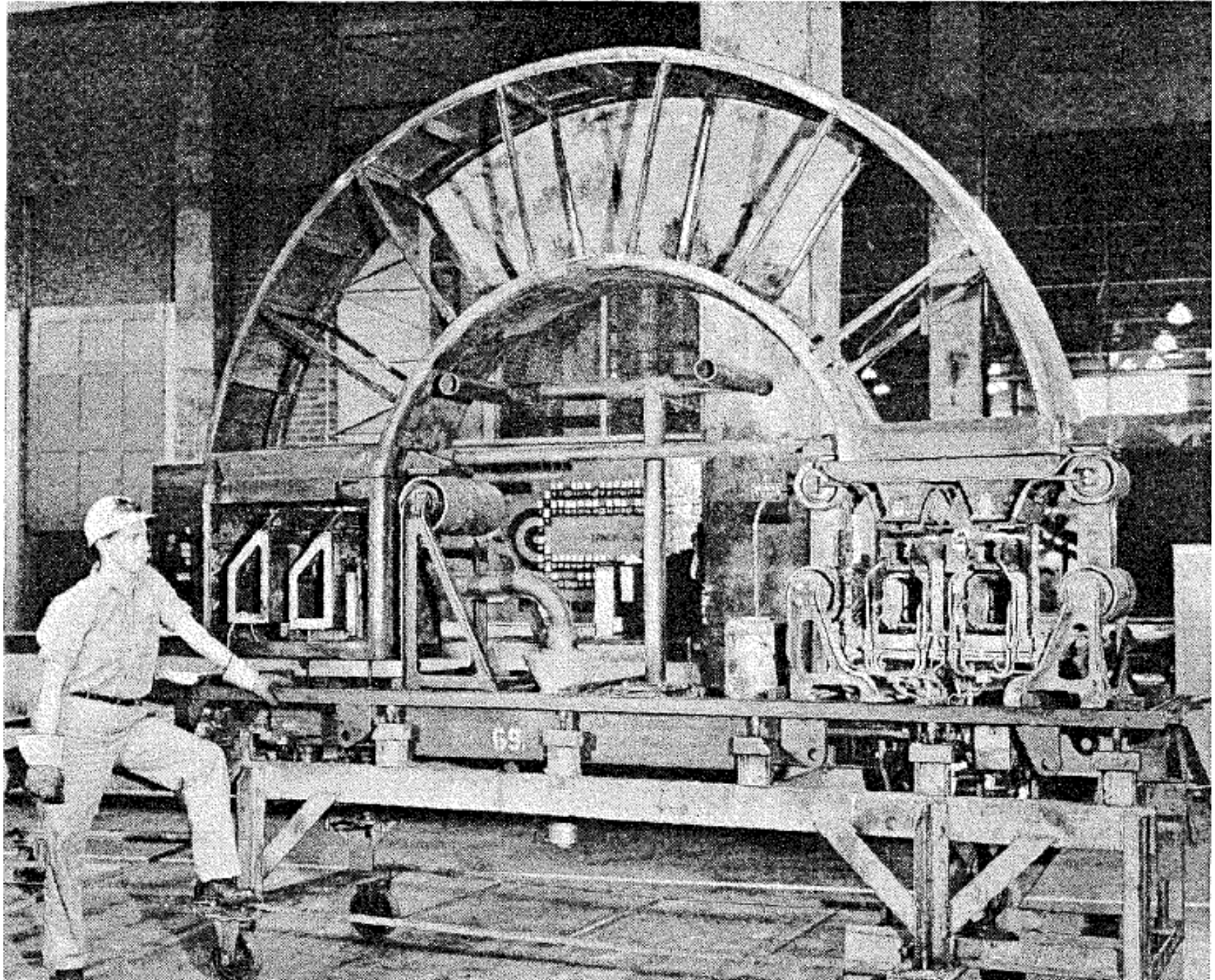
Figure 1.7 Mass spectrometer from A.J. Dempster (1918). A – ion source; B – electromagnet; C – Faraday cup; D – electrometer. (H. Kienitz (ed.), *Massenspektrometrie* (1968), Verlag Chemie, Weinheim. Reproduced by permission of Wiley-VCH.)

Calutron 1942: electromagnetic isotope separation

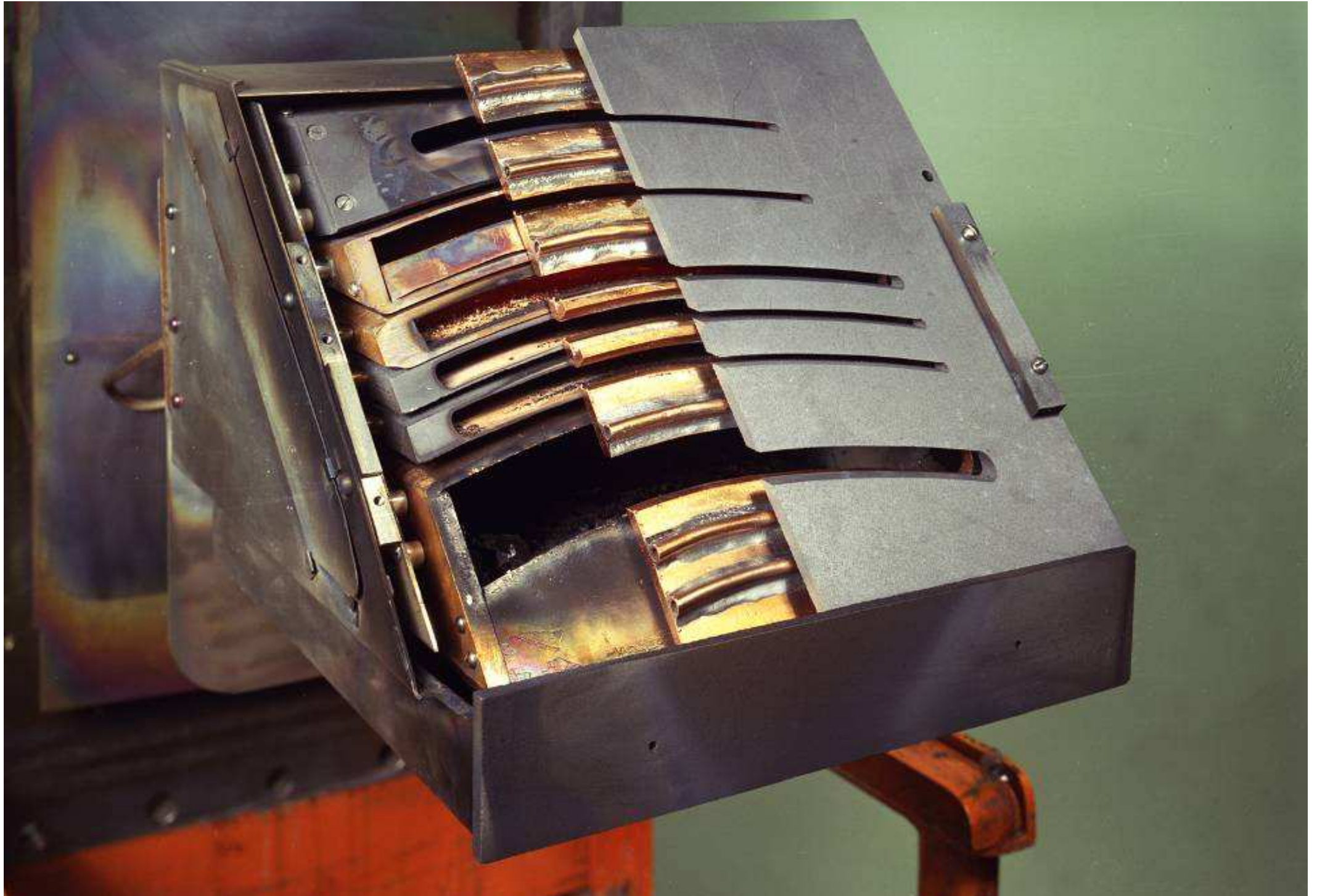
THE E M METHOD OF SEPARATING THE COMPONENTS OF TUBALLOY



Large scale electromagnetic isotope separation

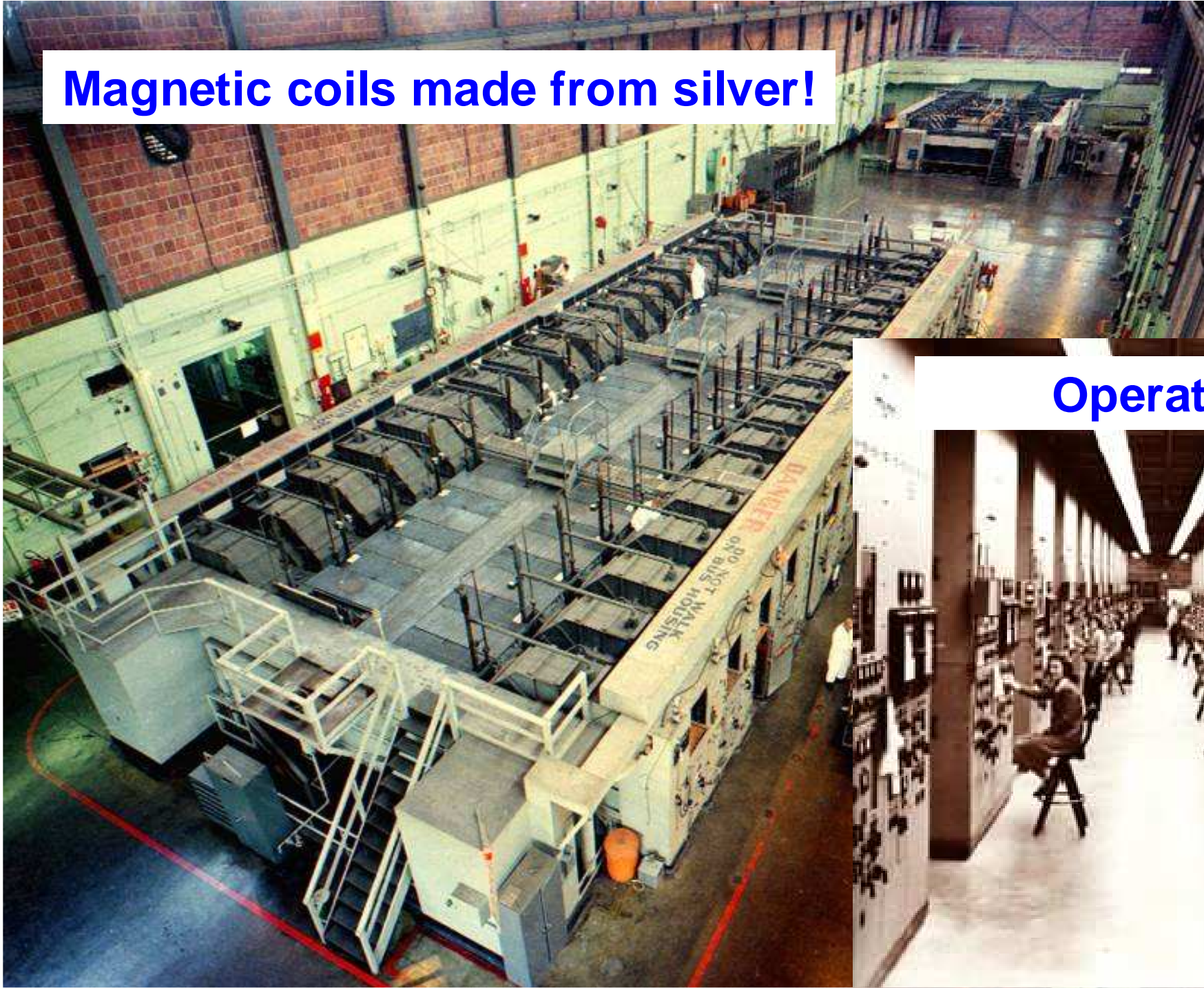


Collector plates of a Calutron



1945: large scale electromagnetic isotope separation

Magnetic coils made from silver!



Operators



1945: “Impact” of electromagnetic isotope separation



Hiroshima: 60 kg of isotopically enriched ^{235}U

Aston 1925: improved mass spectrograph



Improved version gives mass resolution: $\Delta m/m = 1/600$

Accuracy of mass determination: 10^{-4}

Used to study deviations of atomic masses m from A .

Introduced: “packing fraction” = $m/A - 1$

Systematic investigation of nuclear binding energies

Carbon isotopes

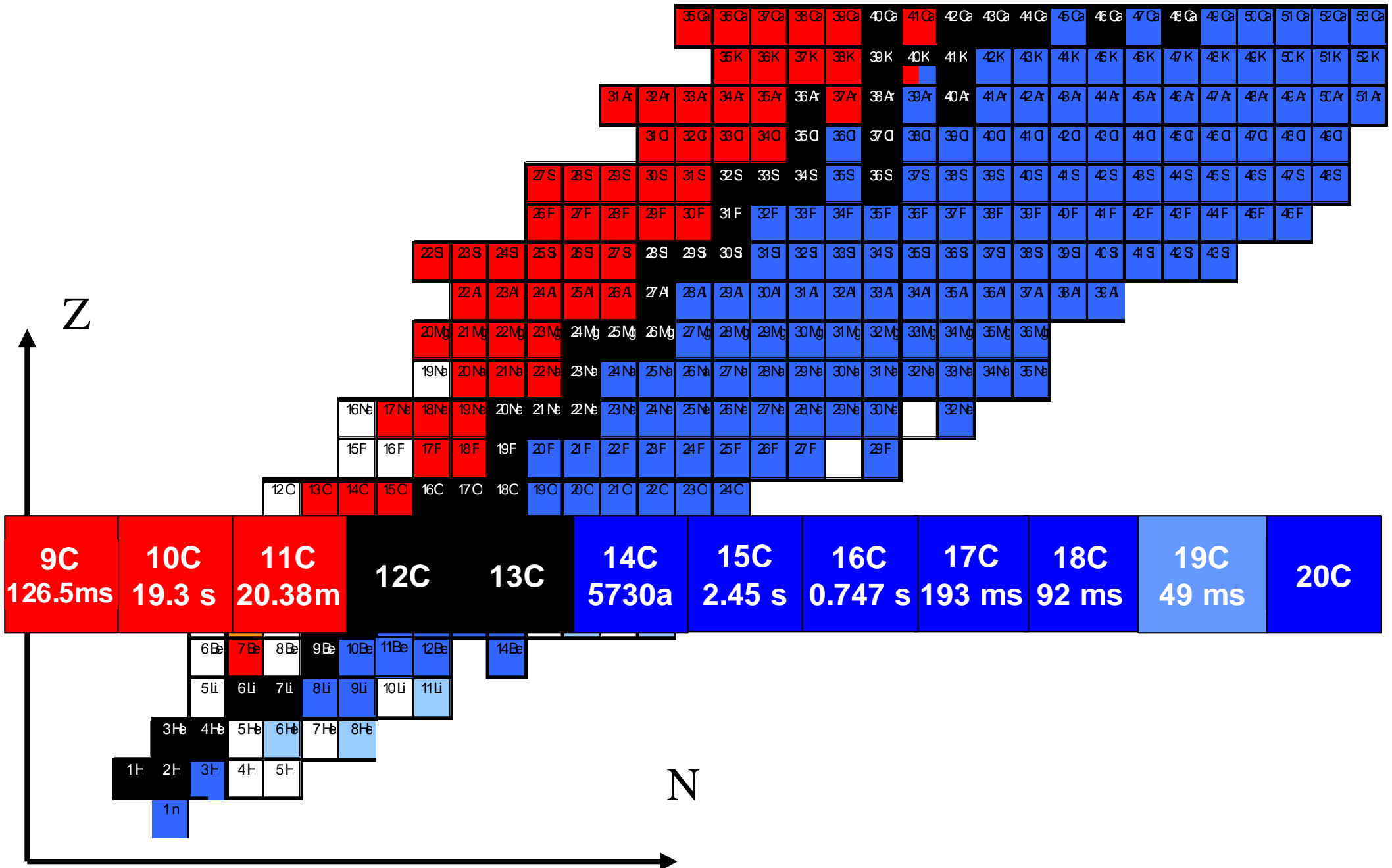
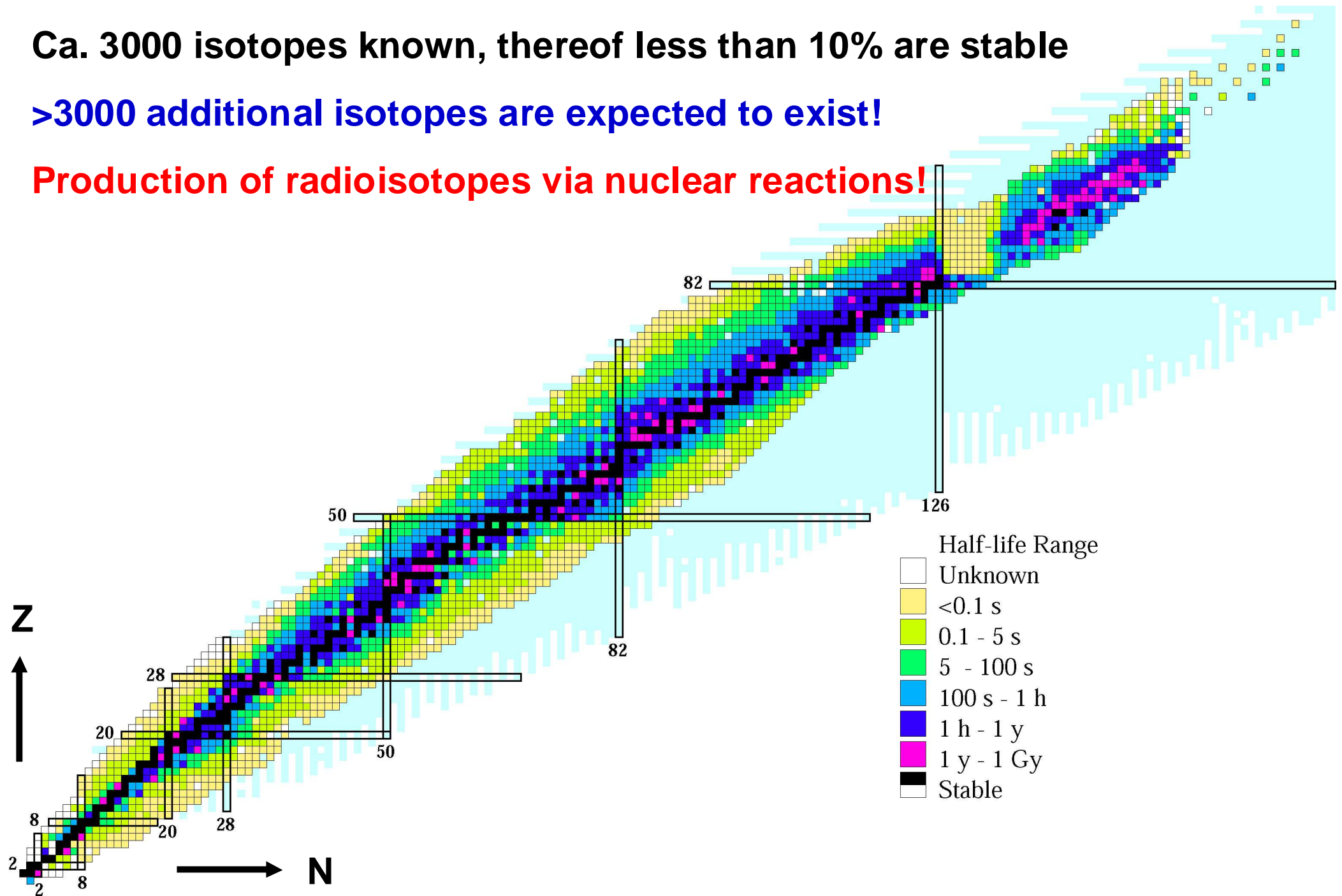


Chart of the nuclides

Ca. 3000 isotopes known, thereof less than 10% are stable

>3000 additional isotopes are expected to exist!

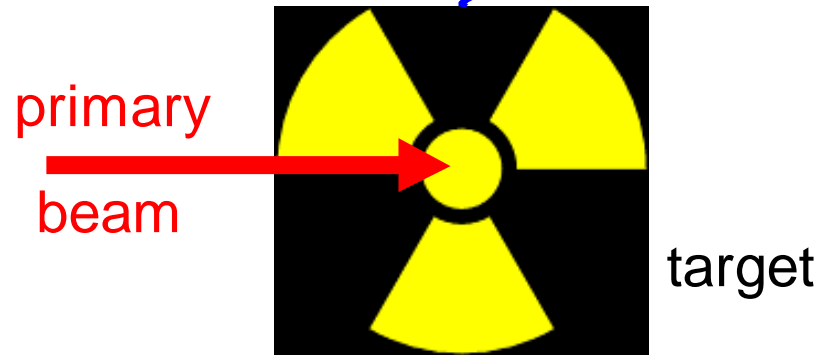
Production of radioisotopes via nuclear reactions!



Why “ion beams”?

Production:

high radiation environment



Detection:

low radiation background



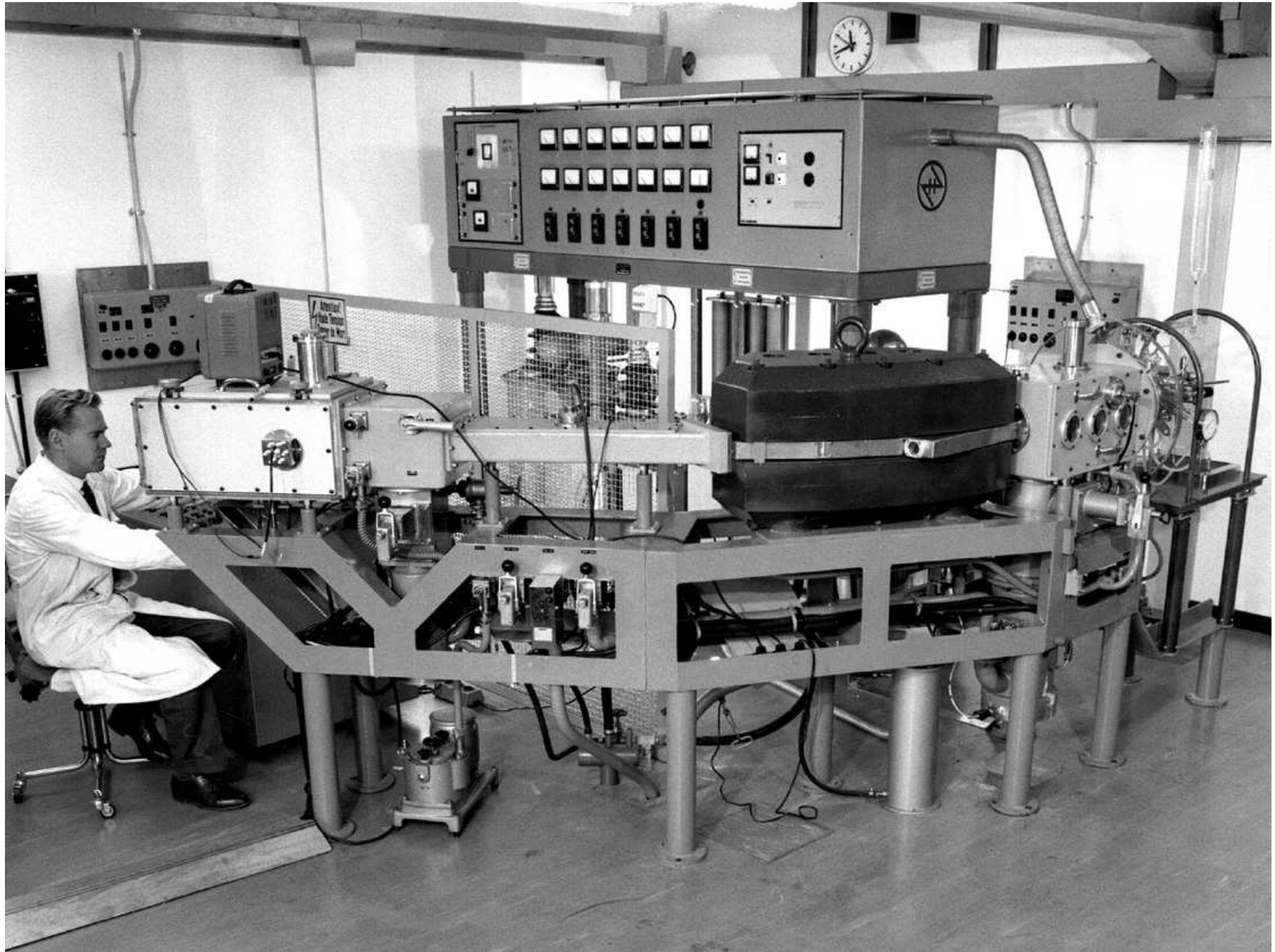
Transport methods:

- carry (“SRAFAP”)
- drive (*G.T. Seaborg and W.D. Loveland, The Elements beyond Uranium, John Wiley & Sons, 1990*)
- transport shuttle with pressurized air
- transport in gas-jet
- pump through vacuum system
- **send as ion beam**

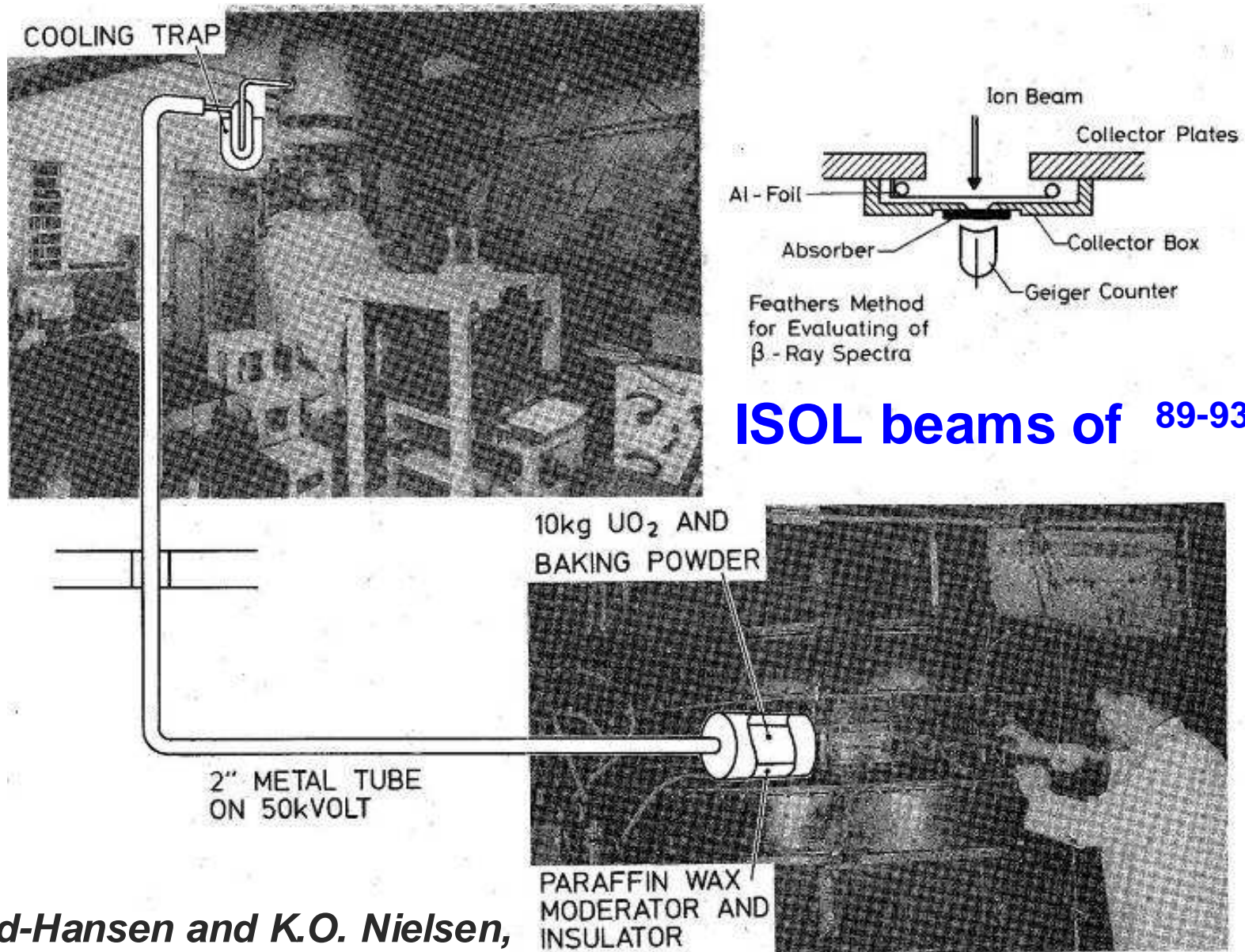
Irradiations of targets



Off-line mass separator



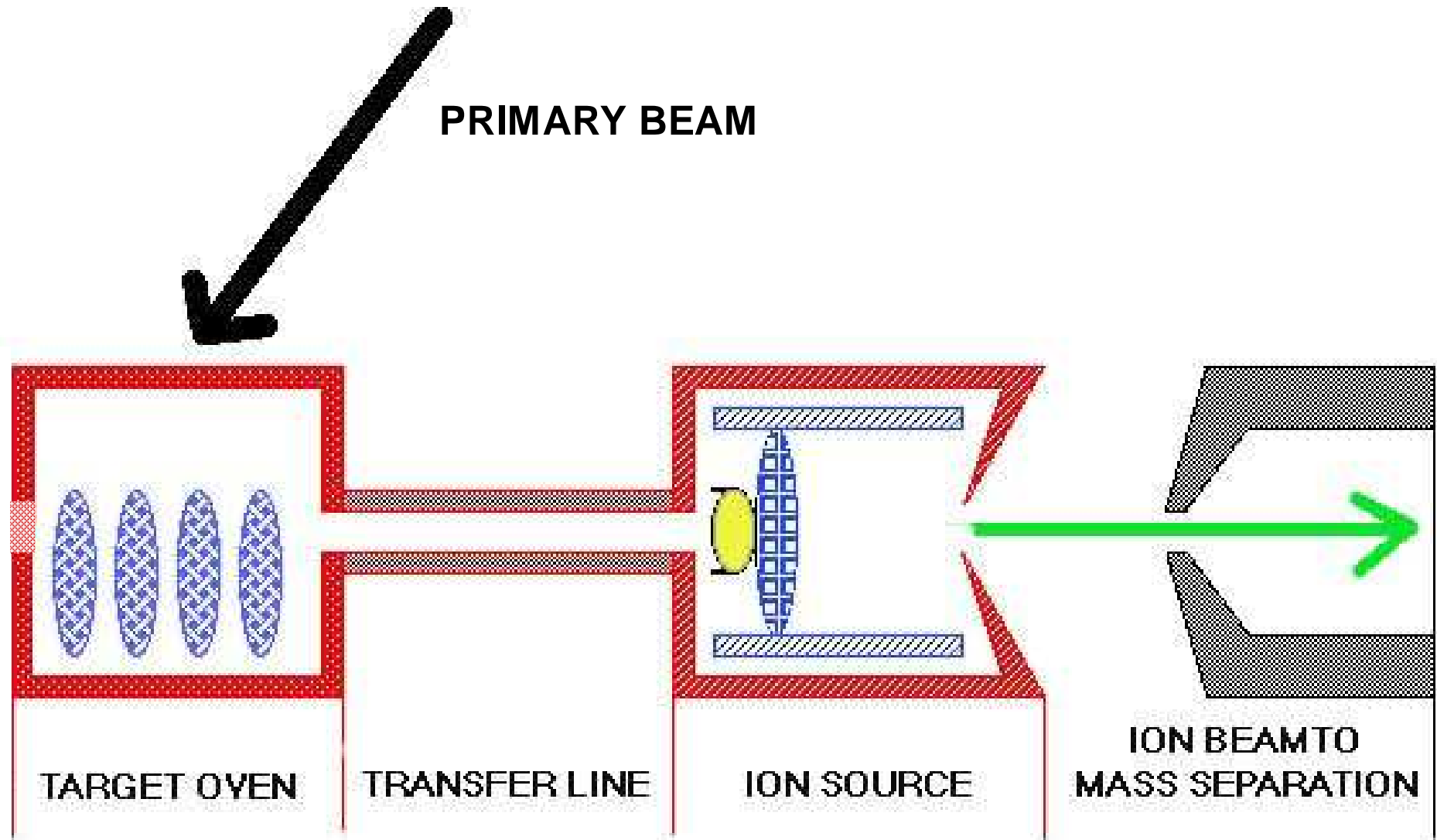
1951: first ISOL experiment at Niels Bohr Institute



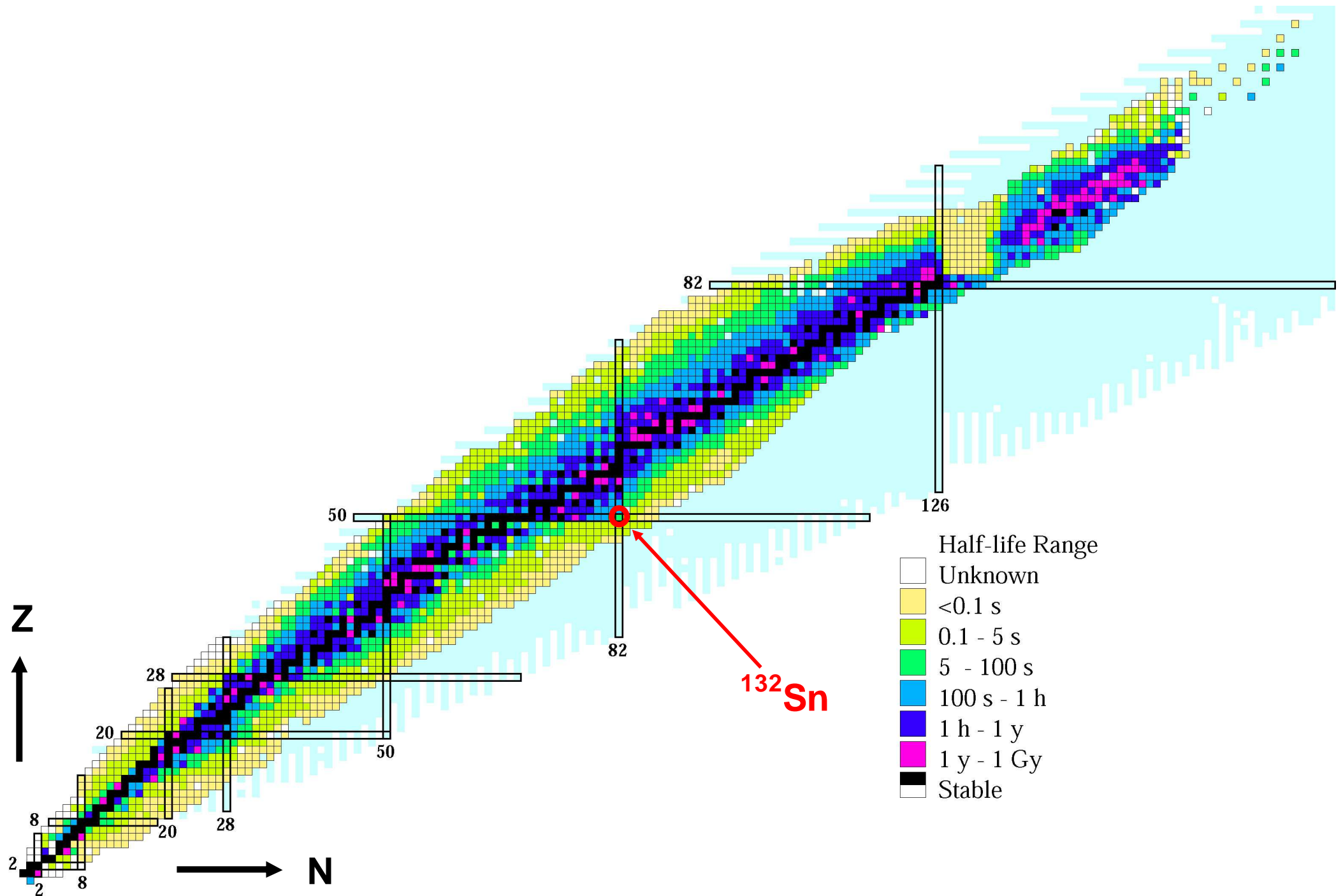
O. Kofoed-Hansen and K.O. Nielsen,

Mat. Fys. Medd. Dan. Vid. Selsk. 26, Nr. 7 (1951).

Isotope Separation On-Line



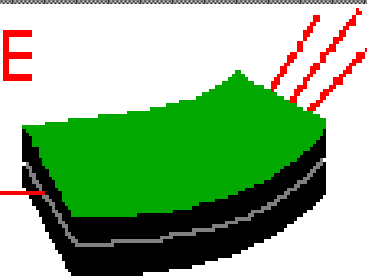
Isotope selection



Isotope selection with ISOL method

- Ionization to $q = 1+$
- Acceleration to 60 keV
- Mass selection by magnetic deflection
- $B\rho = p/q \propto \sqrt{A}$

ISOLDE
CERN

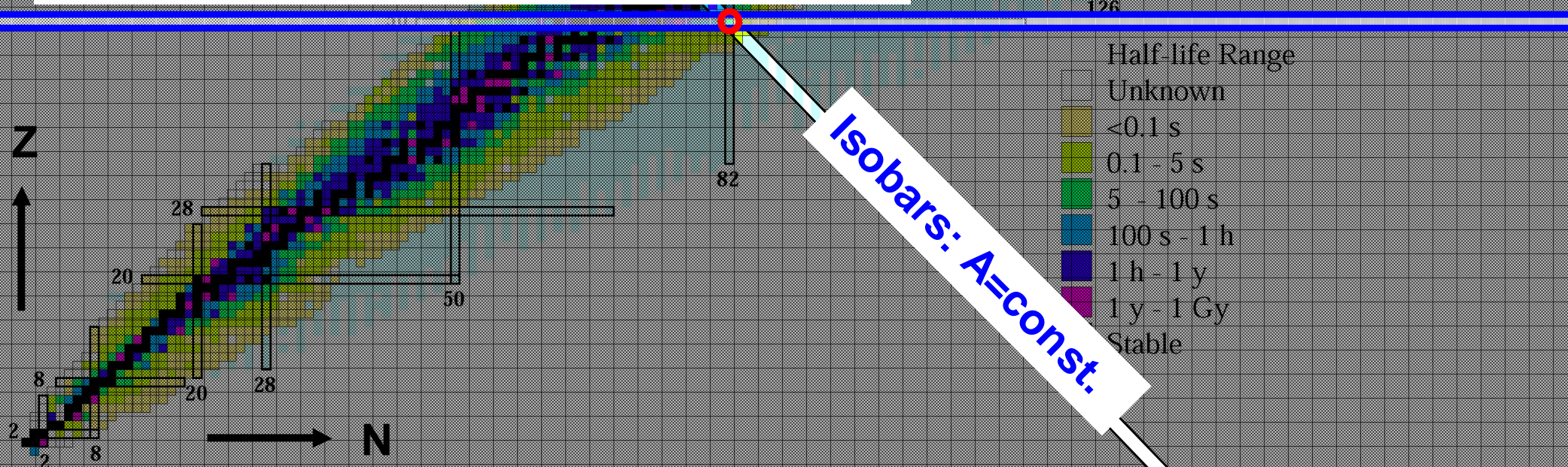


Z selection by chemically selective step

Z



N



Half-life Range

Unknown

<0.1 s

0.1 - 5 s

5 - 100 s

100 s - 1 h

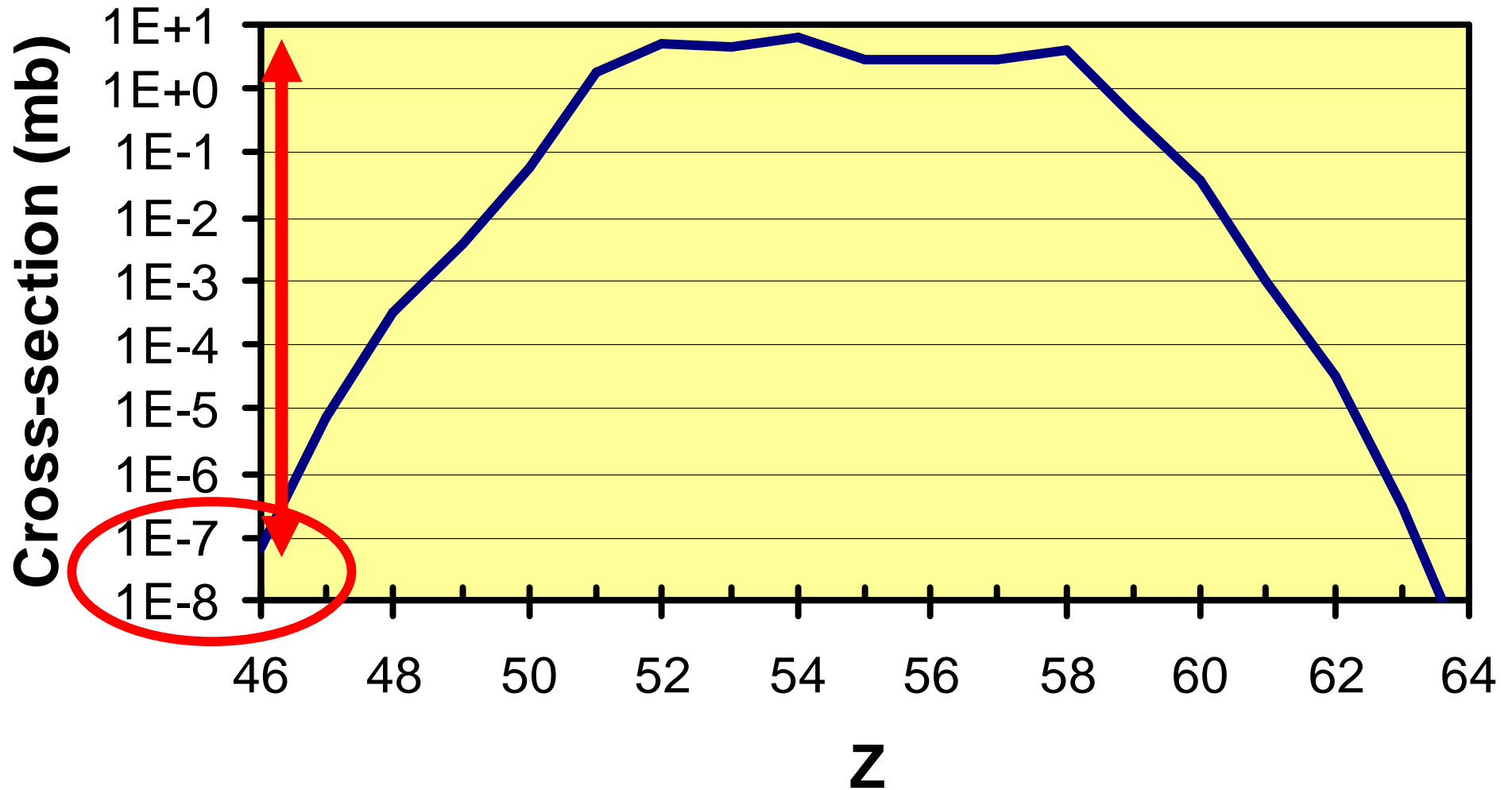
1 h - 1 y

1 y - 1 Gy

Stable

Isobars: $A=\text{const.}$

The challenge of the extremes!



1. low cross-sections \Rightarrow optimize efficiency
2. enormous production of isobars \Rightarrow optimize selectivity
3. short half-lives \Rightarrow optimize rapidity

Optimize event rate

All steps of the separation chain need to be optimized!

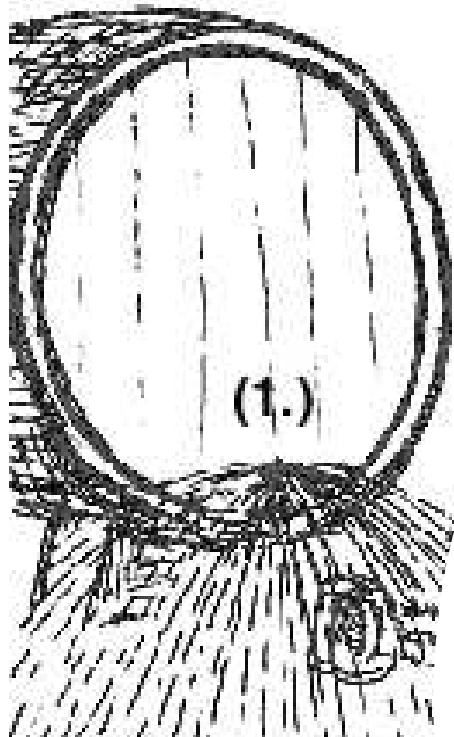
$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$

In-target production

Efficiency

Optimize RIB intensity

All steps of the separation chain need to be optimized!



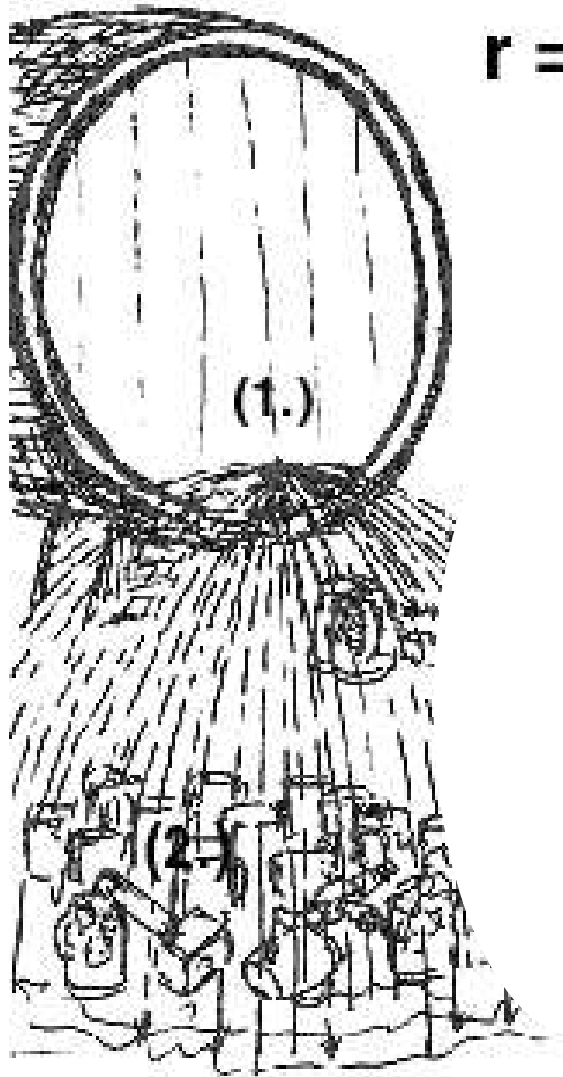
$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$

powerful accelerator

⇒ accelerator technology

Optimize RIB intensity

All steps of the separation chain need to be optimized!



$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$

high production cross-sections

⇒ nuclear physics

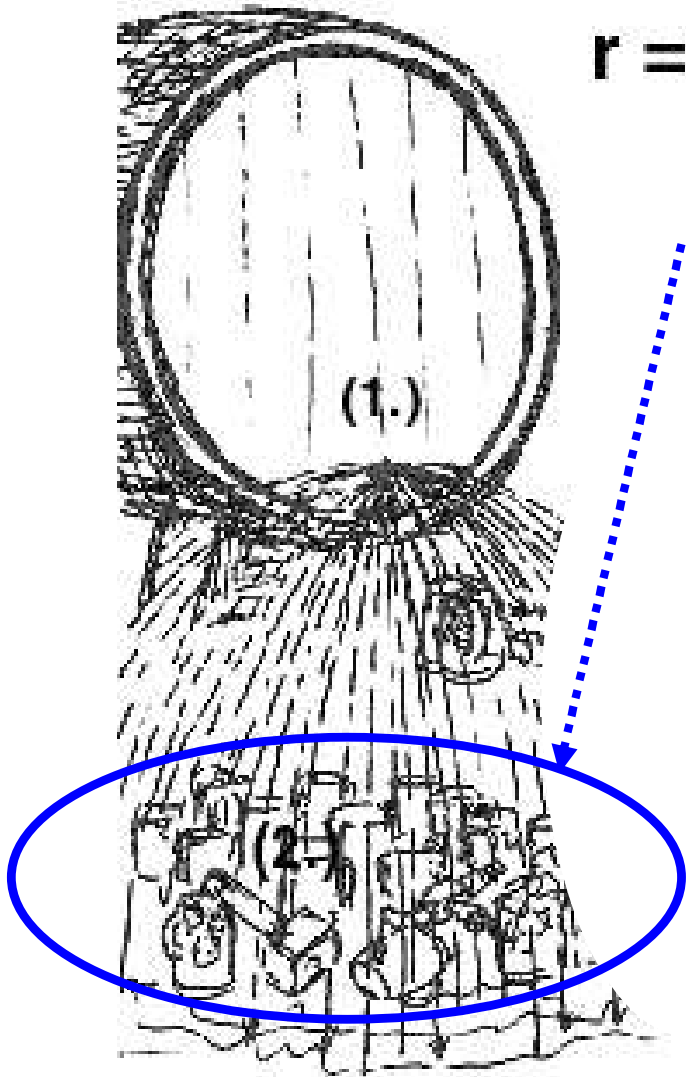
Optimize RIB intensity

All steps of the separation chain need to be optimized!

$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$

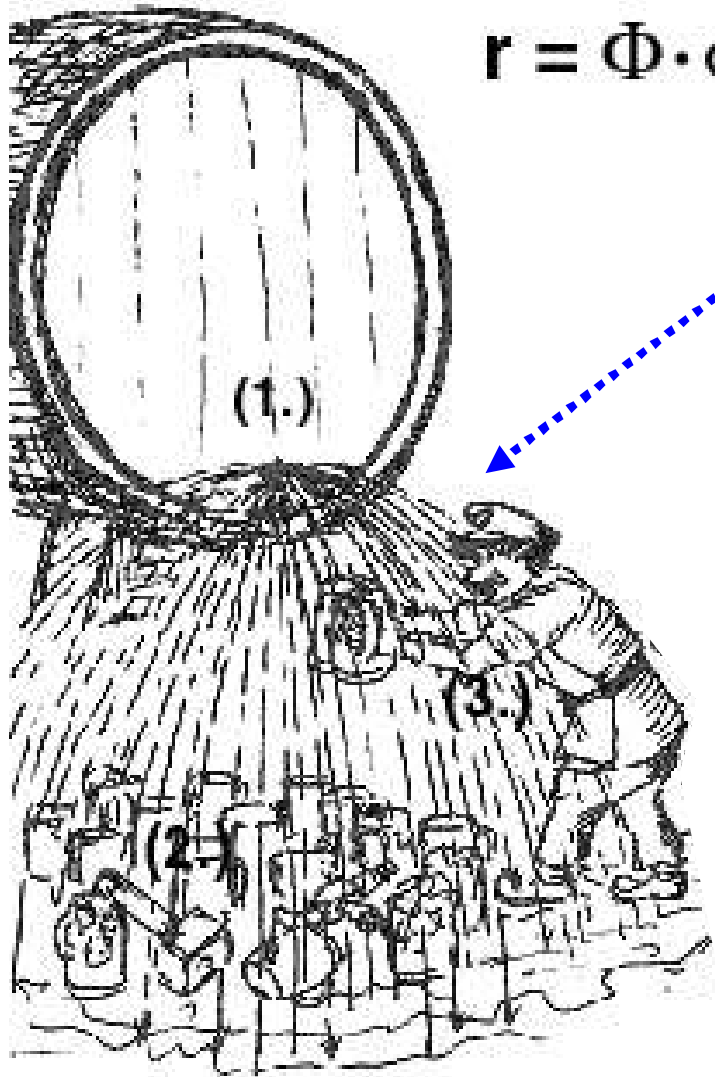
reliable “thick” targets

⇒ materials science



Optimize RIB intensity

All steps of the separation chain need to be optimized!



$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$

Extraction efficiency from target determined by:

- bulk diffusion

⇒ solid state physics

- surface desorption

⇒ surface chemistry

- effusion

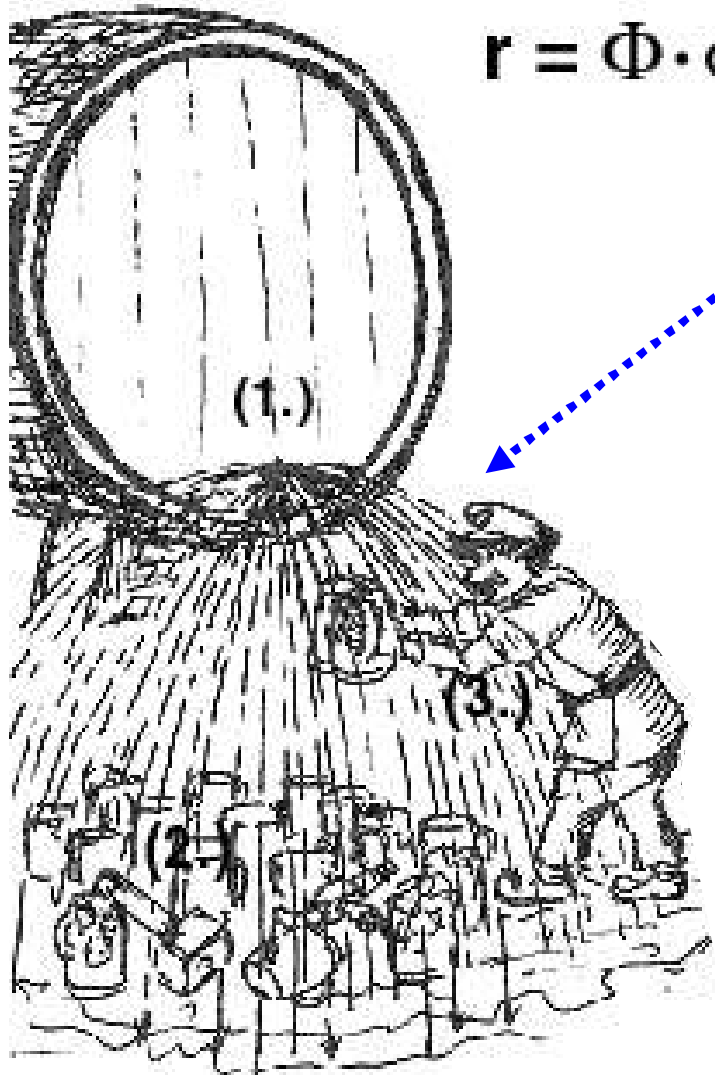
⇒ gas phase chemistry

strongly element dependent!

Optimize RIB intensity

All steps of the separation chain need to be optimized!

$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$



high ionization and extraction efficiency

⇒ ion source technology

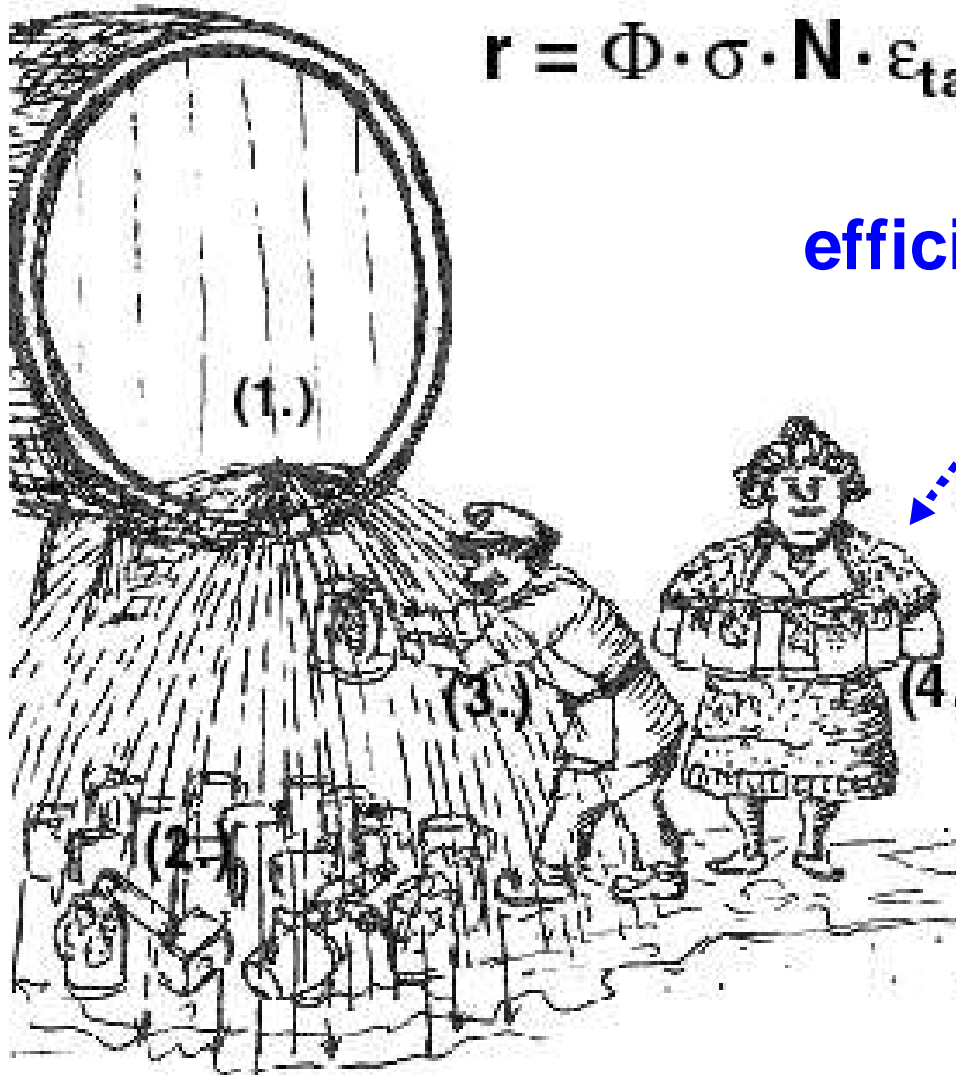
Optimize RIB intensity

All steps of the separation chain need to be optimized!

$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$

efficient transport of RIB

⇒ ion optics



Optimize RIB intensity

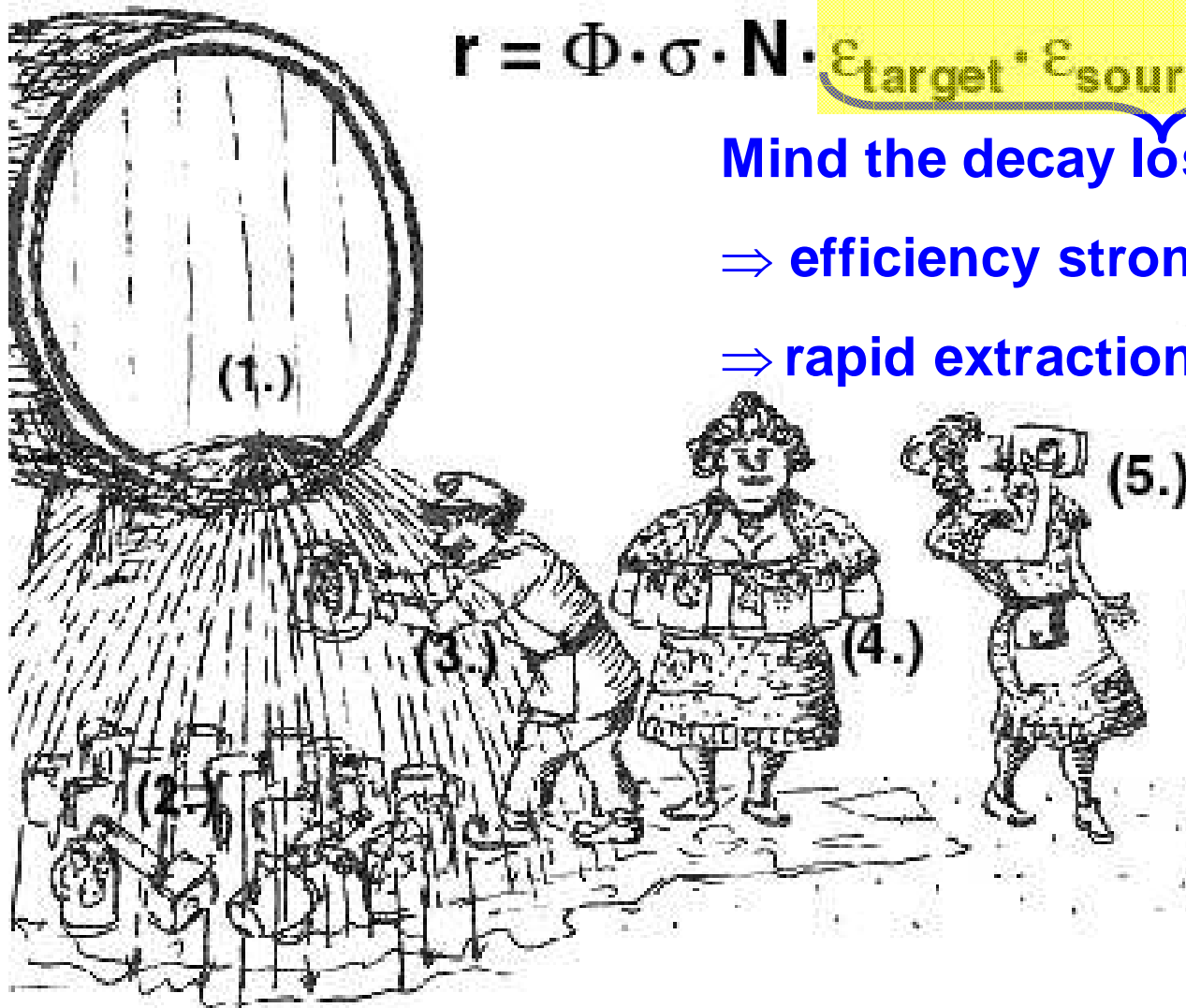
All steps of the separation chain need to be optimized!

$$r = \Phi \cdot \sigma \cdot N \cdot \underbrace{\epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}}}_{\text{Mind the decay losses during delays}} \cdot \epsilon_{\text{det}}$$

Mind the decay losses during delays

⇒ efficiency strongly half-life dependent

⇒ rapid extraction required!

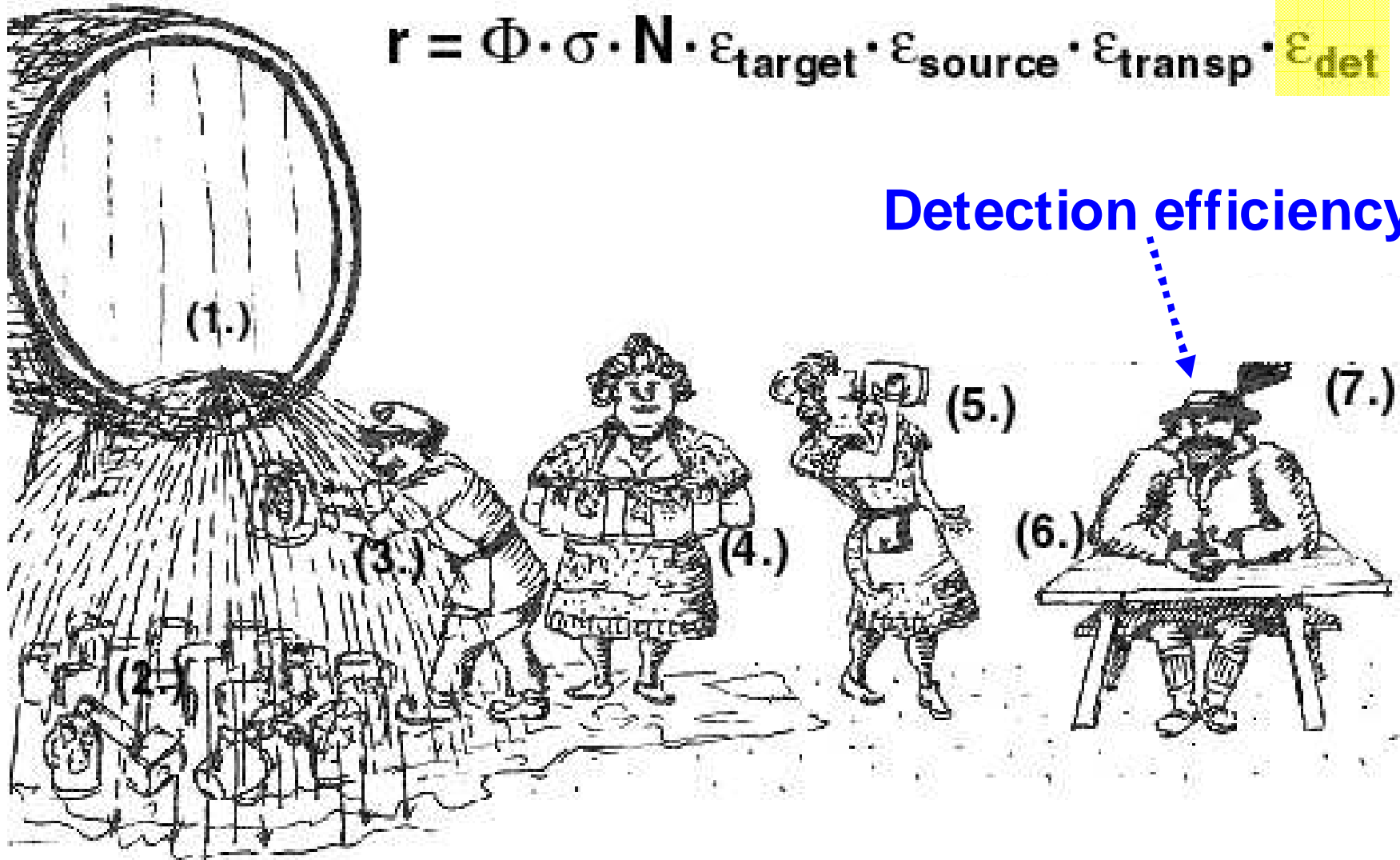


Optimize RIB intensity

All steps of the separation chain need to be optimized!

$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$

Detection efficiency



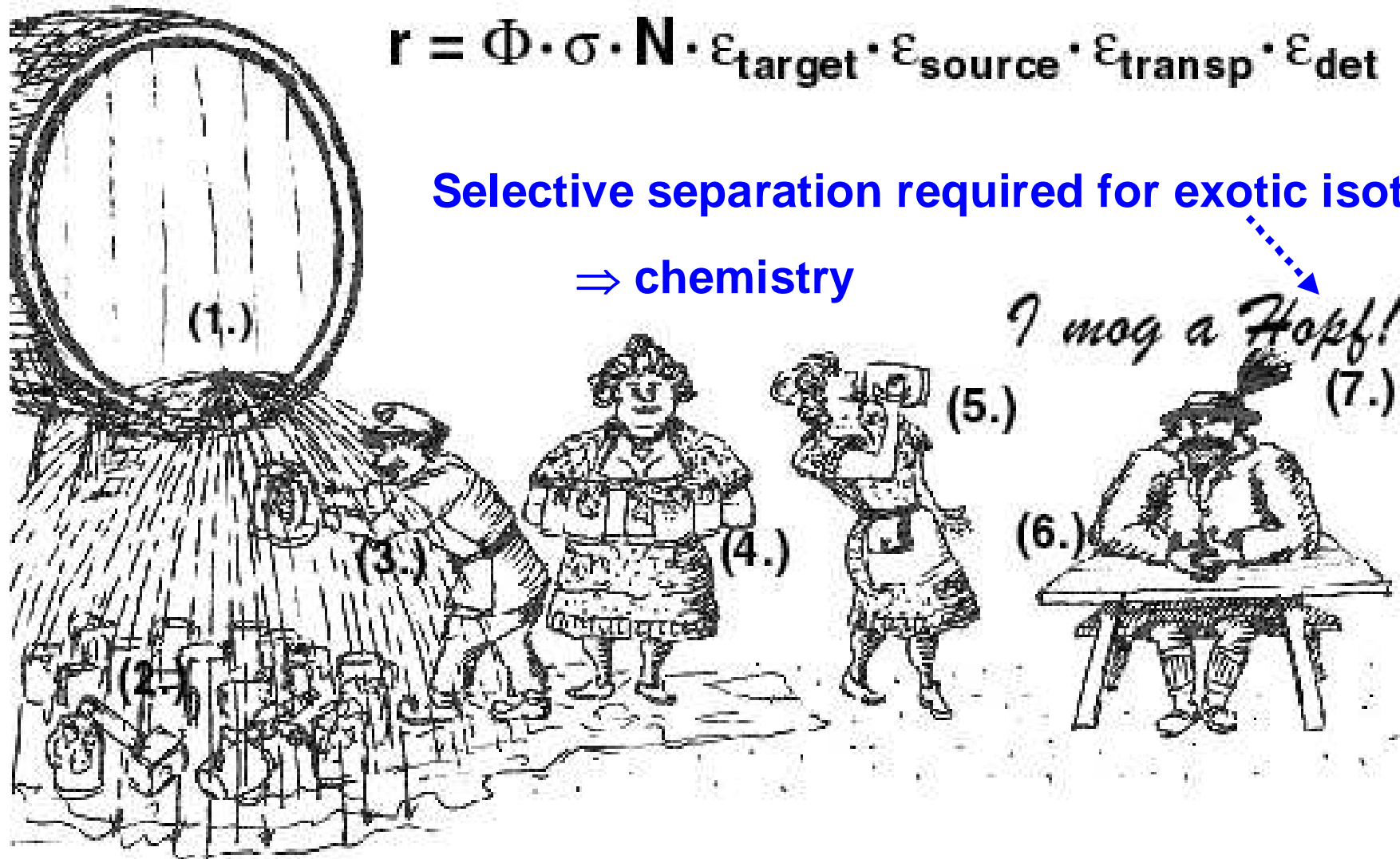
Optimize RIB intensity and purity

All steps of the separation chain need to be optimized!

$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$

Selective separation required for exotic isotopes!

⇒ chemistry

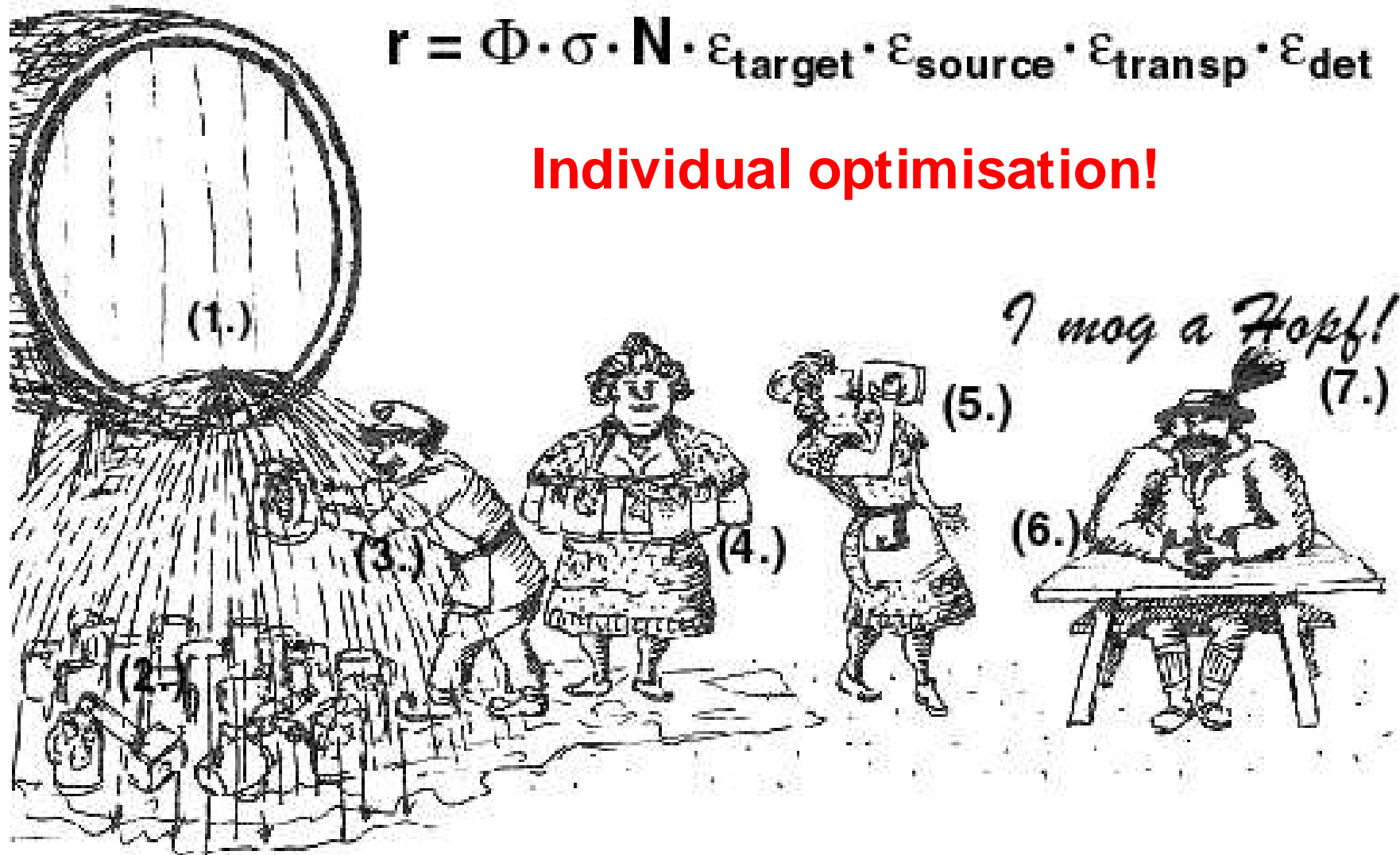


Optimize RIB intensity

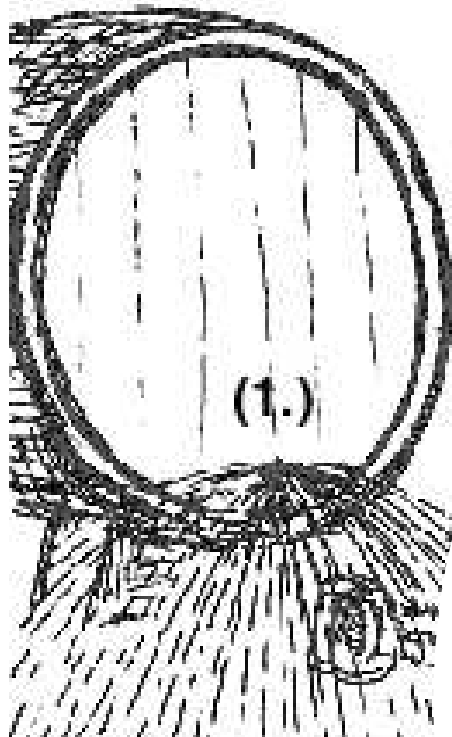
Factors are highly correlated and isotope dependent!

$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$

Individual optimisation!



Particle accelerators



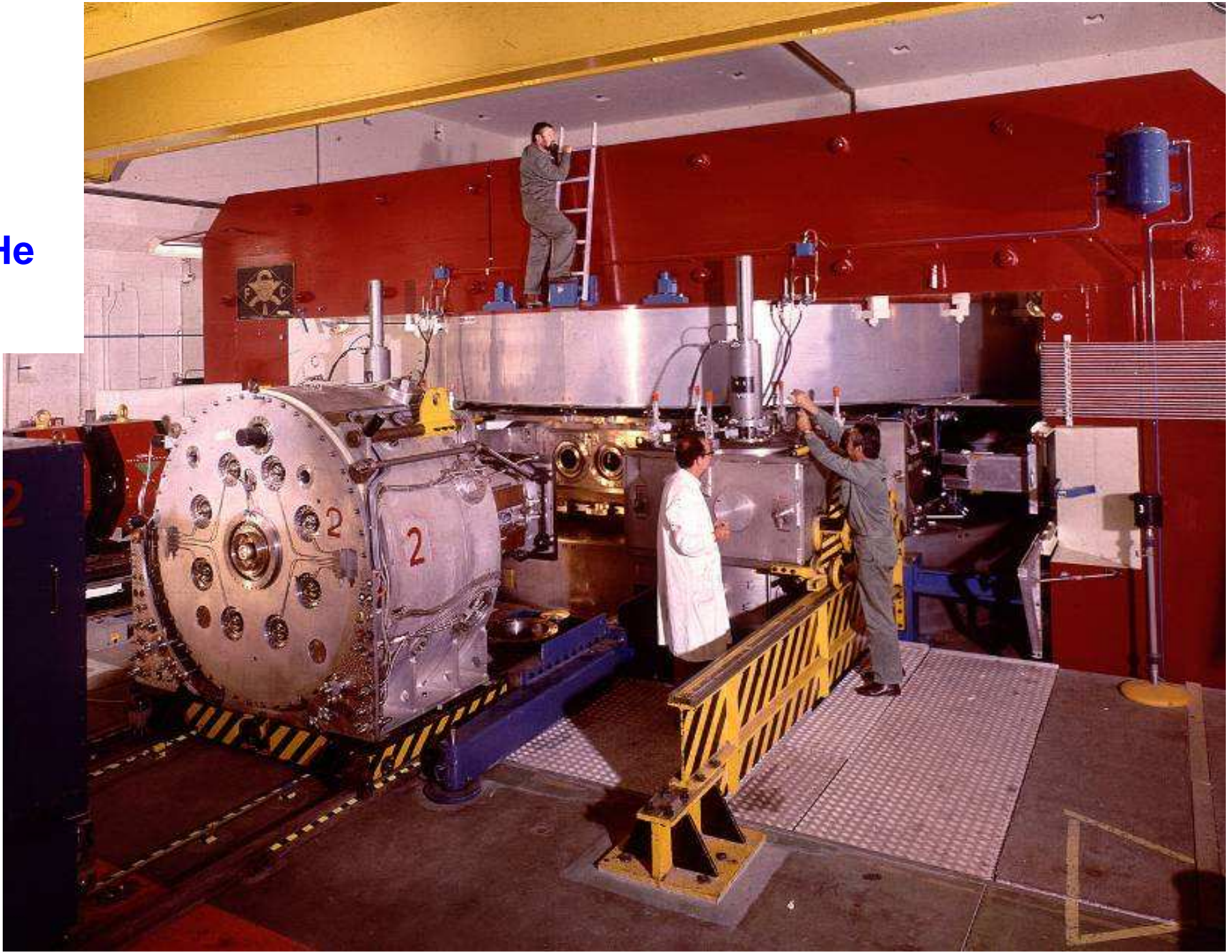
$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$

CERN synchrocyclotron 1957-1990

600 MeV p
up to 4 μ A

910 MeV ^3He

1 GeV ^{12}C



CERN accelerator structure

CERN-PS Booster Synchrotrons

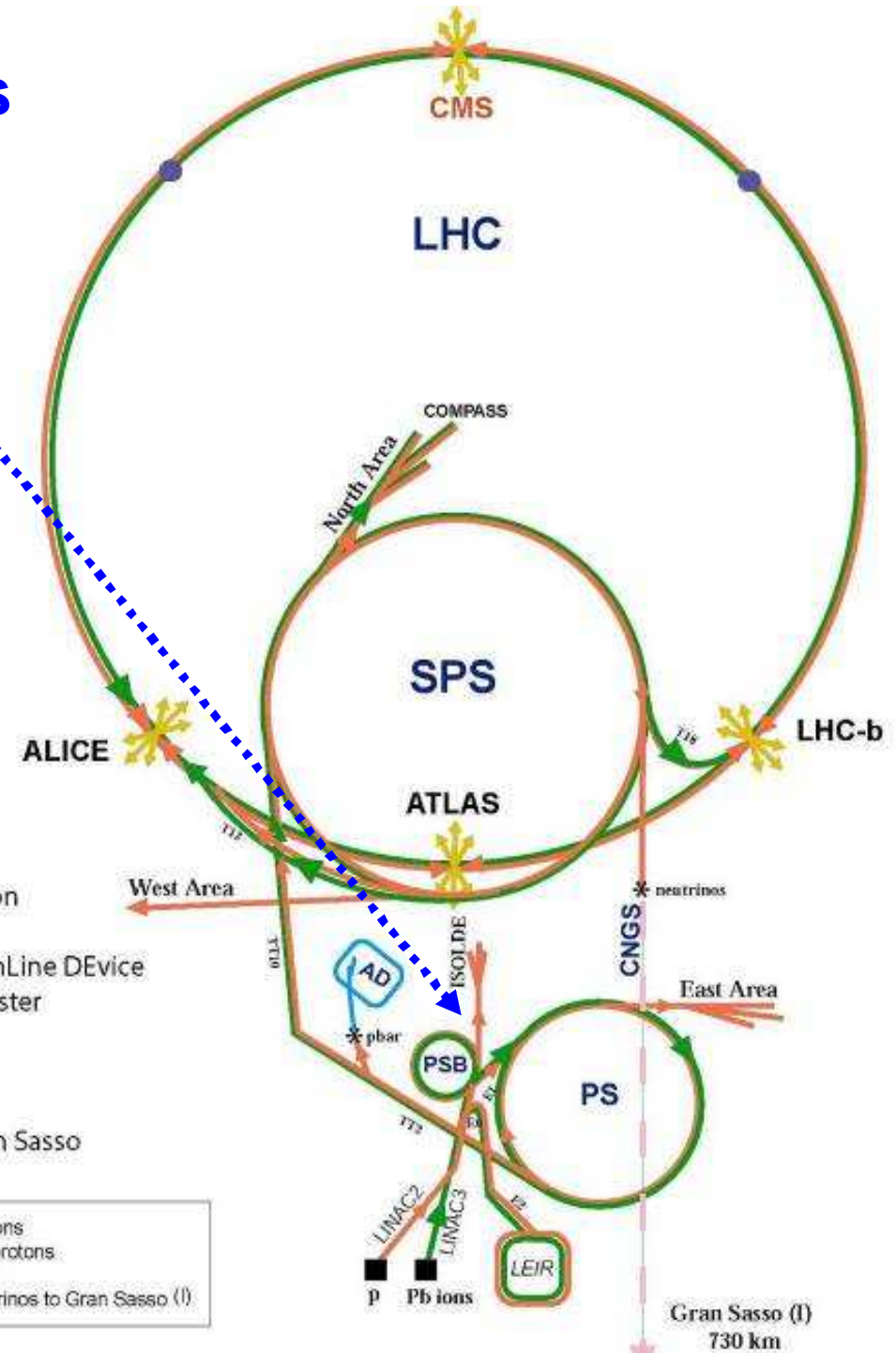
$$E_p = 1.4 \text{ GeV}$$

$$3 \cdot 10^{13} \text{ protons/pulse} = 5 \mu\text{C}$$

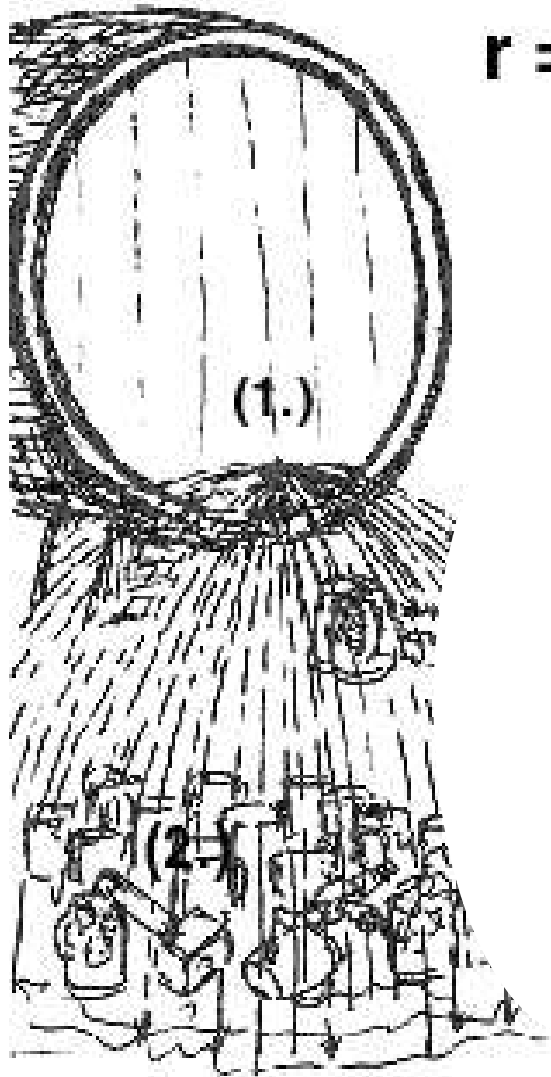
$$I_{\text{average}} = 4 \mu\text{A}$$

$$P_{\text{average}} = 6 \text{ kW}$$

LHC: Large Hadron Collider
SPS: Super Proton Synchrotron
AD: Antiproton Decelerator
ISOLDE: Isotope Separator OnLine DEvice
PSB: Proton Synchrotron Booster
PS: Proton Synchrotron
LINAC: LINear ACcelerator
LEIR: Low Energy Ion Ring
CNGS: Cern Neutrinos to Gran Sasso

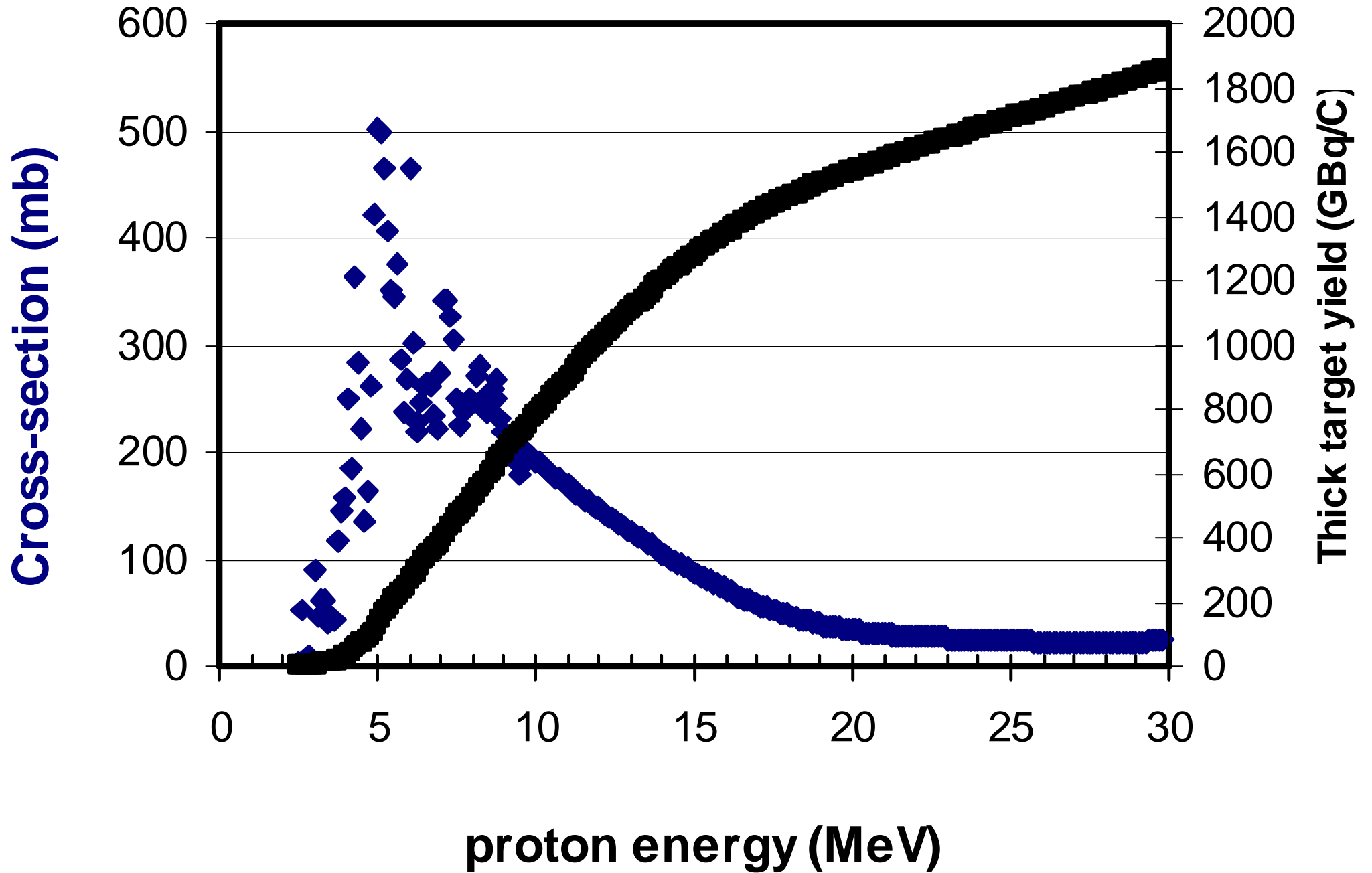


Nuclear reactions



$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$

$^{18}\text{O}(p,n)^{18}\text{F}$ cross-sections

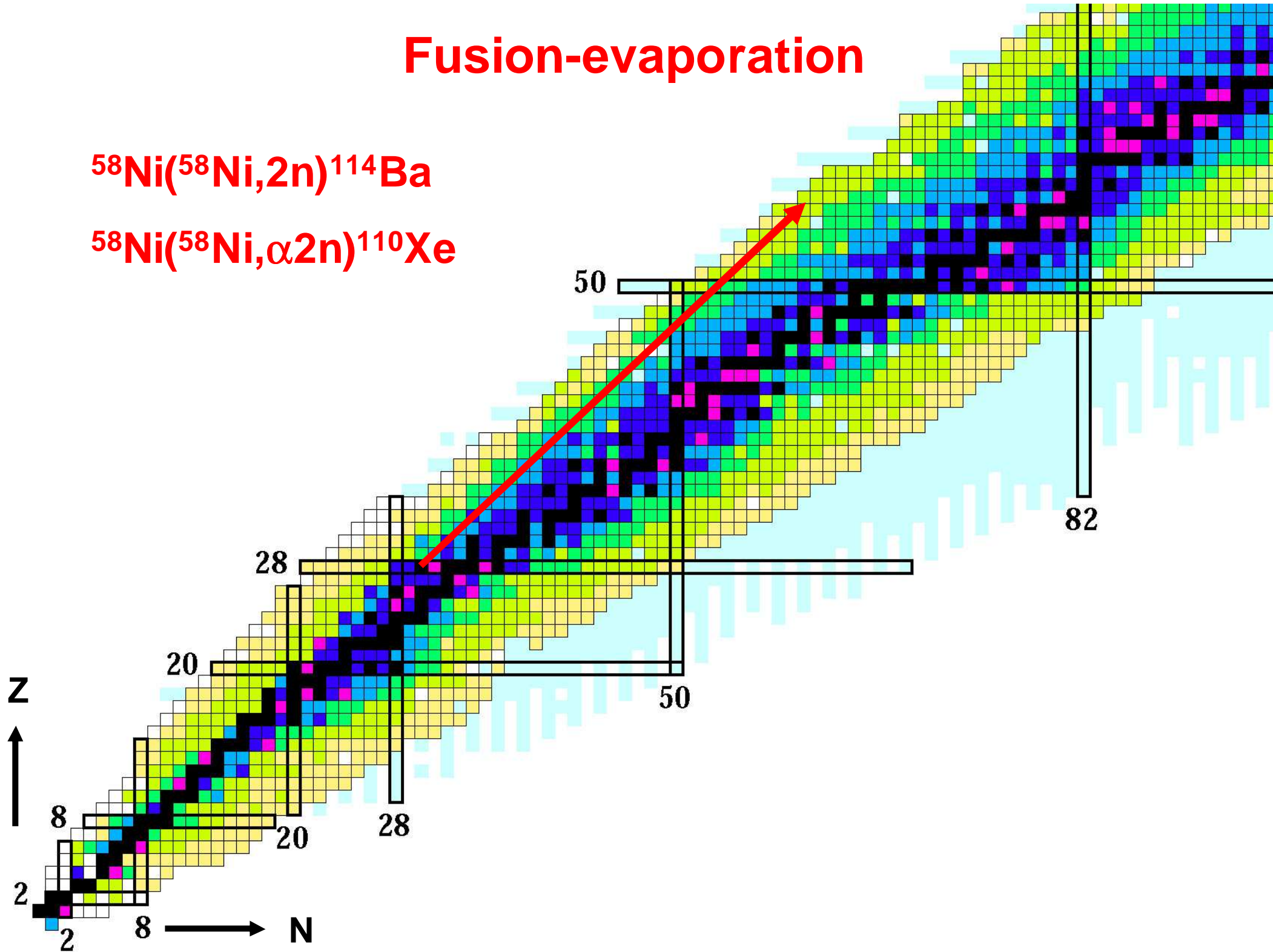


Nuclear reactions

1. Direct reactions

- (p,n) , $({}^3\text{He},n)$, (α,n) , $(n,\alpha),\dots$
- high cross-sections, products relatively close to stability
- driver beams from (low-cost) cyclotrons

Fusion-evaporation



Nuclear reactions

1. Direct reactions

- (p,n) , $({}^3\text{He},n)$, (α,n) , $(n,\alpha),\dots$
- high cross-sections, products relatively close to stability
- driver beams from (low-cost) cyclotrons

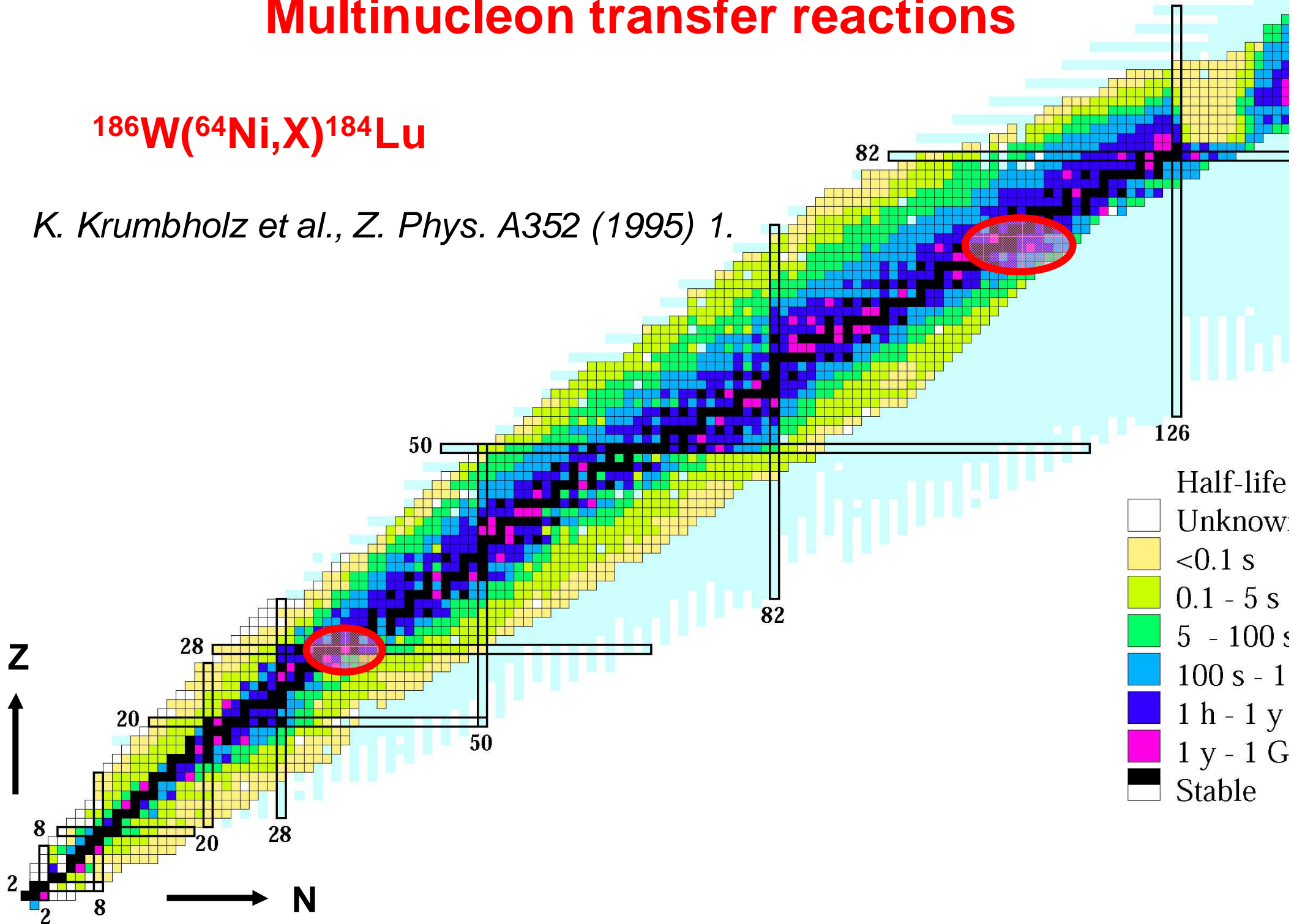
2. Heavy-ion fusion-evaporation

- produces neutron-deficient heavier isotopes
- small energy window in vicinity of Coulomb barrier (some MeV/nucl.)
- requires heavy ion beams \Rightarrow bigger cyclotrons or LINACs

Multinucleon transfer reactions



K. Krumbholz et al., Z. Phys. A352 (1995) 1.



Nuclear reactions

1. Direct reactions

- high cross-sections, products relatively close to stability
- driver beams from (low-cost) cyclotrons

2. Heavy-ion fusion-evaporation

- produces neutron-deficient heavier isotopes
- small energy window in vicinity of Coulomb barrier (some MeV/nucleon)
- requires heavy ion beams \Rightarrow bigger cyclotrons or LINACs

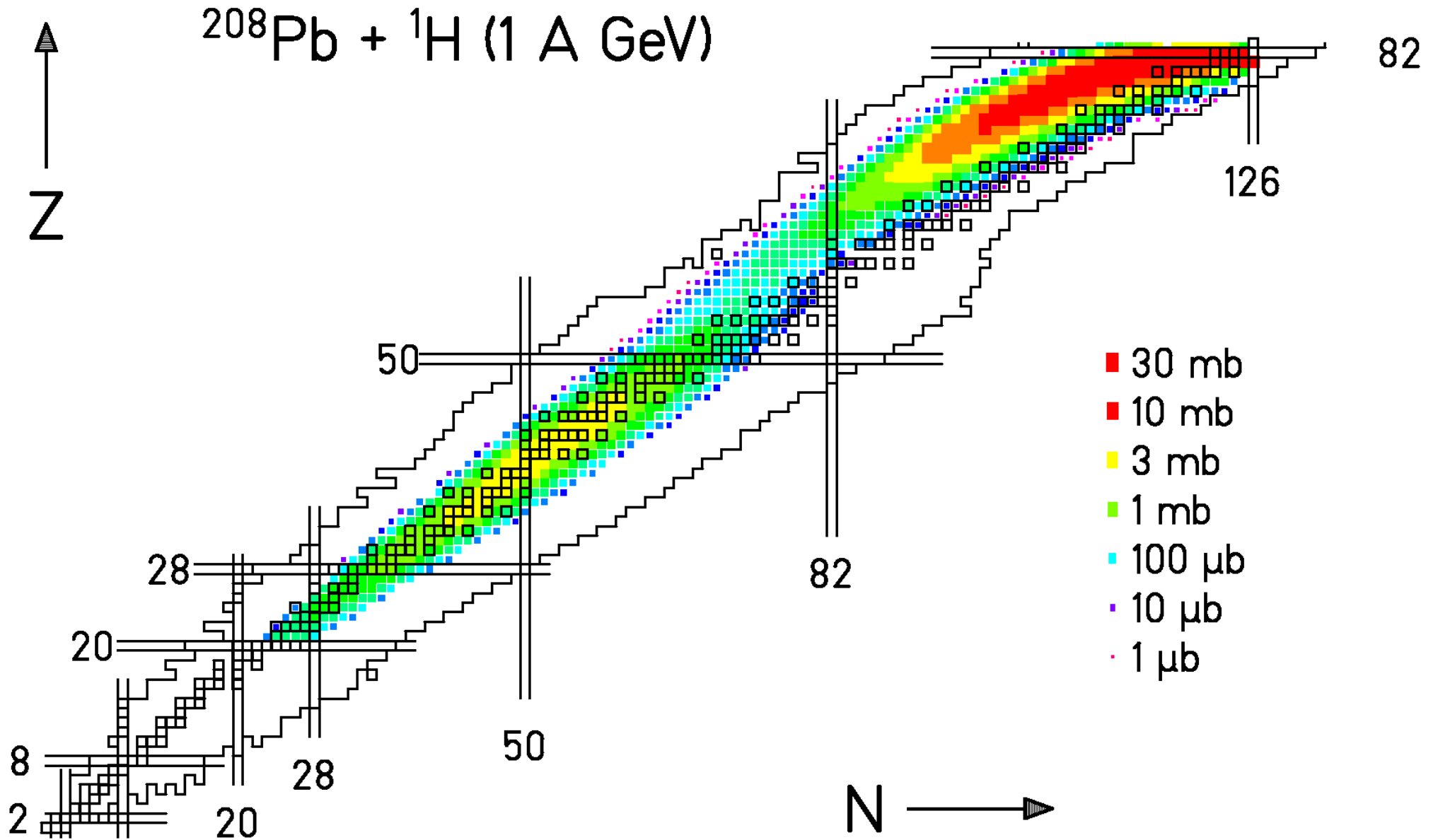
3. Deep inelastic collisions (multi-nucleon transfer)

- products close to target, mass-flow towards stability
- light to heavy ion beams (tens of MeV/nucleon)
- only method to reach neutron-rich isotopes with $N_{\text{product}} > N_{\text{target}} + 1$

4. Spallation

- intranuclear cascade heats nucleus
- evaporation of preferentially neutrons \Rightarrow neutron-deficient products
- high cross-sections for products close to target
- requires protons of >100 MeV \Rightarrow big p cyclotron, synchrotron or LINAC

Spallation + Fragmentation + Fission



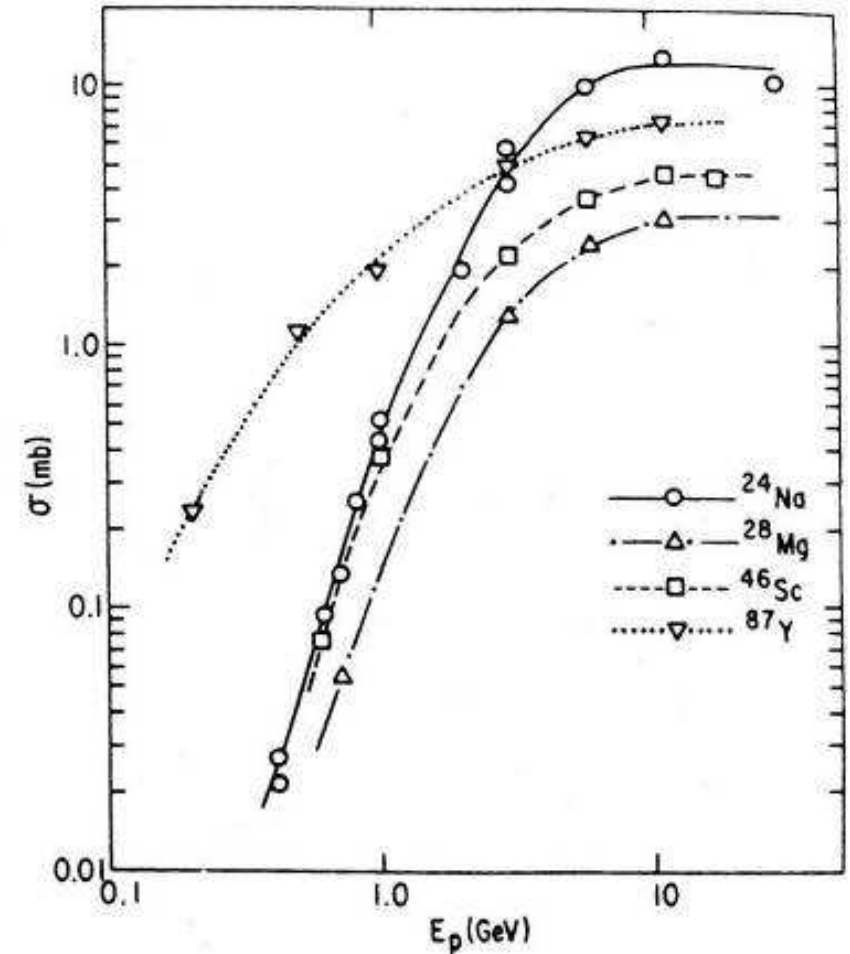
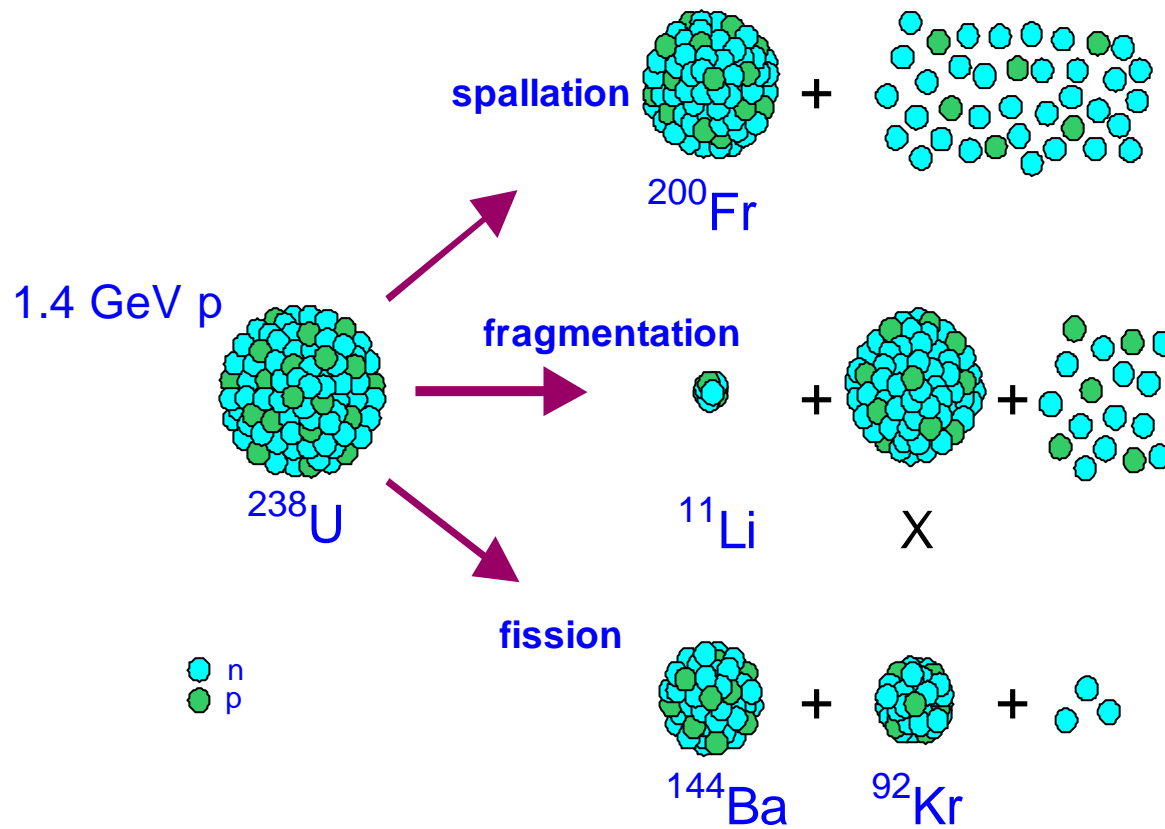
T. Enqvist et al., Nucl. Phys. A686 (2001) 481.

Nuclear reactions

5. Fragmentation

- many cross-sections show little energy dependence in the region 40-2000 MeV/nucleon
- target fragmentation needs e.g. high energy protons (see spallation)
- projectile fragmentation needs high energy heavy ions
⇒ huge cyclotron, synchrotron or LINAC

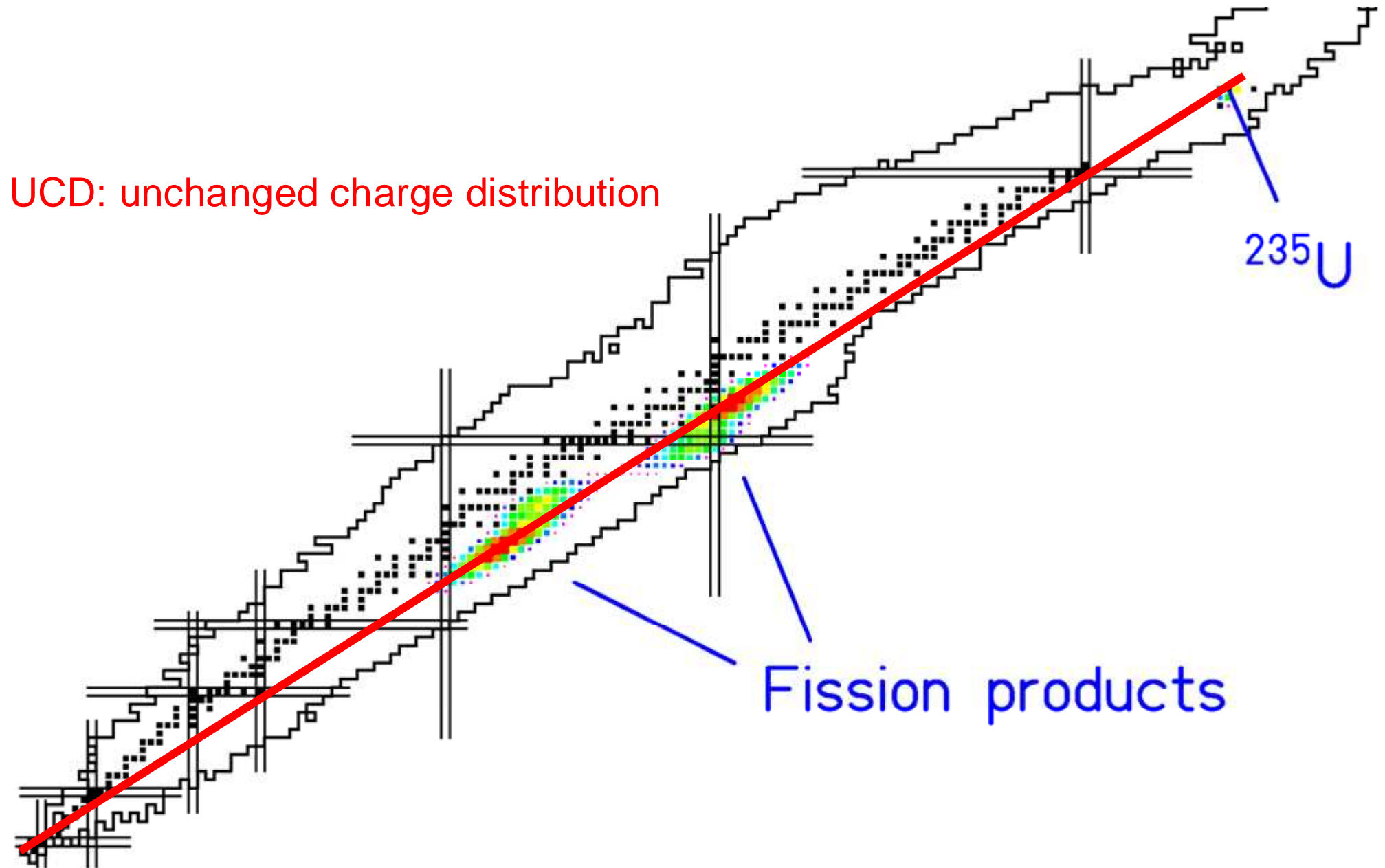
High energy nuclear reactions



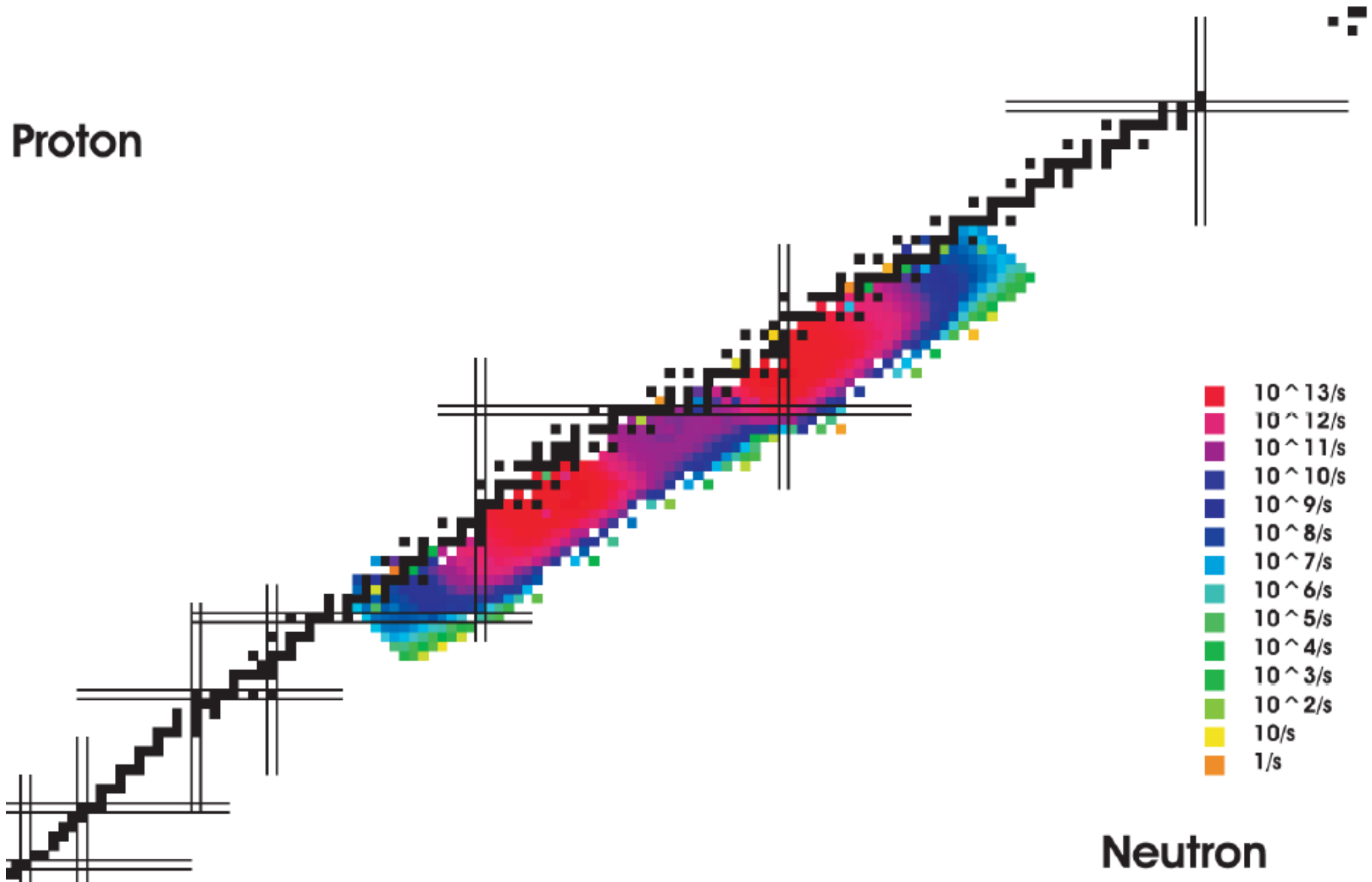
$$\sigma_i = 44.9 A^{0.7} \text{ mb}$$

R. Silberberg and C.H. Tsao, Phys. Rep. 191 (1990) 351.

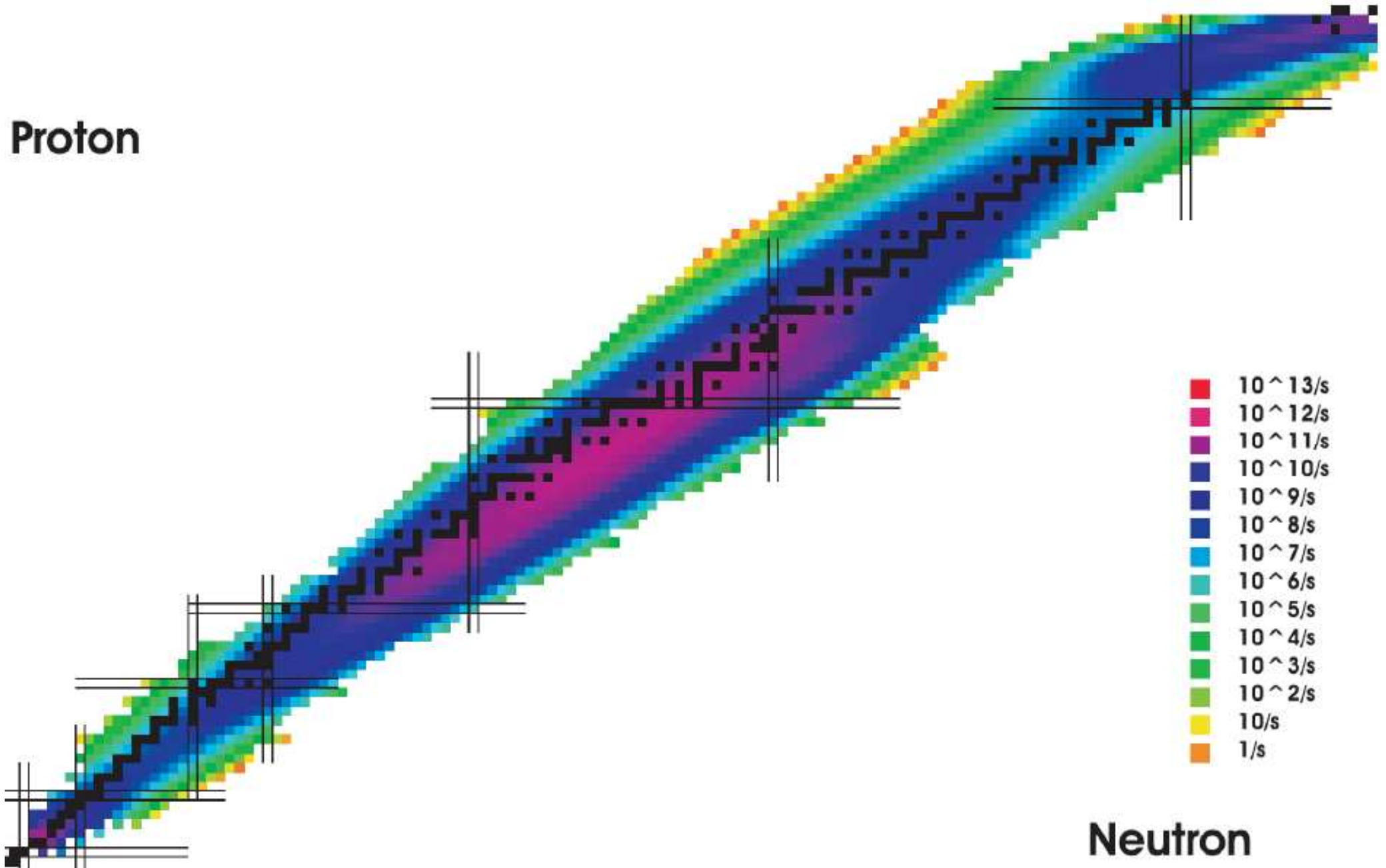
Low-energy fission



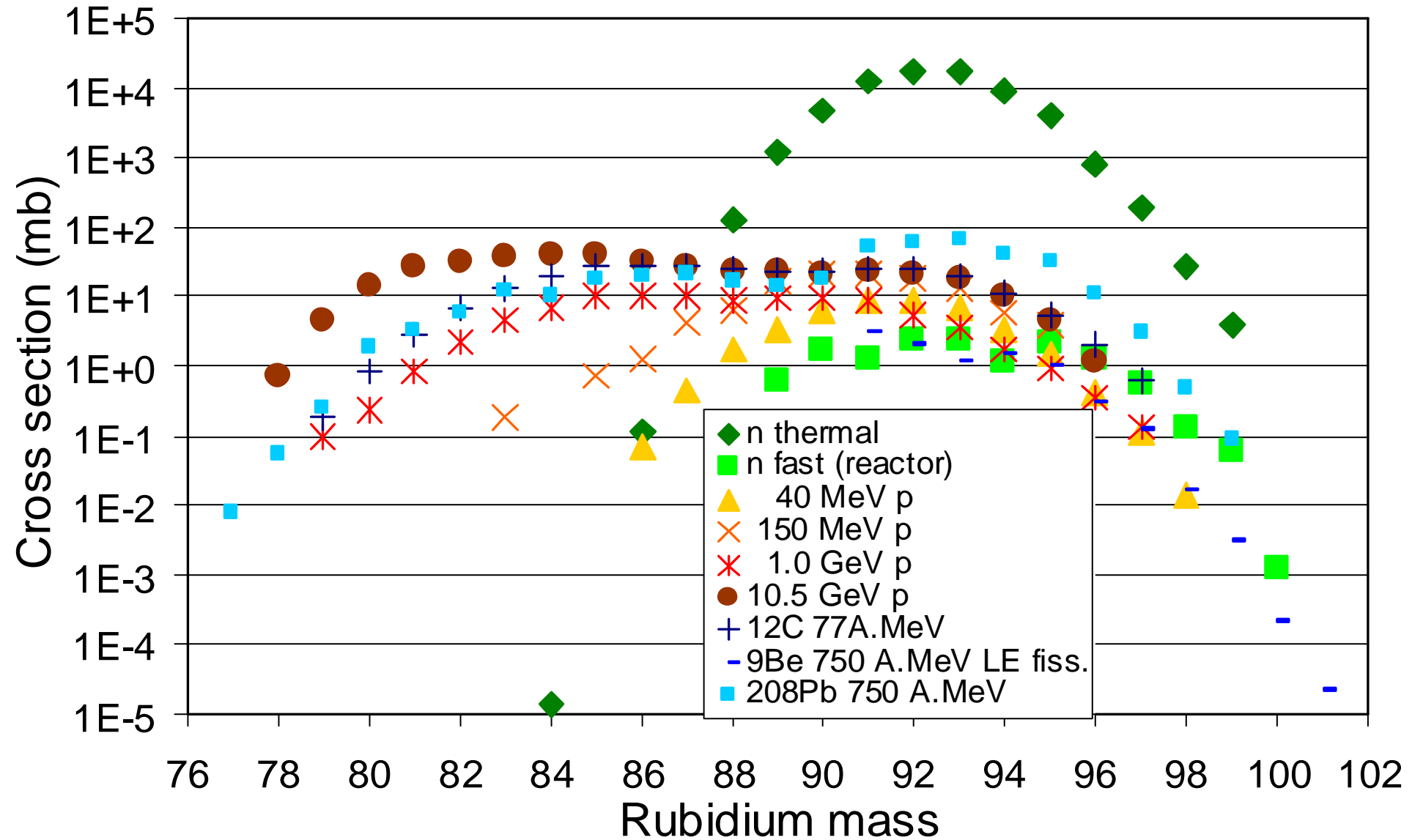
“Low-energy” fission ($^{238}\text{U}(\gamma, f)$ from 50 MeV e^-)



High-energy fission (500 MeV p on ^{238}U)



Rubidium cross-sections



Nuclear reactions

5. Fragmentation

- many cross-sections show little energy dependence in the region 40-2000 MeV/nucleon
- target fragmentation needs high energy protons (see spallation)
- projectile fragmentation needs high energy heavy ions
⇒ huge cyclotron, synchrotron or LINAC

6. Fission

- induced by: “time” (**spontaneous**), **neutrons**, **photons**, **protons**, **heavy ions**, antiprotons, pions, post fusion-evaporation, beta-decay/EC
- highest cross-sections for thermal neutrons
- with increasing excitation energy symmetric and far asymmetric fission is favored, but the products get in average less neutron-rich!
- driver accelerators: reactors, medium-energy (some MeV to tens MeV) deuterons from cyclotron or LINAC, microtron or LINAC for electron beams,...

Radioactive ion beam facilities for fission products

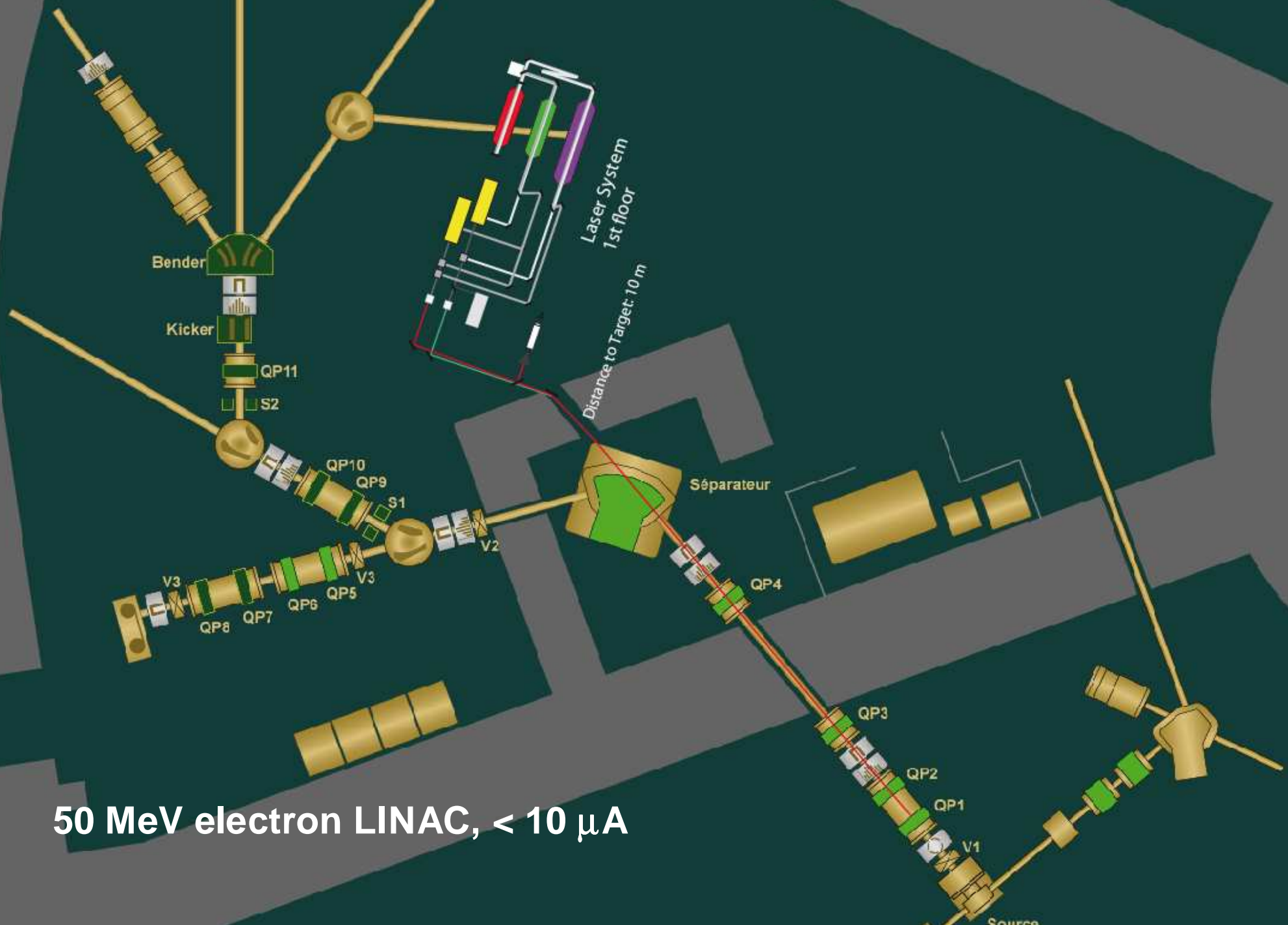
Previous, **presently operating** and **future** RIB facilities using fission:

$^{252}\text{Cf}(\text{sf})$	CARIBU
$^{235}\text{U}(n_{\text{th}},\text{f})$	OSTIS, OSIRIS, LOHENGRIN, TRIGA-SPEC, CARR-ISOL, PIAFE, MAFF, PIK-ISOL
$^{238}\text{U}(\text{p},\text{f})$	ISOLDE, IRIS, LISOL, JYFL, HRIBF, TRIAC, ISAC-II, SPES
$\text{W}(\text{p},\text{xn}..) > ^{238}\text{U}(\text{n},\text{f})$	ISOLDE, IRIS, ISAC-II, EURISOL
$^{12}\text{C}(\text{d},\text{n}) > ^{238}\text{U}(\text{n},\text{f})$	PARRNe, SPIRAL-II
$^2\text{H}(\text{d},\text{n}) > ^{238}\text{U}(\text{n},\text{f})$	SPIRAL-II
$^9\text{Be}(\text{d},\text{n}) > ^{238}\text{U}(\text{n},\text{f})$	PARRNe
$^7\text{Li}(\text{d},\text{n}) > ^{238}\text{U}(\text{n},\text{f})$	FRIB
$\text{W}(\text{e}^-, \gamma) > ^{238}\text{U}(\gamma, \text{f})$	ALTO, DRIBS, HRIBF upgrade, ISAC upgrade
$^1\text{H}, ^9\text{Be}..^{208}\text{Pb}(^{238}\text{U},\text{f})$	GSI-FRS, RIKEN, FRIB, FAIR

ISOL facilities using fission

Facility	Location	Target		Driver beam			Fiss. rate per s
			g/cm ²	Type	MeV	uA	
TRIGA-SPEC	Uni Mainz, D	249Cf	3E-4	(n,f)	3E-8	"0.03"	2E+08
CARIBU	Argonne, US	252Cf	nr	sf	nr	nr	1E+09
ALTO	Orsay, F	238U	40	(g,f)	50	10	8E+10
TRIAC	Tokai, JP	238U	1	(p,f)	36	3	1E+11
IGISOL	Jyväskylä, FIN	238U	0.12	(p,f)	30	10	1E+11
HRIBF	Oak Ridge, US	238U	2.1	(p,f)	42	10	4E+11
ISOLDE	CERN, CH	238U	50	(p,f)	1400	2	2E+12
CARR-ISOL	Beijing, CN	235U	3E-2	(n,f)	3E-8	"32"	7E+12
SPES	Legnaro, I	238U	2.5	(p,f)	40	200	1E+13
ISAC2	Vancouver, CAN	238U	(40)	(g,f)	50	10000	5E+13
SPIRAL2	Caen, F	238U	(40)	d>(n,f)	40	5000	<1E14

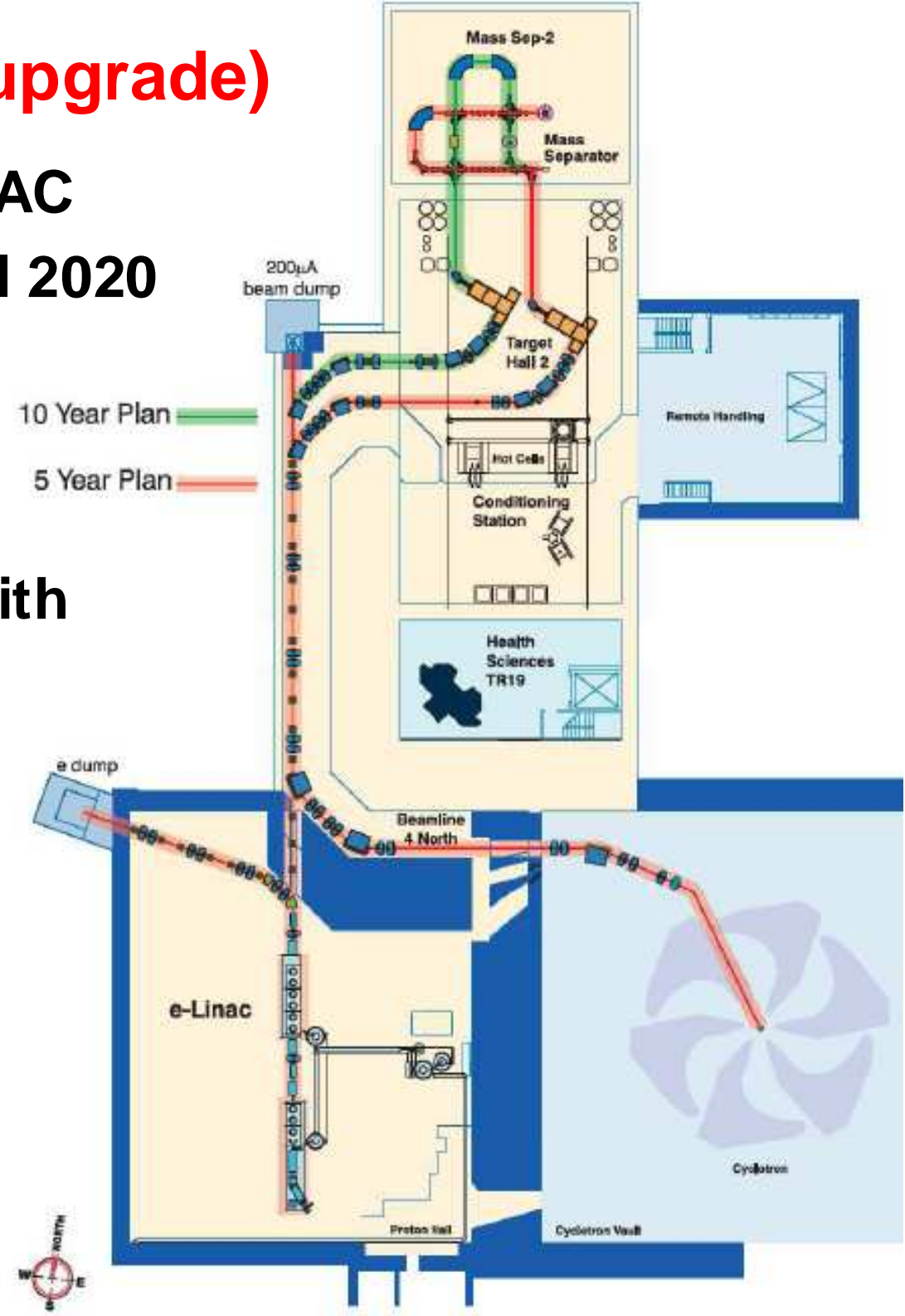
ALTO @ IPN Orsay



50 MeV electron LINAC, $< 10 \mu\text{A}$

ISAC @ TRIUMF (fission upgrade)

- 50 MeV, 10 mA electron LINAC
- <100 kW till 2015, 500 kW till 2020
- aim $4.6E13$ fissions/s with liquid Hg converter
- but also 500 MeV protons with maximum $10 \mu\text{A}$ on UC_x



SPIRAL2 facility at GANIL

GANIL/SPIRAL 1 today

SP2 Beam time: 44 weeks/y
GANIL Beam time: 35 weeks/y
ISOL RIB Beams: 28-33 weeks/y
GANIL+SP 2 Users: 700-800/y

DESIR Facility
low energy RIB

CIME cyclotron RIB at 1-20 AMeV
(up to 9 AMeV for fiss. fragments)

HRS+RFQ Cooler

S3 separator-
spectrometer

**RIB Production Cave
Up to 10^{14} fiss./sec.**

LINAC: 33MeV p, 40 MeV d, 14.5 A MeV HI

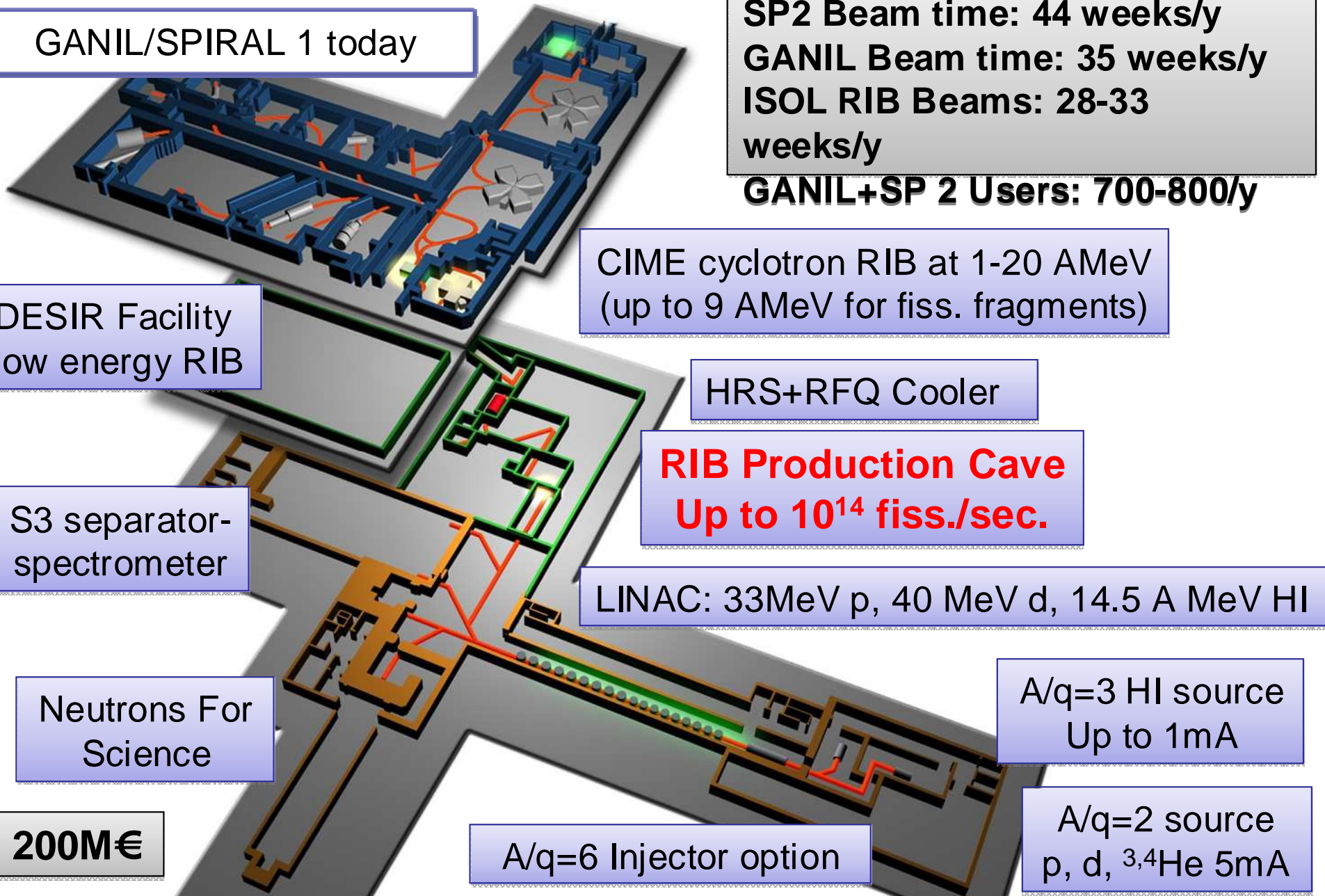
Neutrons For
Science

A/q=3 HI source
Up to 1mA

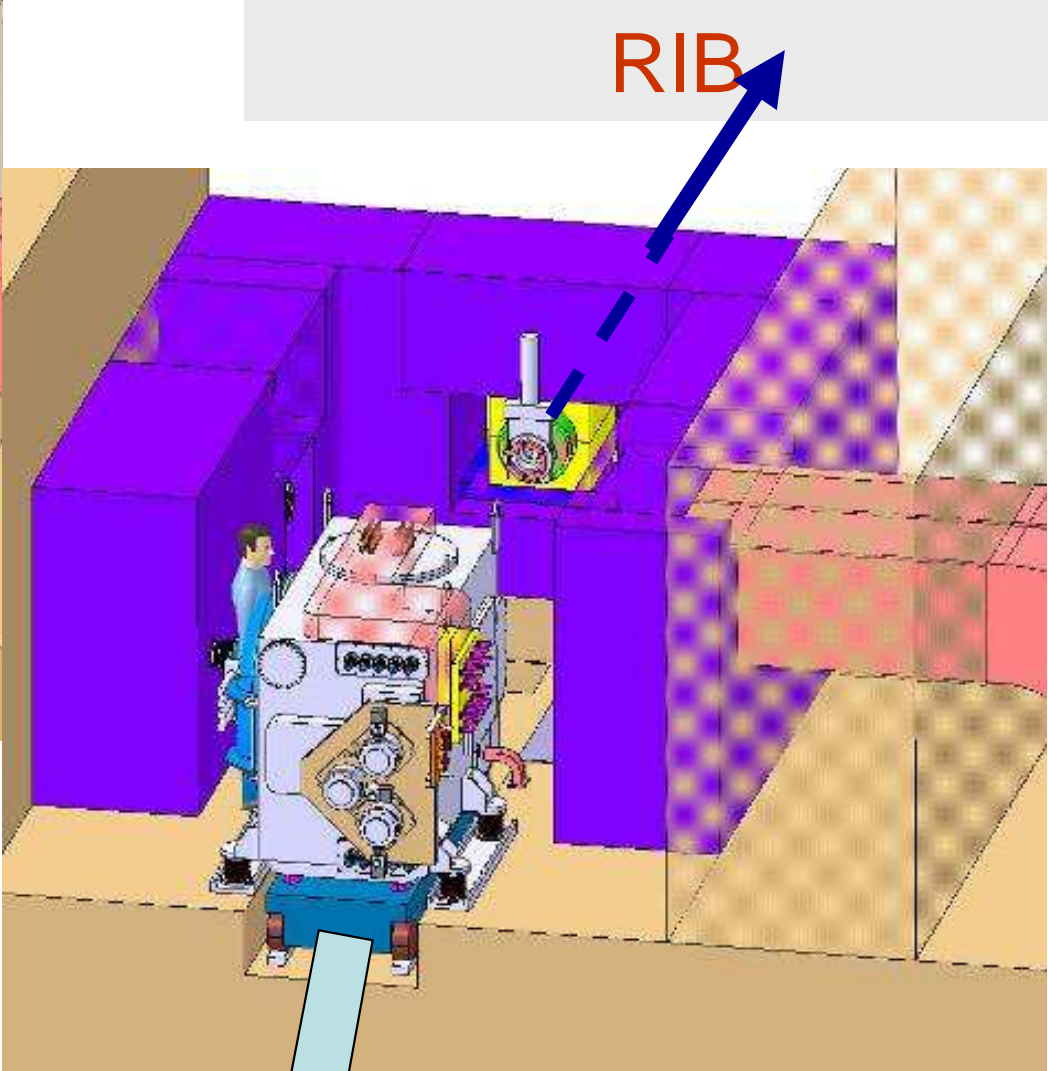
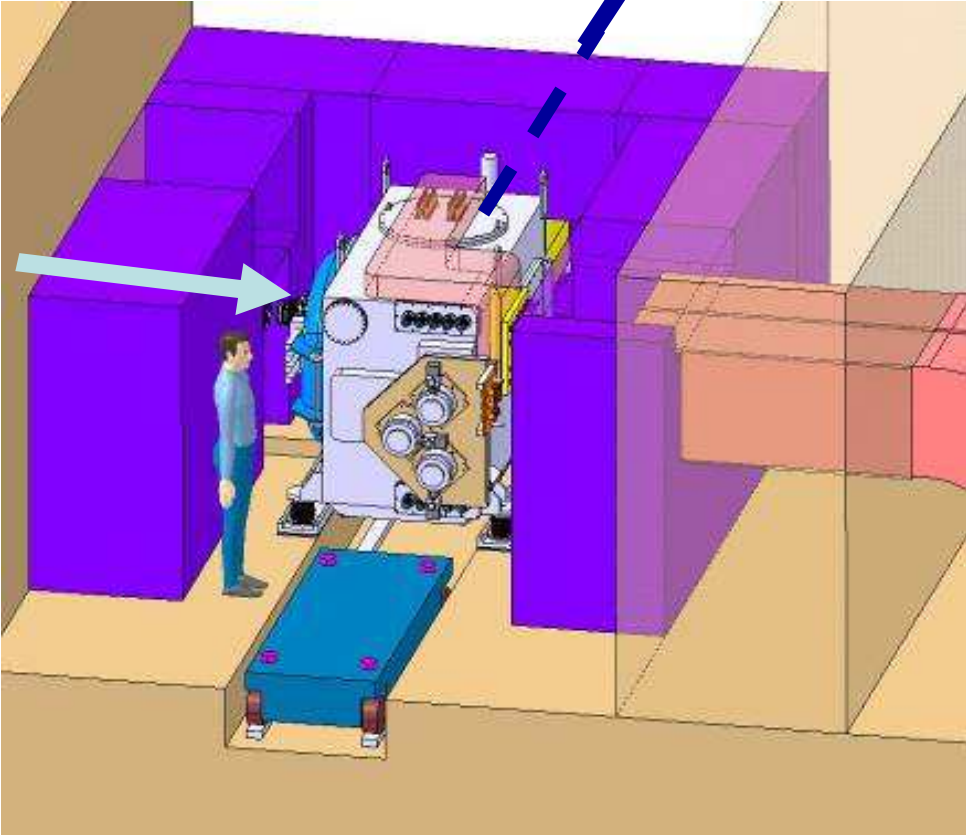
Cost: 200M€

A/q=6 Injector option

A/q=2 source
p, d, ^3He , ^4He 5mA



SPIRAL2 RIB production module



Maintenance and Storage cells



EURISOL



1-2.2 GeV, multi-MW proton driver

Several direct target stations (ca. 100 kW)

One Hg spallation + fission target station (>1 MW, i.e. $1E15$ fissions/s)

Multiple user operation in parallel

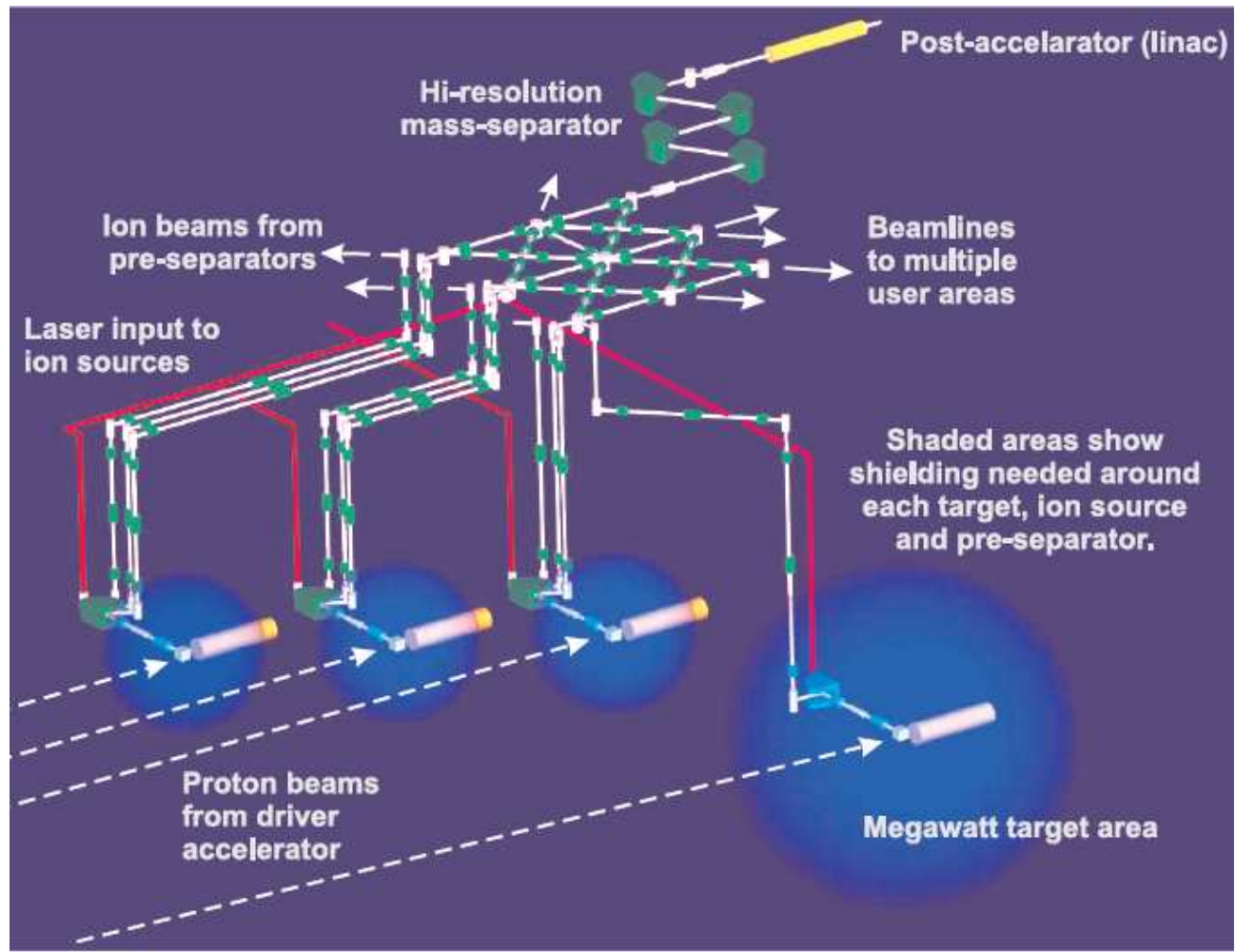
Low-energy beam area

**Post-acceleration with
LINAC up to ca. 10 A.MeV**

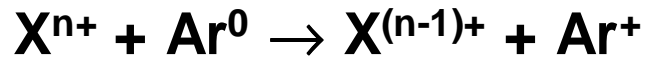
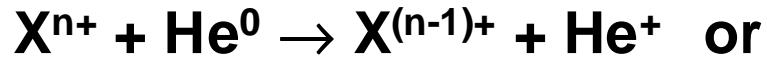
**Post-acceleration to
ca. 100 A.MeV
with LINAC or cyclotron**

**Fragmentation of
post-accelerated RIBs**

Commissioning: > 2020?



IGISOL method

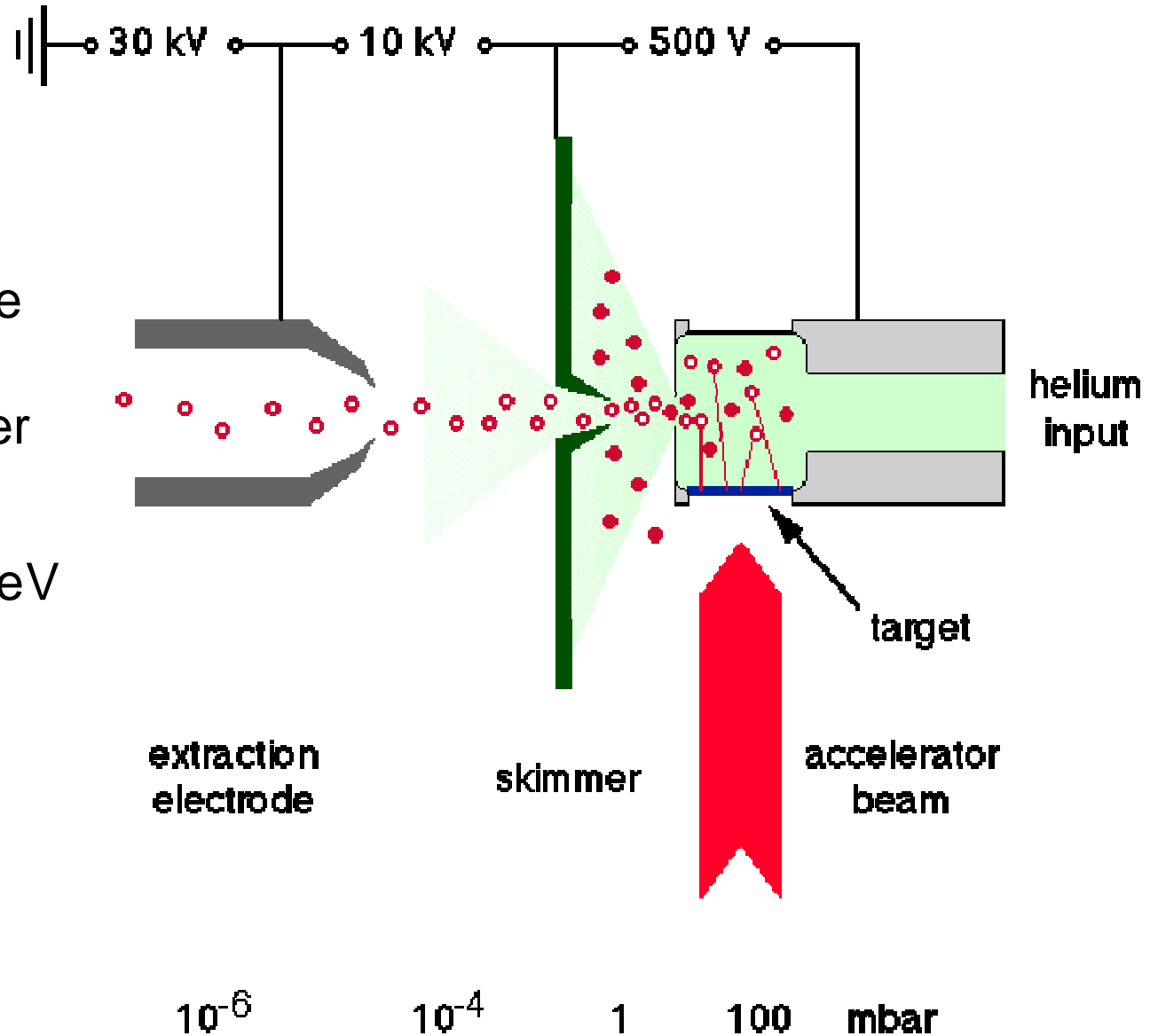


rapid reduction of ionic charge state to 2+ or 1+ by charge exchange reactions with buffer gas

IP(He)=24.6 eV, IP(Ar)=15.8 eV



remains in 1+ or 2+ charge state until charge exchange reaction with impurity molecule (O₂, N₂,...) occurs

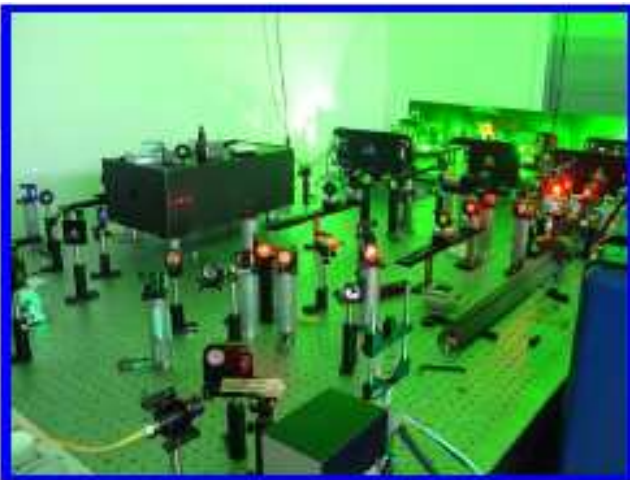


Volatility of the elements

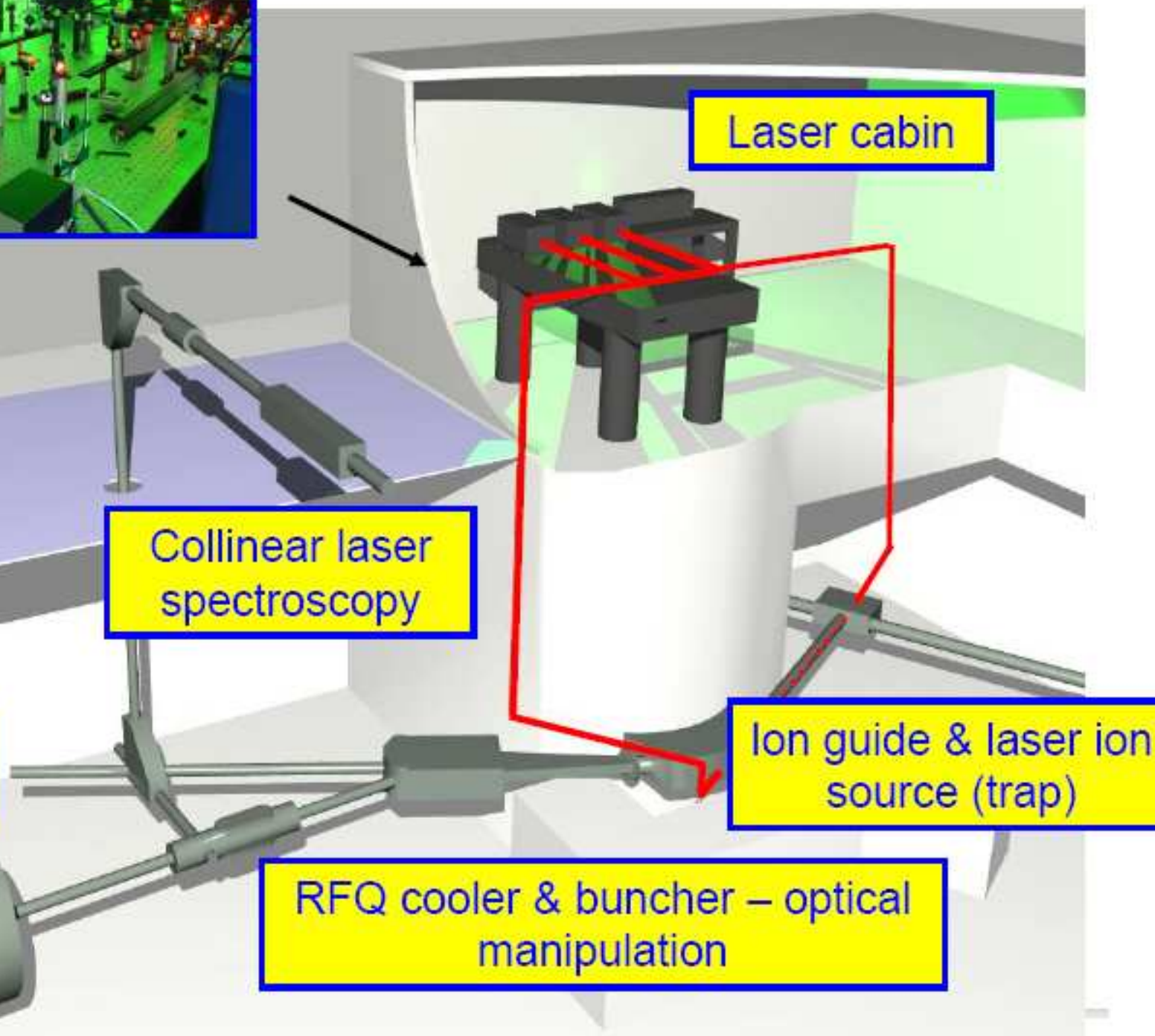
1	T (p vapor > 0.01 mbar) < 100 °C																2						
H	T (p vapor > 0.01 mbar) < 400 °C																He						
3	4	T (p vapor > 0.01 mbar) < 1000 °C																5	6	7	8	9	10
Li	Be	T (p vapor > 0.01 mbar) < 2000 °C																B	C	N	O	F	Ne
11	12	T (p vapor > 0.01 mbar) > 2000 °C																13	14	15	16	17	18
Na	Mg																	Al	Si	P	S	Cl	Ar
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36						
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr						
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54						
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe						
55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86						
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn						
87	88	89	104	105	106	107	108	109	110	111	112												
Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt															

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

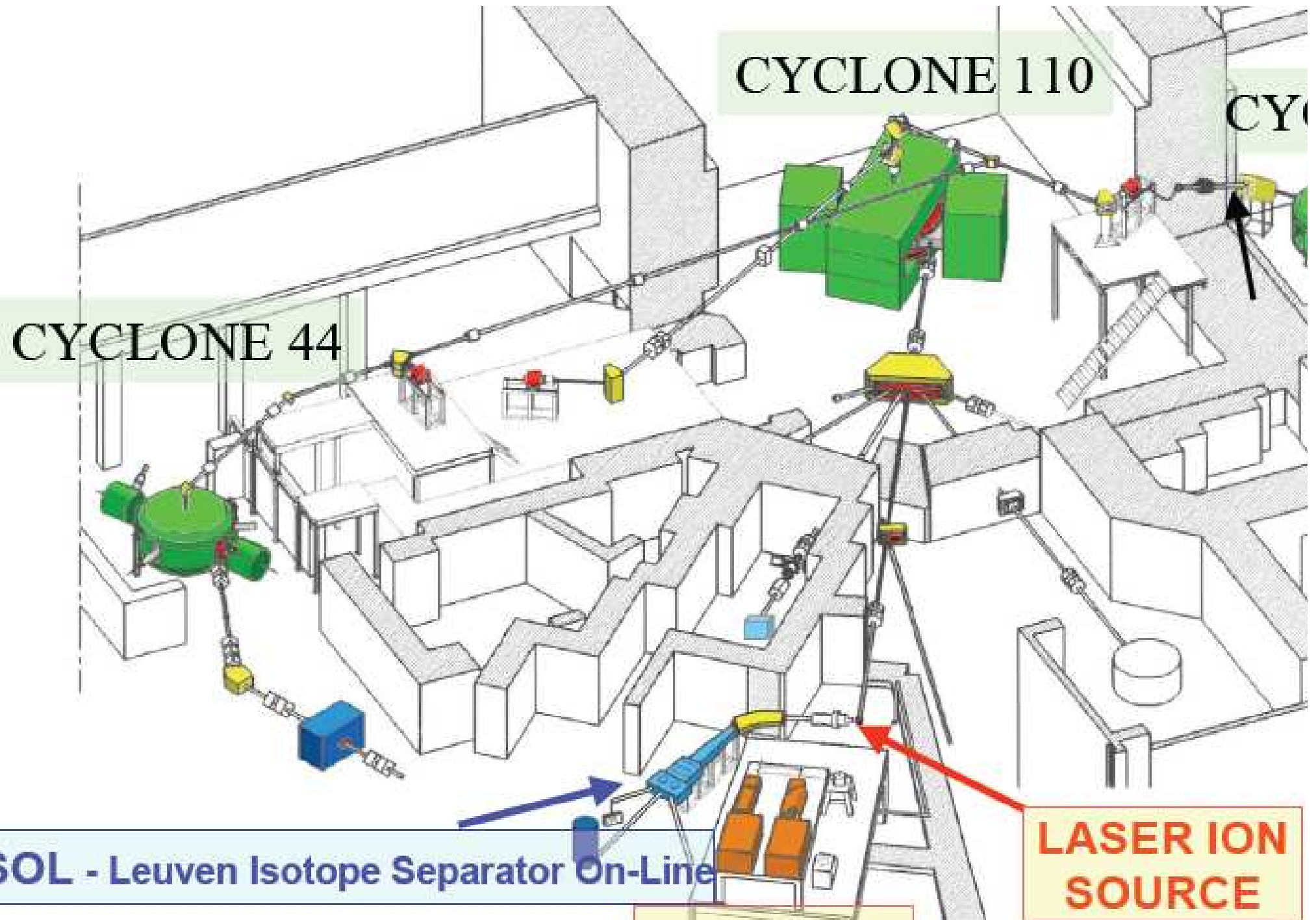
The IGISOL facility at JYFL



Mass & decay spectroscopy



LISOL

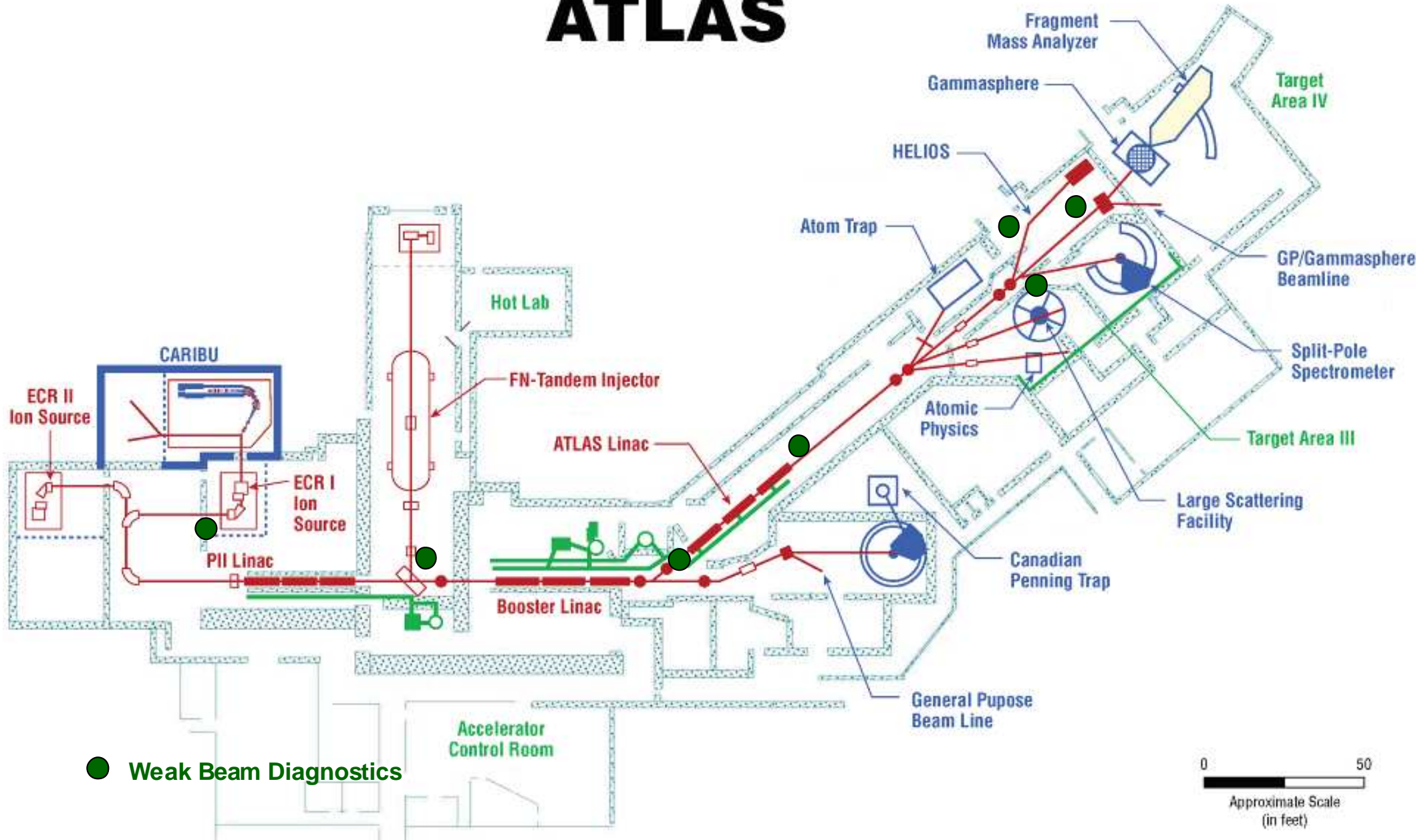


LISOL - Leuven Isotope Separator On-Line

LASER ION SOURCE

CARIBU: Radioactive Beams from $^{252}\text{Cf}(\text{sf})$

ATLAS



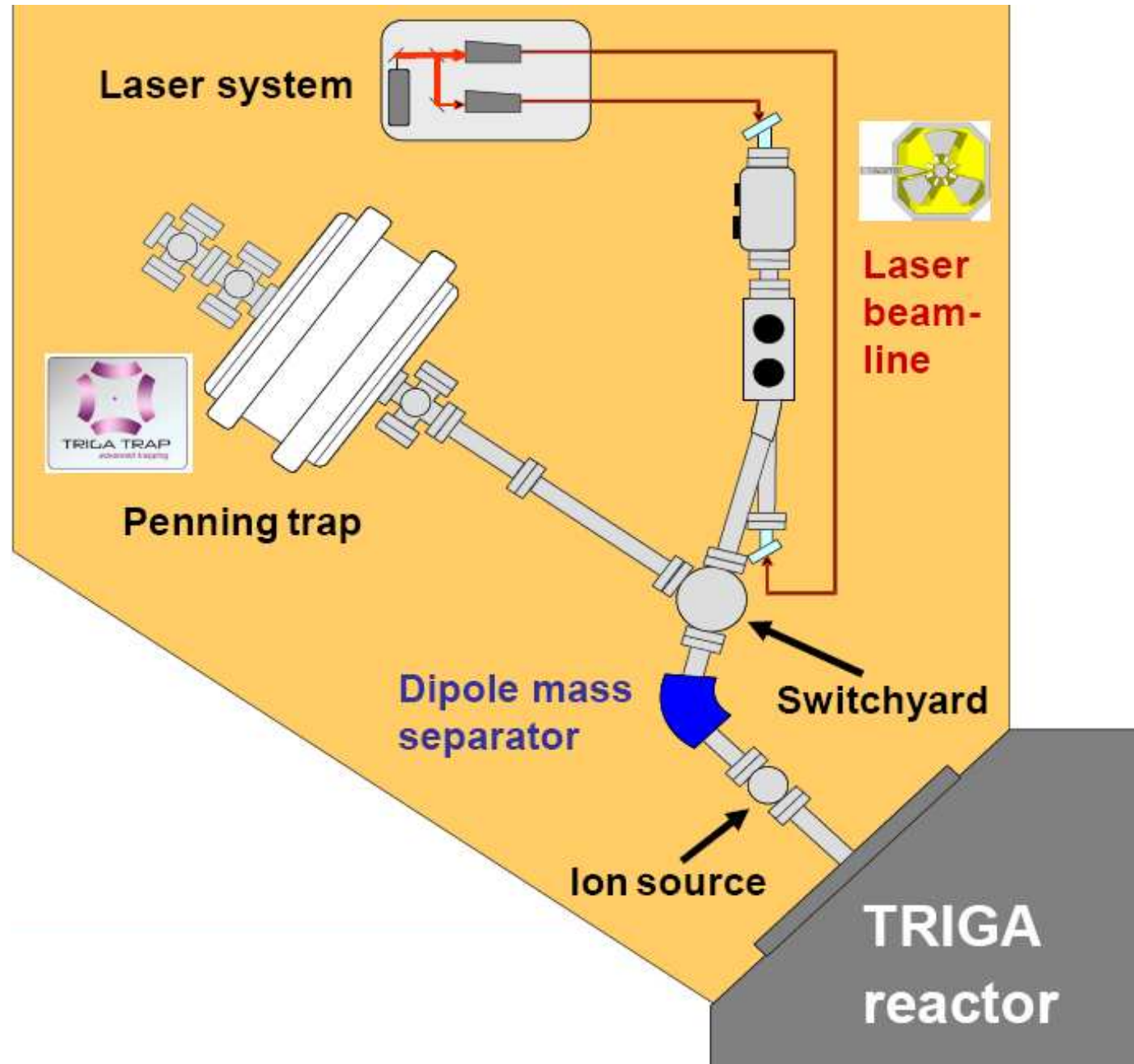
● Weak Beam Diagnostics

0 50
Approximate Scale
(in feet)

1 Ci ^{252}Cf source (1E9 fiss./s) expected end of 2009

TRIGA-SPEC at Mainz reactor

- 0.5 mg ^{235}U or 0.5 mg ^{239}Pu or 0.3 mg ^{249}Cf
- $1.8\text{E}11$ n./cm²/s
- $2\text{E}8$ fiss./s
- start 2009



ISOL@CARR

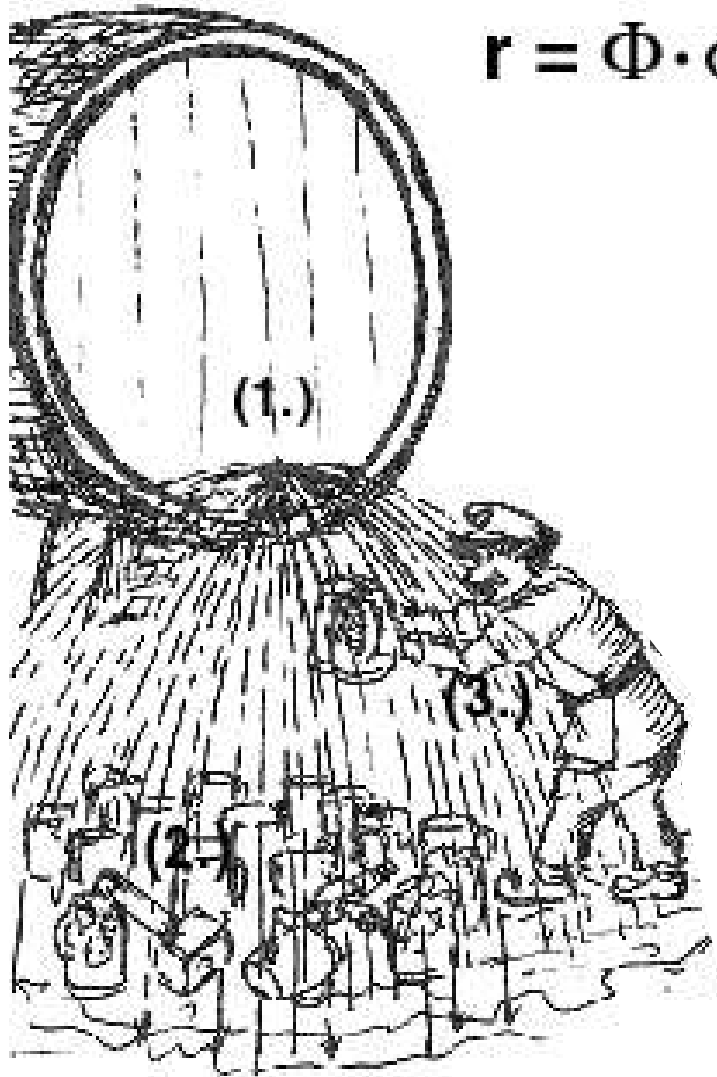
- 60 MW China Advanced Research Reactor (CARR).
- Expected start of operation in 2009.
- gas-jet ISOL system
- 25 mg ^{235}U (90%) in $\Phi=2\text{E}14$ n./cm²/s
- 7E12 fiss./s



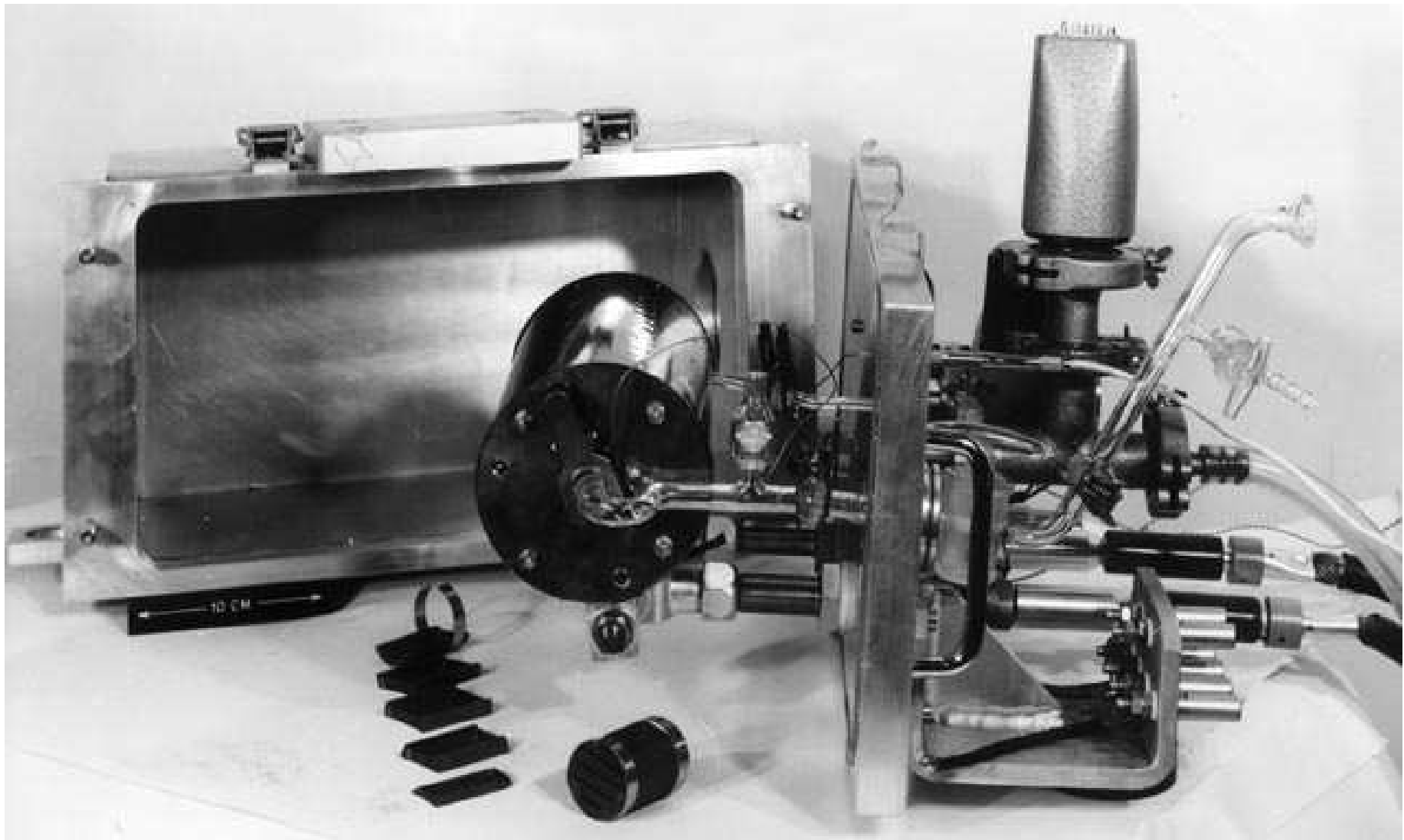
Optimize RIB intensity

All steps of the separation chain need to be optimized!

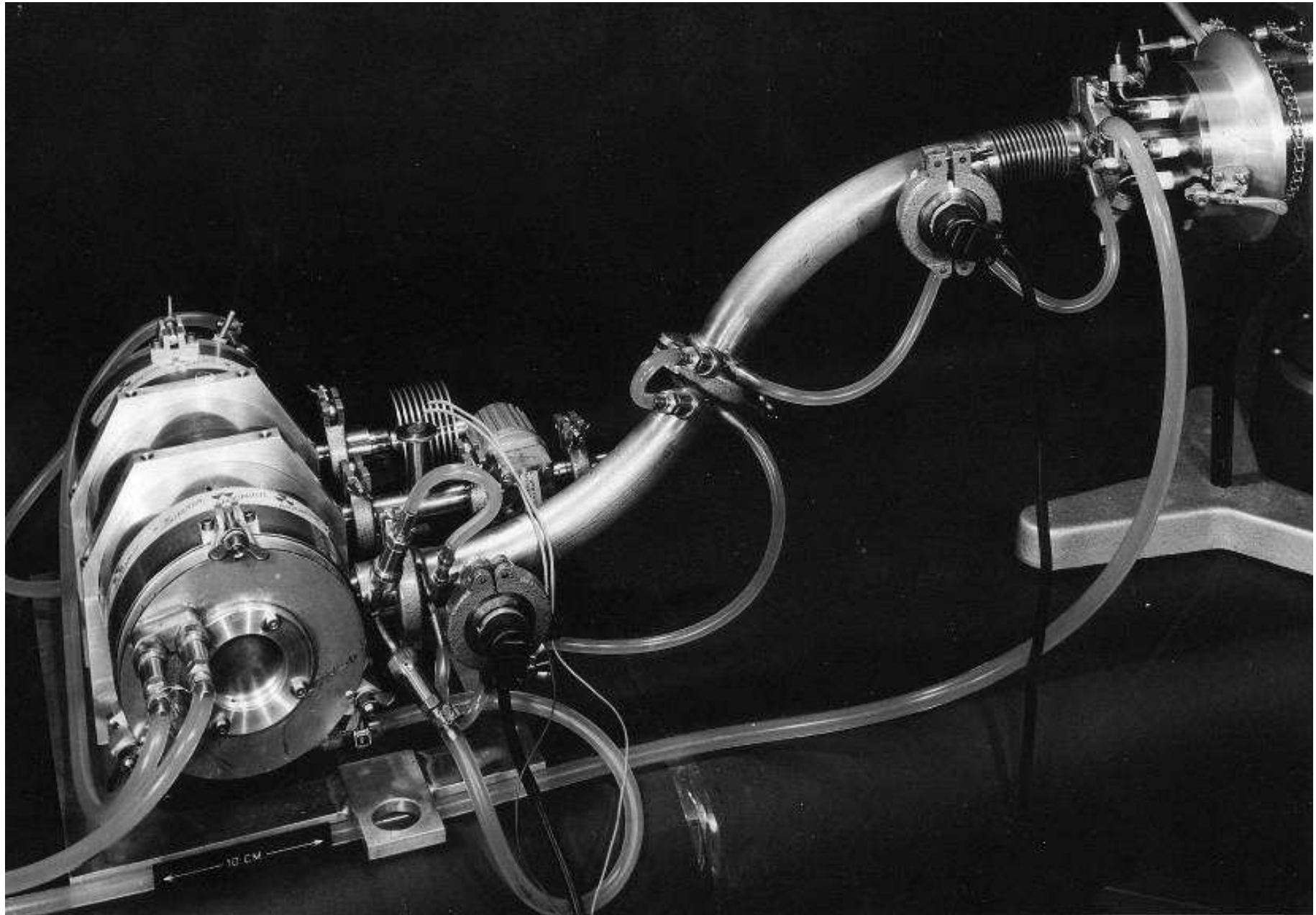
$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$



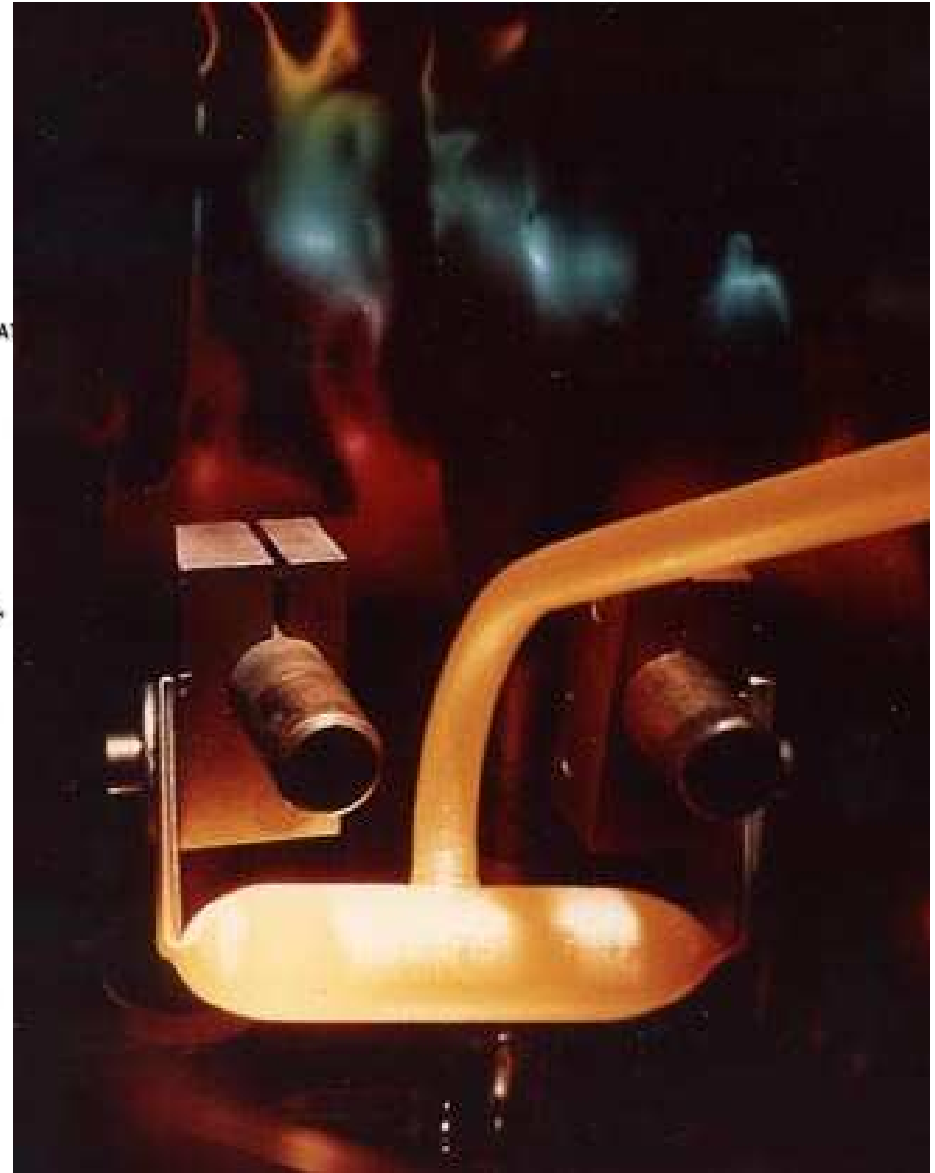
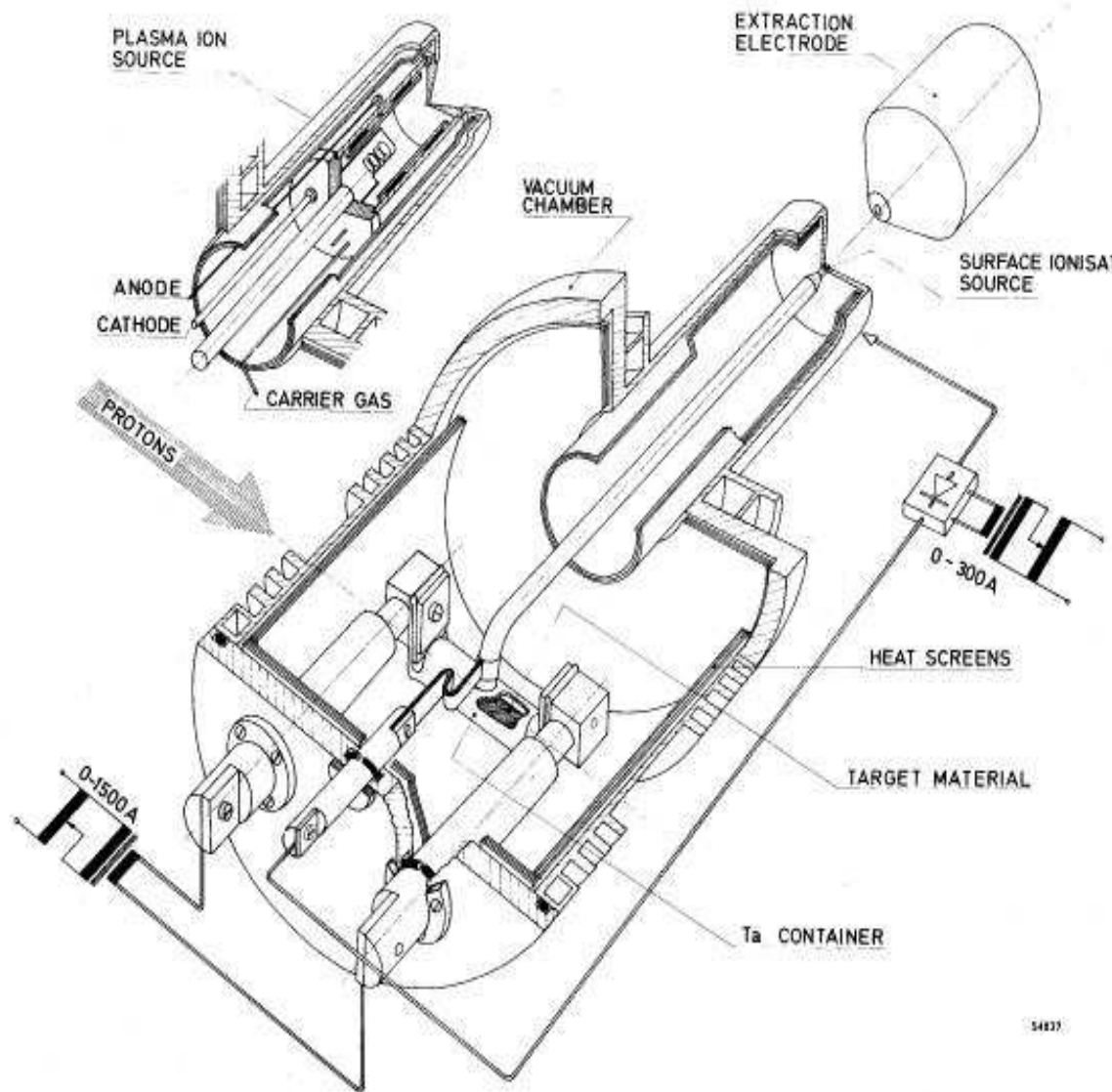
Target (1967)



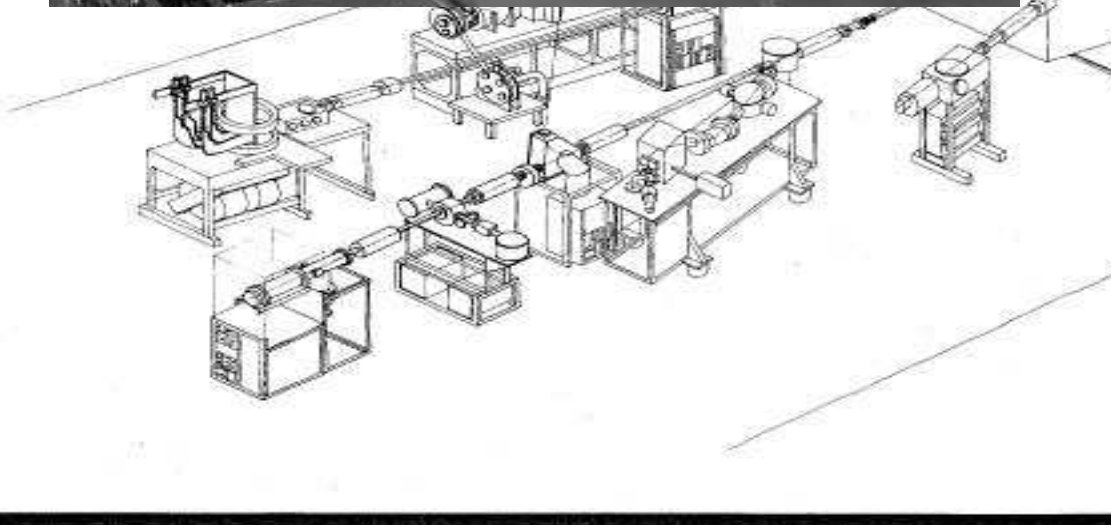
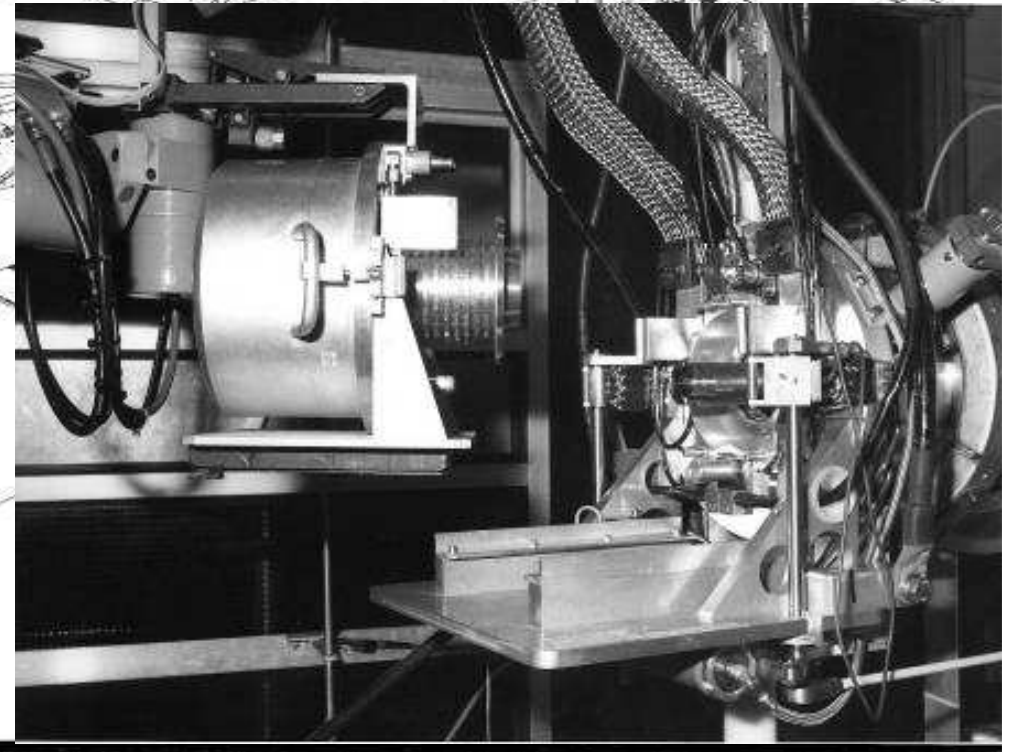
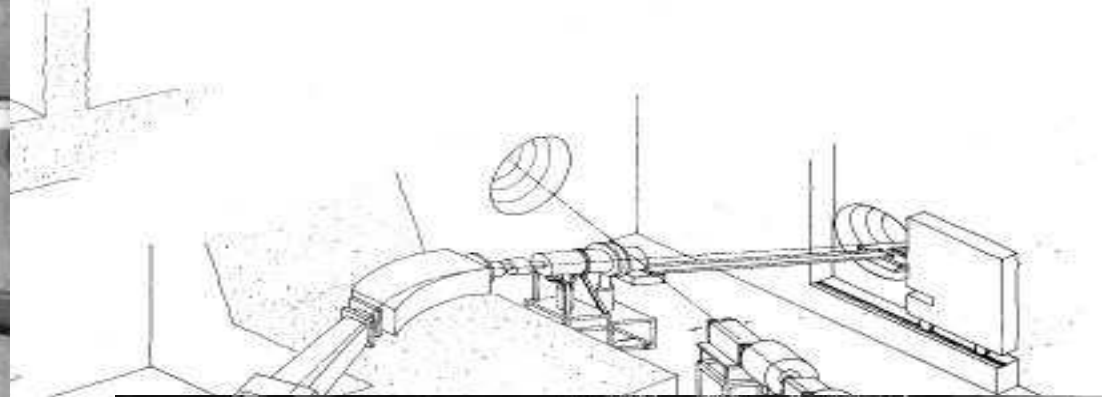
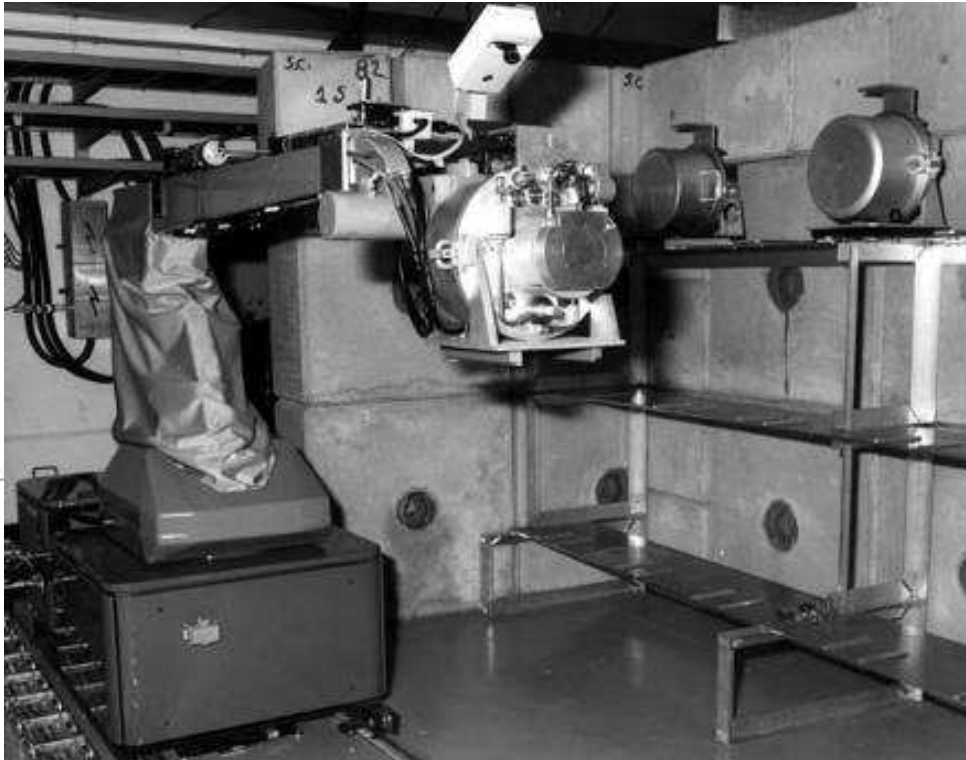
Target and ion source (1968)



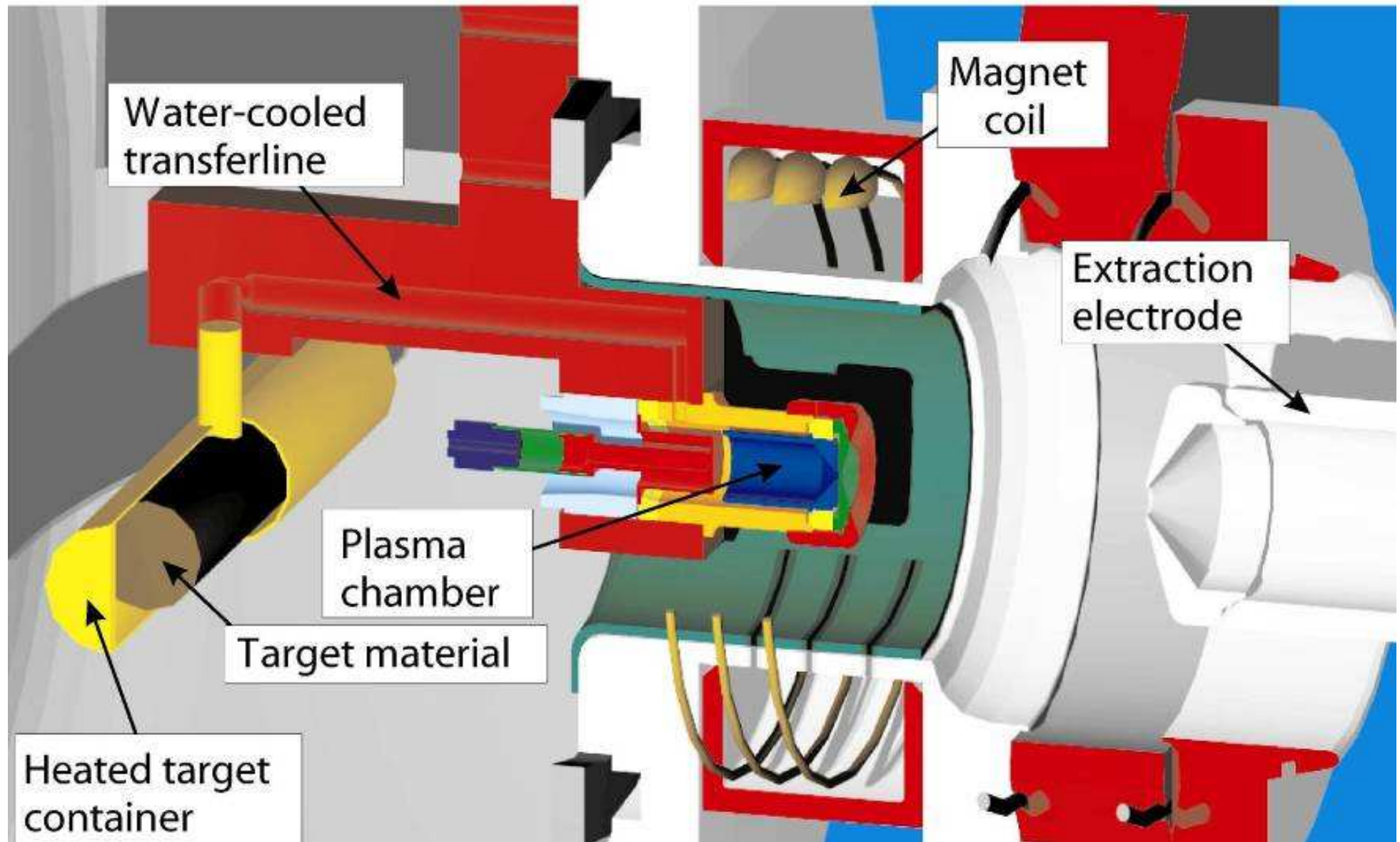
Compact target and ion source (1974)



Robot handling



ISOLDE target and ion source unit



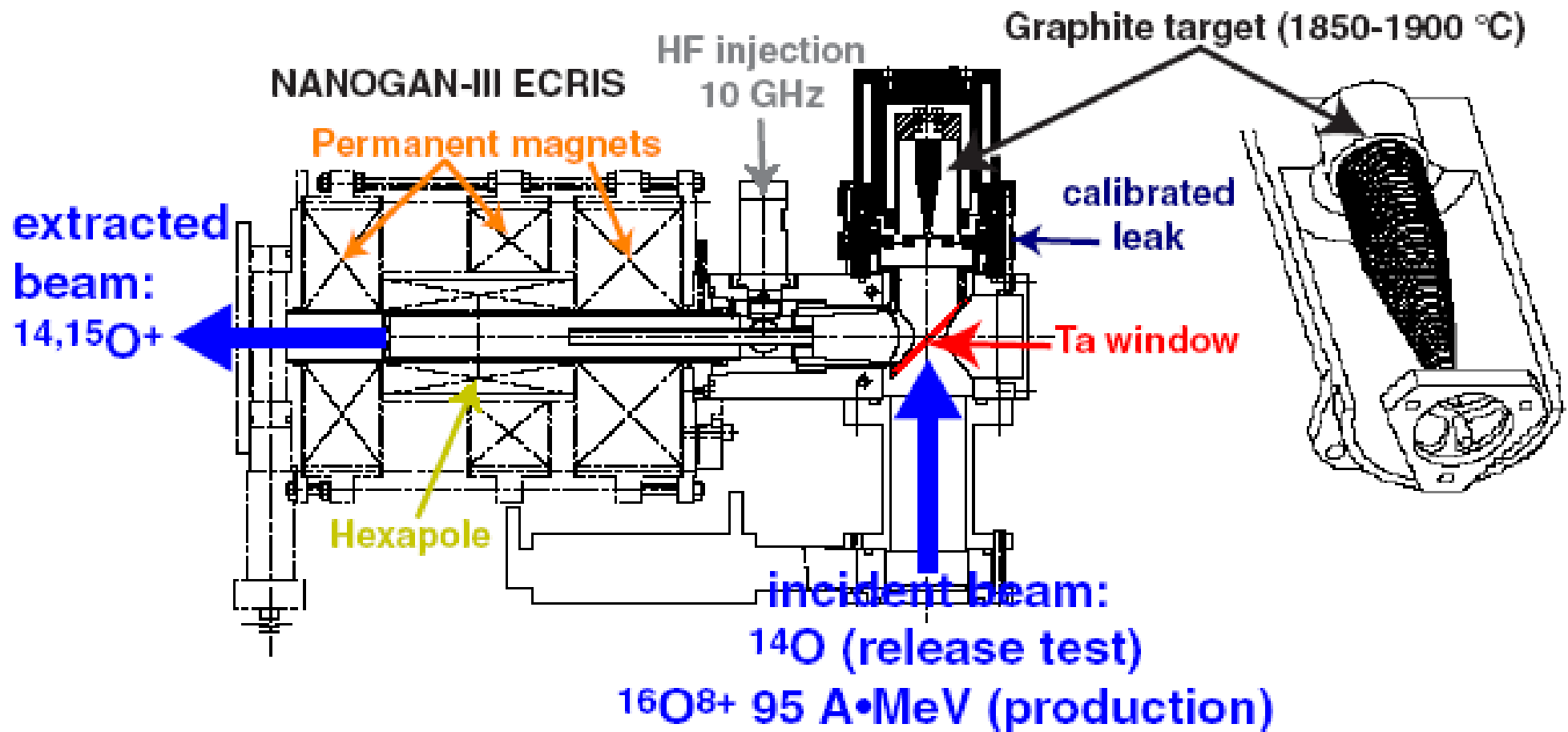
Historical development

Miniaturisation ⇒ **faster release**

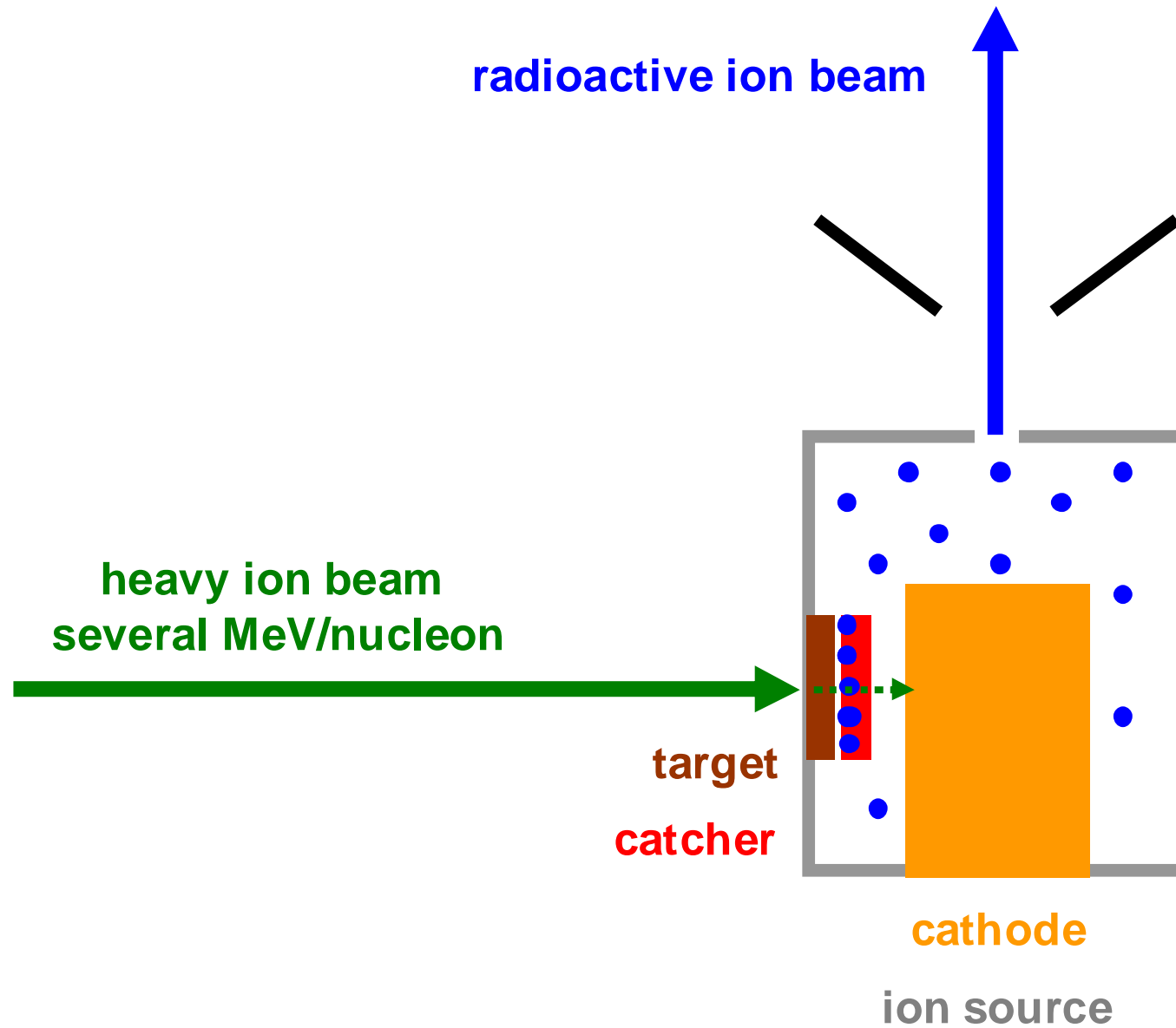
Standardisation ⇒ **easier mass-production**

Remote handling ⇒ **higher activities**

SPIRAL target and ion source unit



GSI-ISOL target and ion source unit



Variants of ISOL facilities

- 1a protons on thick (heavy) target:** fragmentation, spallation, fission
ISOLDE-CERN (1.4 GeV), IRIS-PNPI (1 GeV), ISAC-TRIUMF (0.5 GeV)
- 1b direct reactions in thick target**
CRC Louvain-la-Neuve, HRIBF Oak Ridge, TRIAC Tokai
- 1c fission in thick target**
OSIRIS (Studsvik), HRIBF Oak Ridge, TRIAC Tokai, SPIRAL2 (GANIL)
- 2 projectile fragmentation in thick (carbon) target**
SPIRAL (GANIL), DRIBS (Dubna), EXCYT (LNS Catania)
- 3 fusion-evap. or multinucleon transfer in thin target plus solid catcher**
GSI-ISOL, UNIRIB (ORNL), DOLIS (Daresbury), LISOL (Leuven),
IMP Lanzhou, MASHA (Dubna), SPIRAL2 (GANIL)
- 4 fusion-evap., direct reaction or fission in thin target plus gas catcher
(Ion Guide ISOL = IGISOL)**
IGISOL (Jyväskylä), LISOL (Leuven), ...
- 5 liquid helium catcher**
JYFL Jyväskylä, KVI Groningen

The life cycle of an ISOLDE target







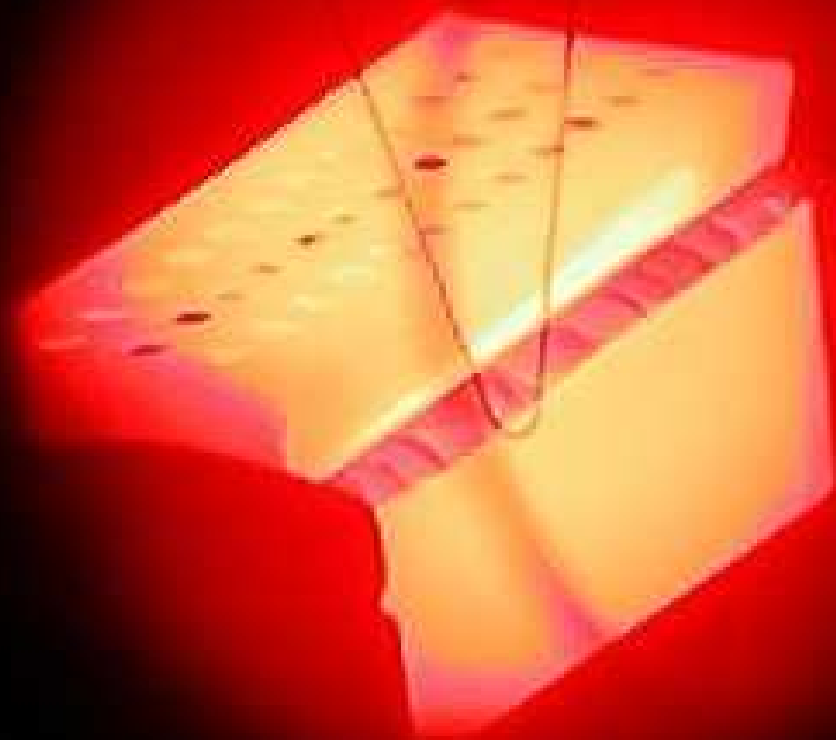


CHEF DE US
BERNARD







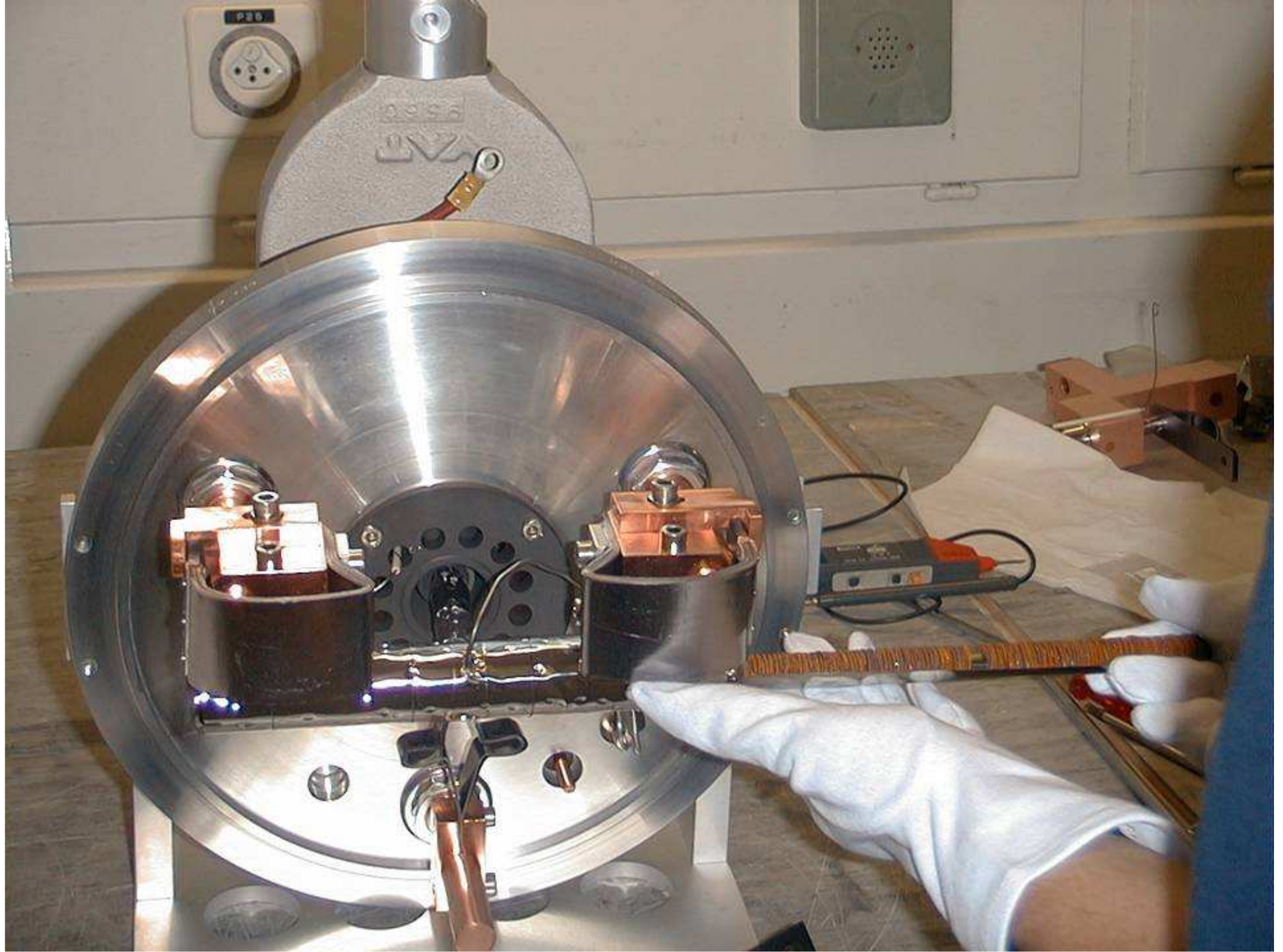











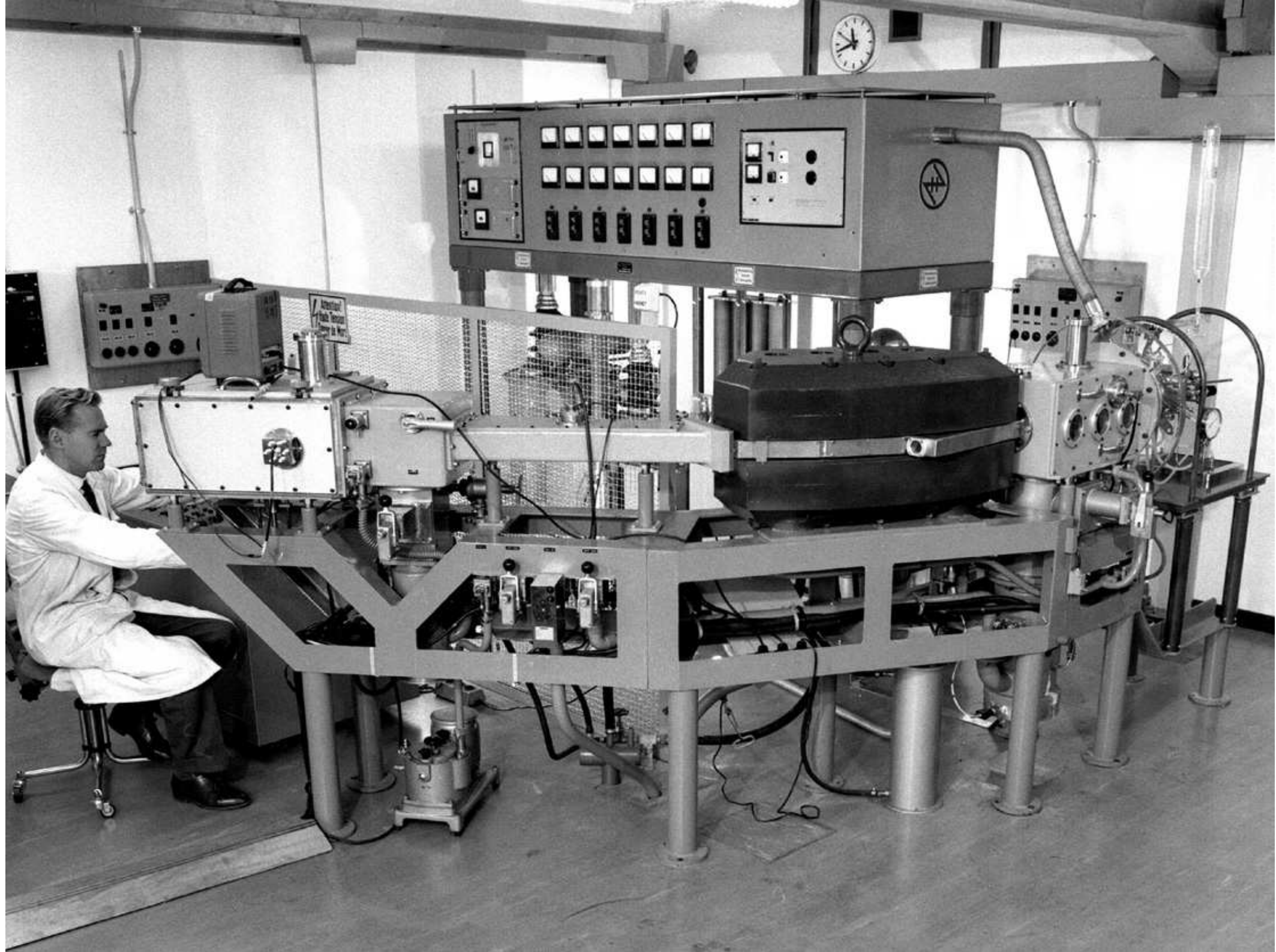




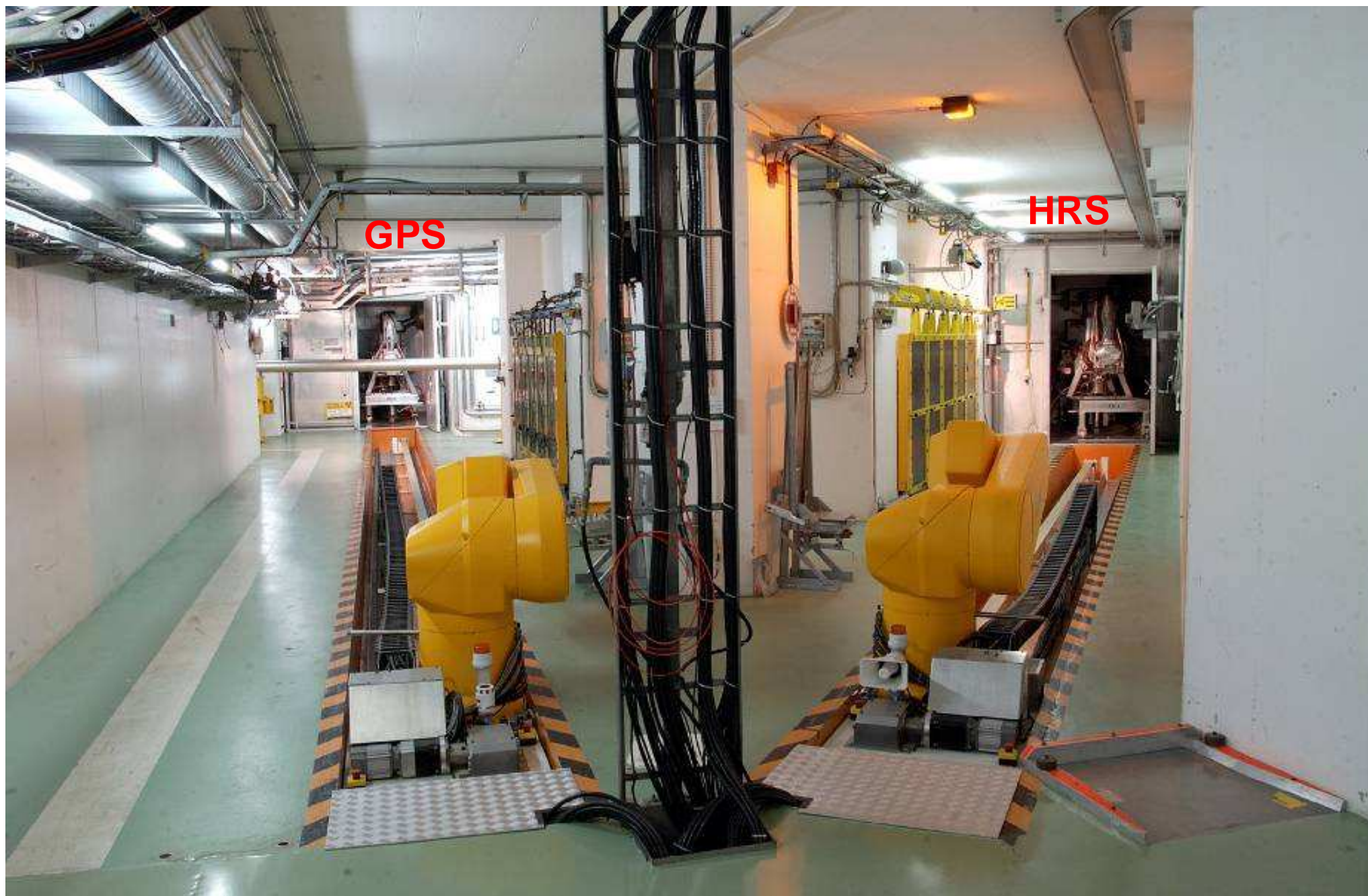
UR D'URGENCE
AGENCY STOP

RADIATION 
contamination of
U+Th (inside)

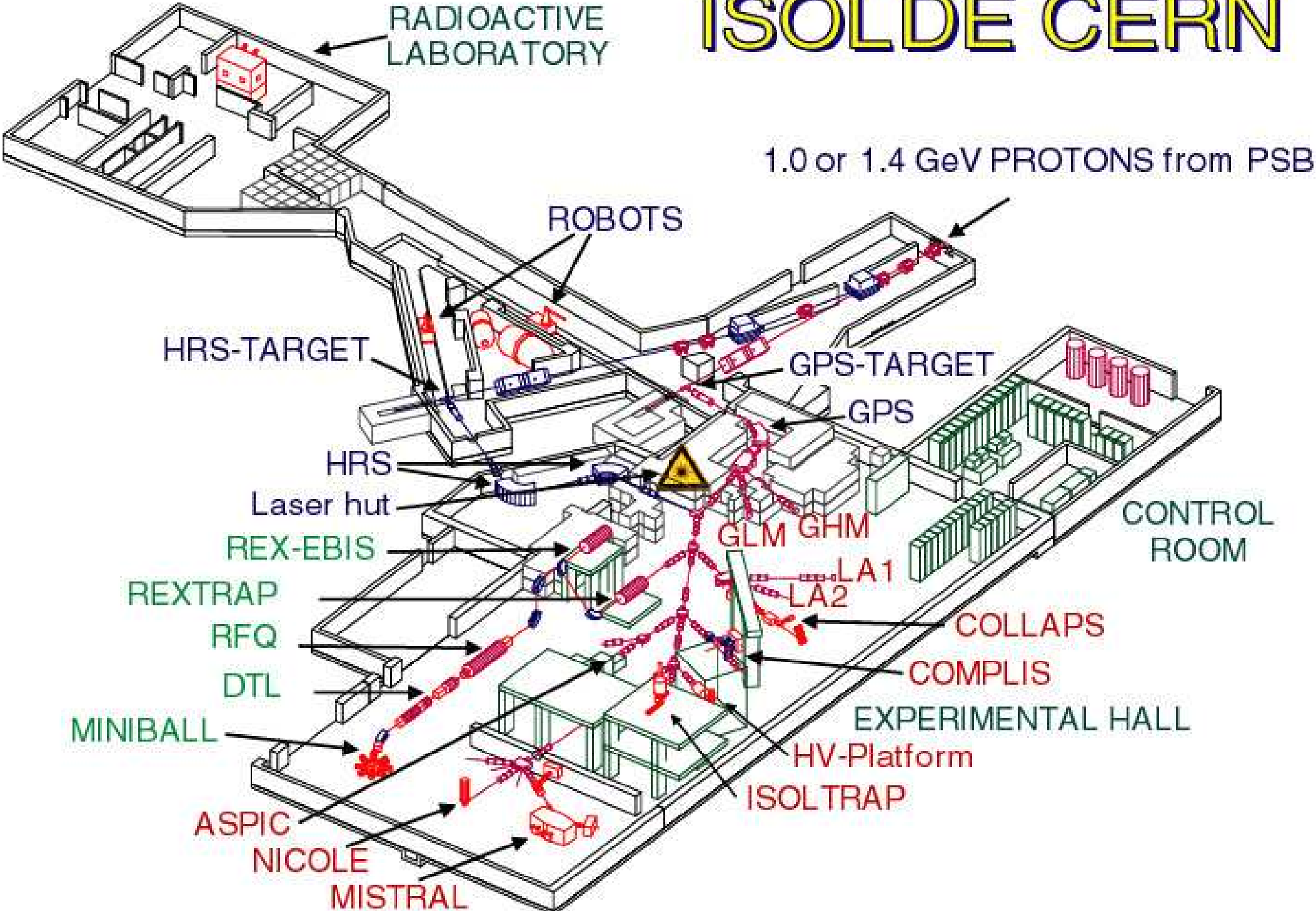




ISOLDE front-ends

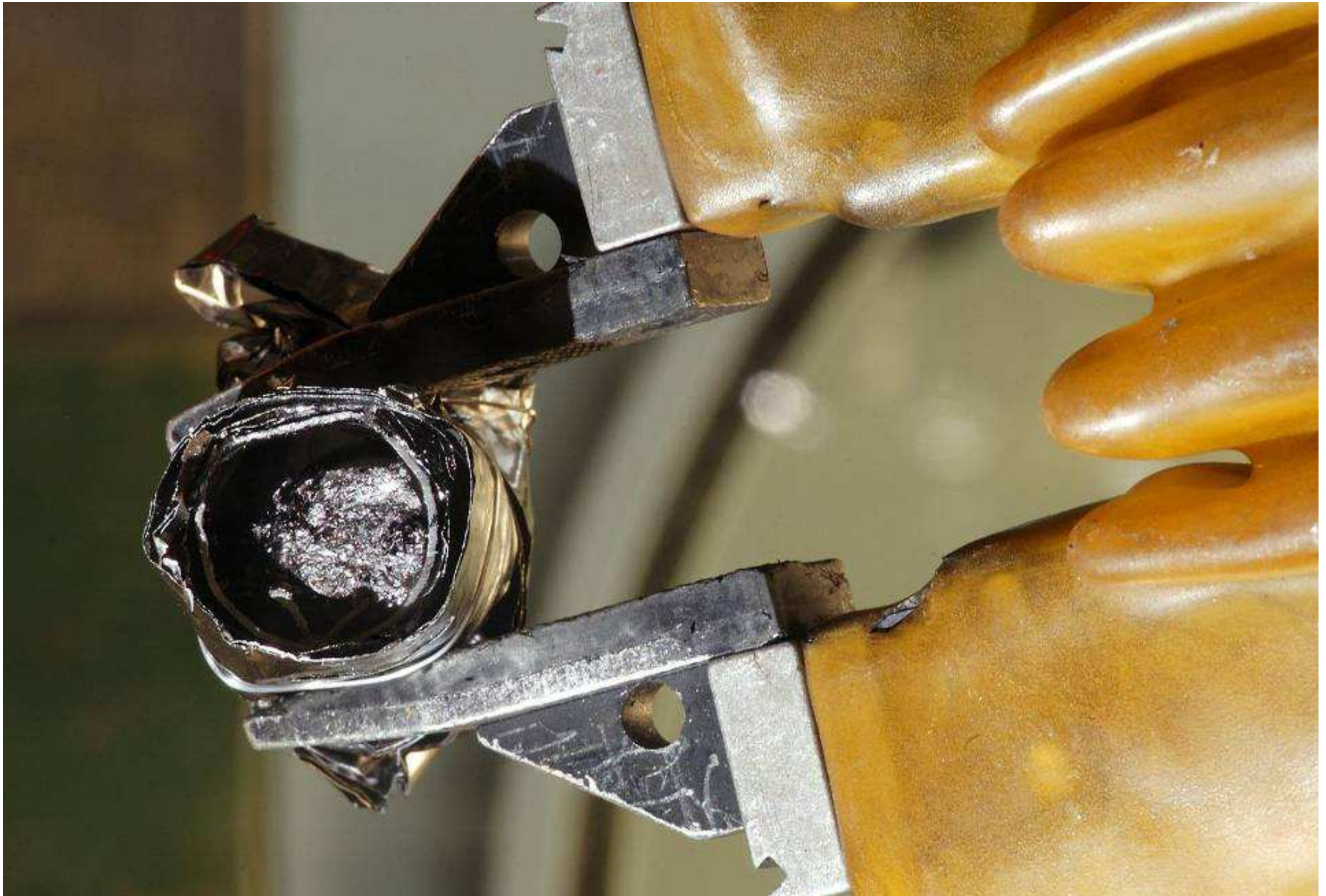


ISOLDE CERN



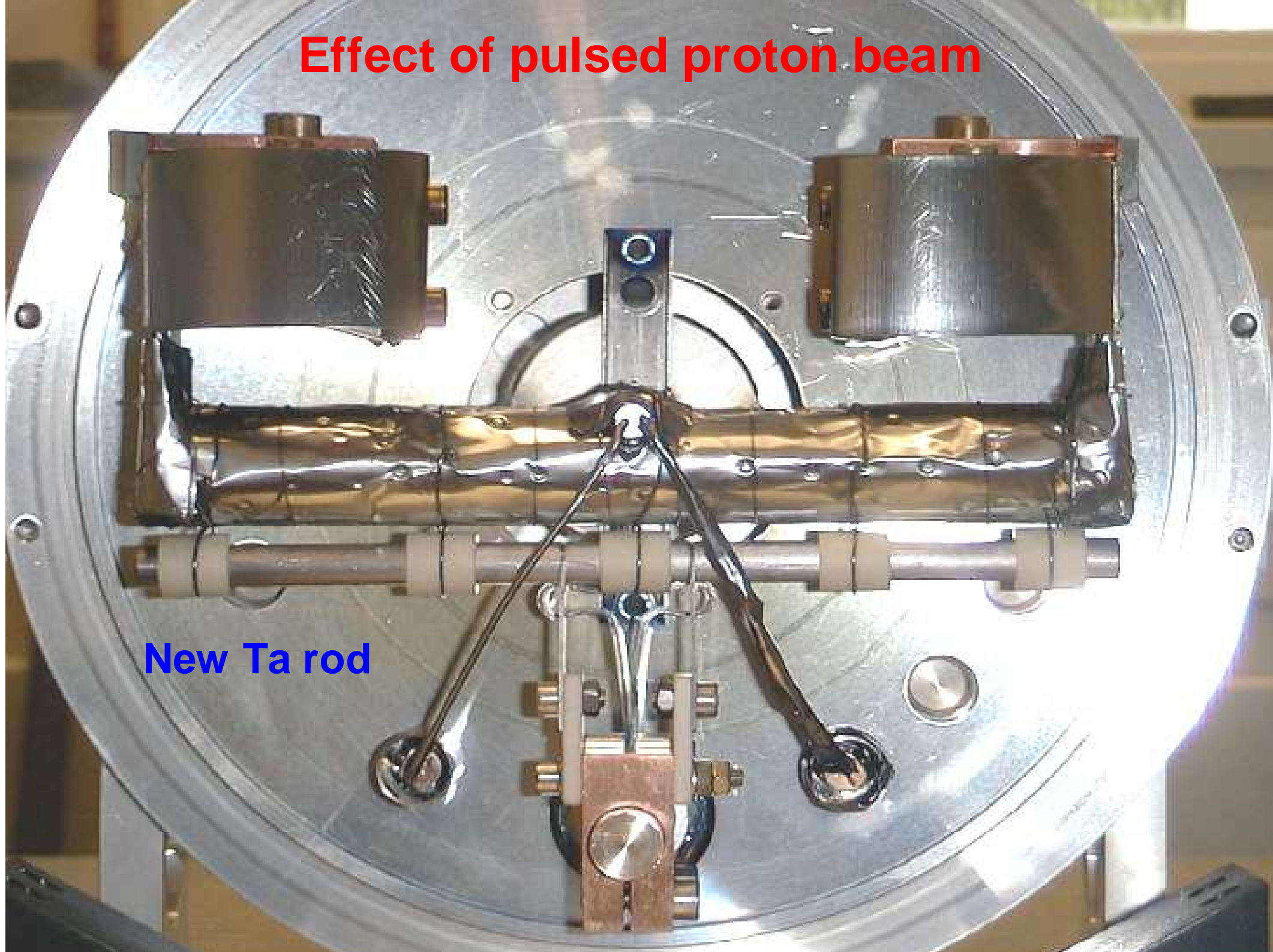


Postmortem examination of used targets

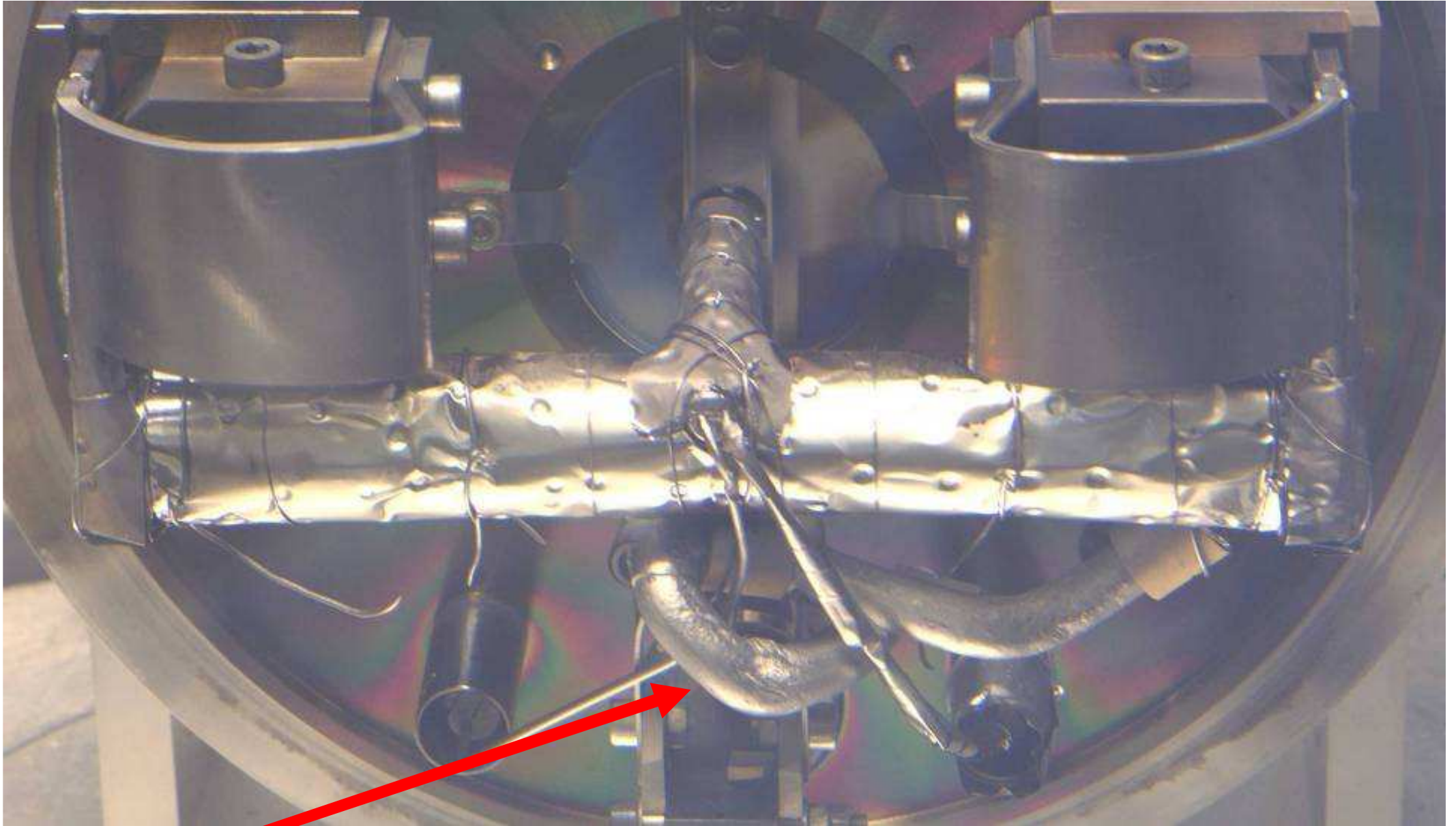


Effect of pulsed proton beam

New Ta rod



Effect of pulsed proton beam



**Ta rod after irradiation with ca. 200000 proton pulses
(1 Coulomb)**

ISOL targets

Target materials:

1. molten metals: Ge, Sn, La, Pb, Bi, U,...
2. solid metals: Ti, Zr, Nb, Mo, Ta, W, Th,...
3. carbides: Al_4C_3 , SiC, VC, ZrC, LaC_x , ThC_x , UC_x ,...
4. oxides: MgO, Al_2O_3 , CaO, TiO_x , ZrO_2 , CeO_x , ThO_2 ,...
5. others: graphite, borides, silicides, sulfides, zeolithes,...

Target dimensions:

target container: 20 cm long, 2 cm diameter

target thickness 2—200 g/cm², **10—100% of bulk density**

micro-dimensions of foils, fibers or pressed powder: **1—30 μm**

U.K. for the ISOLDE Collaboration, Radiochimica Acta 89 (2001) 749.

Diffusion characteristics

Bad diffusion hosts (narrow and/or stiff crystal lattice):

Re, diamond, SiC,...

Good diffusion hosts (wide crystal lattice):

Ti, Zr, Hf (fcc metals), Nb, Ta, graphite,
polycrystalline oxides (in particular fibers!)

Characteristic diffusion length:

$$d = (2 n D t)^{1/2} \quad n=1 \text{ (foil), } n=2 \text{ (fiber), } n=3 \text{ (sphere)}$$

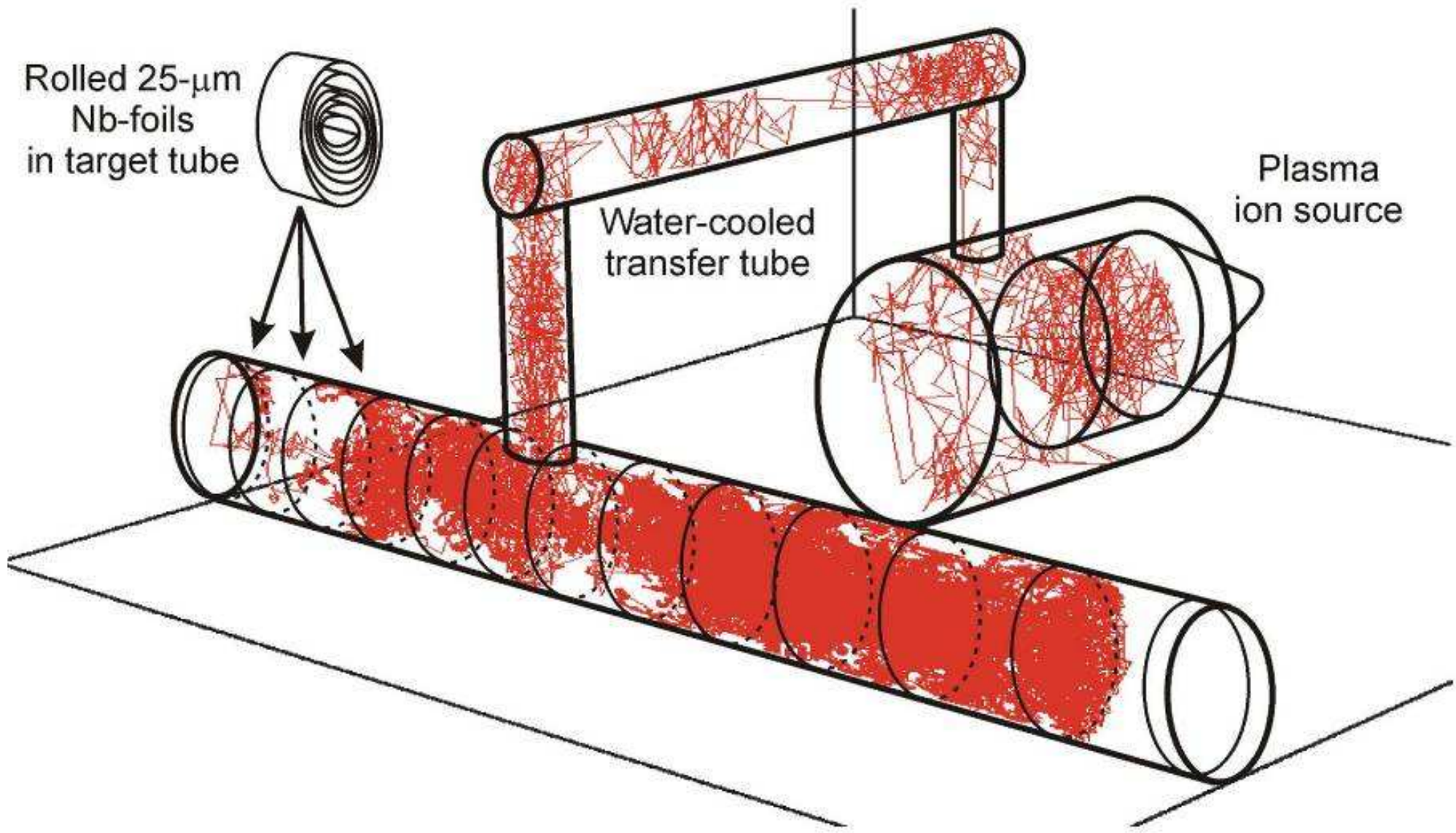
Maximize D and minimize diffusion path:

⇒ thin metal foils (2 μm ... 30 μm)

⇒ fine powders (μm)

⇒ thin fibers (some μm)

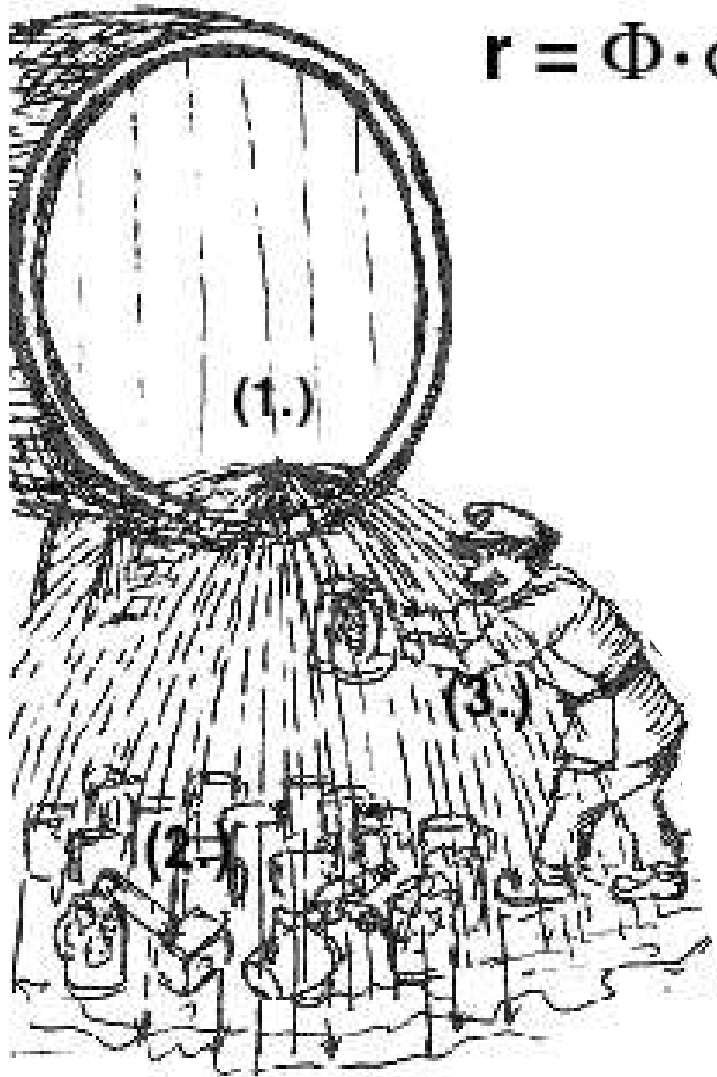
Effusion: random walk release



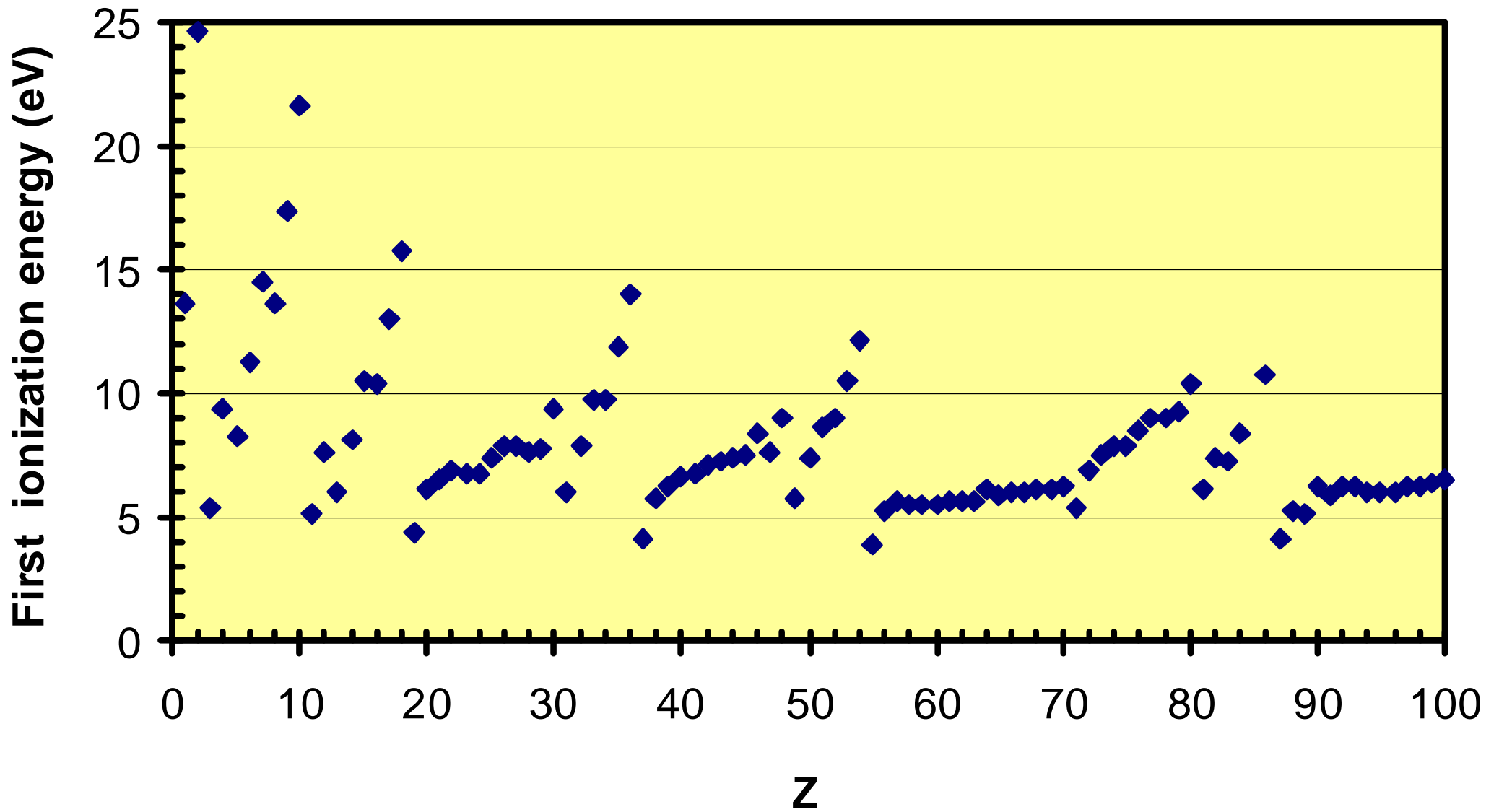
Optimize RIB intensity

All steps of the separation chain need to be optimized!

$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$

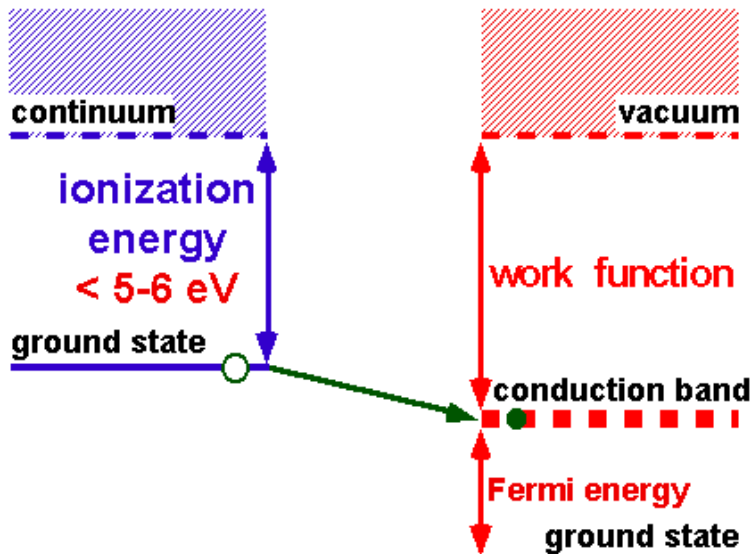
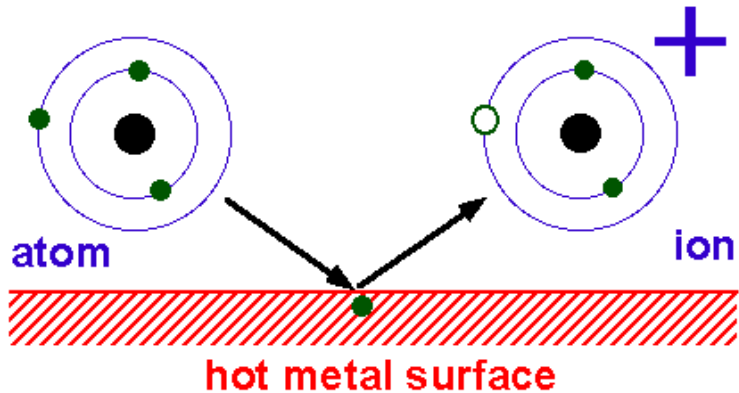


The first ionization energy of the elements

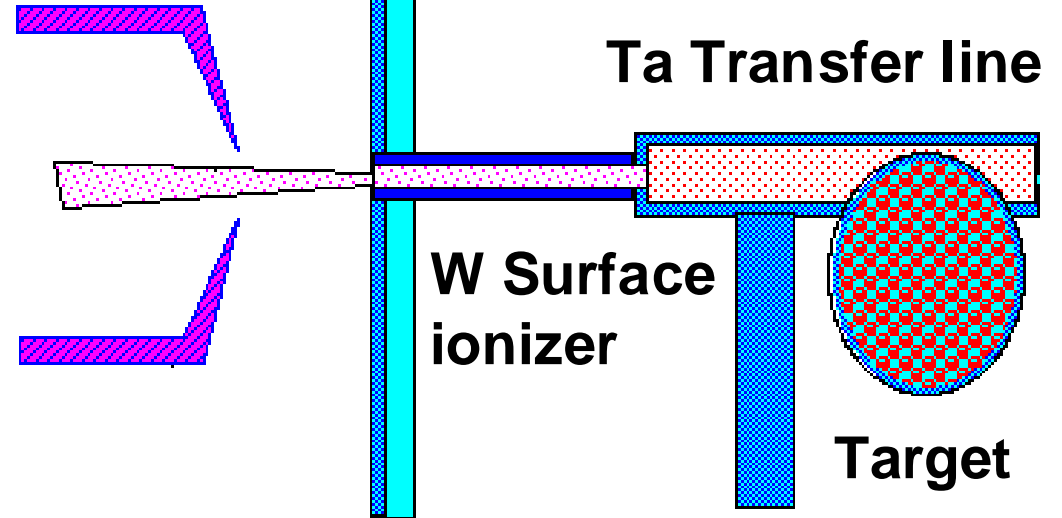


Positive surface ionization source

Surface Ionization



Extraction
electrode



$$\alpha_s = g_+/g_0 \exp(-(IP - \Phi)/kT)$$

$$\varepsilon_s = \alpha_s / (1 + \alpha_s)$$

Saha-Langmuir equation

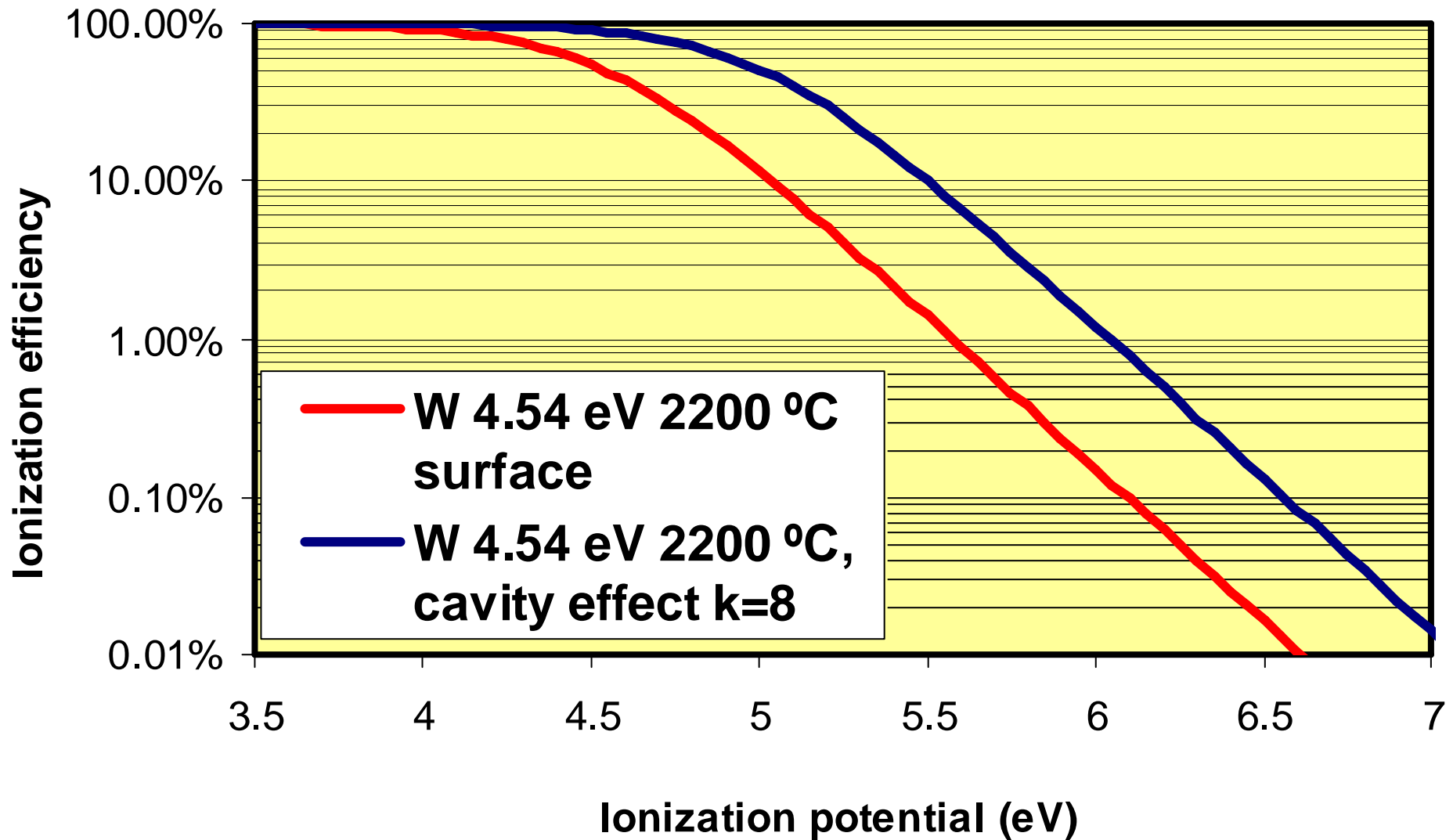
ε_s surface ionization efficiency

Φ work function of surface

IP ionization potential of atom

$g=2J+1$ stat. factor ($g_0=2$, $g_+=1$ for alkalis)

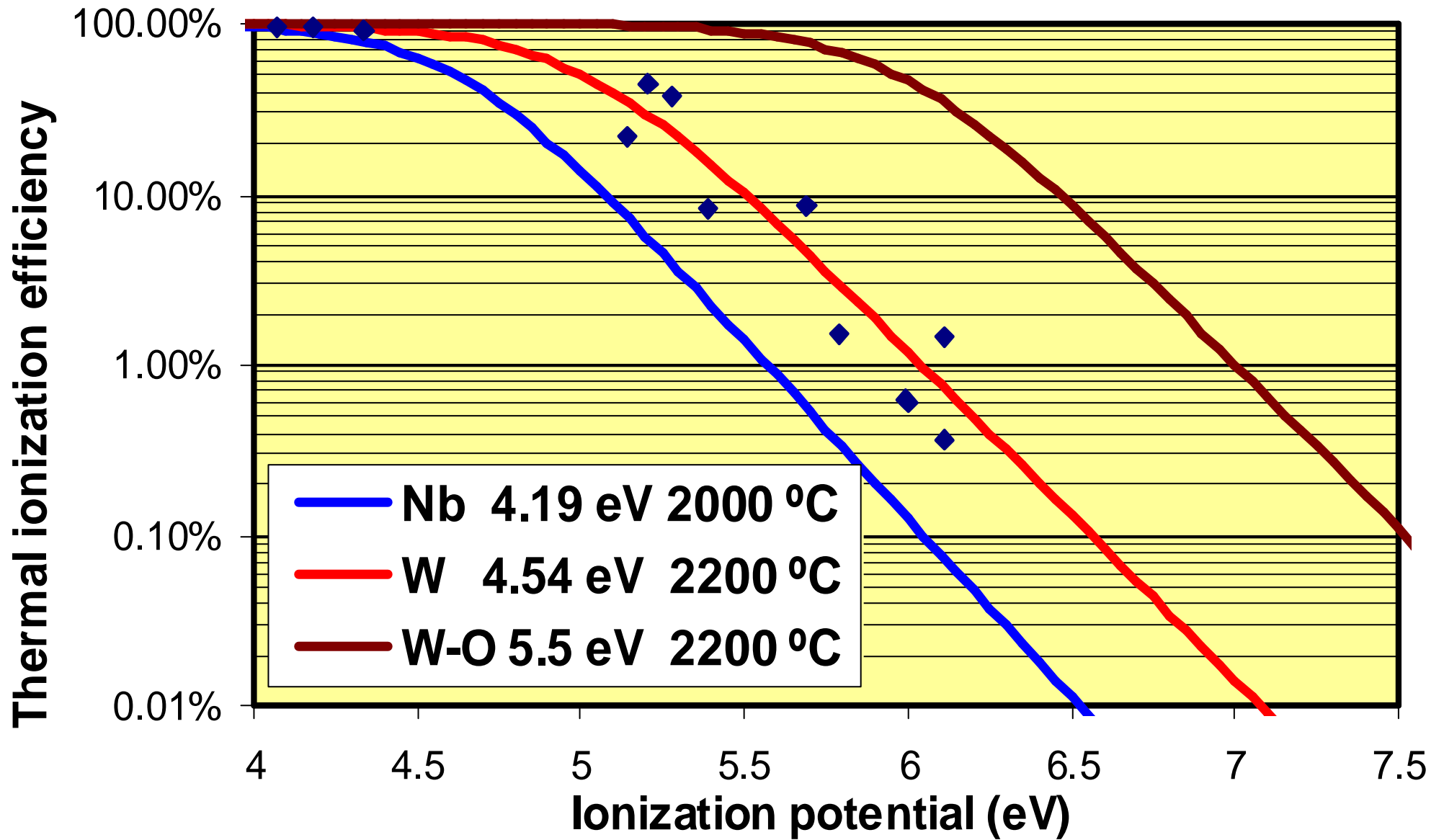
Surface ionization versus thermal ionization



$$\epsilon_{th} = 1 / (1 + g_0 / g_+ / k \exp((IP - \Phi) / kT))$$

Thermal ionization efficiency
in realistic ionizer cavity

Thermal ionization efficiency in realistic cavity



Watch out for surface contaminations!

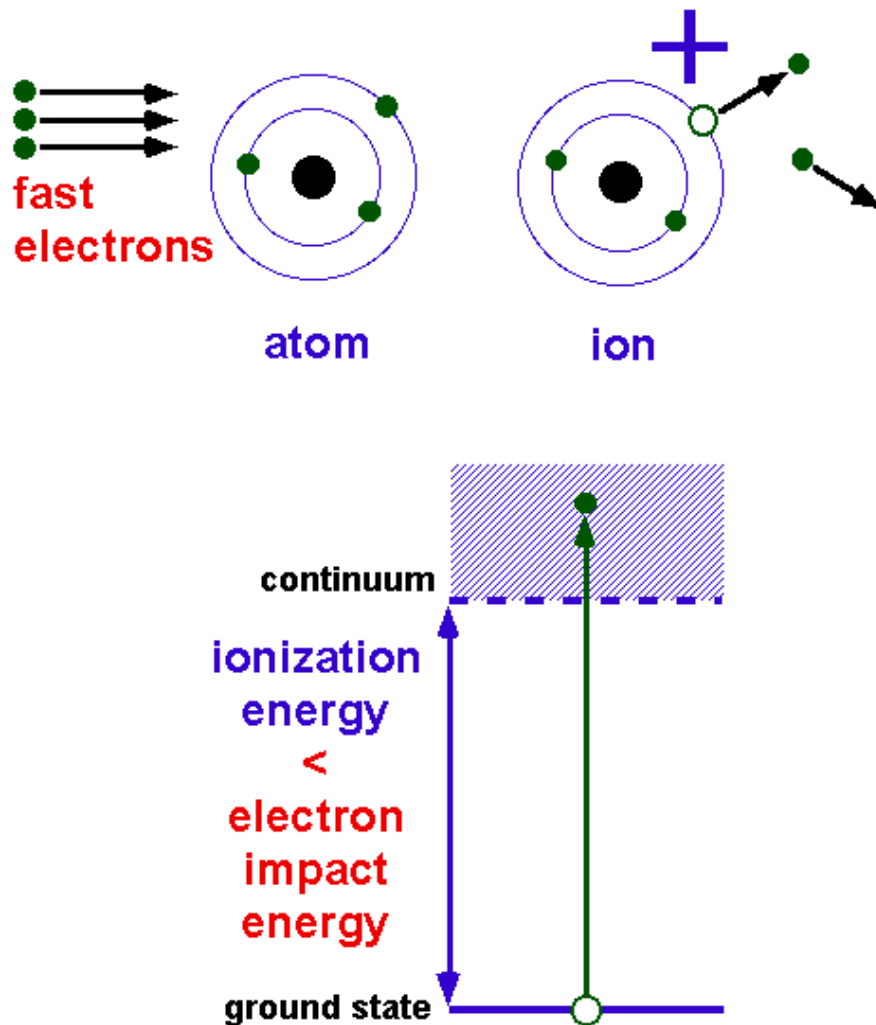
Ionization potentials of the elements

1 H	Ionization potential: < 5 eV																2 He
3 Li	4 Be	Ionization potential: 5.0 - 5.8 eV										5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg	Ionization potential: 5.8 - 6.5 eV										13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110	111	112						

58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

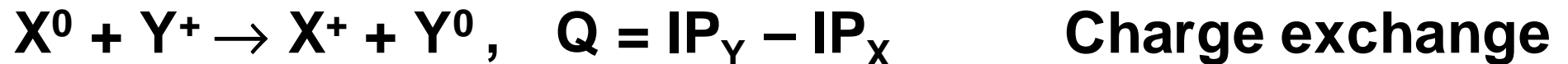
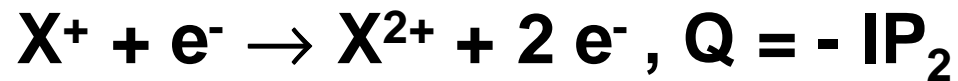
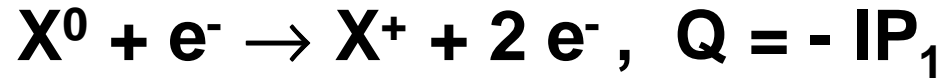
Ingredients of a plasma ion source

Ionization by electron impact

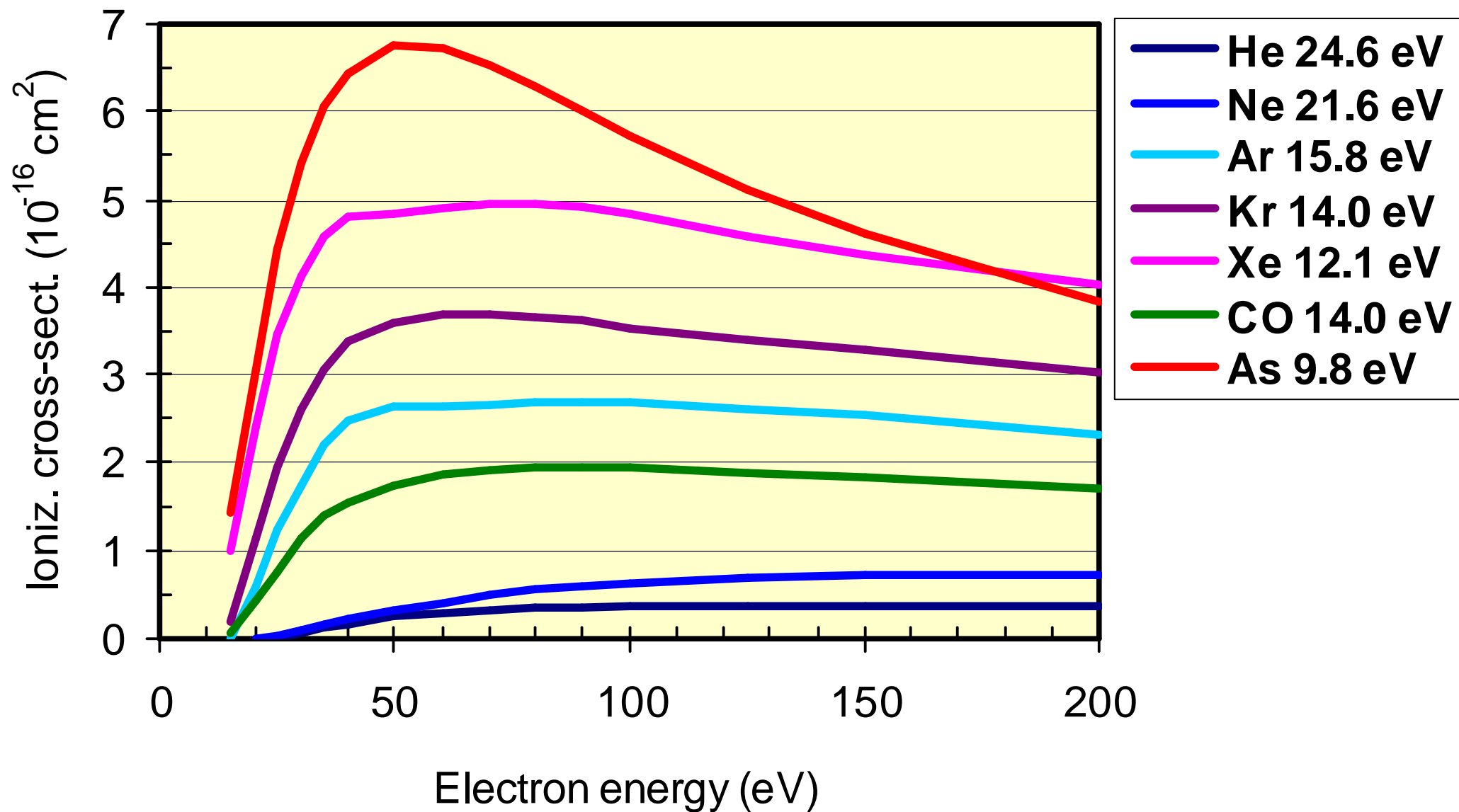


- Fast electrons:
 - A) Thermionic emission + accelerating field
 - B) RF heating
 - Atom confinement: plasma chamber
 - Electron “recycling”: magnetic field
 - Ion extraction system
- $$I[\text{A}] = A^* T[\text{K}]^2 \exp(-\Phi[\text{eV}]/kT[\text{K}])$$
- $$A^* = 120 \text{ A cm}^{-2} \text{ K}^{-2}$$
- $$\nu_{\text{cyc}}[\text{GHz}] = 28 B[\text{T}]$$
- $$r[\text{mm}] = 0.35 E_e[\text{eV}]^{1/2}/B[\text{T}]$$

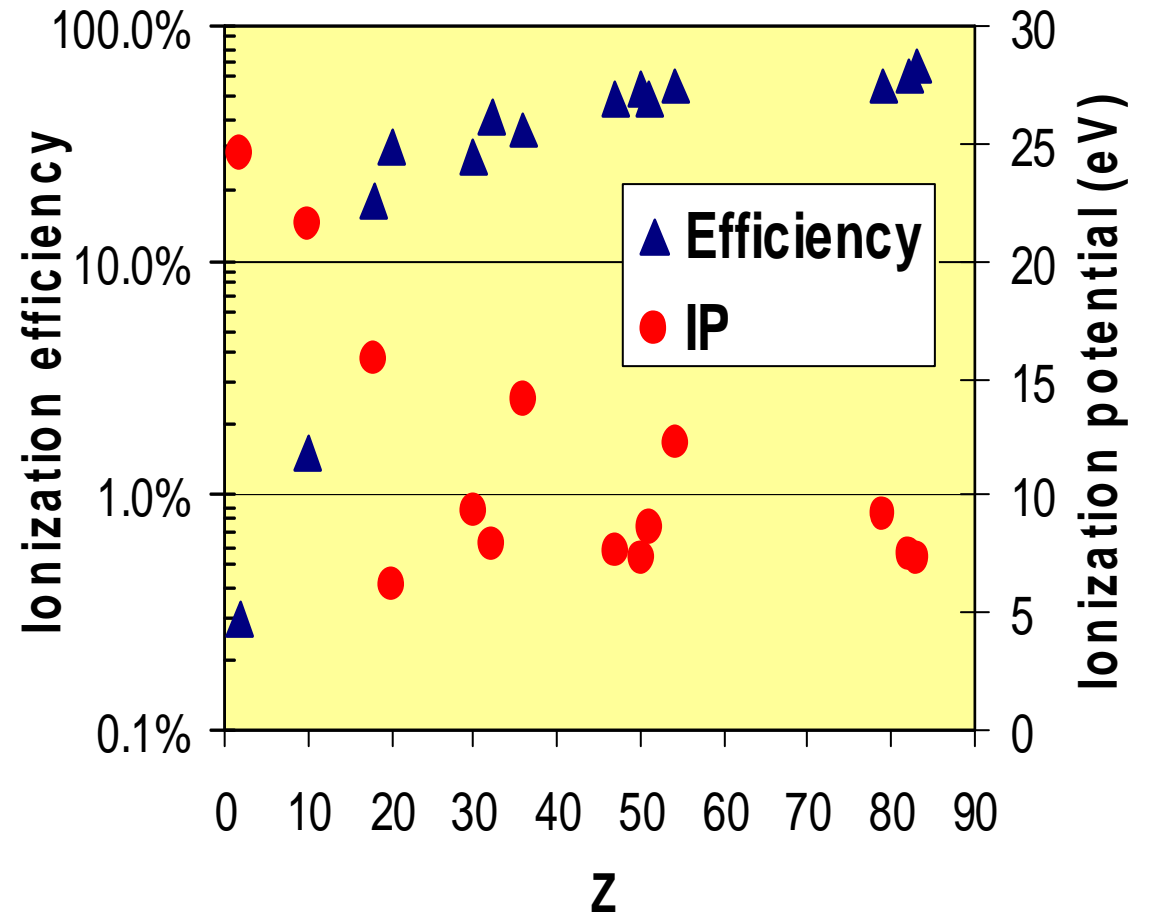
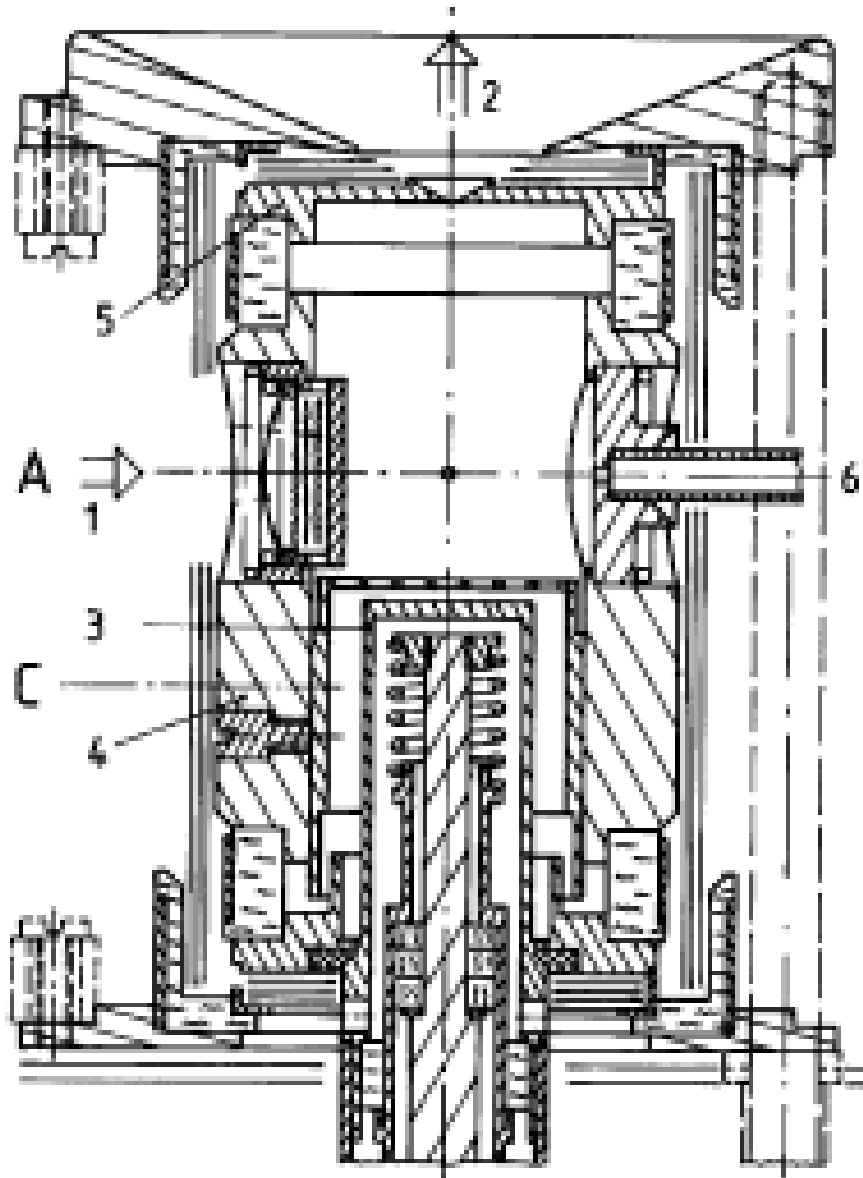
Ionization and neutralization



Electron impact ionization cross-sections



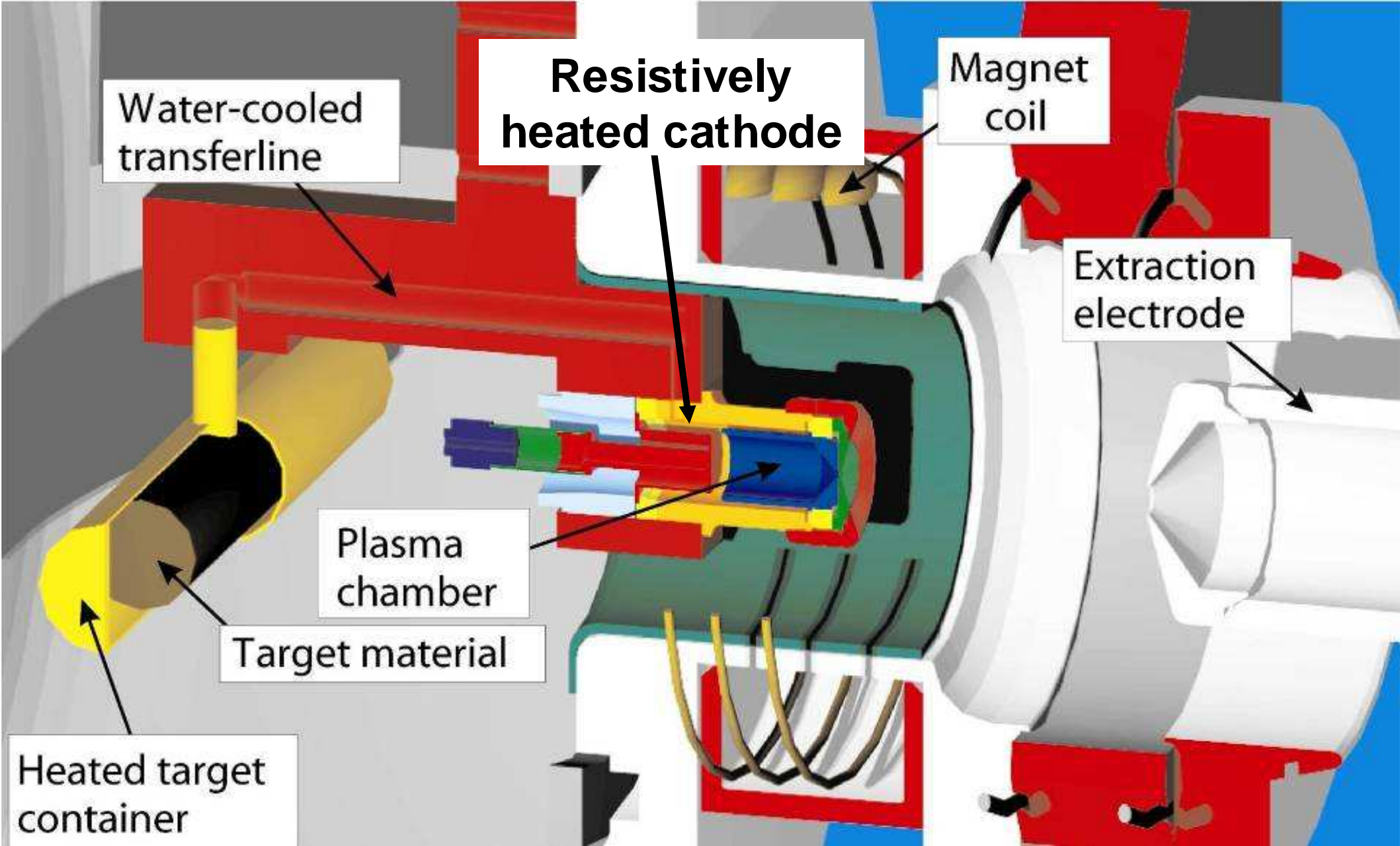
Forced Electron Beam Ion Arc Discharge (FEBIAD)



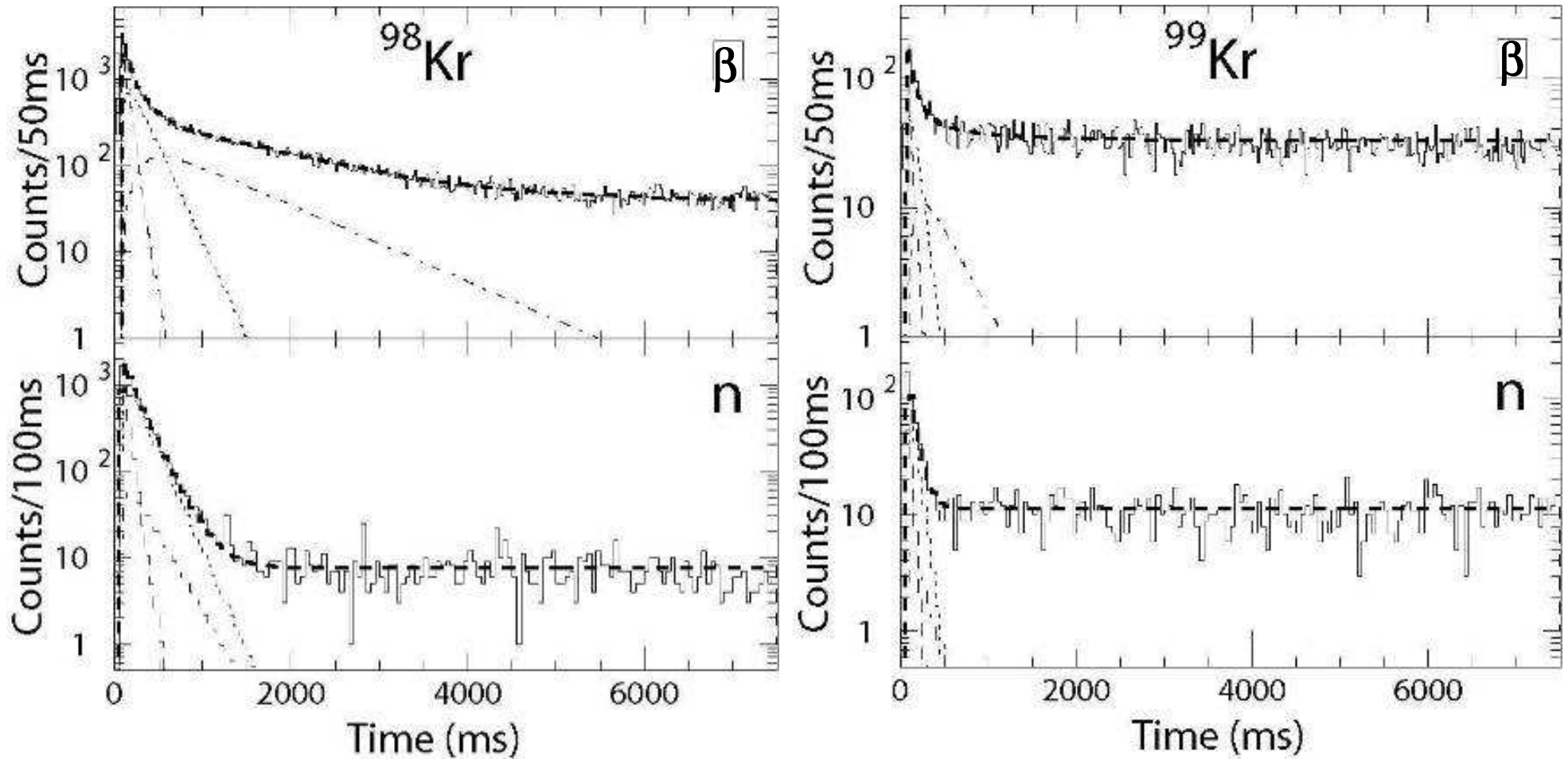
FEBIAD ion sources are excellent for heavier elements!

R. Kirchner, Rev. Sci. Instr. 67 (1996) 928.

ISOLDE "FEBIAD"



2001: $^{94-99}\text{Kr}$ decay studied at ISOLDE



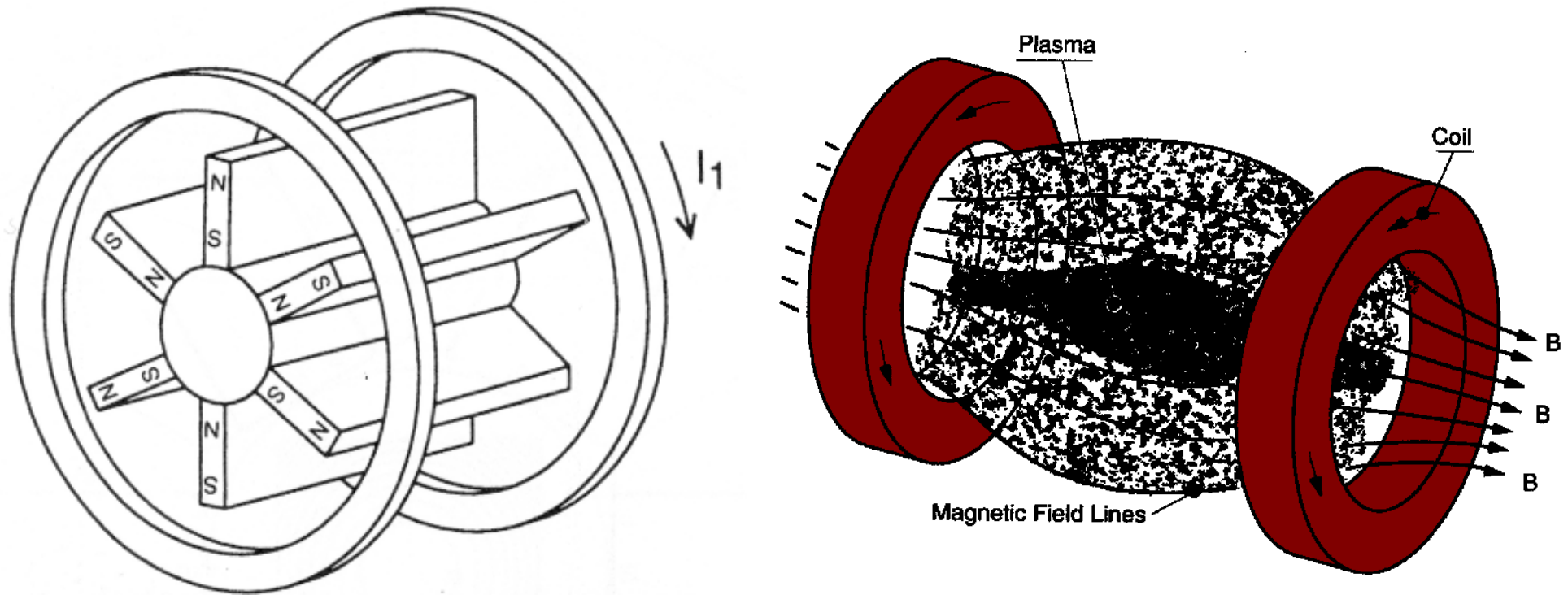
U.C. Bergmann et al., Nucl. Phys. A 714 (2003) 21.

Volatility of the elements

1	<p>T (p vapor > 0.01 mbar) < 100 °C</p> <p>T (p vapor > 0.01 mbar) < 400 °C</p> <p>T (p vapor > 0.01 mbar) < 1000 °C</p> <p>T (p vapor > 0.01 mbar) < 2000 °C</p> <p>T (p vapor > 0.01 mbar) > 2000 °C</p>																2
H																	He
3	4											5	6	7	8	9	10
Li	Be											B	C	N	O	F	Ne
11	12											13	14	15	16	17	18
Na	Mg											Al	Si	P	S	Cl	Ar
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
87	88	89	104	105	106	107	108	109	110	111	112						
Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt									

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Electron Cyclotron Resonance Ion Source (ECRIS)



radial plasma confinement by magnetic multipole field

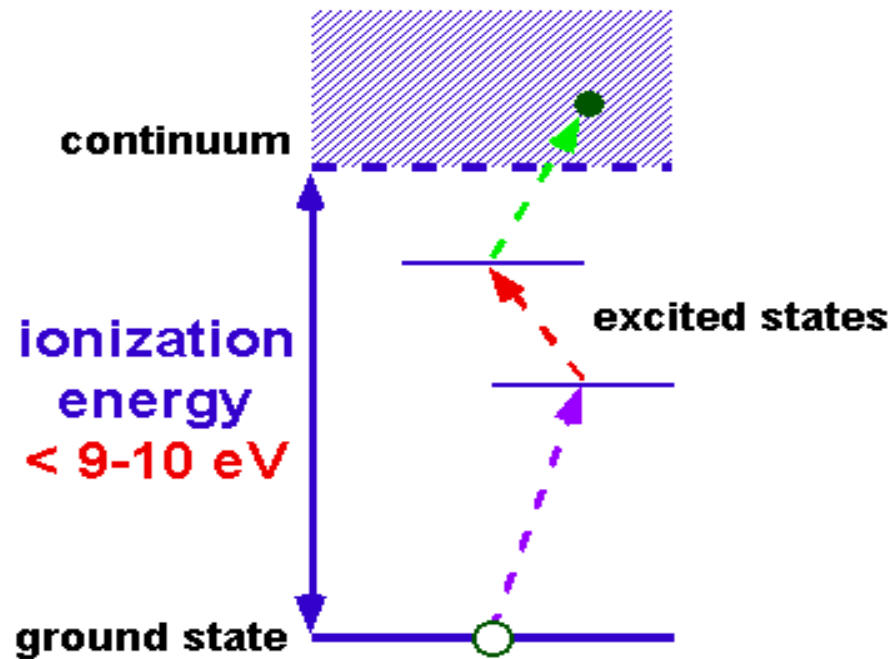
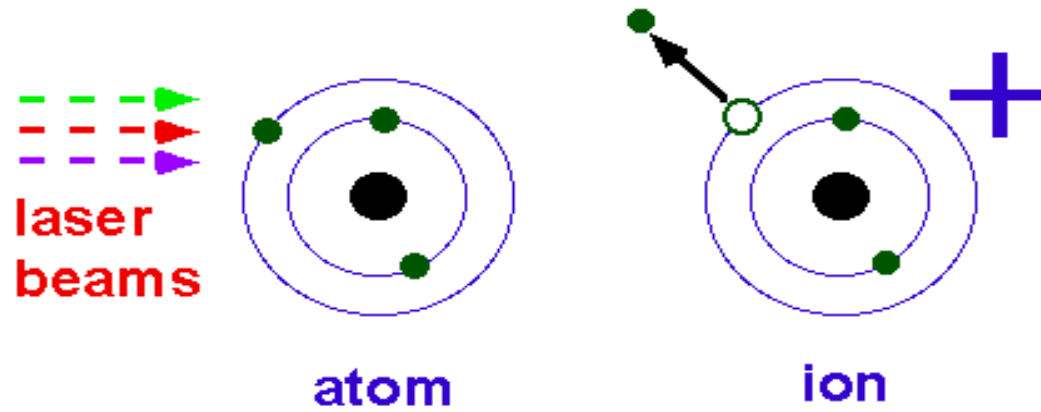
longitudinal plasma confinement by magnetic bottle effect (1+ ECRIS)
or minimum B configuration (n+ ECRIS)

plasma heating by RF (typically 2.45 – 30 GHz)

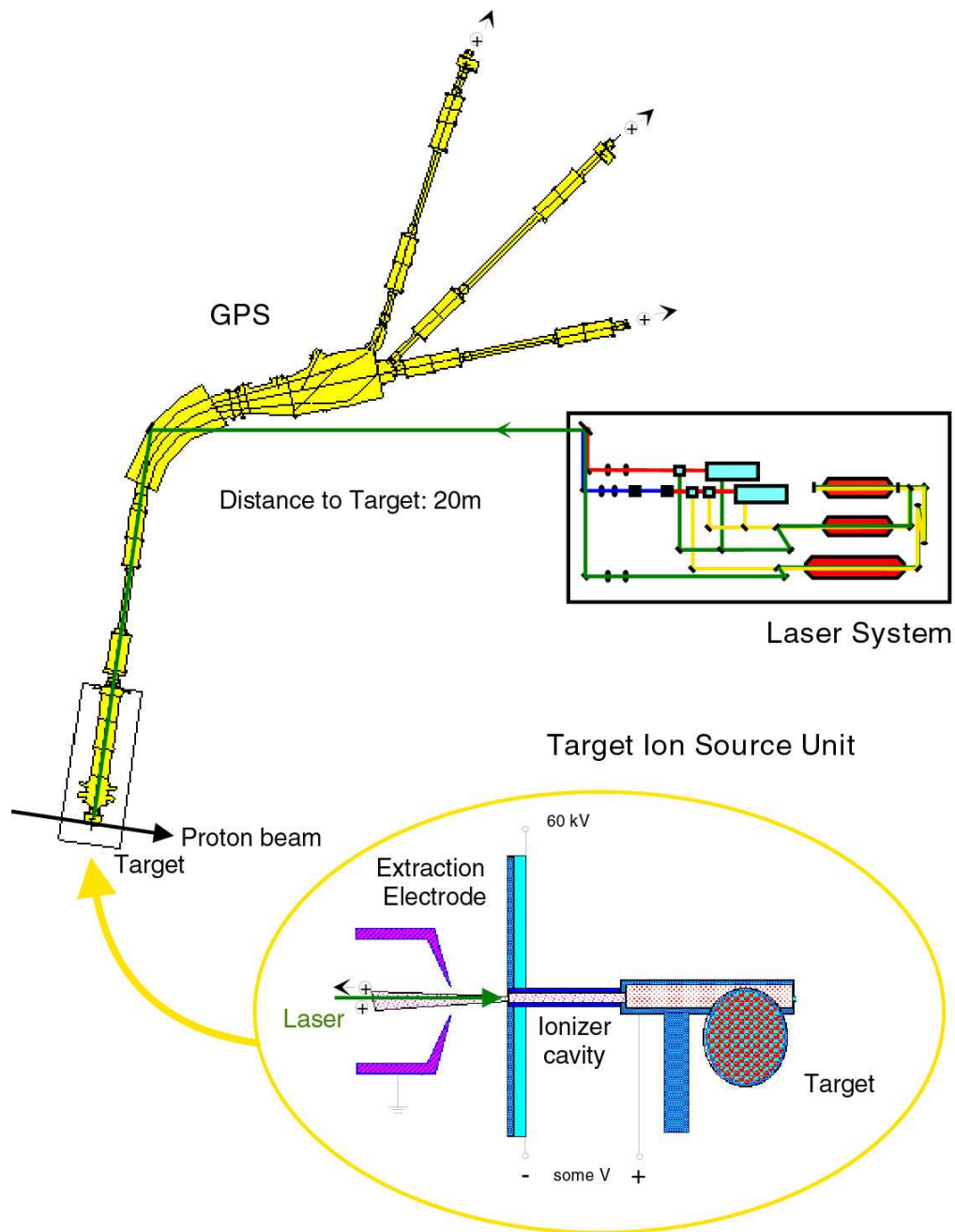
good efficiency for light elements (20% He⁺, 50% C⁺, O⁺, Ar⁺, 90% Xe⁺)

R. Geller, Electron Cyclotron Resonance Ion Sources and ECR Plasmas, IOP, Bristol, 1996.

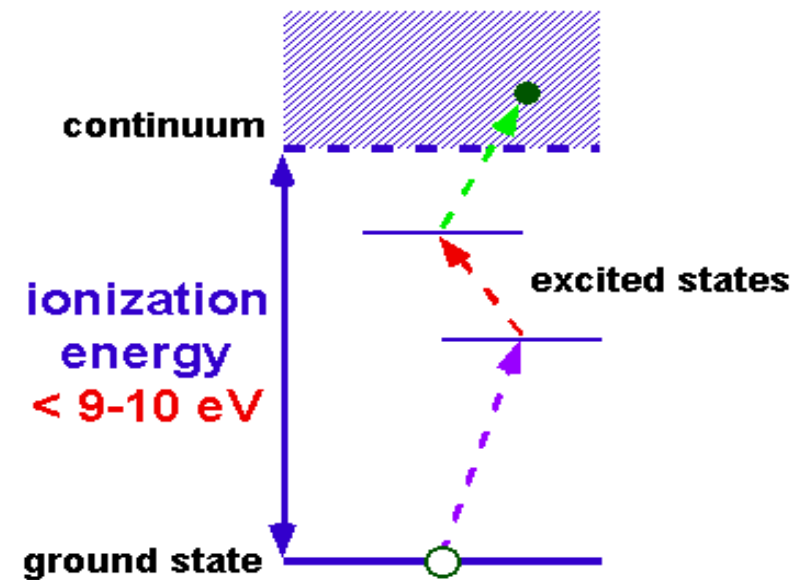
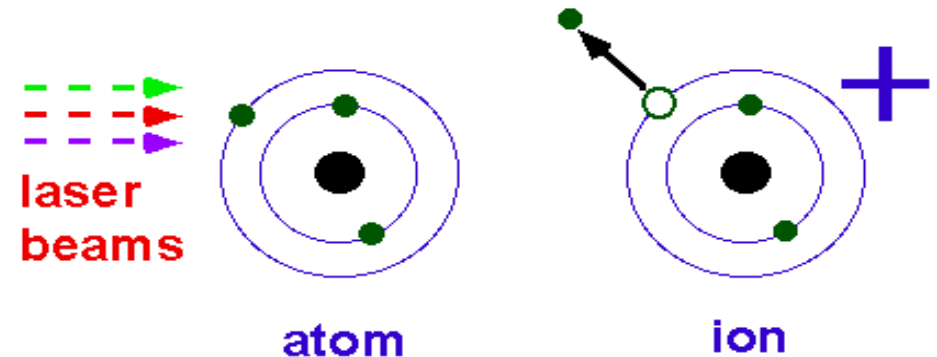
Laser Ionization



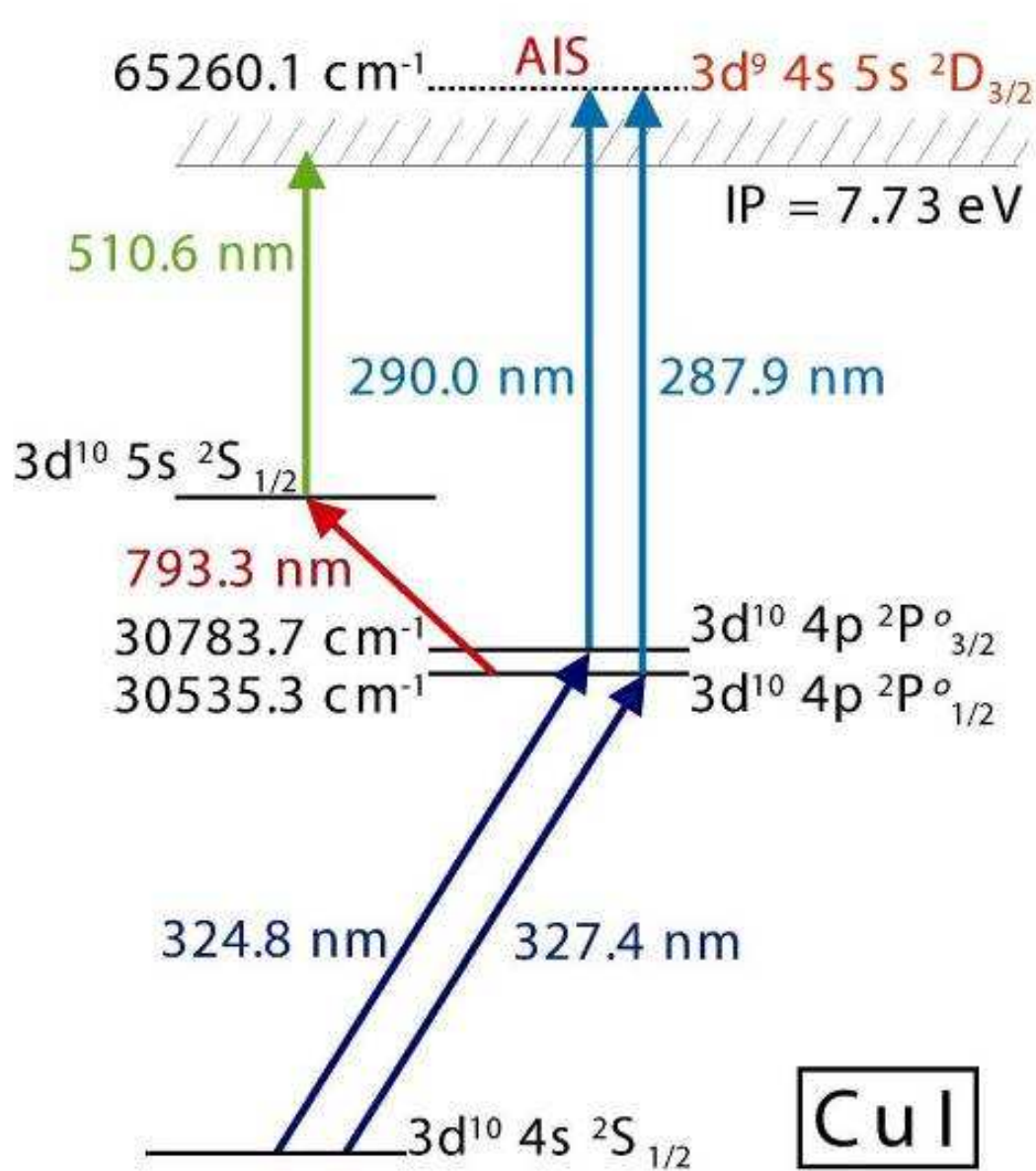
Resonance Ionization Laser Ion Source



Laser Ionization

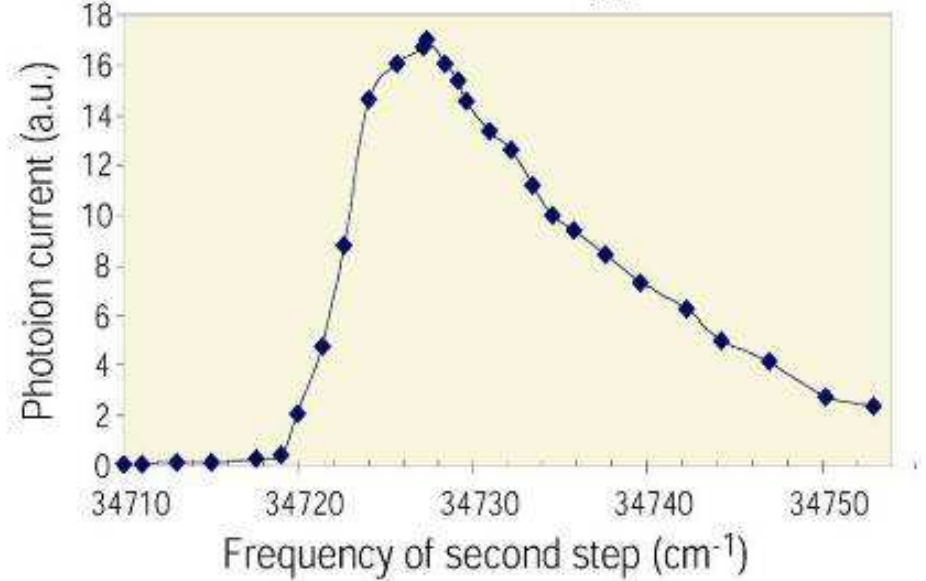


Ionization of Cu

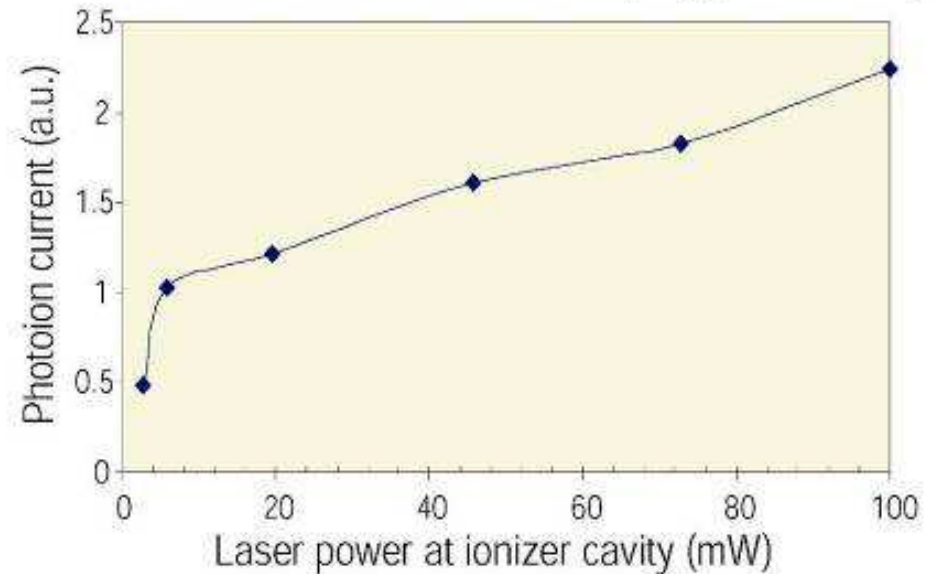


Cu I

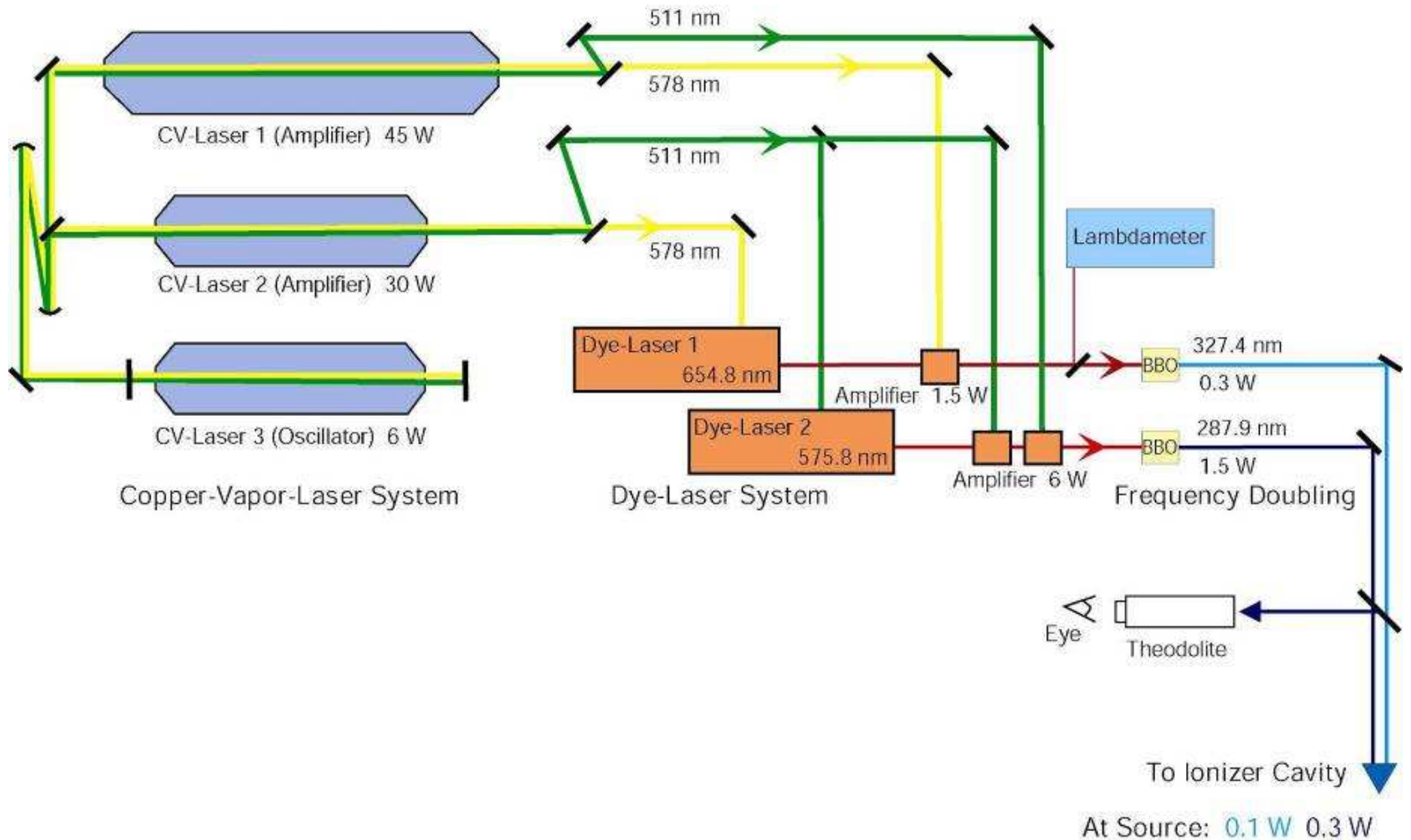
Structure of the $3d^9 4s 5s \ ^2D_{3/2}$ AIS at 65260.1 cm^{-1}



Saturation curve of first step ($\lambda_1=327.4 \text{ nm}$)



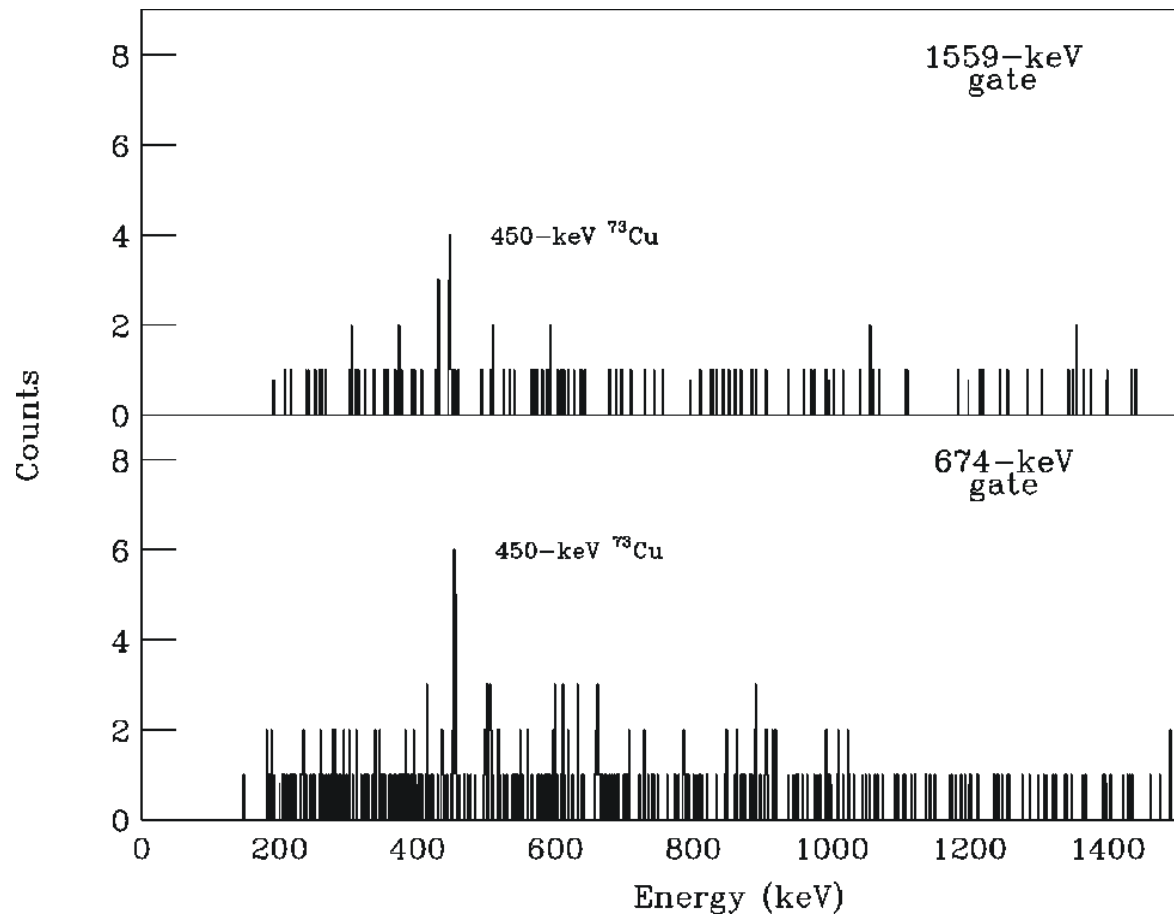
The laser system



ISOL versus In-flight separation

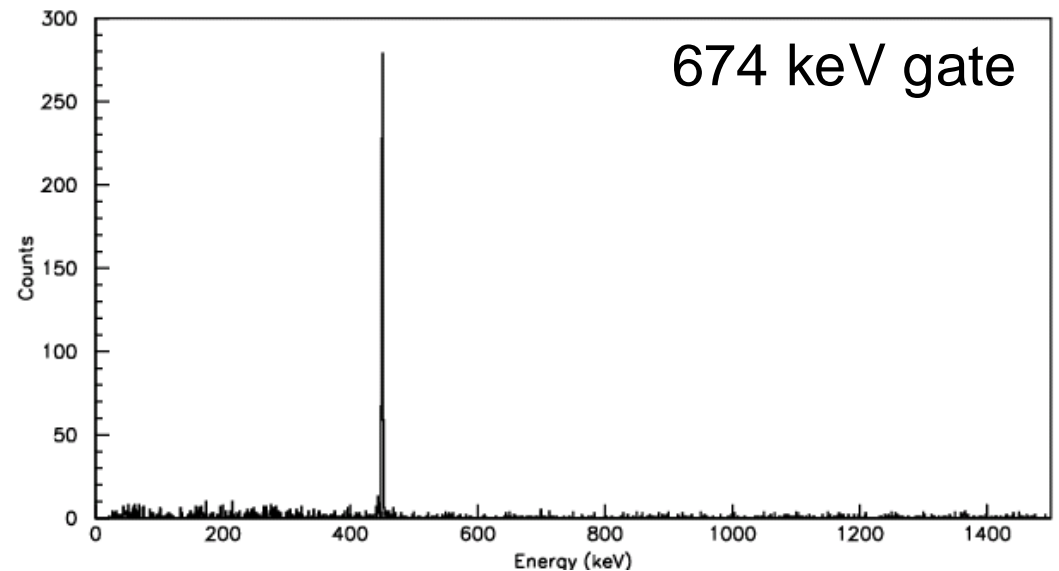
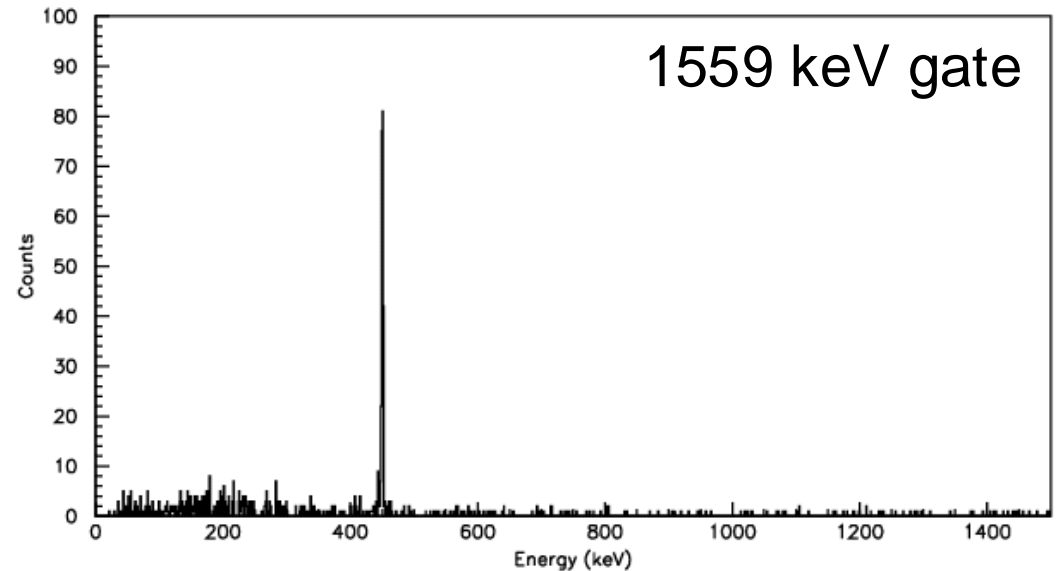
- ^{76}Ge fragmentation at A1200 fragment separator (MSU)
- Ge detectors (80% and 120%) in close geometry
- 24 h data taking for ^{73}Cu

M. Huhta et al., Phys. Rev. C58 (1998) 3187.



ISOL versus In-flight separation

- ISOLDE: 50 g/cm² UC_x/graphite target plus RILIS
- Ge detectors (75% and 64%) in close geometry
- 20 min data taking for ⁷³Cu

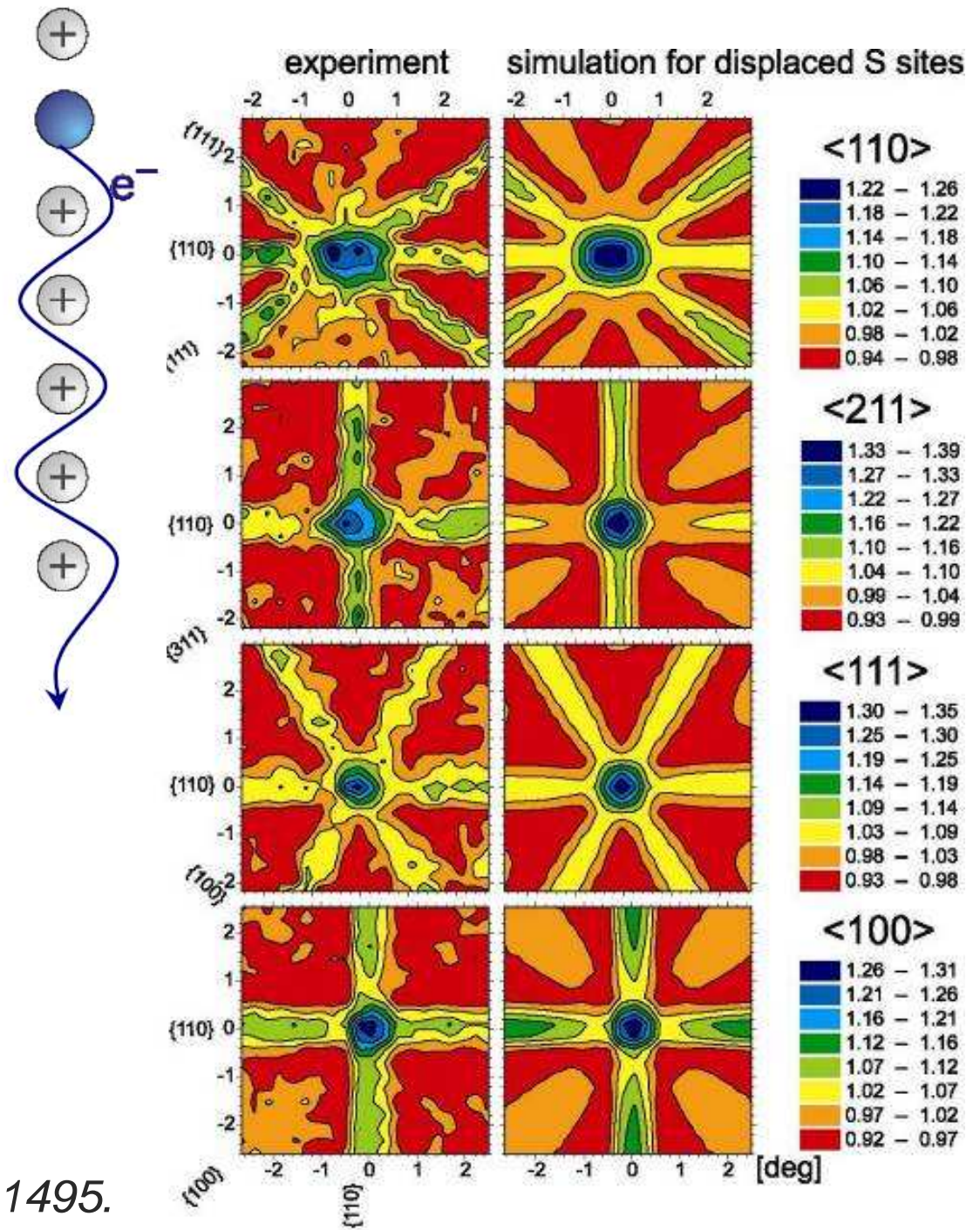
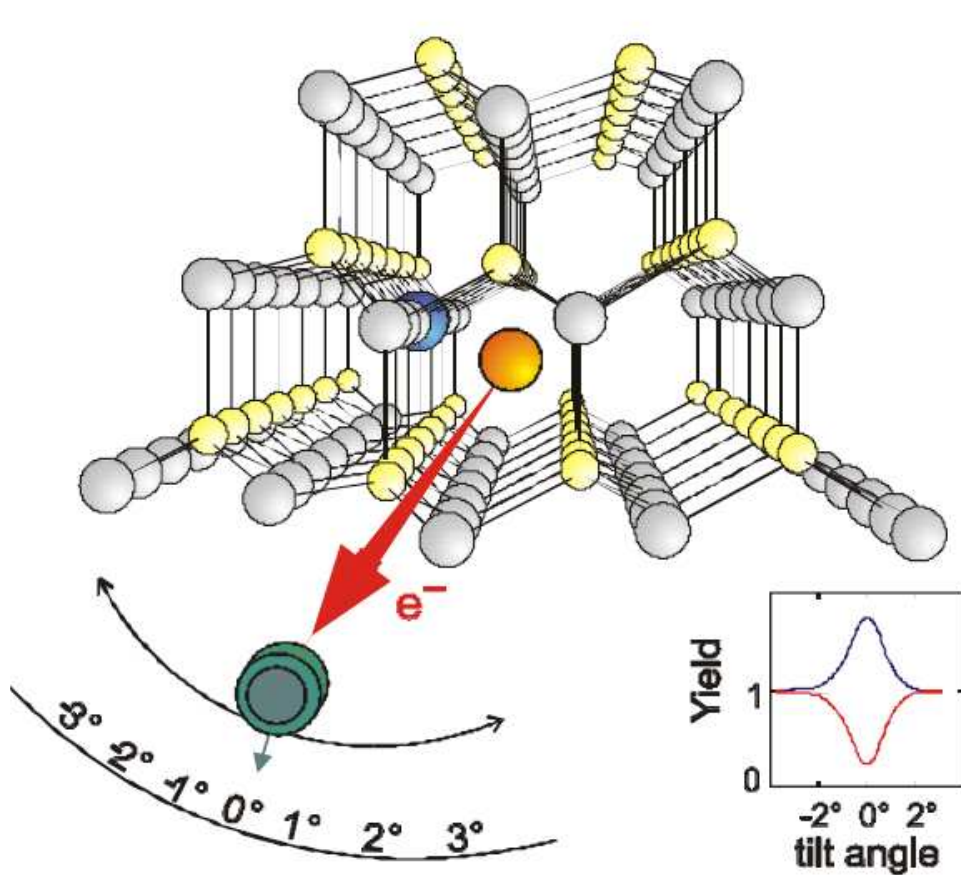


Neutron-rich Mn isotopes from UC_x/graphite target

Ge 64 64 s β ⁺ 3.0; 3.3... γ 427; 667; 128...	Ge 65 31 s β ⁺ 4.6; 5.2... γ 650; 62; 809; 191... βp 1.28...	Ge 66 2.3 h ε β ⁺ 0.7; 1.1... γ 382; 44; 169; 273...	Ge 67 18.7 m β ⁺ 3.0; 3.2... γ 167; 1473...	Ge 68 270,82 d ε no β ⁺ no γ	Ge 69 39.0 h ε β ⁺ 1.2... γ 1107; 574; 872; 1338...	Ge 70 21,23 α 3.0	Ge 71 11,43 d ε no γ	Ge 72 27.66 α 0.9	Ge 73 7.73 α 15	Ge 74 35,94 α 0.14 + 0.28	Ge 75 47 s 89 m β ⁻ 1.2... γ 265; 199...	Ge 76 7.44 1,53 · 10 ²¹ a α 0.09 + 0.06
Ga 63 31,4 s β ⁺ 4.5... γ 637; 627; 193; 650...	Ga 64 2,62 m β ⁺ 2.9; 6.1... γ 982; 808; 3366; 1387; 2195...	Ga 65 15 m β ⁺ 2.1; 2.2... γ 115; 61; 153; 752...	Ga 66 9.4 h β ⁺ 4.2... γ 1039; 2752; 834; 2190; 4296...	Ga 67 78,3 h ε no β ⁺ γ 93; 185; 300...	Ga 68 67,63 m β ⁺ 1.9... γ 1077; (1833...)	Ga 69 60,108 α 1.85	Ga 70 21,15 m β ⁻ 1.7... ε γ (1040; 176)	Ga 71 39,892 α 4.7	Ga 72 14,1 h β ⁻ 1.0; 3.2... γ 834; 2202; 630; 2508...	Ga 73 4,86 h β ⁻ 1.2; 1.5... γ 297; 53; 326... e ⁻	Ga 74 9,5 s 8,1 m β ⁻ 2.6; 4.9... γ 556; 2354; 608...	Ga 75 2,1 m β ⁻ 3.3... γ 253; 575... g
Zn 62 9,13 h ε β ⁺ 0.7 γ 41; 597; 548; 508...	Zn 63 38,1 m β ⁺ 2.3... γ 670; 962; 1412...	Zn 64 48.6 α 0.77	Zn 65 244,3 d ε; β ⁺ 0.3 γ 1115... α 66	Zn 66 27,9 α 1.0	Zn 67 4,1 α 6.9	Zn 68 18,8 α 0.072 + 0.8	Zn 69 13,8 h 56 m h 430 β ⁻ 0.9... γ (574; 1018...)	Zn 70 0.6 α 0.0061 + 0.093	Zn 71 3,9 h 2,4 m β ⁻ 1.0; 2.5... γ 386; 487; 510; 820...	Zn 72 46,5 h β ⁻ 0.3... γ 145; 192... e ⁻	Zn 73 5,8 s 23,5 s h 196 β ⁻ 4.0... γ 218; 941; 498...	Zn 74 96 s β ⁻ 2.1; 2.3... γ 49; 144; 193... m; g
Cu 61 3,4 h β ⁺ 1.2... γ 283; 656; 67; 1186...	Cu 62 9,74 m β ⁺ 2.9... γ (1173...)	Cu 63 69,17 α 4.5	Cu 64 12,700 h ε; β ⁺ 0.6 β ⁺ 0.7 γ (1346) α - 270	Cu 65 30,83 α 2.17	Cu 66 5,1 m β ⁻ 2.6... γ 1039; (834...) α 140	Cu 67 61,9 h β ⁻ 0.4; 0.6... γ 185; 93; 91...	Cu 68 3,8 m 30 s h 526 85; 111... β ⁻ 0.7 1.9 γ 1077; 520...	Cu 69 3,0 m β ⁻ 2.5... γ 1007; 834; 531... g	Cu 70 42 s 5 s β ⁻ 3.3; 4.5... γ 850; 962; 1252... β ⁻ 5.2; γ 685...	Cu 71 19,5 s β ⁻ γ 490; 595; 587... g; m	Cu 72 6,6 s β ⁻ γ 652; 1005; 1658; 847...	Cu 73 3,9 s β ⁻ γ 450; 199; 502; 307...
Ni 60 26,223 α 2.9	Ni 61 1,140 α 2.5	Ni 62 3,634 α 1.5	Ni 63 100 a β ⁻ 0.07 no γ α 24	Ni 64 0,926 α 1.5	Ni 65 2,52 h β ⁻ 2.1... γ 1482; 1115; 366... α 22	Ni 66 54,6 h β ⁻ 0.2 no γ	Ni 67 21 s β ⁻ 3.8... γ (1837; 1115; 822...)	Ni 68 29 s β ⁻ γ 758; 84 g	Ni 69 11,4 s β ⁻ γ 1871; 690; 1213; 1463...	Ni 70 6,0 s β ⁻ γ 1036; 78	Ni 71 2,56 s β ⁻ γ 534; 2016	Ni 72 1,57 s β ⁻ γ 376; 94
Co 59 100 α 20.7 + 16.5	Co 60 10,5 m 5,272 a h 59 β ⁻ 0.3 1.5... e ⁻ β ⁻ 1.0... γ 1302; 1173... α 58	Co 61 1,65 h β ⁻ 1.2... γ 67; 909...	Co 62 14,0 m 1,5 m β ⁻ 2.9... γ 1173; 1103; 2902; 2003... β ⁻ 4.1... γ 1173; 1128...	Co 63 27,5 s β ⁻ 3.6... γ 87; 982...	Co 64 0,3 s β ⁻ 7.0... γ 1346; 931	Co 65 1,14 s β ⁻ 6.0... γ 1142; 311; 964...	Co 66 0,23 s β ⁻ 7.0... γ 1425; 1246; 471...	Co 67 0,42 s β ⁻ 6.6 γ 694	Co 68 0,18 s β ⁻	Co 69 0,27 s β ⁻	Co 70 0,15 s β ⁻	Co 71 0,21 s β ⁻
Fe 58 0,28 α 1.3	Fe 59 44,503 d β ⁻ 0.5; 1.6... γ 1099; 1292... α < 10	Fe 60 1,5 · 10 ⁶ a β ⁻ 0.1 m	Fe 61 6,0 m β ⁻ 2.6; 2.8... γ 1205; 1027; 298...	Fe 62 68 s β ⁻ 2.5 γ 506 g	Fe 63 6,1 s β ⁻ 6.7... γ 995; 1427; 1239...	Fe 64 2,0 s β ⁻ γ 311	Fe 65 0,45 s β ⁻	Fe 66 0,44 s β ⁻	Fe 67 0,47 s β ⁻	Fe 68 0,1 s β ⁻	Fe 69 0,17 s β ⁻	44
Mn 57 1,5 m β ⁻ 2.6... γ 14; 122; 492	Mn 58 65,3 s 3,0 s β ⁻ 3.9... γ 811; 1523... h 72 e ⁻	Mn 59 4,6 s β ⁻ 4.4; 4.8... γ 726; 473; 571...	Mn 60 1,77 s 51 s β ⁻ 5.7... 6.4... γ 824; 1990; γ 272	Mn 61 0,71 s 623 ms β ⁻ 6.4... γ 628; 207...	Mn 62 0,88 s 671 ms β ⁻ γ 877; 942; 1299; 1815...	Mn 63 0,25 s 275 ms β ⁻ > 3.7 γ 356	Mn 64 0,14 s 89 ms β ⁻	Mn 65 0,11 s 88 ms β ⁻	Mn 66 0,09 s 66 ms β ⁻	Mn 67 42 ms	Mn 68 28 ms	Mn 69 14 ms

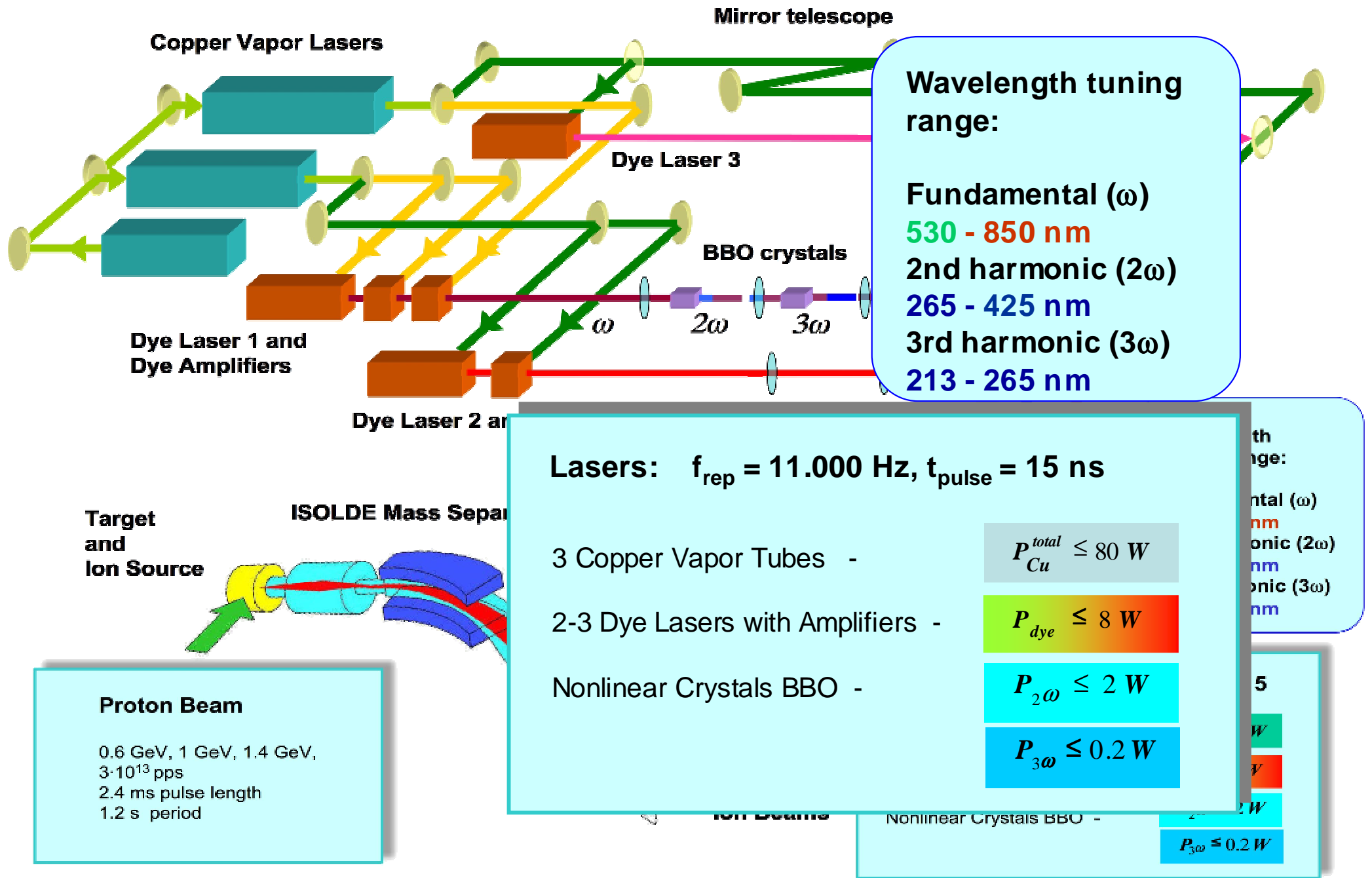
M. Hannawald et al., Phys. Rev. Lett. 82 (1999) 1391.

^{67}Cu for nuclear solid state physics



- Lattice location of Cu in Si studied by emission channeling
- 5 samples with each 5 min implantation of ^{67}Cu (ca. 100 ppm doping)

RILIS Laser setup



Wavelength tuning range:

Fundamental (ω)
 530 - 850 nm
 2nd harmonic (2ω)
 265 - 425 nm
 3rd harmonic (3ω)
 213 - 265 nm

Lasers: $f_{rep} = 11.000 \text{ Hz}$, $t_{pulse} = 15 \text{ ns}$

3 Copper Vapor Tubes - $P_{Cu}^{total} \leq 80 \text{ W}$

2-3 Dye Lasers with Amplifiers - $P_{dye} \leq 8 \text{ W}$

Nonlinear Crystals BBO - $P_{2\omega} \leq 2 \text{ W}$

$P_{3\omega} \leq 0.2 \text{ W}$

Proton Beam

0.6 GeV, 1 GeV, 1.4 GeV,
 $3 \cdot 10^{13}$ pps
 2.4 ms pulse length
 1.2 s period

Wavelength tuning range:

Fundamental (ω)
 530 - 850 nm
 2nd harmonic (2ω)
 265 - 425 nm
 3rd harmonic (3ω)
 213 - 265 nm

Power limits:

$P_{Cu}^{total} \leq 80 \text{ W}$

$P_{dye} \leq 8 \text{ W}$

$P_{2\omega} \leq 2 \text{ W}$

$P_{3\omega} \leq 0.2 \text{ W}$

Elements ionized with CVL pumped dye lasers

elements ionized with ISOLDE RILIS

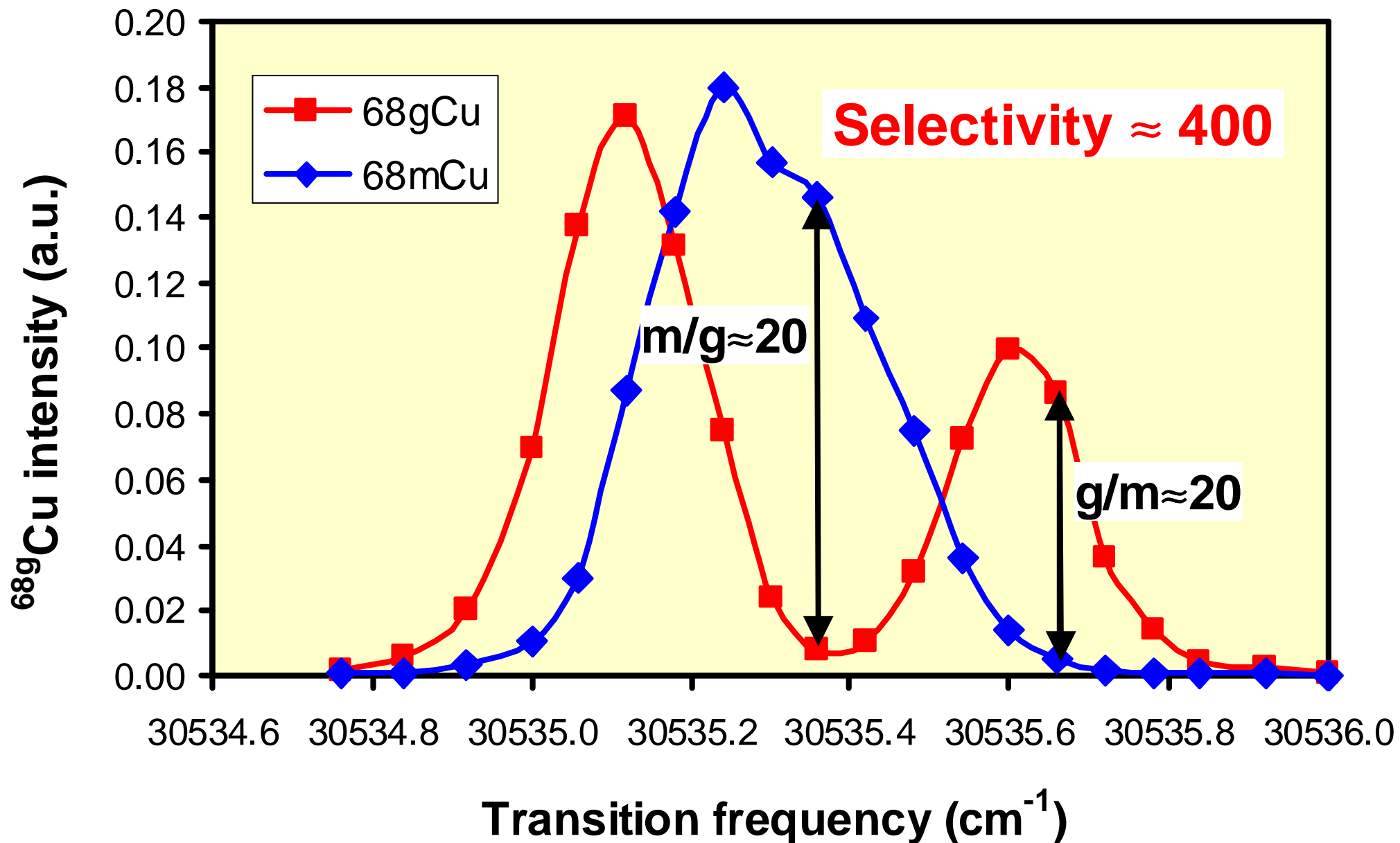
tested ionization scheme

possible ionization scheme (untested)

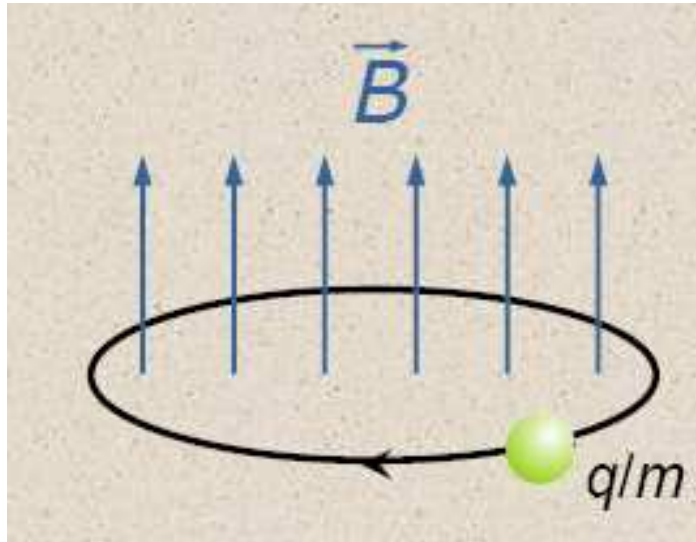
1 H																	2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110	111	112						

58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

Isomer separation



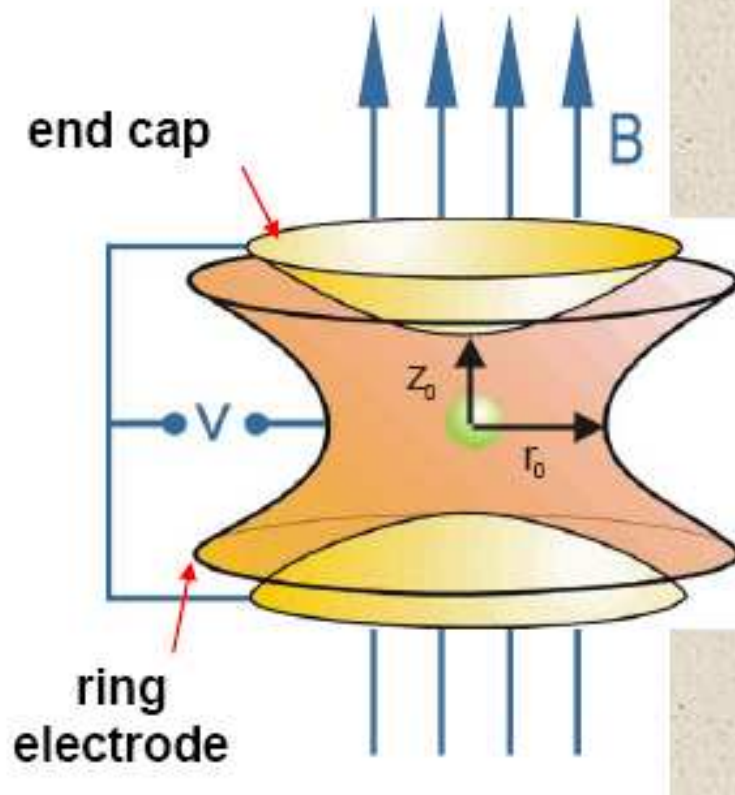
Mass measurements with Penning traps



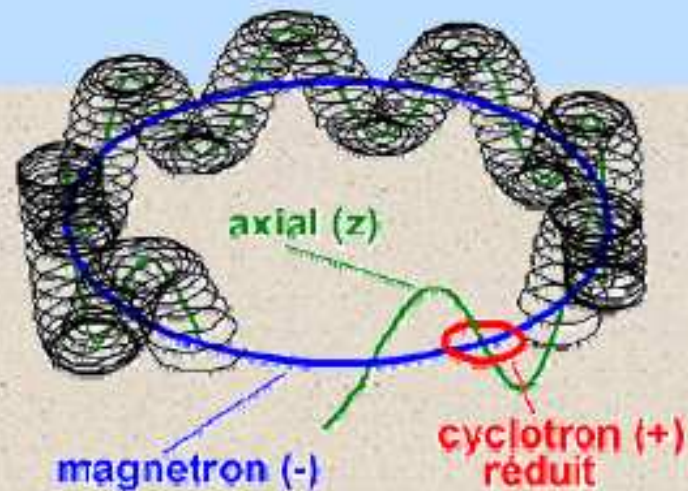
Cyclotron frequency:
$$f_c = \frac{1}{2\pi} \cdot \frac{q}{m} \cdot B$$

PENNING trap

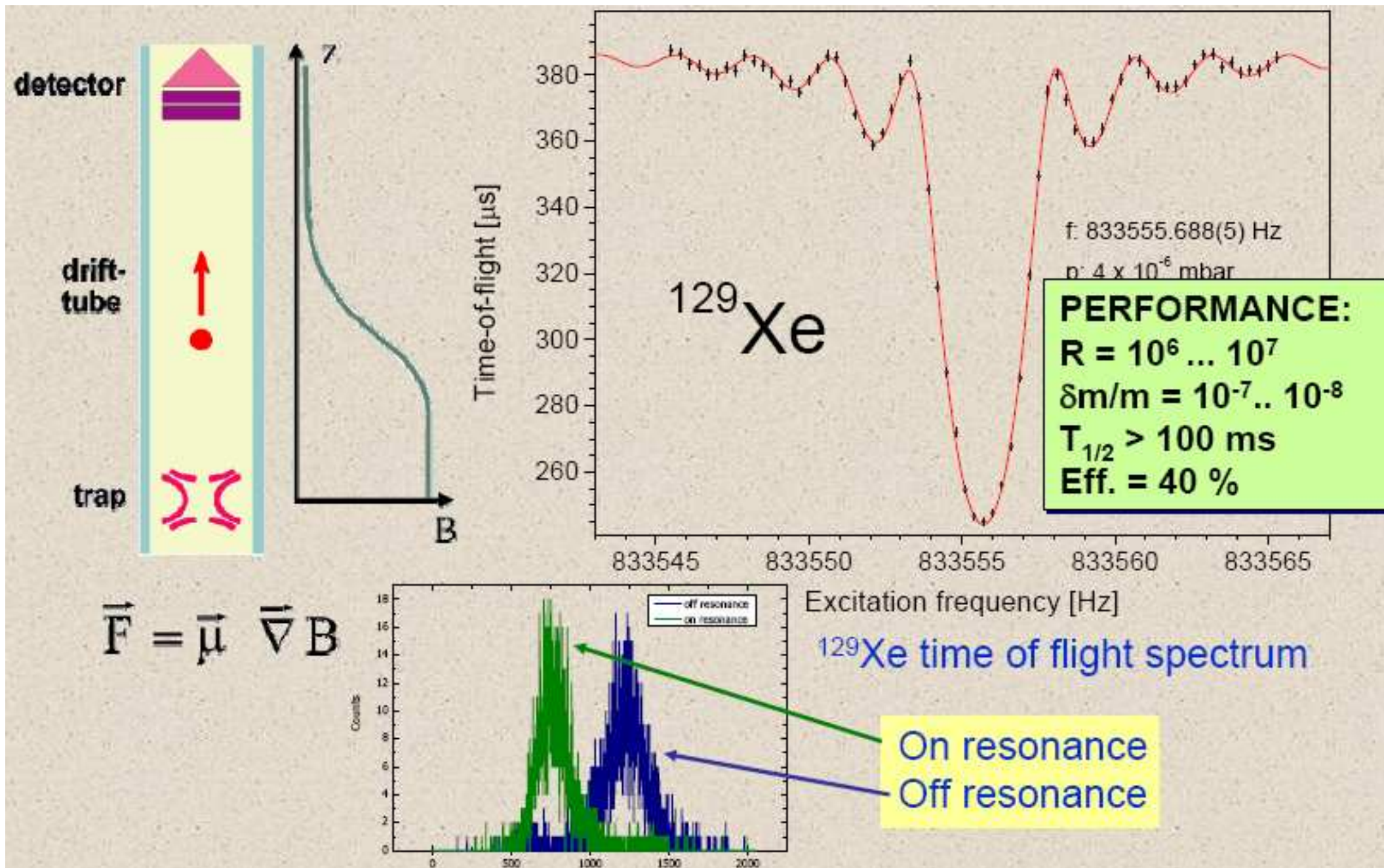
- Strong homogeneous magnetic field
- Weak electric 3D quadrupole field



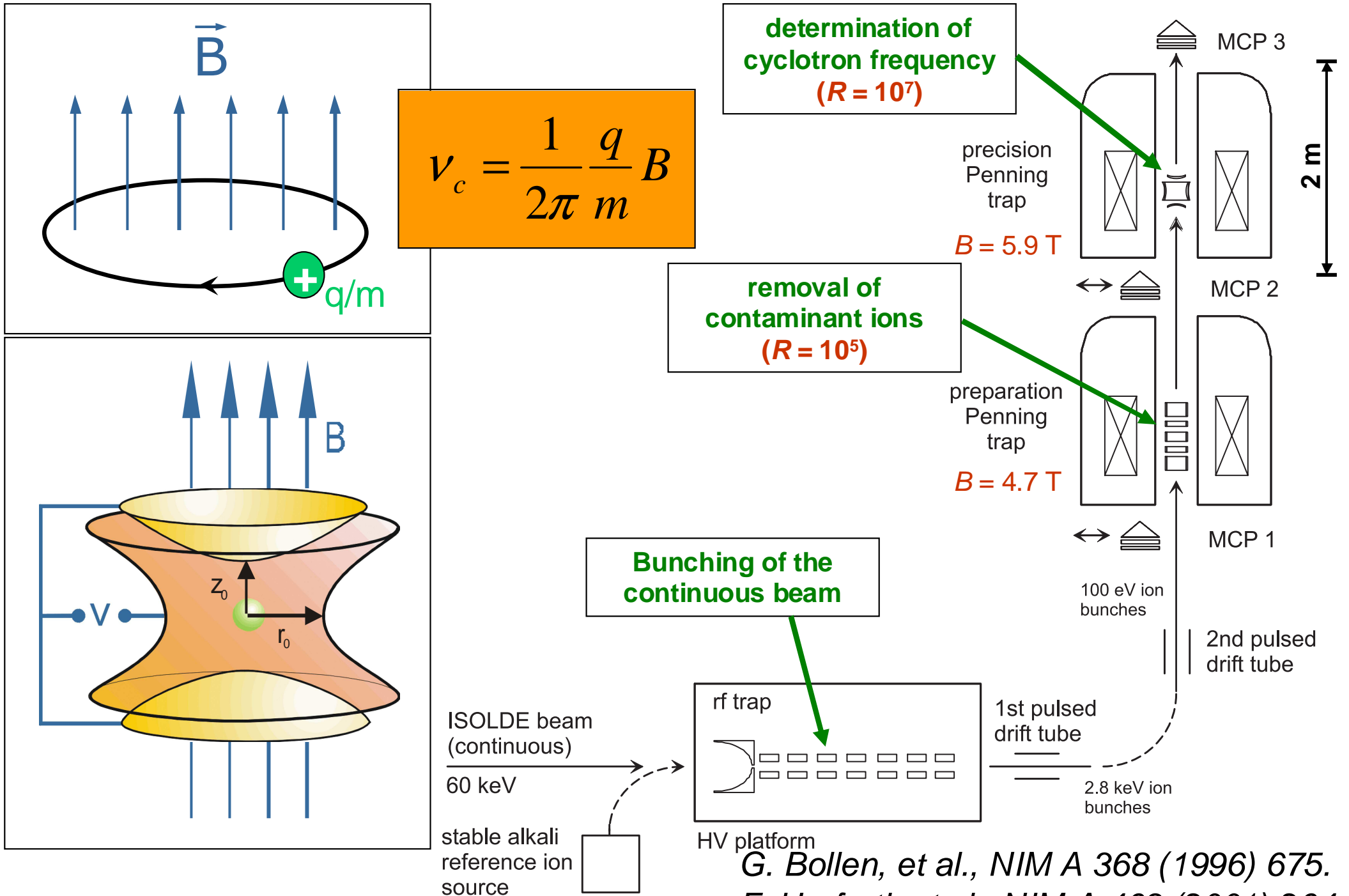
$$\omega_+ + \omega_- = \omega_c = \frac{q}{m} B$$



Resonance frequency measurement via TOF method

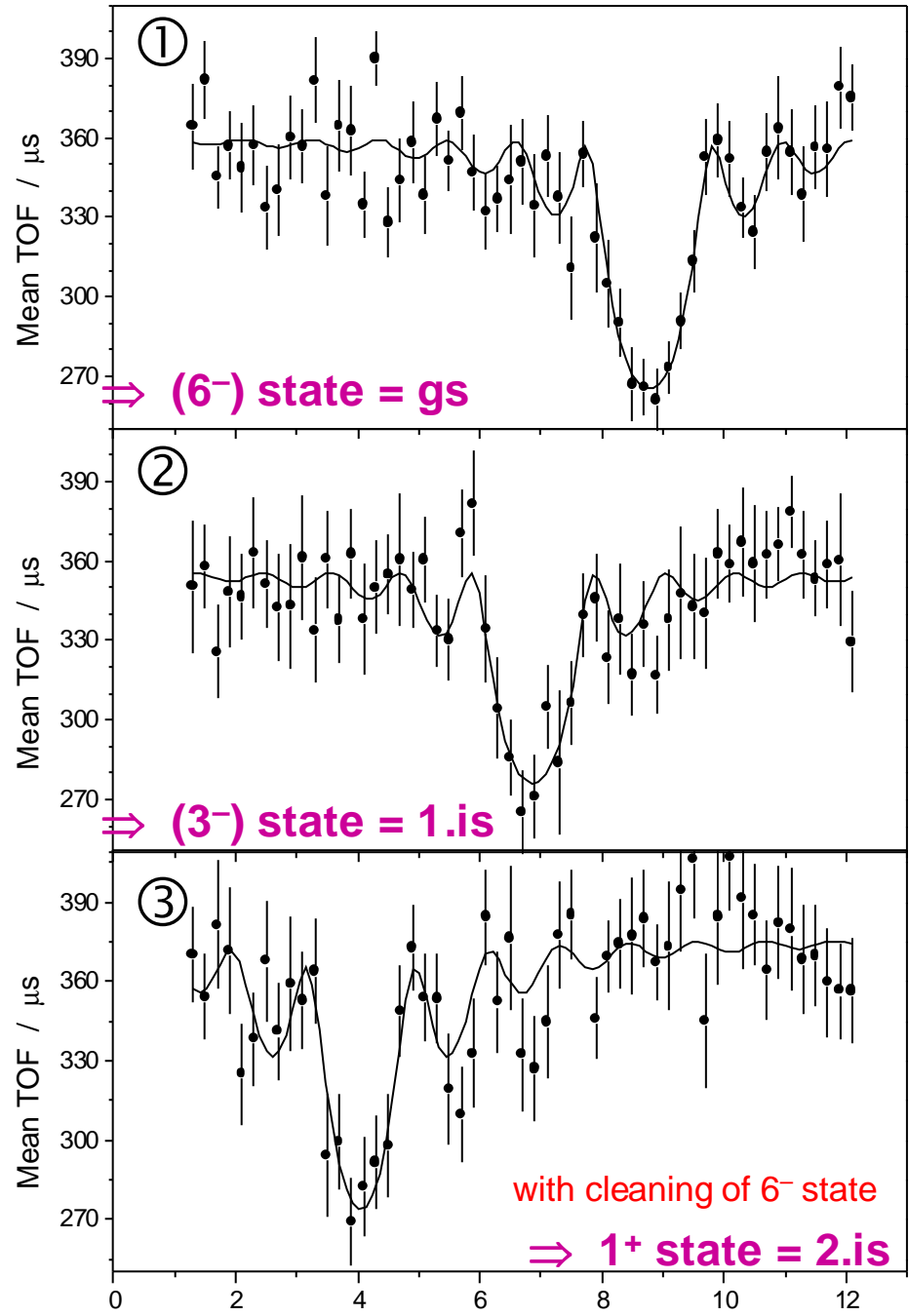
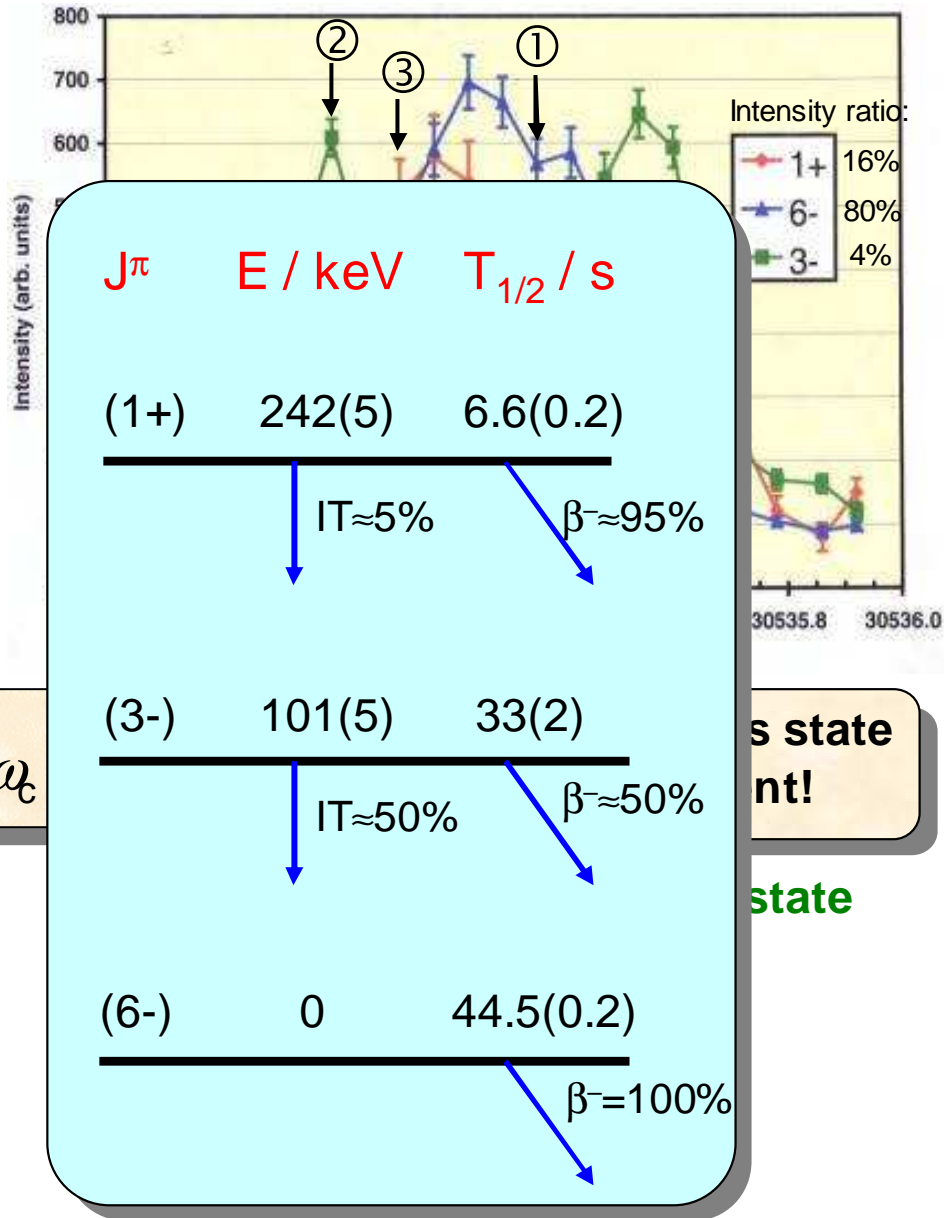


Mass measurements with ISOLTRAP



G. Bollen, et al., NIM A 368 (1996) 675.
 F. Herfurth et al., NIM A 469 (2001) 264.

Solving the ^{70}Cu mass puzzle



CERN accelerator structure

CERN-PS Booster Synchrotrons

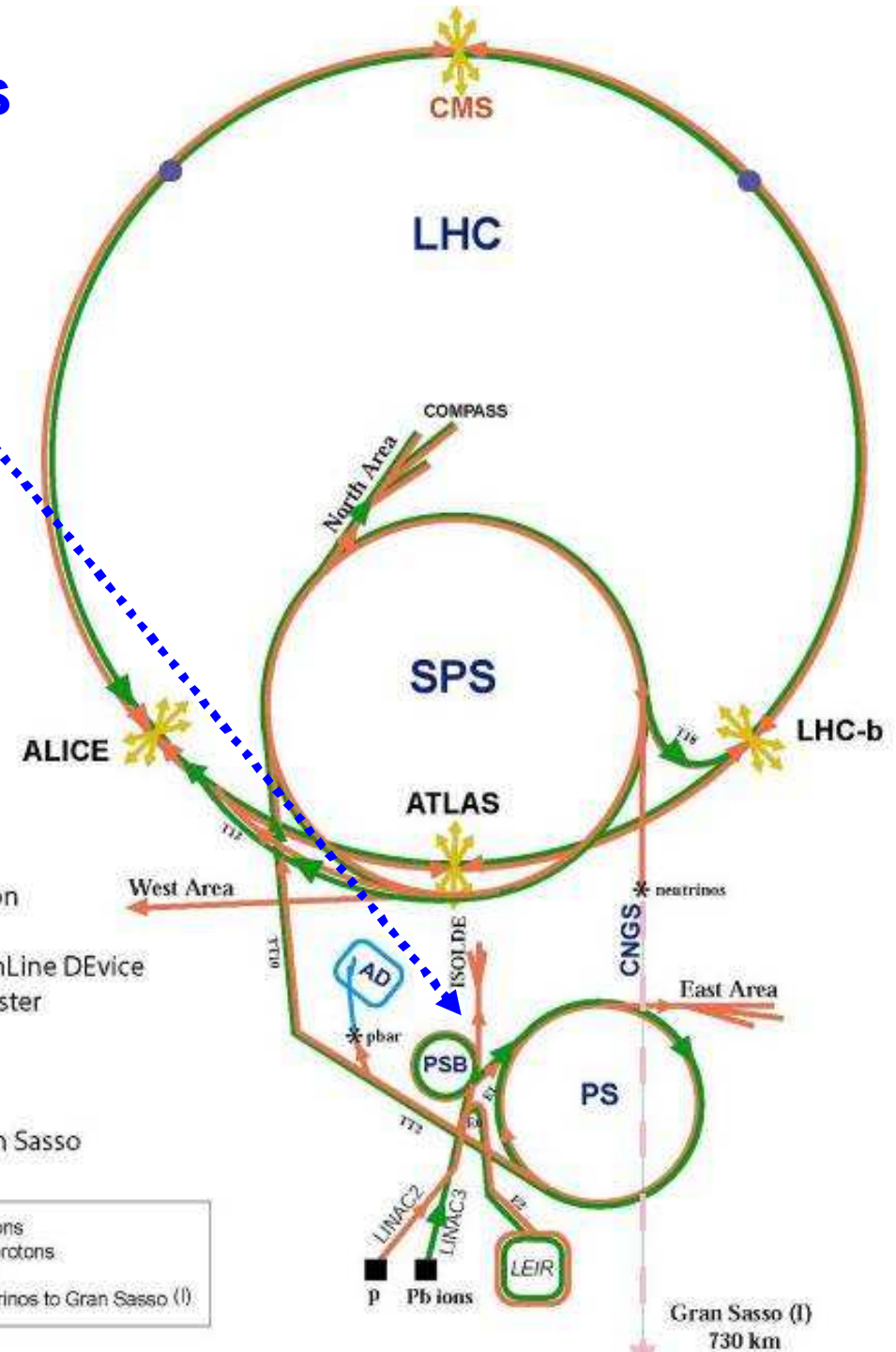
$$E_p = 1.4 \text{ GeV}$$

$$3 \cdot 10^{13} \text{ protons/pulse} = 5 \mu\text{C}$$

$$I_{\text{average}} = 4 \mu\text{A}$$

$$P_{\text{average}} = 6 \text{ kW}$$

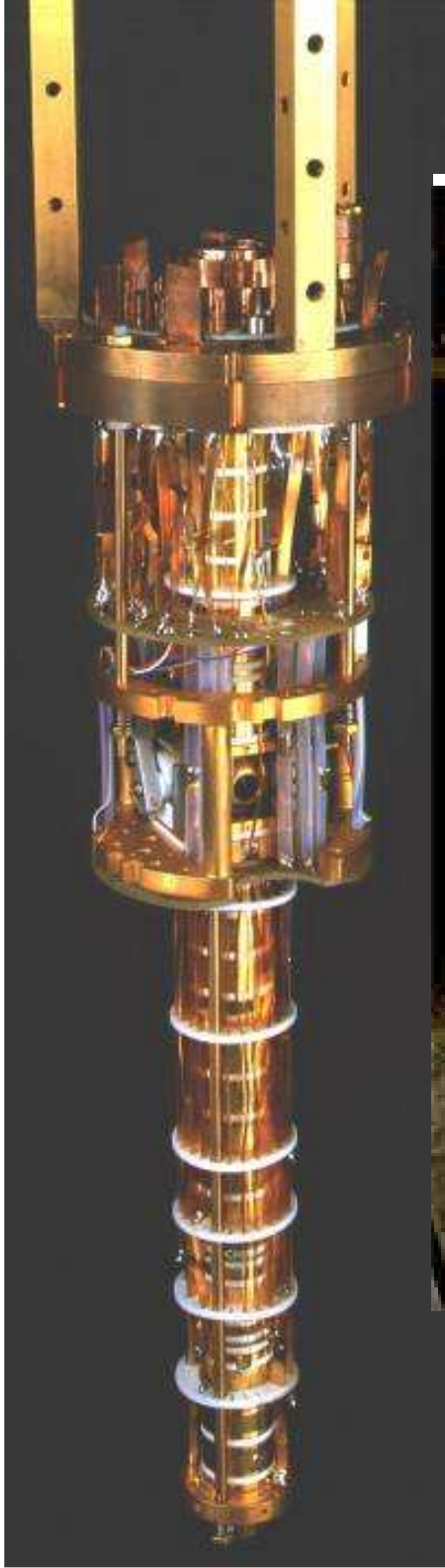
LHC: Large Hadron Collider
SPS: Super Proton Synchrotron
AD: Antiproton Decelerator
ISOLDE: Isotope Separator OnLine DEvice
PSB: Proton Synchrotron Booster
PS: Proton Synchrotron
LINAC: LINear ACcelerator
LEIR: Low Energy Ion Ring
CNGS: Cern Neutrinos to Gran Sasso



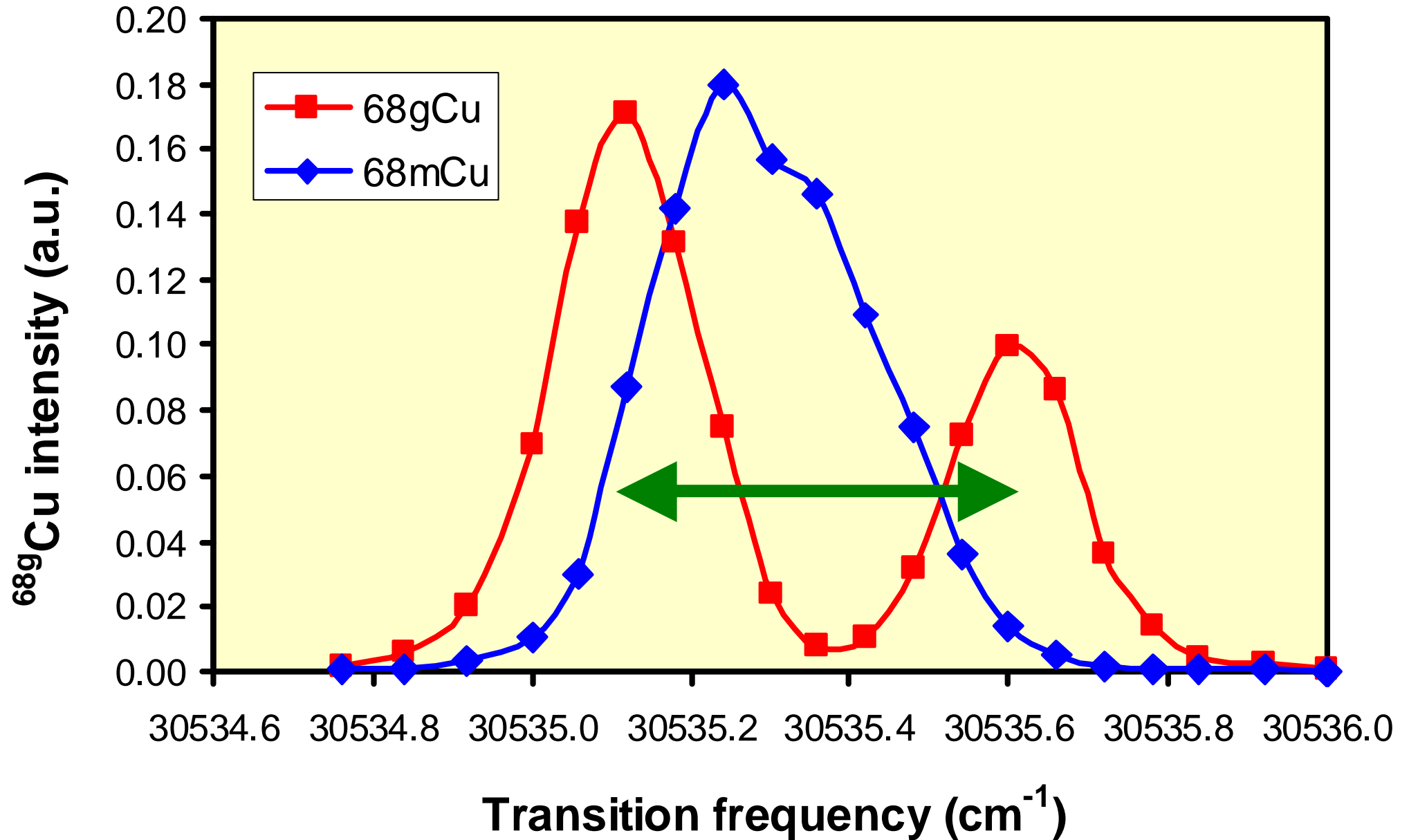
Antiproton traps at CERN



G. Gabrielse et al., Phys. Rev. Lett. 100 (2008) 113001.
M. Amoretti et al., Phys. Lett. B 583 (2004) 59.
D. Brown, R. Howard et al., "Angels and Demons" (2009).



Measurement of magnetic moments



hyperfine splitting proportional to g-factor

Nuclear properties from laser spectroscopy

Hyperfine splitting in electronic transitions:

$$\Delta E_{HFS} = \frac{A}{2} * K + \frac{B}{4} * \frac{\frac{3}{2} K(K+1) - 2I(I+1)J(J+1)}{I(2I-1)J(2J-1)}$$

$$K = \frac{F(F+1) - J(J+1) - I(I+1)}{J(J+1)(2I+1)}$$

$$A = \frac{\mu \cdot H_e(0)}{IJ} \Rightarrow \text{magnetic dipole moment} \quad \mu \quad (J > 0)$$

$$B = \frac{eQ_s}{\left. \frac{d\Phi}{dr} \right|_{(0)}} \Rightarrow \text{electric quadrupole moment} \quad Q \quad (J > 1/2)$$

Isotope shift between two nuclei:

$$\delta \nu_{IS}^{A,A'} = (K_{NMS} + K_{SMS}) * \frac{M_{A'} - M_A}{M_{A'} M_A} + F_{el.} * \delta \langle r^2 \rangle^{A'A}$$

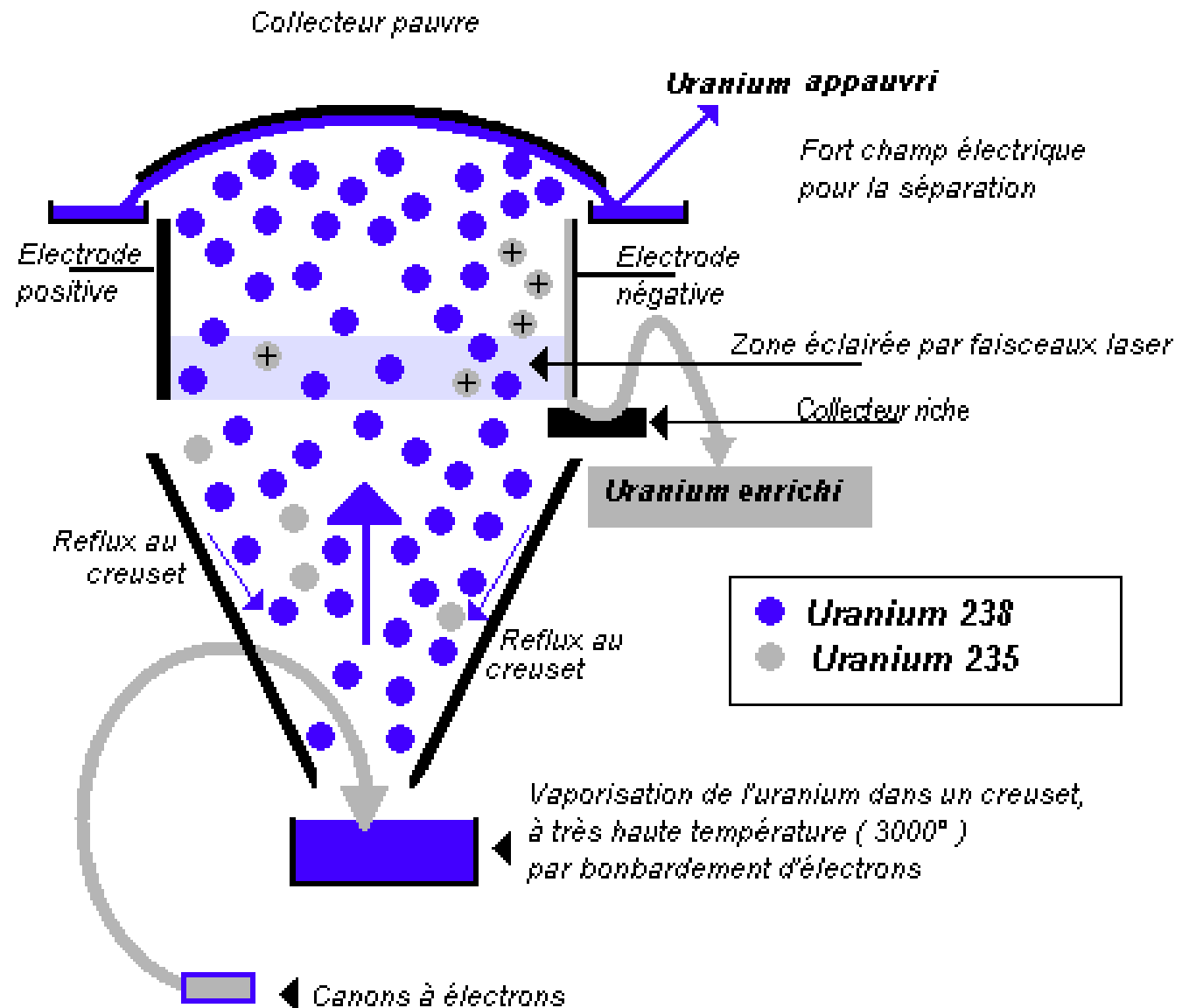
\Rightarrow nuclear charge radii

$\delta \langle r^2 \rangle^{A'A}$

Atomic vapor laser isotope separation (AVLIS)

Shéma du procédé SILVA

large scale application of resonant laser ionization!



Surface ionized background

1 H	Ionization potential: < 5 eV																2 He
3 Li	4 Be	Ionization potential: 5.0 - 5.8 eV										5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg	Ionization potential: 5.8 - 6.5 eV										13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110	111	112						

58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

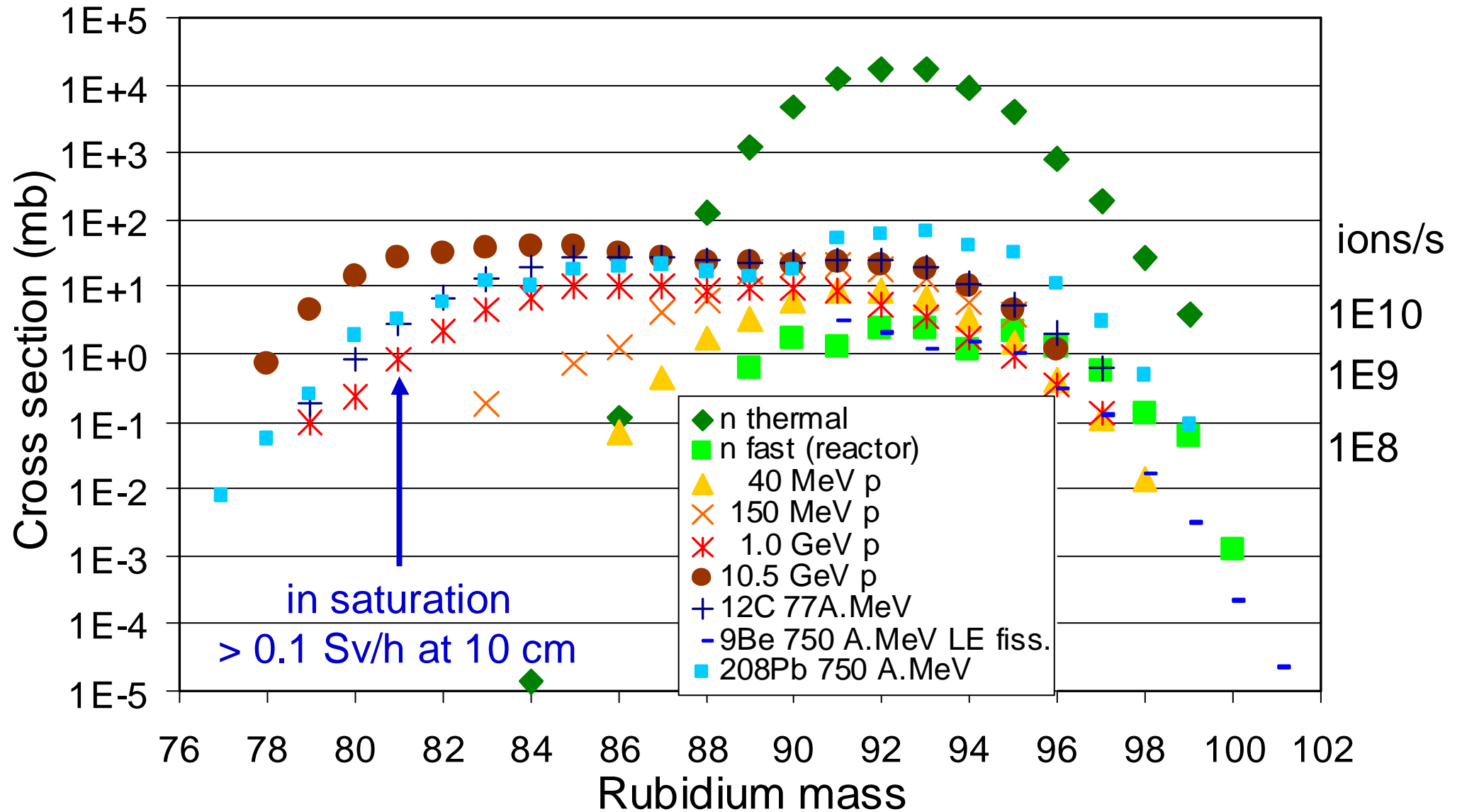
ISOLDE beams around N=50

Sr 78 2,65 m	Sr 79 2,3 m	Sr 80 1,8 h	Sr 81 22,2 m	Sr 82 25,34 d	Sr 83 5,0 s 30,4 h	Sr 84 0,56	Sr 85 67,7 m 64,9 d	Sr 86 9,86	Sr 87 2,81 h 7,01	Sr 88 82,58	Sr 89 50,5 d	Sr 90 28,84 a	Sr 91 9,5 h	Sr 92 2,71 h
Rb 77 3,9 m	Rb 78 5,7 m 17,7 m	Rb 79 23,0 m	Rb 80 30 s	Rb 81 20,3 m 4,58 h	Rb 82 6,3 h 1,27 m	Rb 83 86,2 d	Rb 84 20,5 m 32,8 d	Rb 85 72,165	Rb 86 1,02 m 18,7 d	Rb 87 27,835	Rb 88 17,8 m	Rb 89 15,2 m	Rb 90 4,2 m 2,6 m	Rb 91 58 s
Kr 76 14,6 h	Kr 77 1,24 h	Kr 78 0,35	Kr 79 50 s 34,9 s	Kr 80 2,25	Kr 81 13,1 s 2,3 10 ⁵ s	Kr 82 11,6	Kr 83 1,83 h 11,5	Kr 84 57,0	Kr 85 4,40 h 10,78 a	Kr 86 17,3	Kr 87 76,3 m	Kr 88 2,84 h	Kr 89 3,18 m	Kr 90 32,3 s
Br 75 1,6 h	Br 76 1,32 s 16,4 h	Br 77 4,3 s 57,6 h	Br 78 6,46 m	Br 79 4,9 s 90,89	Br 80 4,42 s 17,6 m	Br 81 49,31	Br 82 8,1 m 25,34 h	Br 83 2,40 h	Br 84 6,0 h 31,8 m	Br 85 2,87 m	Br 86 55,1 s	Br 87 55,7 s	Br 88 16,3 s	Br 89 4,40 s

⁸¹Rb background is 150000 times more abundant than ⁸¹Zn!

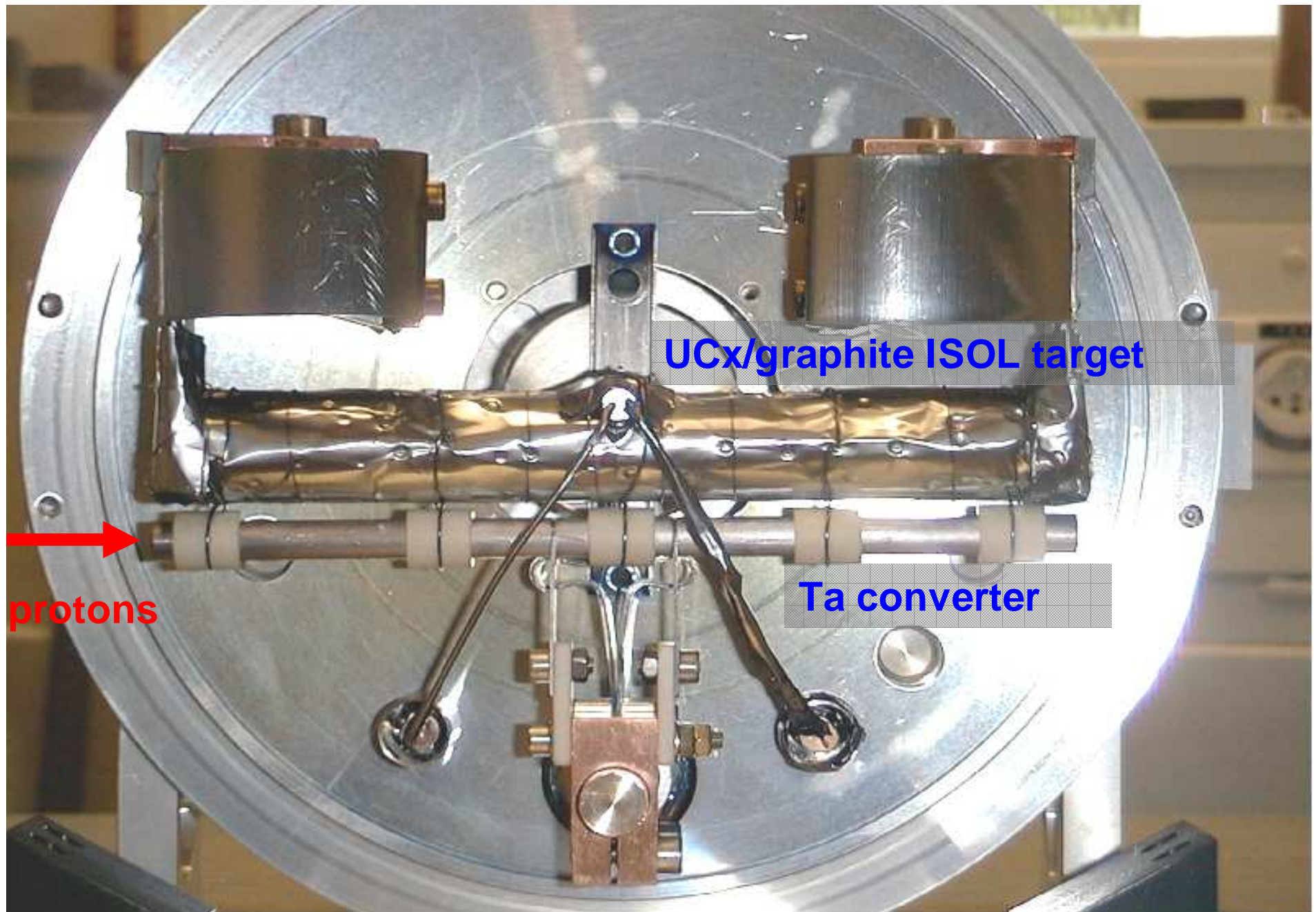
As 73 80,3 d	As 74 17,77 d	As 75 100	As 76 26,4 h	As 77 38,8 h	As 78 1,5 h	As 79 8,2 m	As 80 15,2 s	As 81 34 s	As 82 14,2 s 13,1 s	As 83 13,3 s	As 84 4,5 s	As 85 2,03 s	As 86 0,9 s	As 87 0,73 s
Ge 72 27,66	Ge 73 7,73	Ge 74 33,94	Ge 75 47 s 63 m	Ge 76 7,44	Ge 77 53 s 11,3 h	Ge 78 88 m	Ge 79 36 s 19 s	Ge 80 29,5 s	Ge 81 7,6 s 7,6 s	Ge 82 4,60 s	Ge 83 1,85 s	Ge 84 984 ms	Ge 85 535 ms	Ge 86
Ga 71 39,892	Ga 72 14,1 h	Ga 73 4,86 h	Ga 74 3,3 s 8,1 m	Ga 75 2,1 m	Ga 76 32,6 s	Ga 77 13 s	Ga 78 5,49 s	Ga 79 2,85 s	Ga 80 1,70 s	Ga 81 1,22 s	Ga 82 0,60 s	Ga 83 0,31 s	Ga 84 85 ms	1,327 54
Zn 70 0,6	Zn 71 3,9 h 24 m	Zn 72 46,5 h	Zn 73 5,8 s 20,5 s	Zn 74 96 s	Zn 75 10,2 s	Zn 76 5,6 s	Zn 77 1,05 s 2,28 s	Zn 78 1,47 s	Zn 79 995 ms	Zn 80 537 ms	Zn 81 0,29 s	0,3239	0,5490	1,005
Cu 69 3,0 m	Cu 70 42 s 3 s	Cu 71 19,5 s	Cu 72 6,6 s	Cu 73 3,9 s	Cu 74 1,59 s	Cu 75 1,22 s	Cu 76 1,27 s 641 ms	Cu 77 469 ms	Cu 78 342 ms	Cu 79 188 ms	Cu 80	0,1974		
Ni 68 29 s	Ni 69 11,4 s	Ni 70 6,0 s	Ni 71 2,56 s	Ni 72 1,57 s	Ni 73 0,84 s	Ni 74 0,9 s	Ni 75 0,6 s	Ni 76 - 0,24 s	Ni 77	Ni 78		0,04712	0,1269	

Rubidium cross-sections



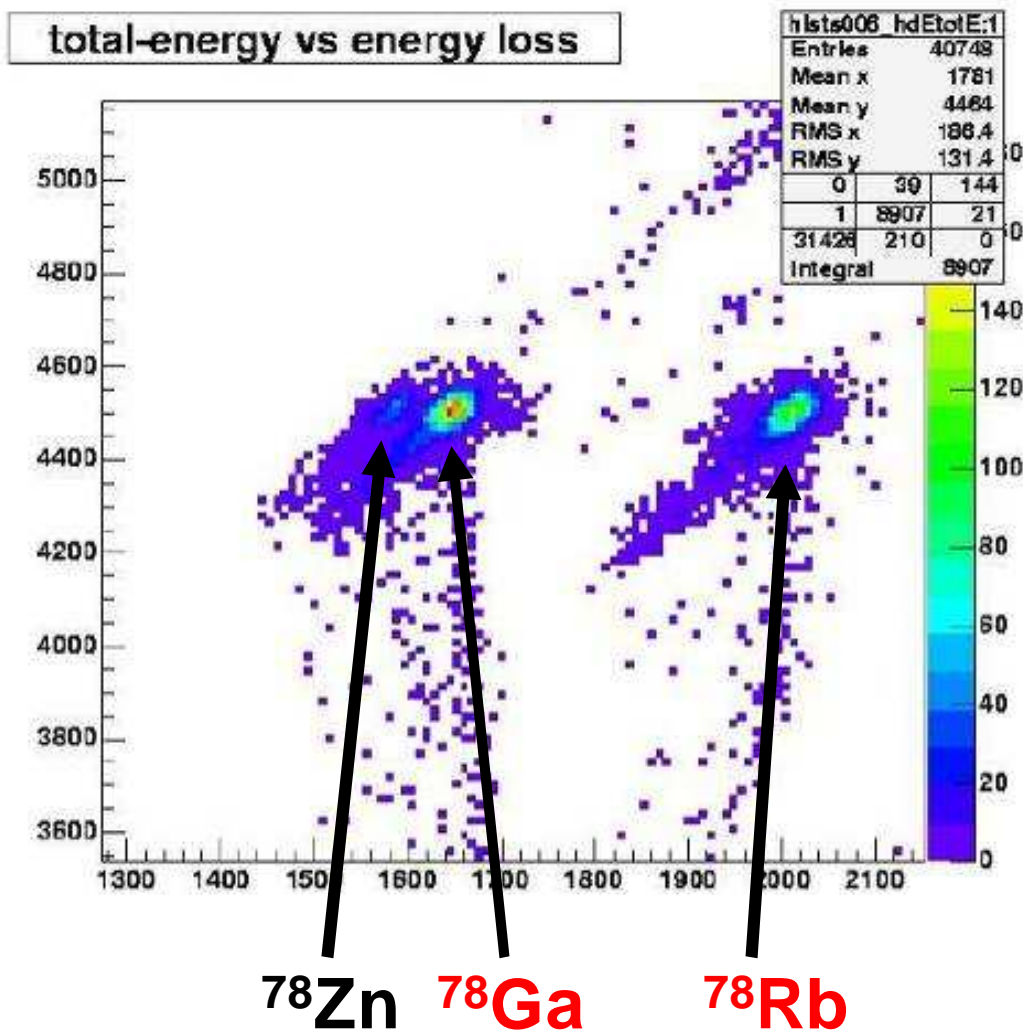
better use neutron-induced fission!

Neutron “converter” geometry

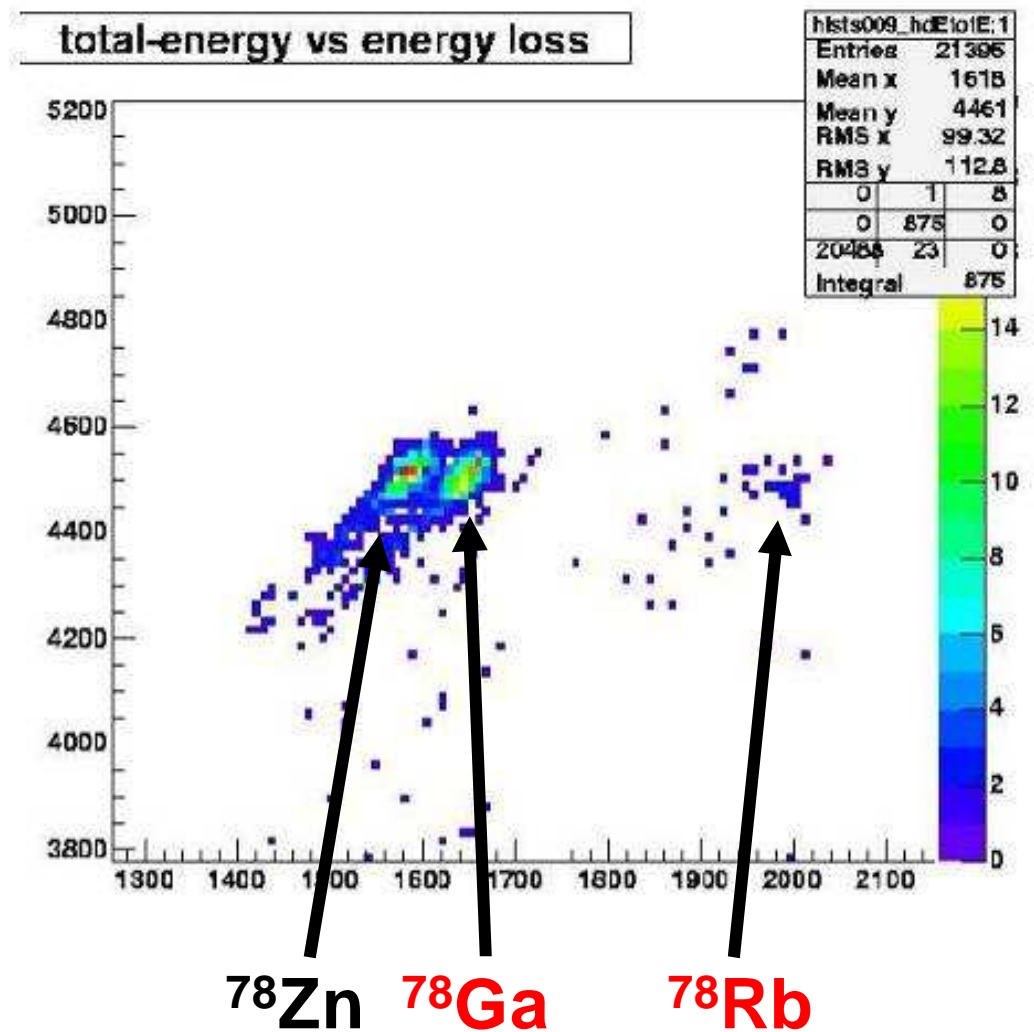


Ga & Rb suppression with the “neutron converter”

proton beam on **target**

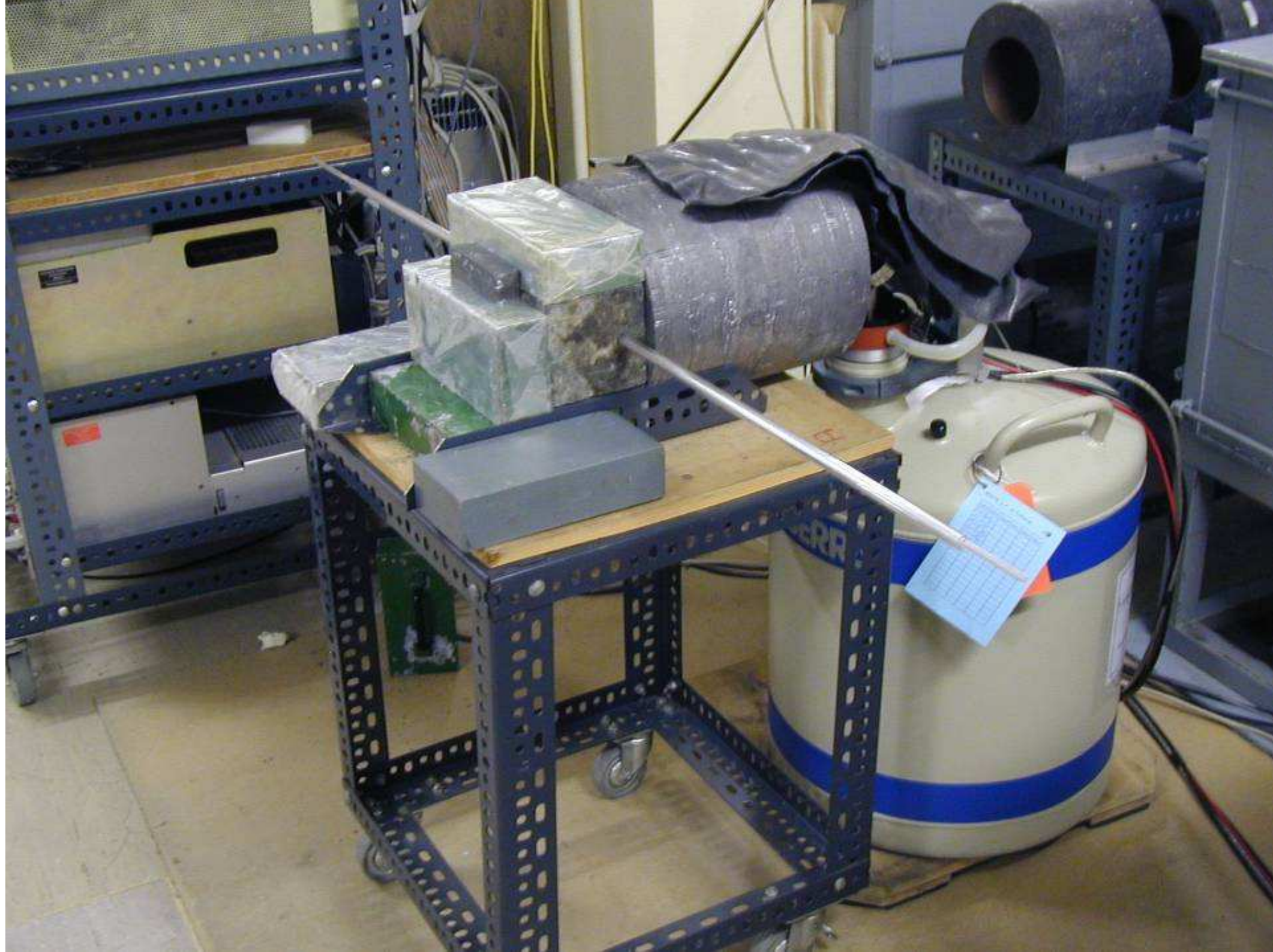


proton beam on **converter**

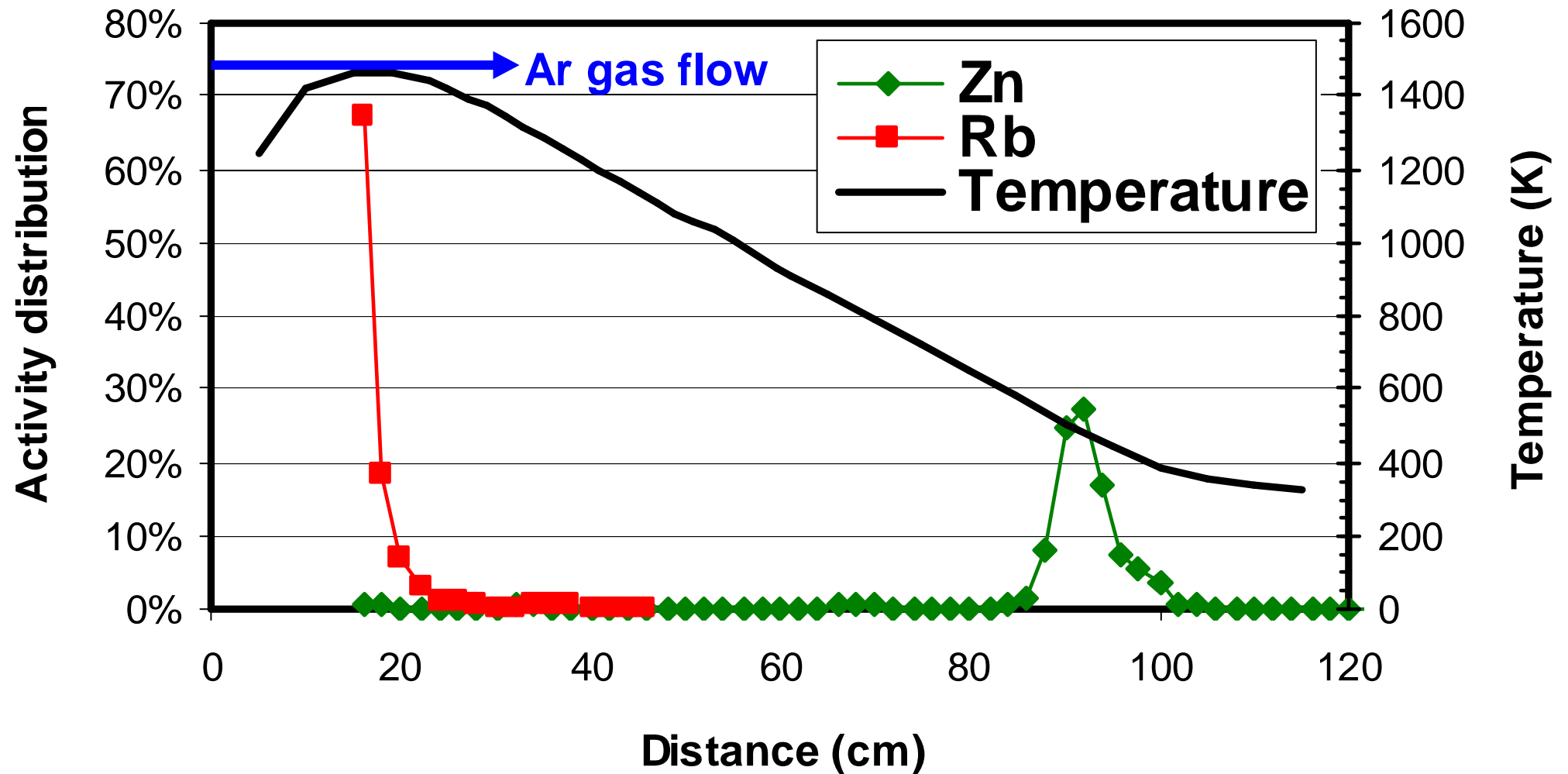


ISOLDE thermochromatography set-up





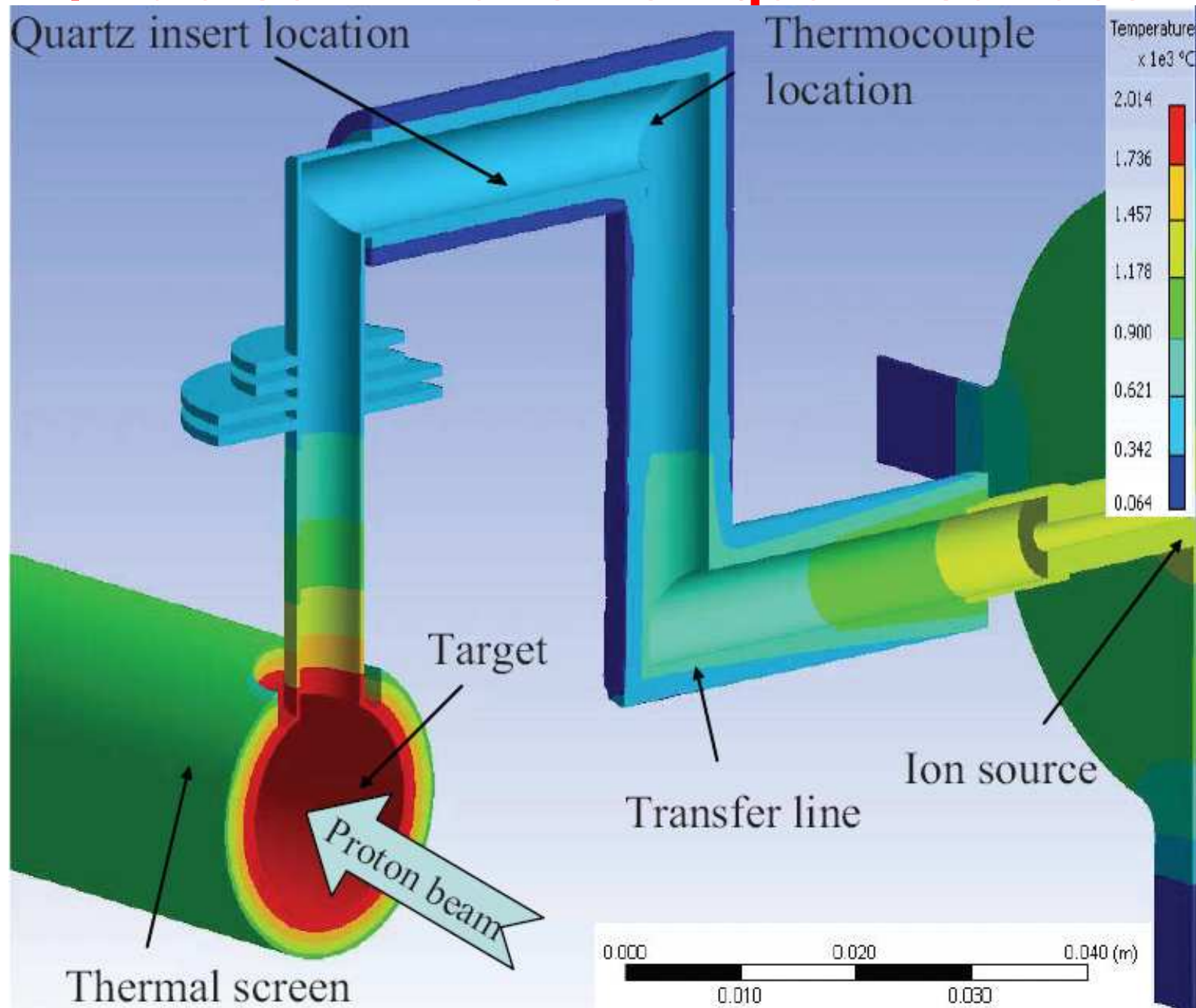
Zn/Rb discrimination on quartz surface!



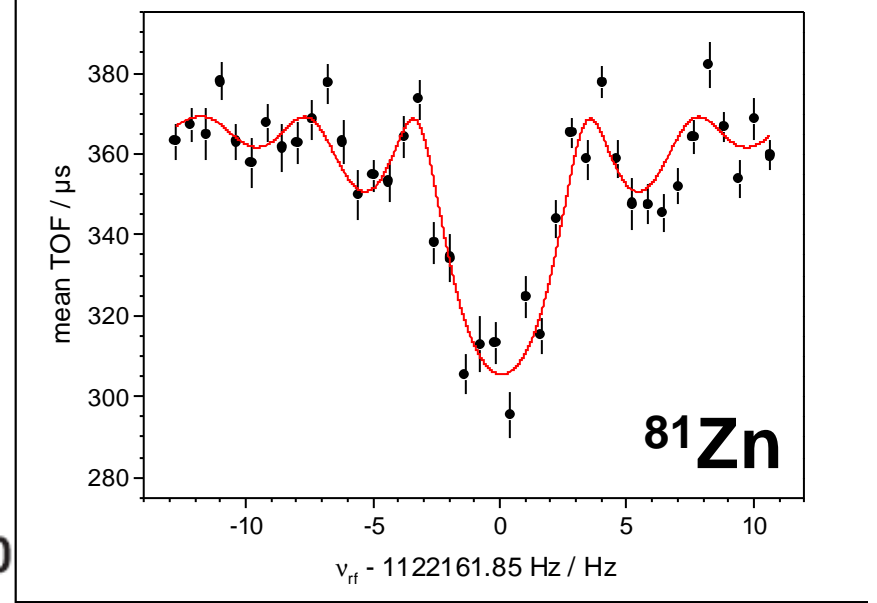
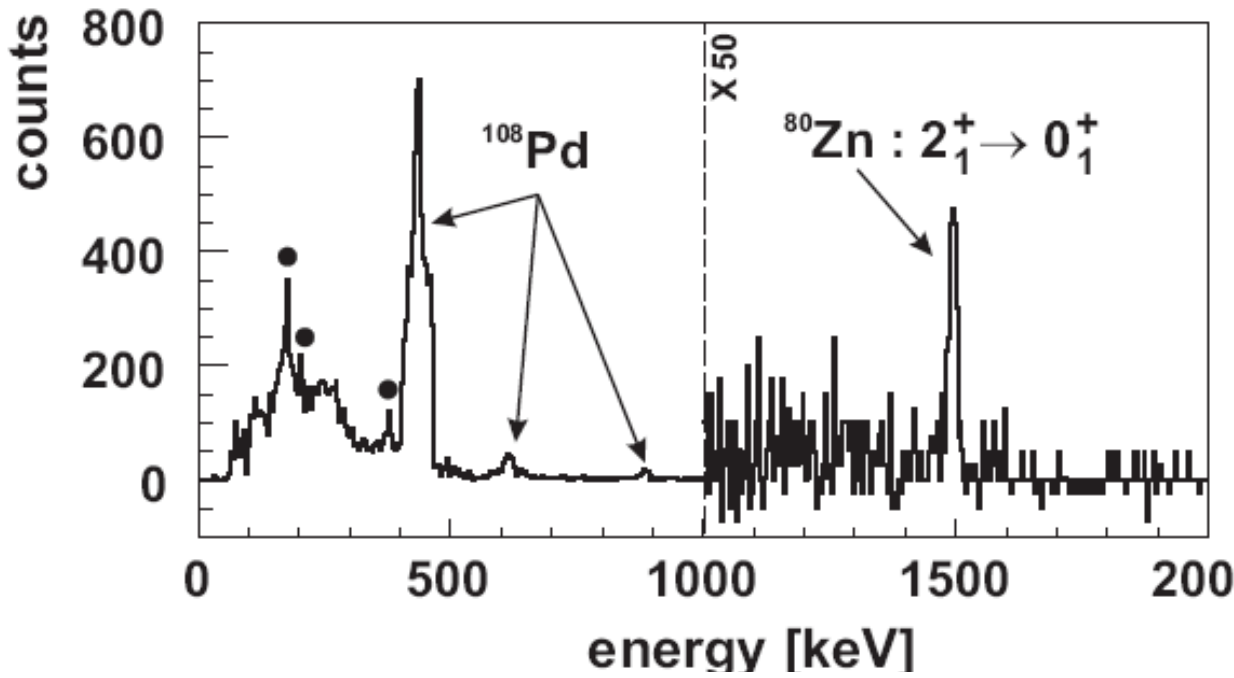
Combination of neutron converter and quartz transfer line provides $^{81}\text{Zn}/^{81}\text{Rb}$ selectivity gain of 100000!

U.K. et al., Nucl. Instr. Meth. B266 (2008) 4229.

Zn/Rb discrimination on quartz surface!



E. Boucquerel et al., Nucl. Instr. Meth. B266 (2008) 4298.

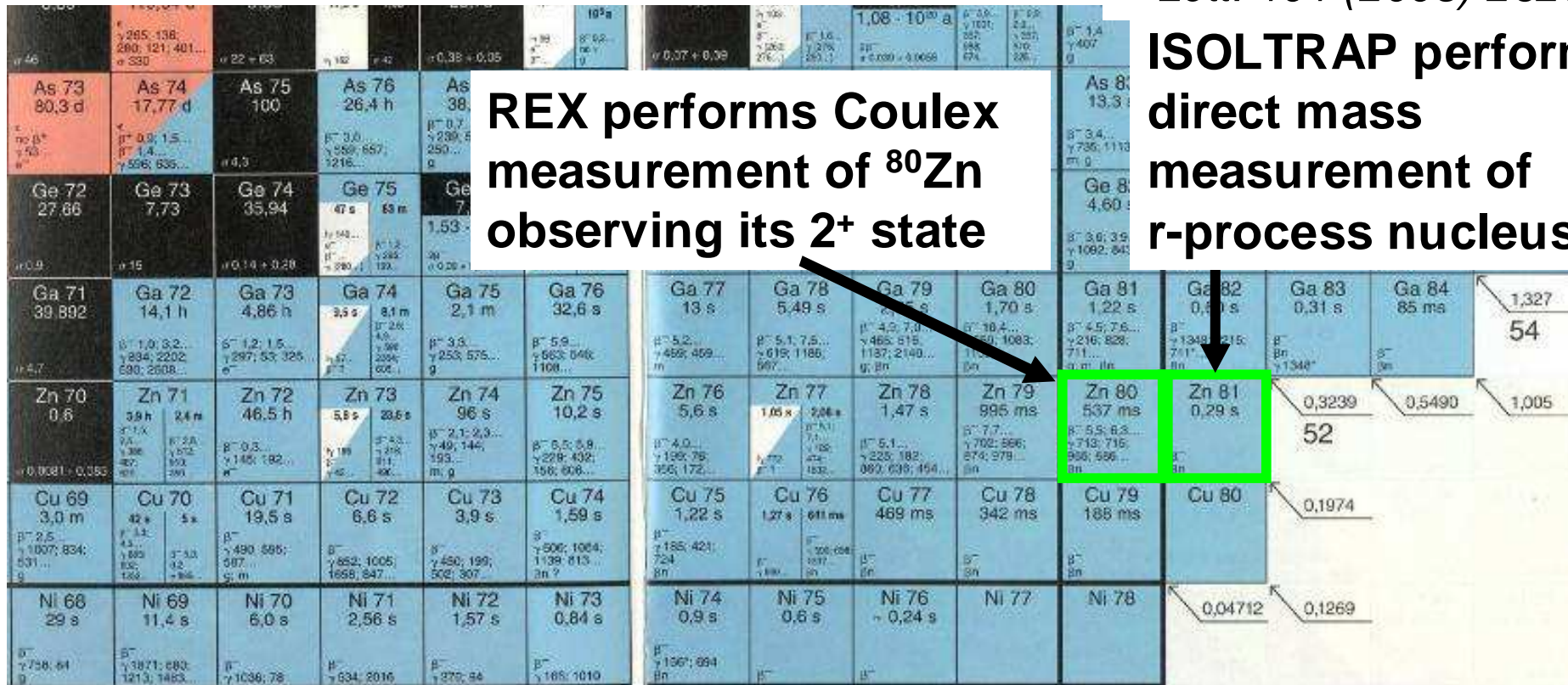


J. Van de Walle et al., Phys. Rev. Lett. 99 (2007) 142501.

S. Baruah et al., Phys. Rev. Lett. 101 (2008) 262501.

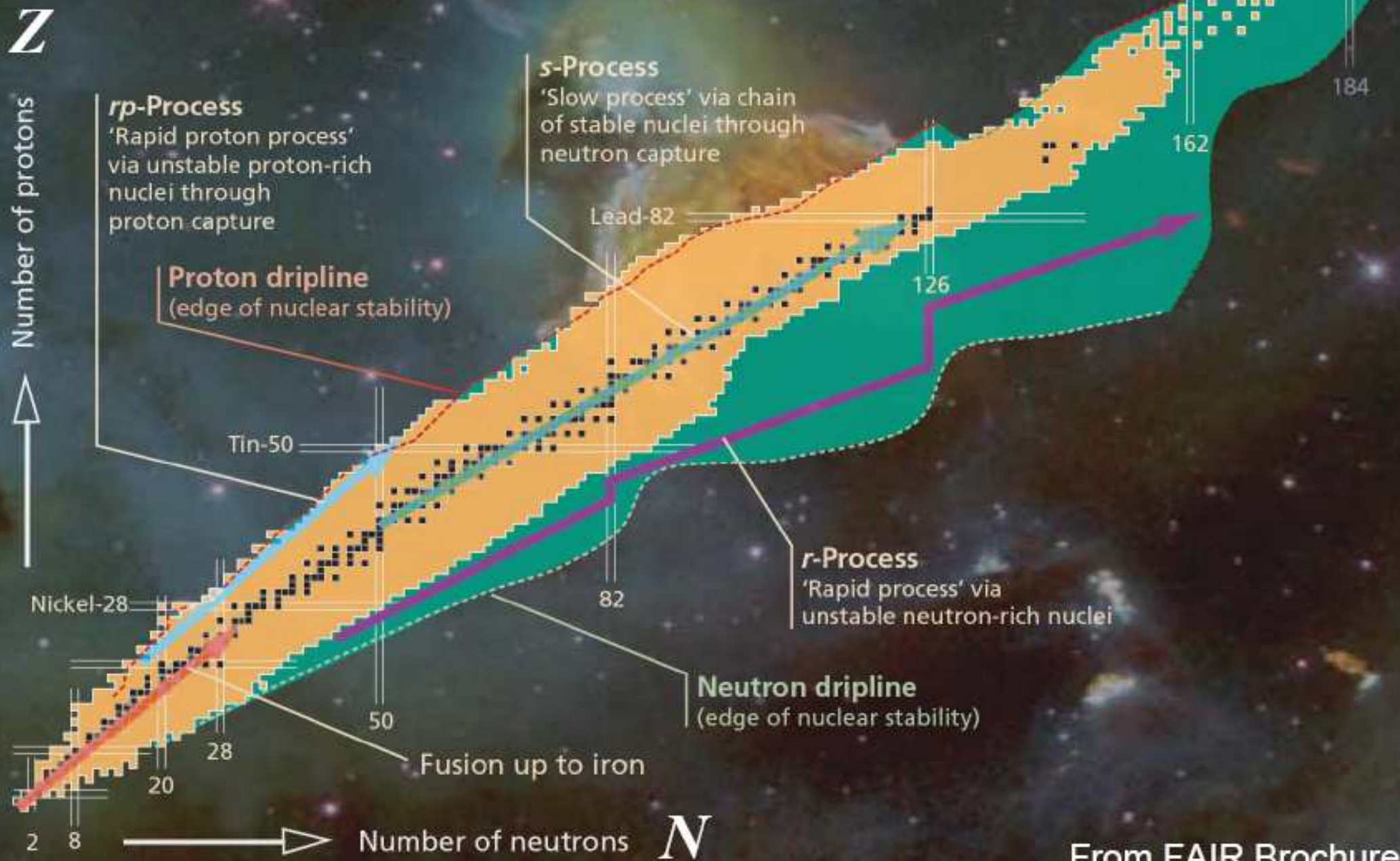
REX performs Coulex measurement of ^{80}Zn observing its 2^+ state

ISOLTRAP performs direct mass measurement of r-process nucleus!



Experimental access to *r*-process nuclides

The landscape of possible nuclei



Elements ionizable with CVL pumped dye lasers

elements ionized with ISOLDE RILIS

tested ionization scheme

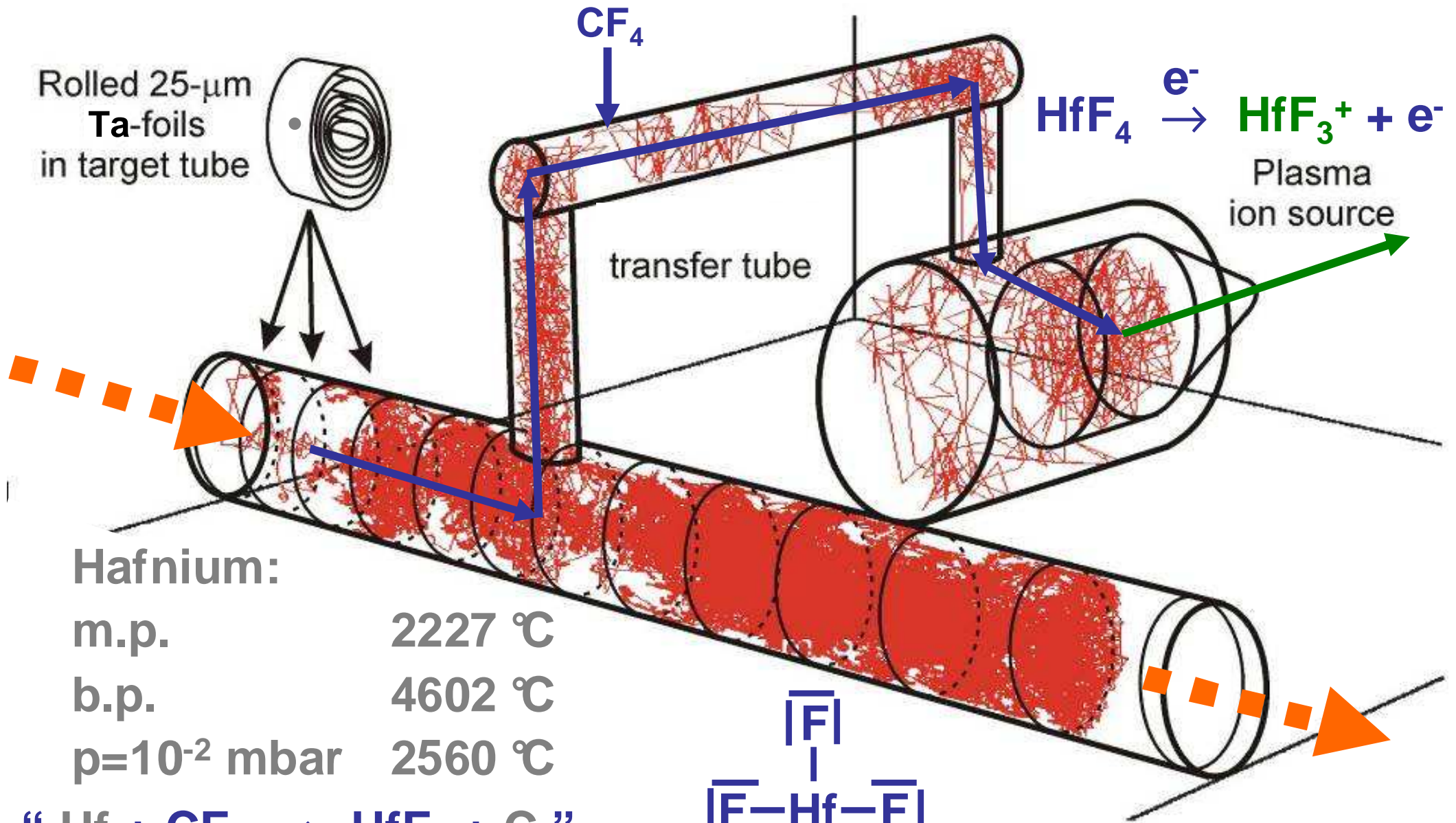
possible ionization scheme (untested)

refractory elements

1 H																	2 He				
3 Li	4 Be															5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg															13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr				
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe				
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn				
87 Fr	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110	111	112										

58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

Hf beams at ISOLDE



Hafnium:

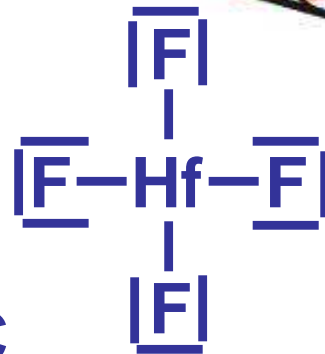
m.p. 2227 $^\circ\text{C}$

b.p. 4602 $^\circ\text{C}$

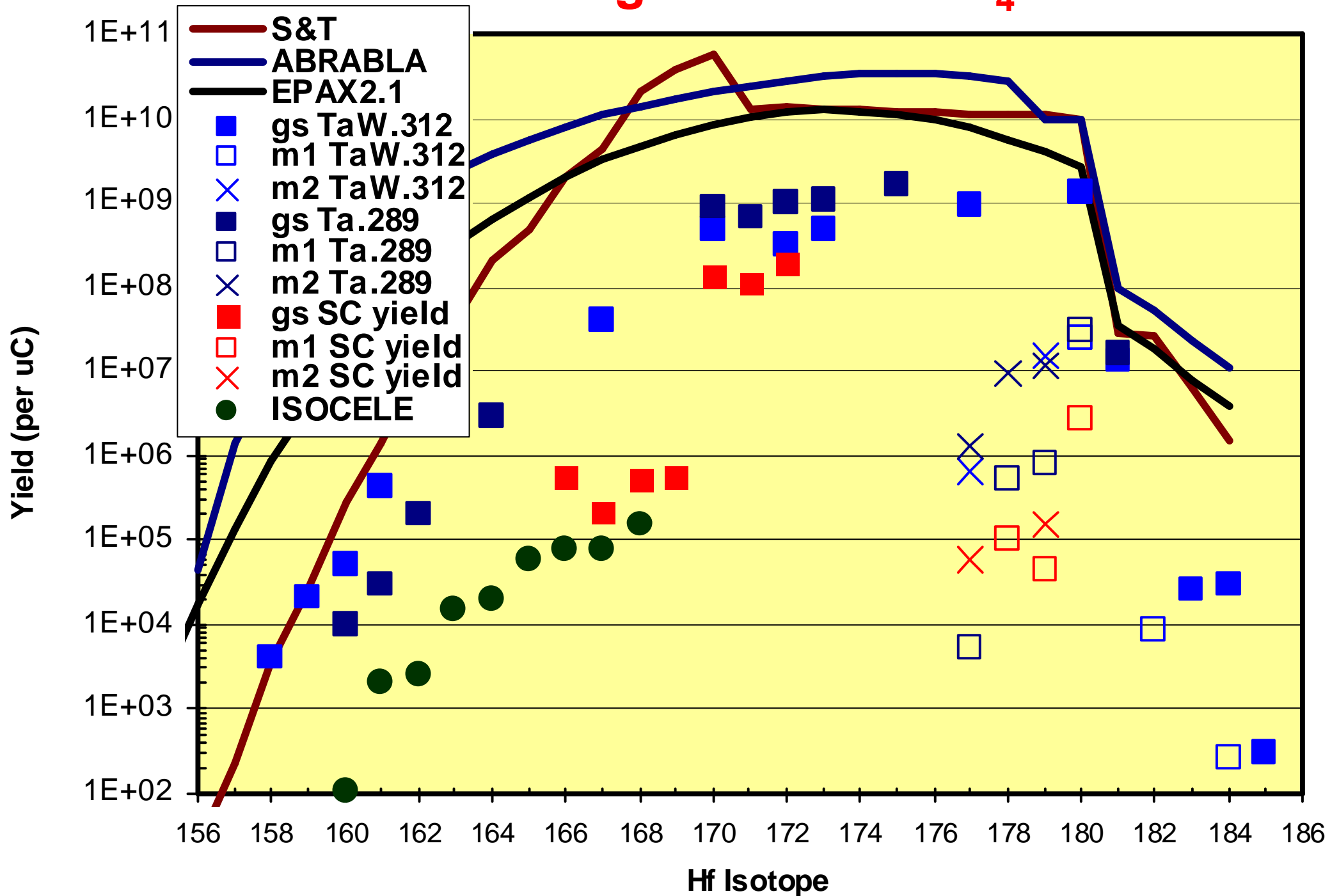
$p=10^{-2}$ mbar 2560 $^\circ\text{C}$



HfF_4 : $p=10^{-2}$ mbar ca. 1200 $^\circ\text{C}$



Ta/W foil target + MK5 + CF₄



Neutron-rich hafnium beams



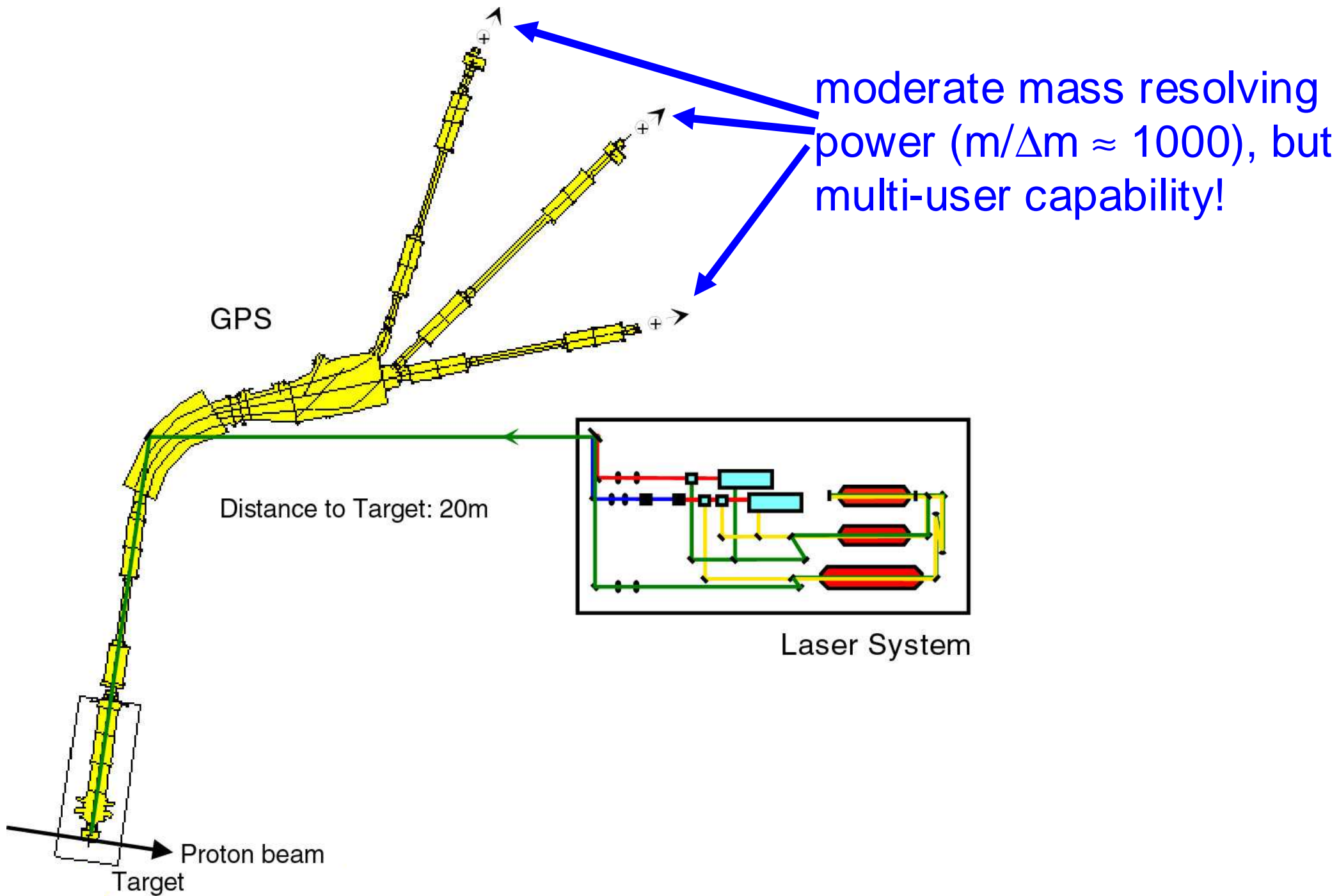
Overview of molecular ISOL beams

		Separation as XF^+ , XCl^+					Separation as XO_x^+					Separation as XCO^+						
		Separation as XF_2^+					Separation as XS^+					Separation as AlX^+						
		Separation as XF_3^+					Separation as HX^+											
		Separation as XF_4^+					Separation as NX^+											
1 H																		2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F		10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl		18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br		36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I		54 Xe
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At		86 Rn
87 Fr	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110	111	112	112	112	112				

58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

U.K. et al., Nucl. Instr. Meth. B266 (2008) 4229.

General Purpose Separator (GPS)



Francium beams

Francium

1 H																	2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89	104	105	106	107	108	109	110	111	112						

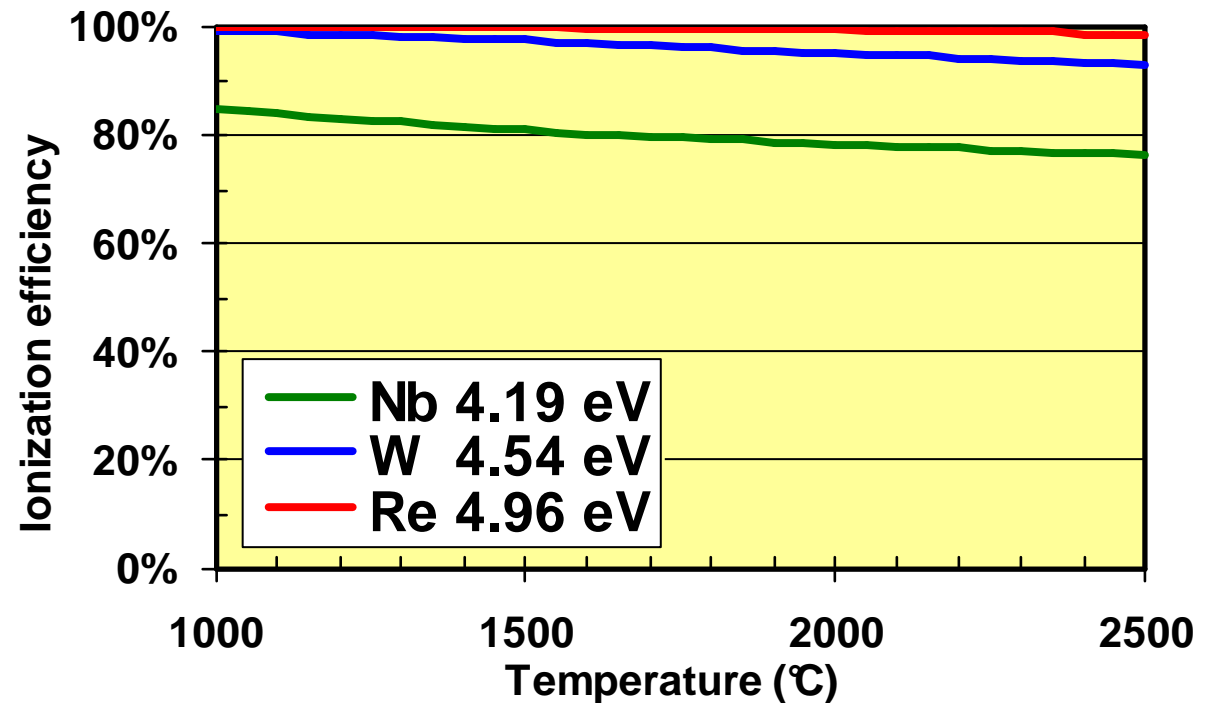
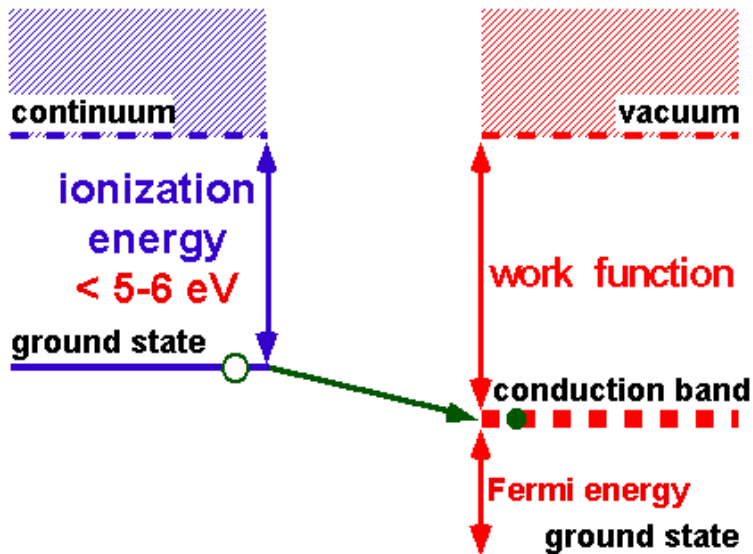
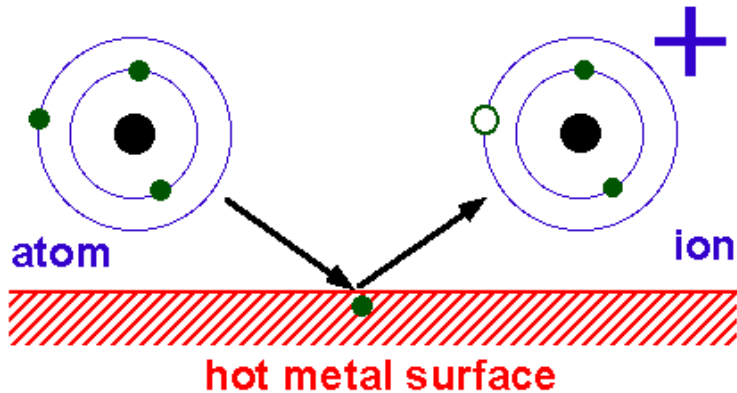
	r (pm)	m.p. (°C)	b.p. (°C)	IP (eV)
Cs	265.4	28.5	671	3.89
Fr	270	27	677	4.07

68 Er	69 Tm	70 Yb	71 Lu
100 Fm	101 Md	102 No	103 Lr

Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
----	----	---	----	----	----	----	----	----	----	----	----	----	----

Positive surface ionization source

Surface Ionization



$$\alpha_s = g_+/g_0 \exp(-(IP - \Phi)/kT)$$

$$\varepsilon_s = \alpha_s / (1 + \alpha_s)$$

Saha-Langmuir equation

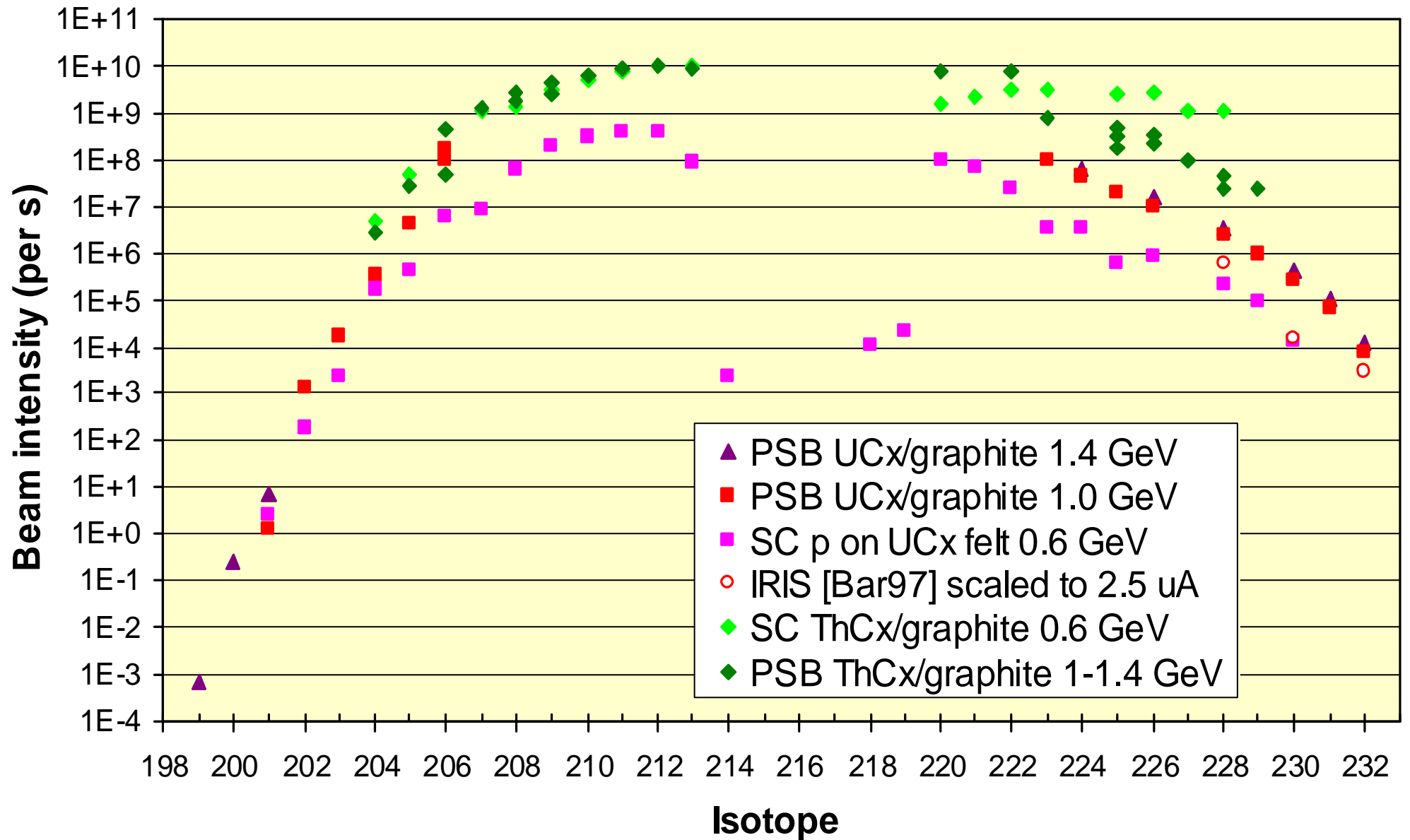
ε_s surface ionization efficiency

Φ work function of surface

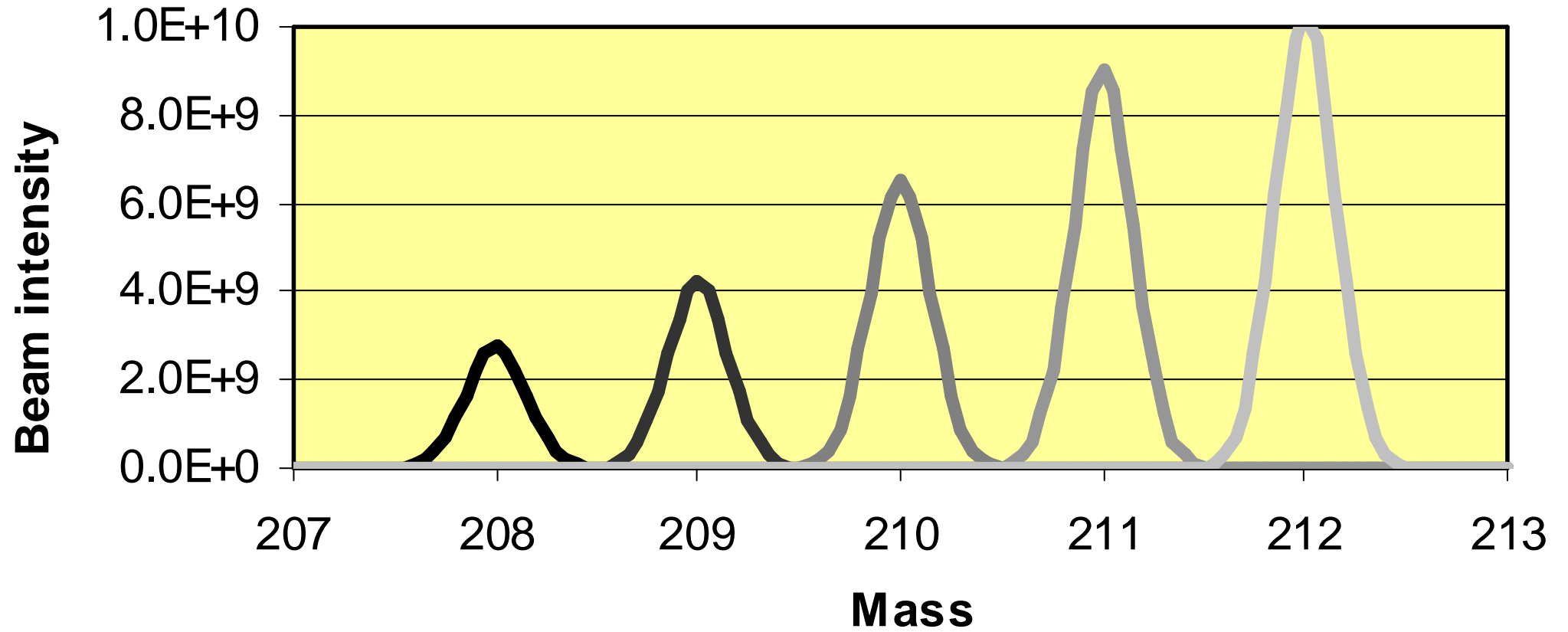
IP ionization potential of atom

$g=2J+1$ stat. factor ($g_0=2, g_+=1$ for alkalis)

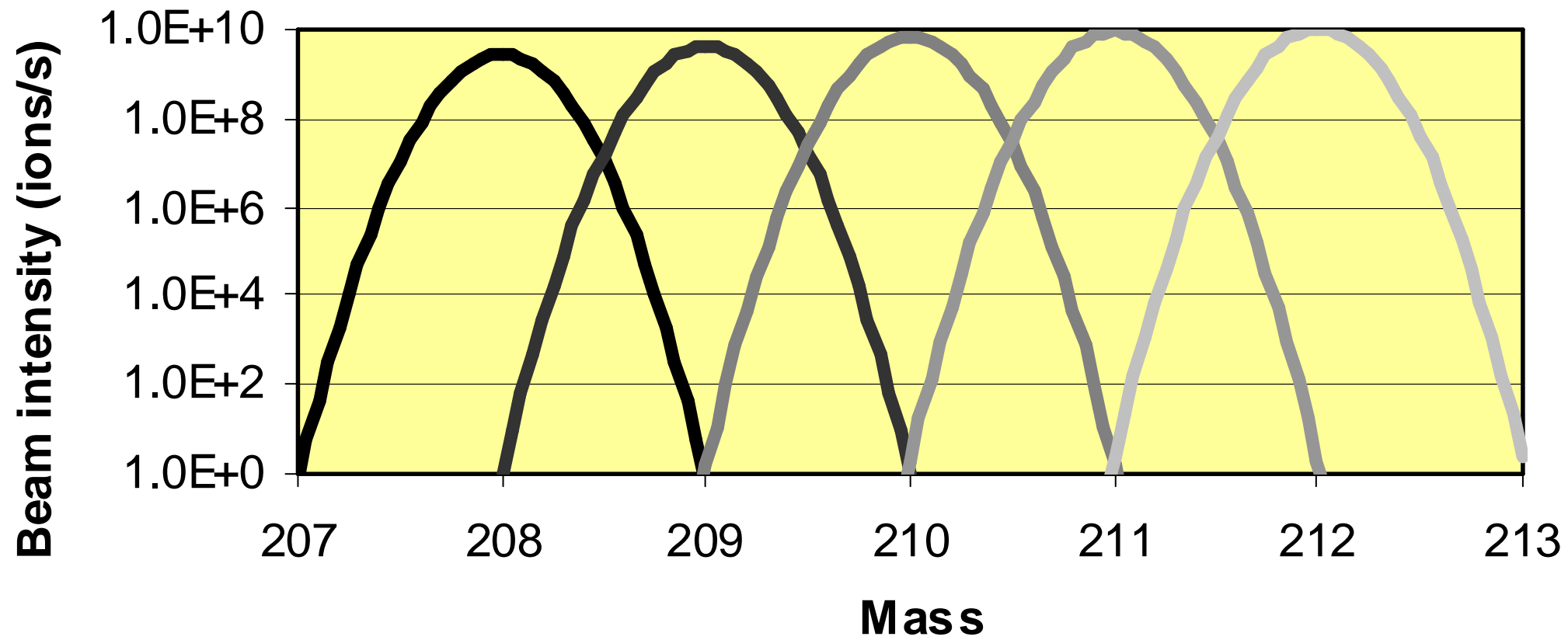
Francium beam intensities



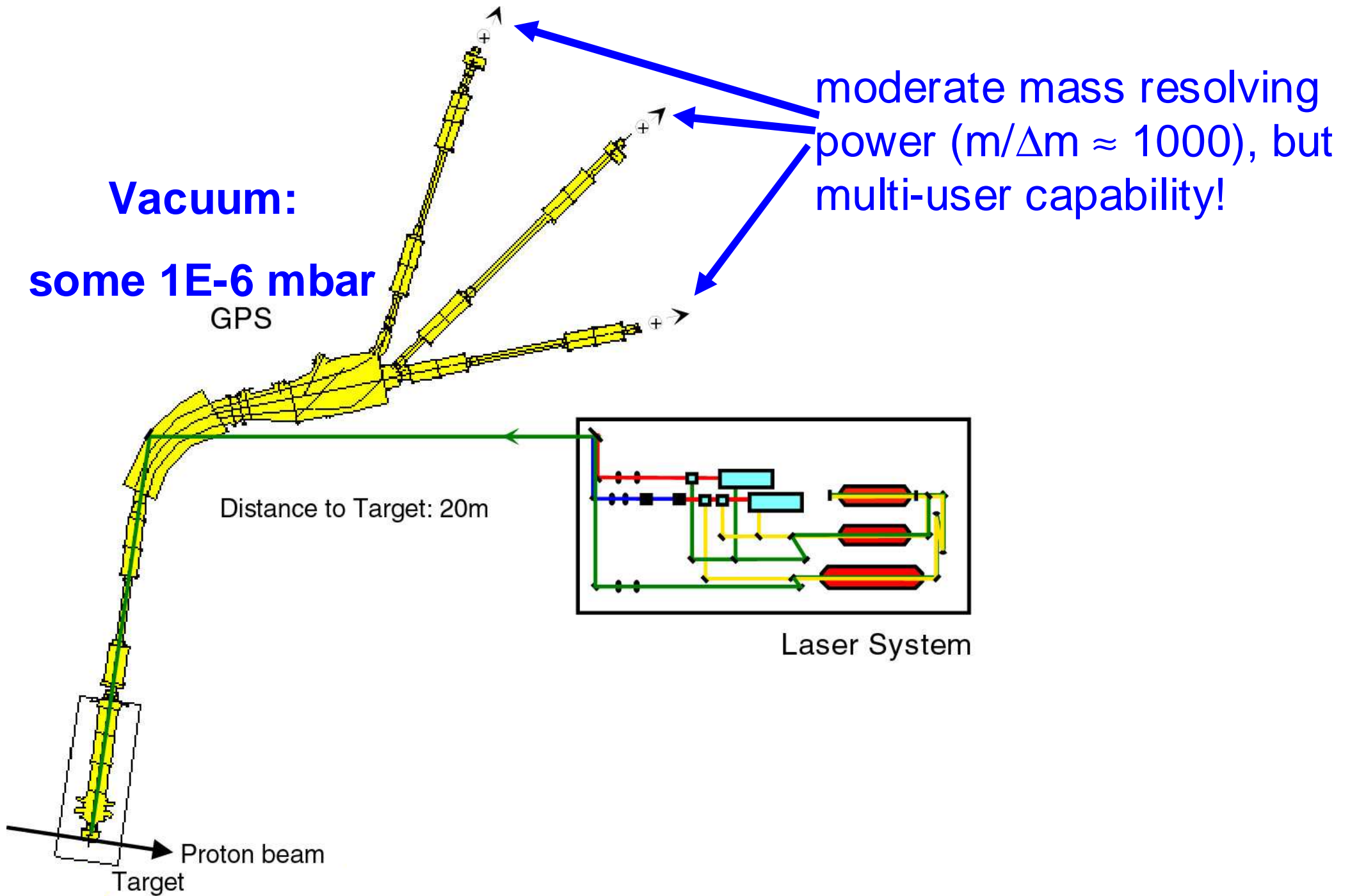
Mass separation



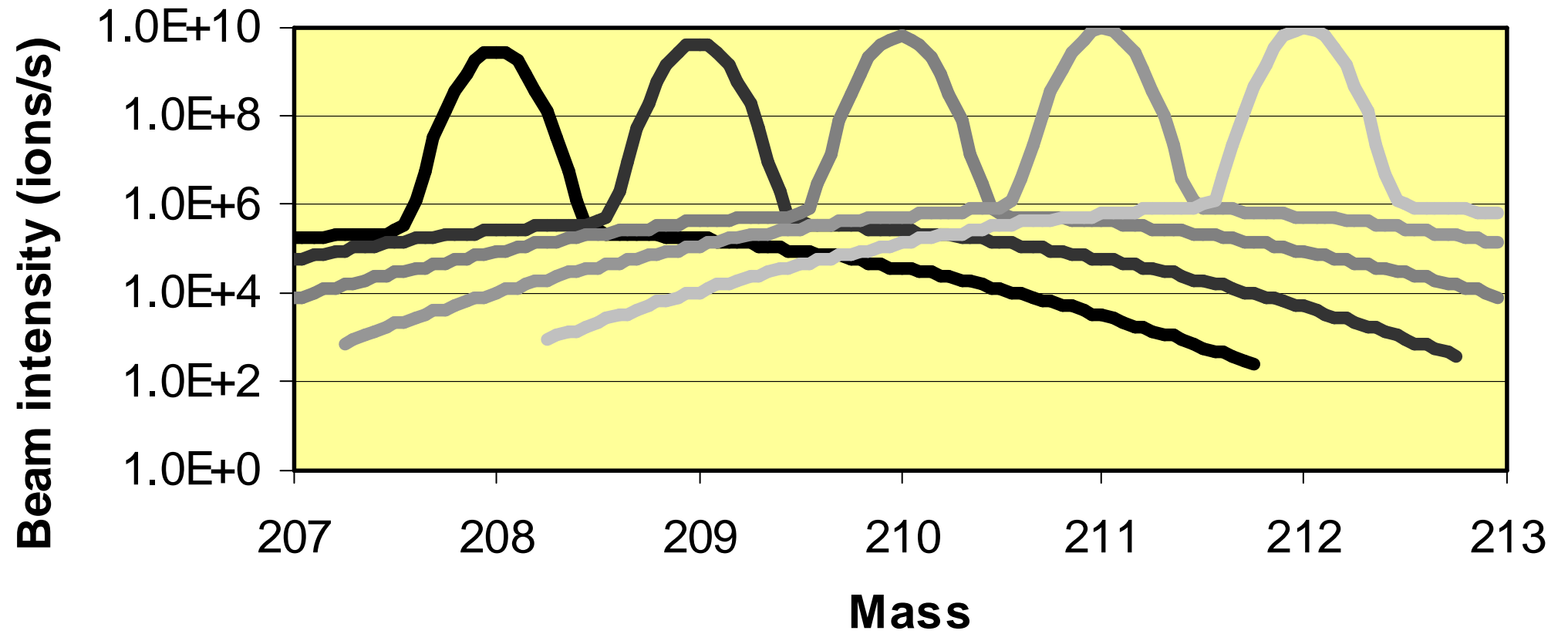
Mass separation



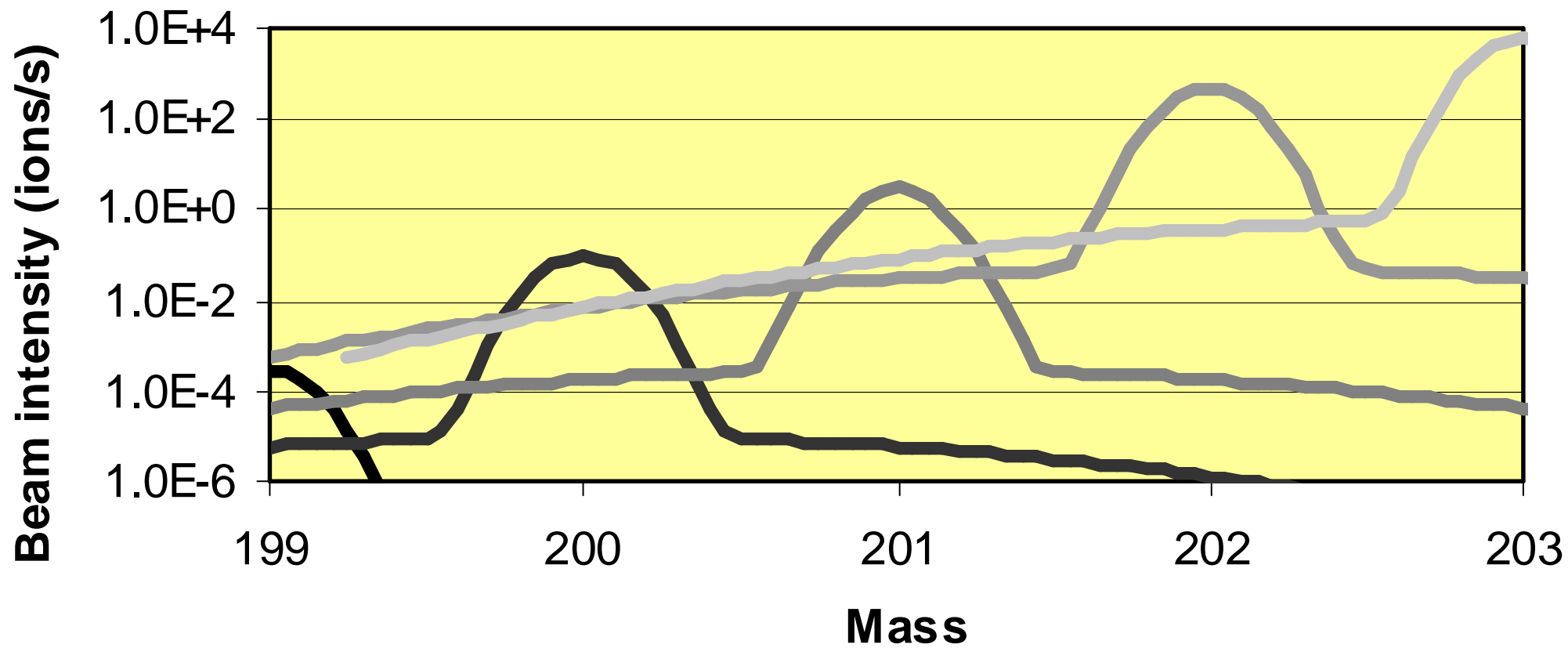
General Purpose Separator (GPS)



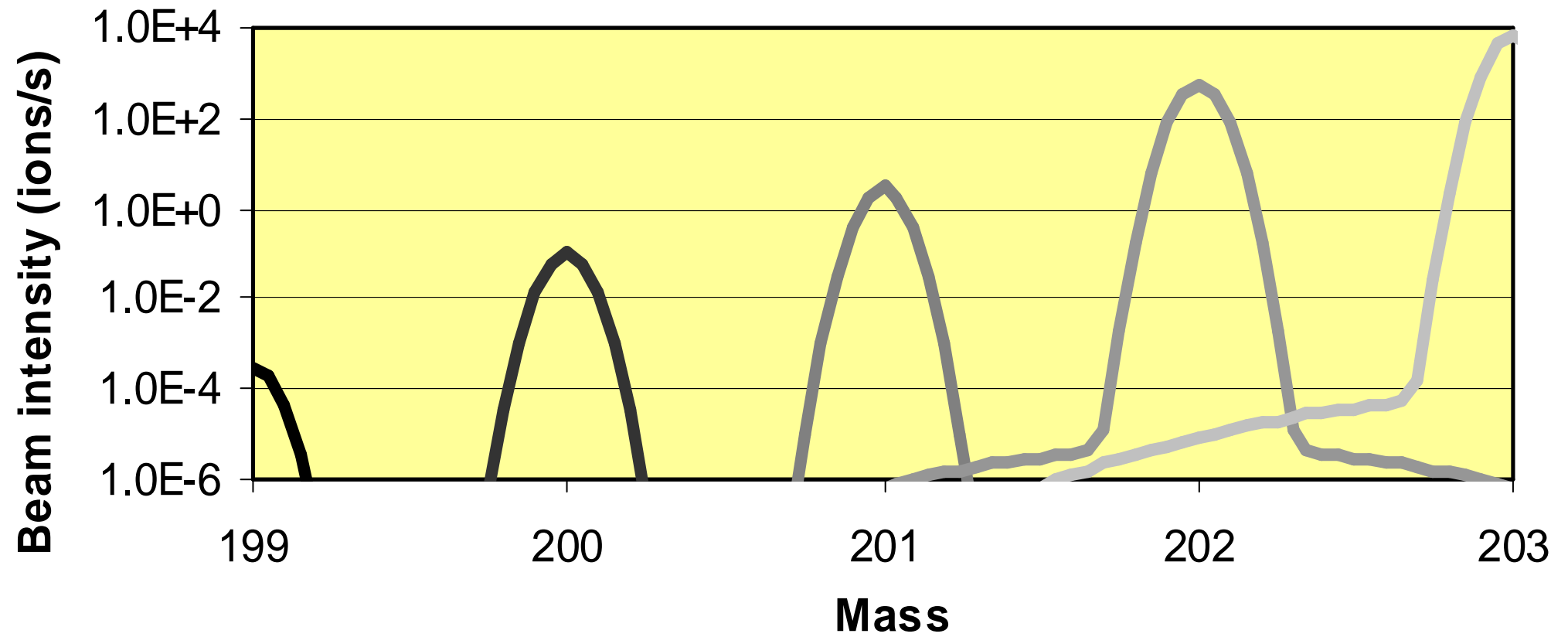
Mass separation



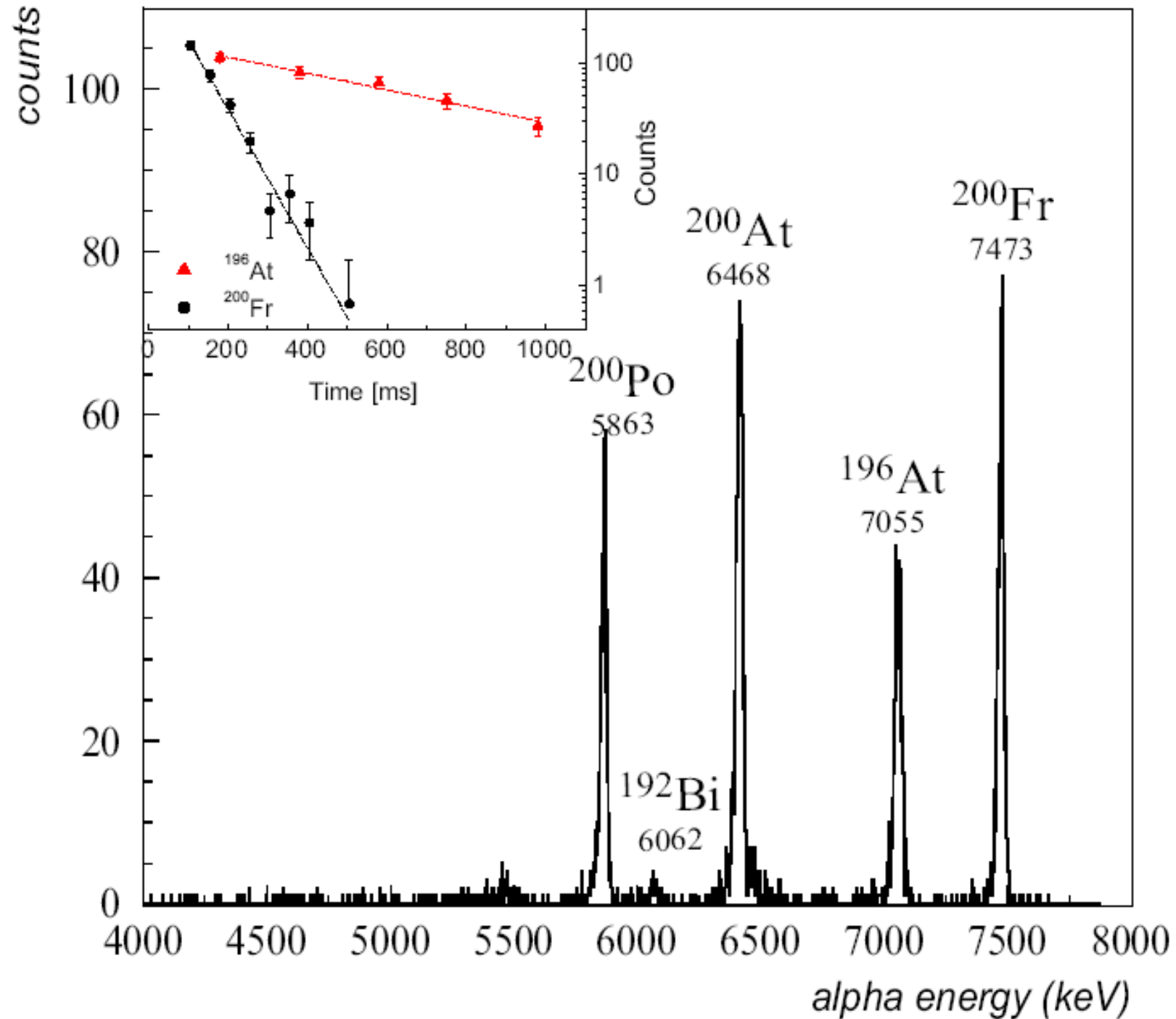
Mass separation



Mass separation

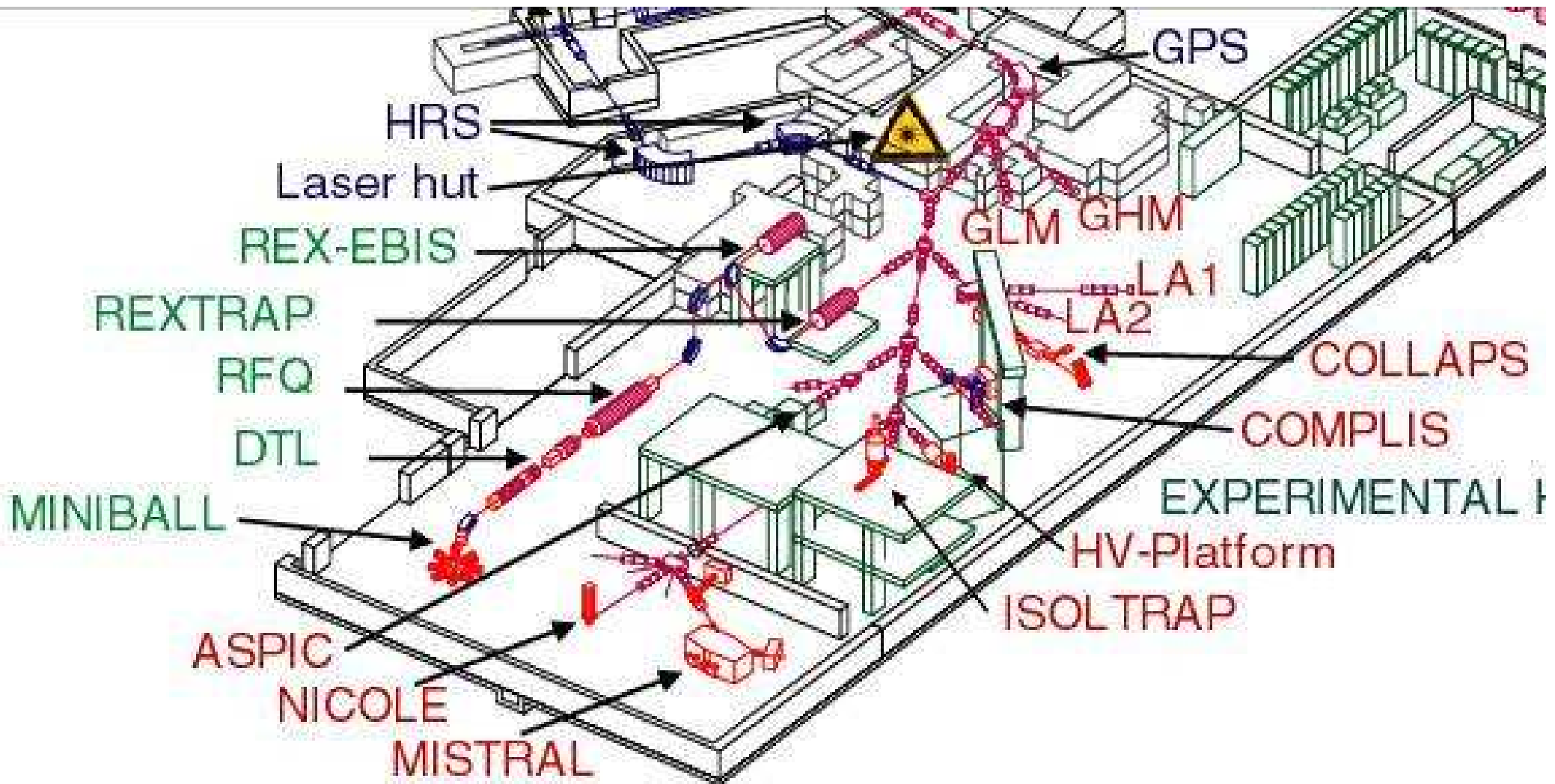


A=200 alpha energy spectrum

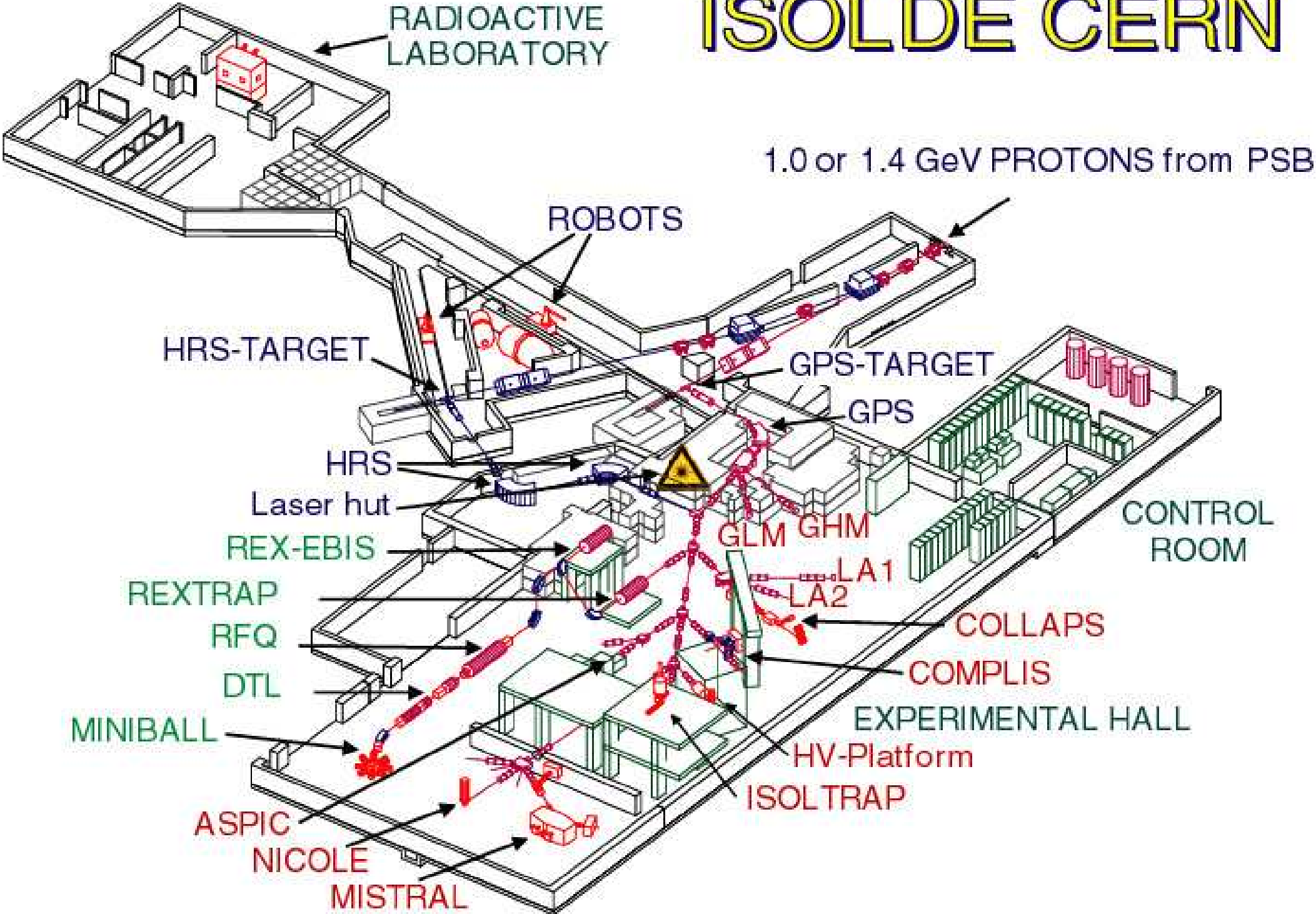


Beam transport

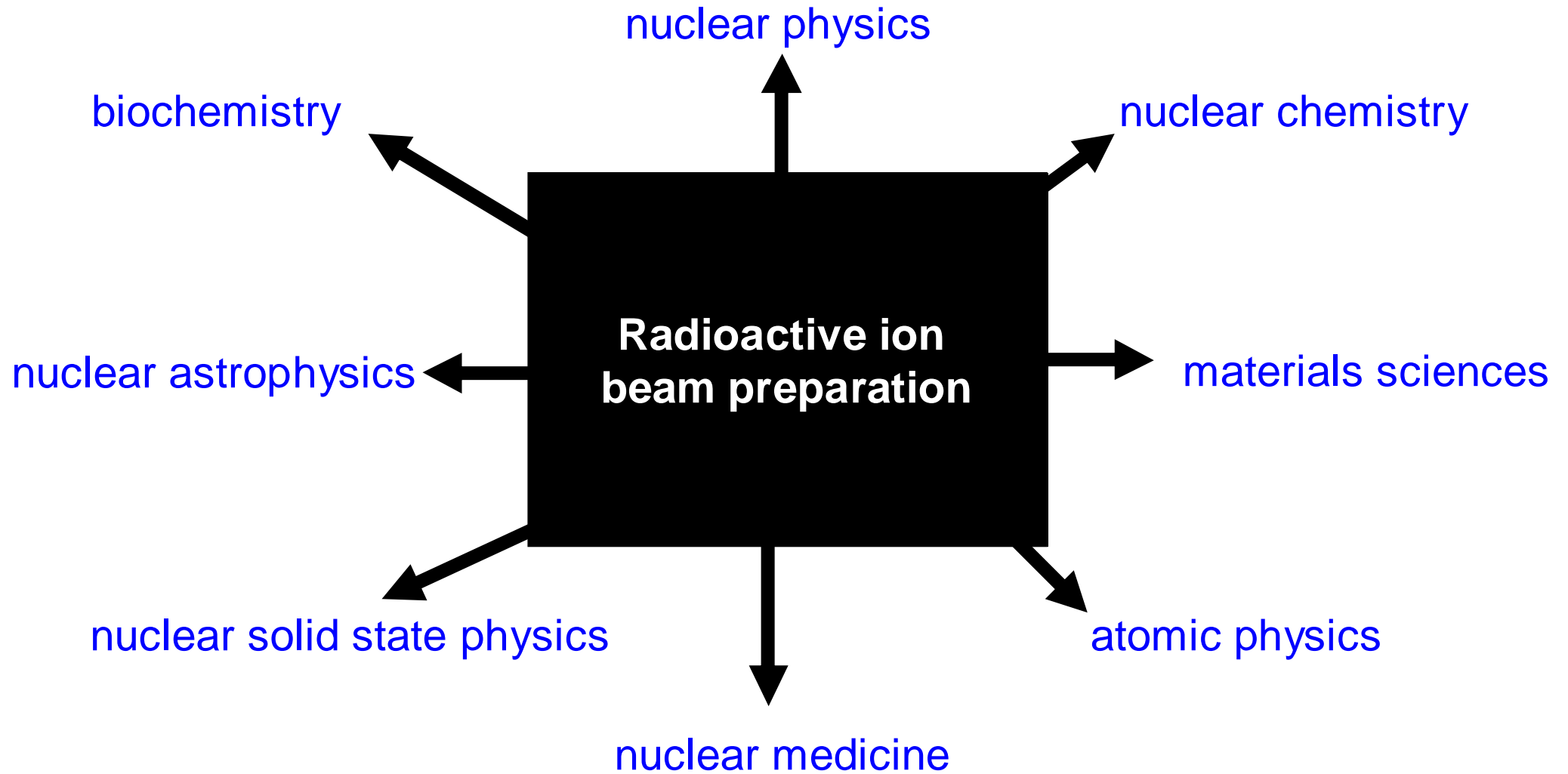
electrostatic beam transport is mass-independent ($E=60$ keV),
but has space charge limit for high beam intensities (>10 μA)
 \Rightarrow high current beams need magnetic beam transport

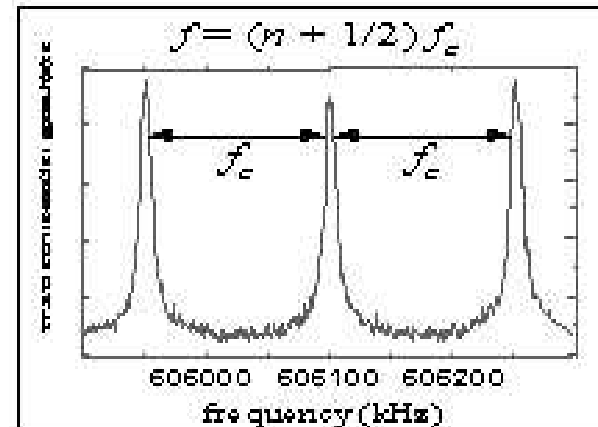
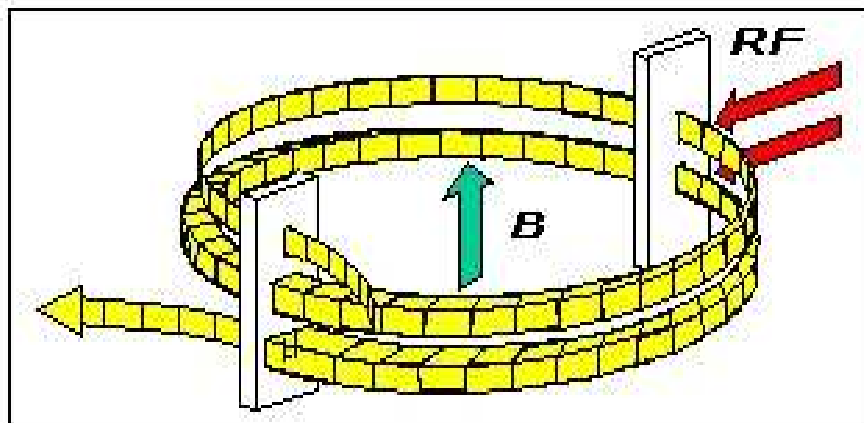
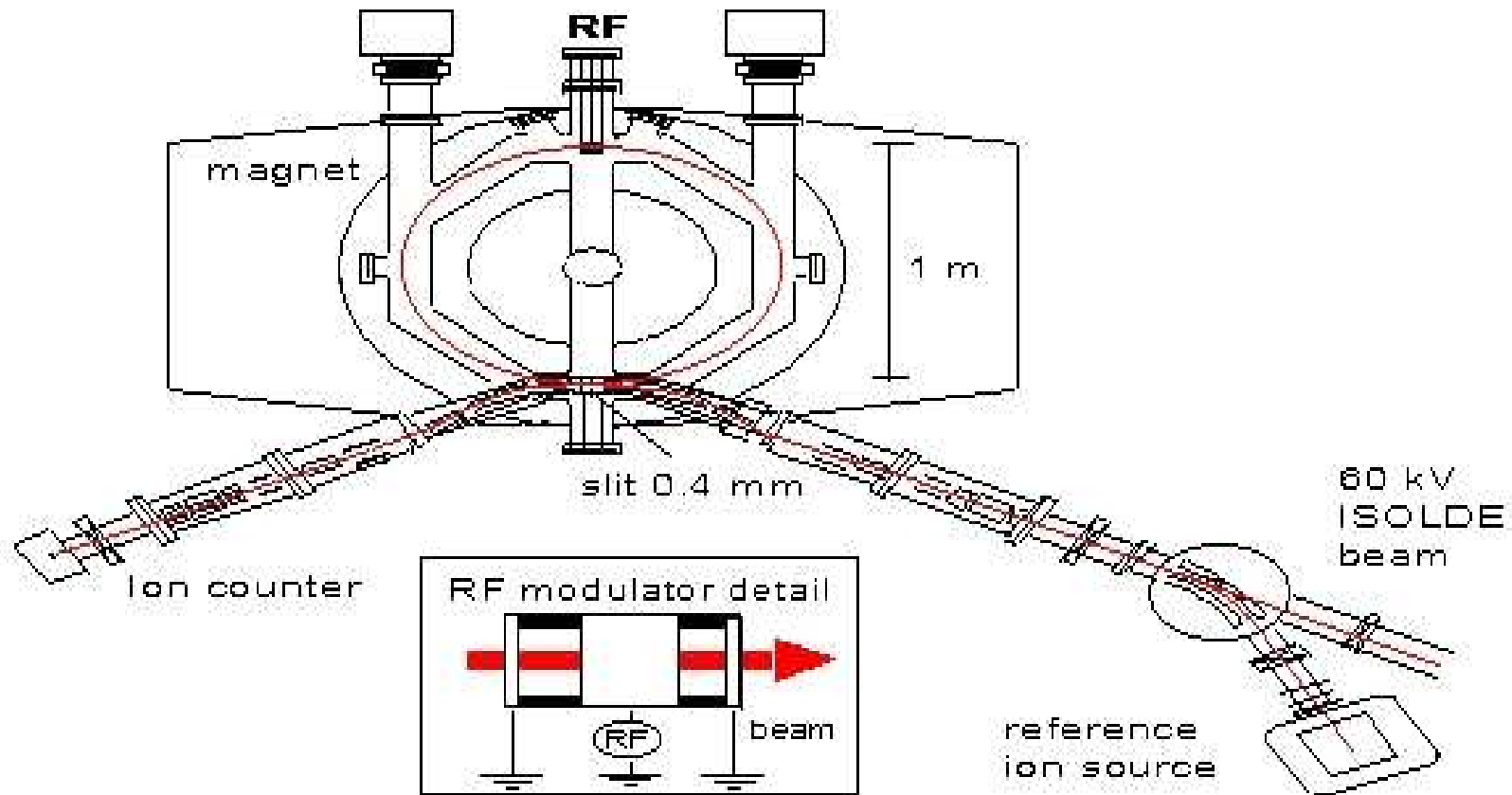


ISOLDE CERN



Applications





Mass measurement of ^{11}Li ($T_{1/2} = 9$ ms)

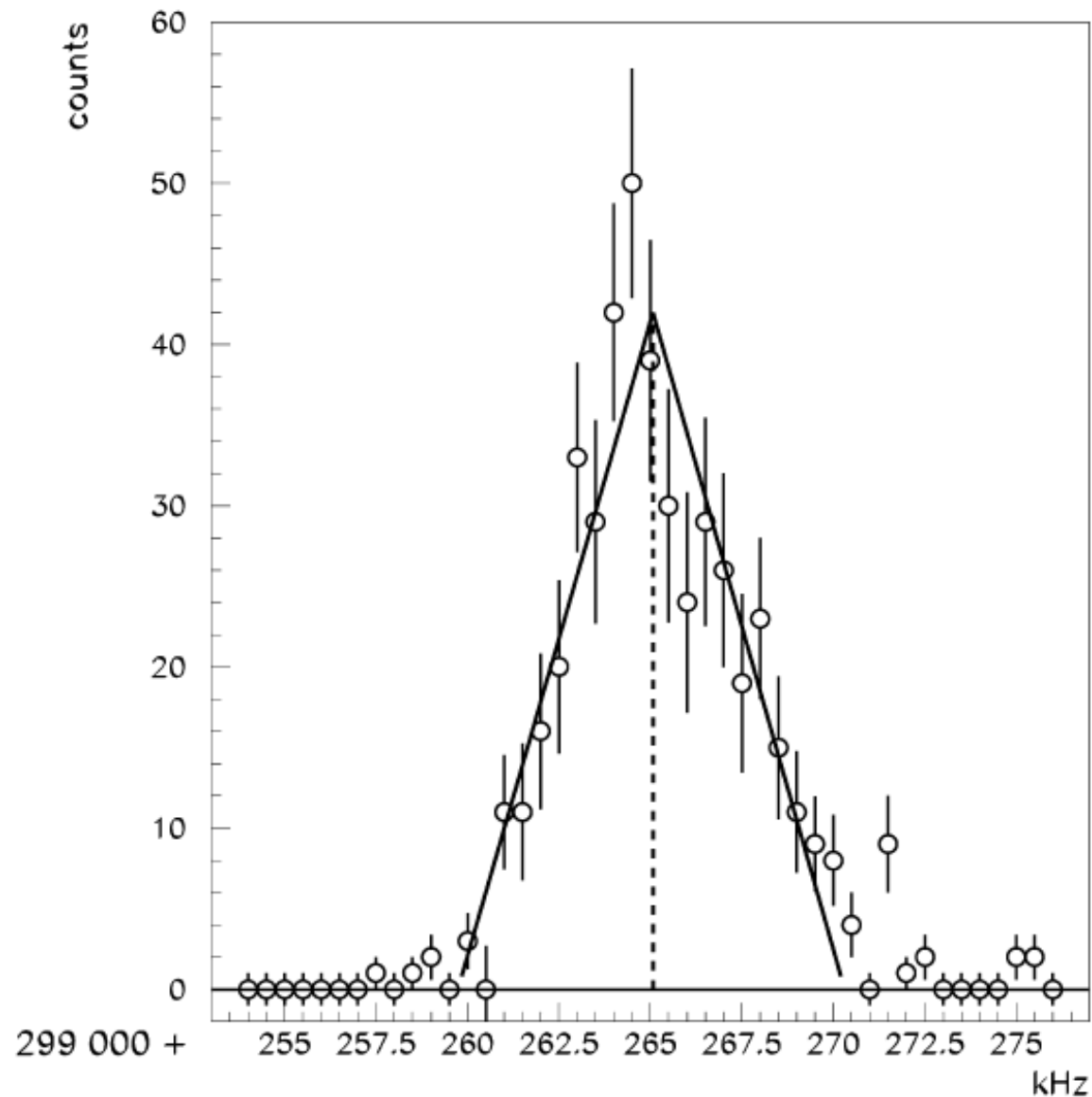
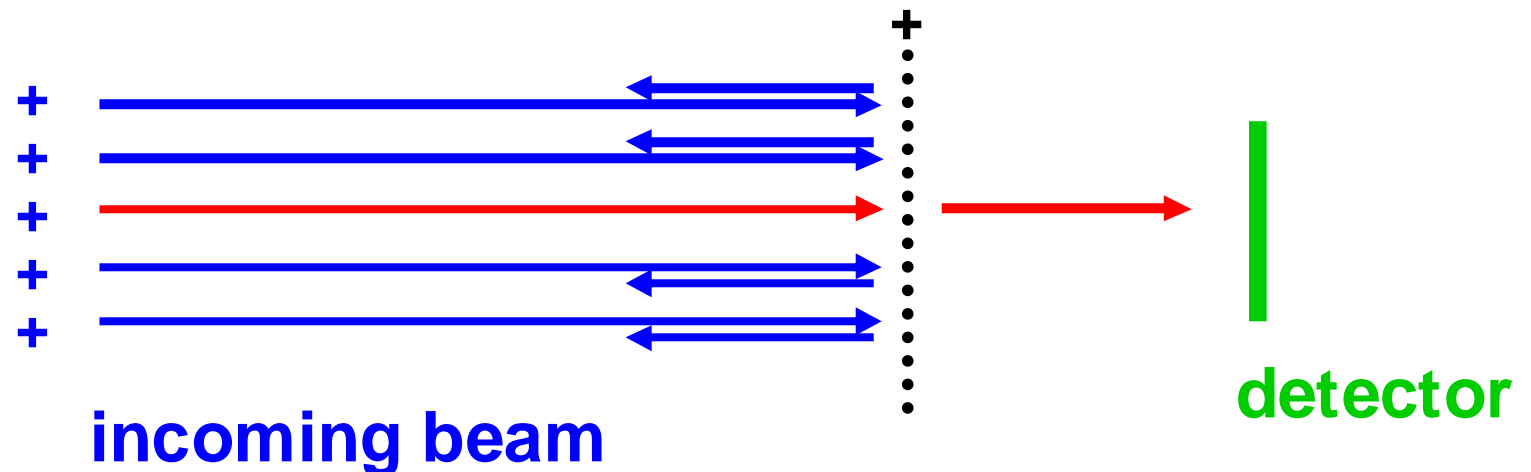


Fig. 2. Example of accumulated transmission peak for ^{11}Li . The center position (299 265 kHz) corresponds to the 917th harmonic of the cyclotron frequency. With a RF power of 100 W, a mass resolving power of $\frac{\Delta M}{M} \sim 57\,000$ was achieved.

C. Bachelet et al., Phys. Rev. Lett. 100 (2008) 182501.

Retardation spectrometer

- electrostatic energy measurement
- charged particles move against electrostatic potential; transmission measured as function of repulsive potential
- analyzes only the energy component perpendicular to the analyzer
- total energy measurement requires perfectly parallel beam



Examples of MAC-E retardation spectrometer

1. **WITCH at ISOLDE: weak interaction studies via recoil detection after EC/ β^+ decay**
2. **ASPECT at ILL: precision spectroscopy of angular correlation between neutron spin and decay protons**
3. **KATRIN in Karlsruhe: precision measurement of beta endpoint in tritium decay for neutrino mass determination**



β -decay and neutrino mass

model independent neutrino mass from β -decay kinematics

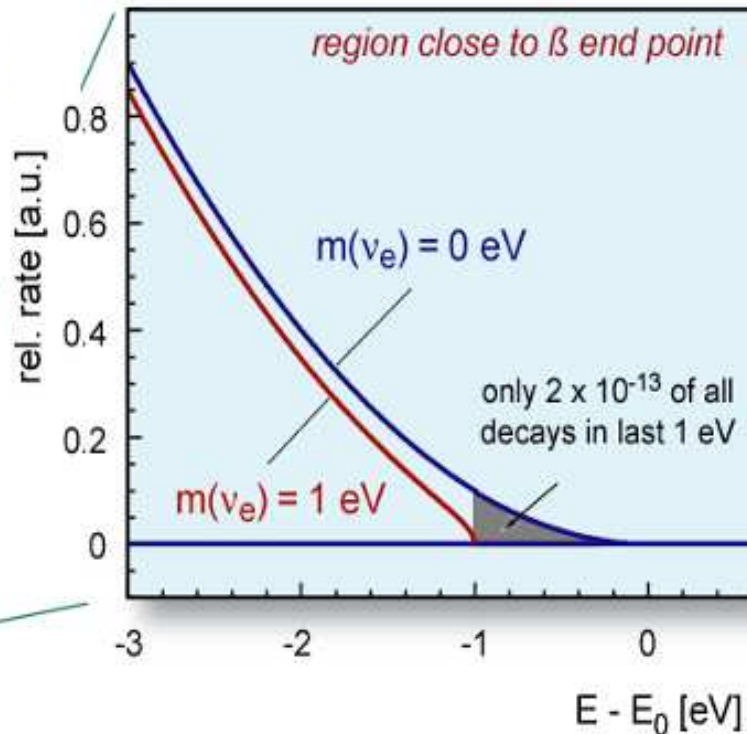
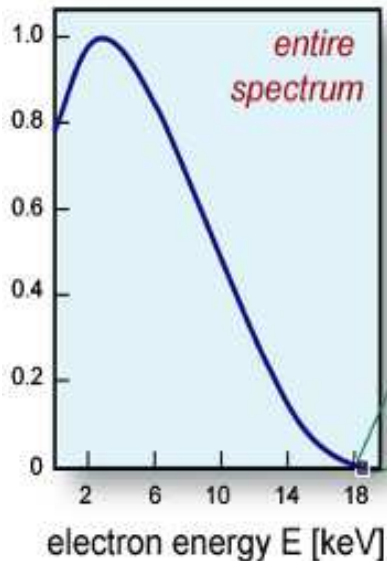
$$\frac{d\Gamma_i}{dE} = C p (E + m_e) (E_0 - E) \sqrt{(E_0 - E)^2 - m_i^2} F(E) \theta(E_0 - E - m_i)$$

$$C = G_F^2 \frac{m_e^5}{2\pi^3} \cos^2 \theta_C |M|^2$$

experimental observable is m_ν^2

$$E_0 = 18.6 \text{ keV}$$

$$T_{1/2} = 12.3 \text{ y}$$



β -source requirements :

- high β -decay rate (short $t_{1/2}$)
- low β -endpoint energy E_0
- superallowed β -transition
- few inelastic scatters of β 's

β -detection requirements :

- high resolution ($\Delta E < \text{few eV}$)
- large solid angle ($\Delta\Omega \sim 2\pi$)
- low background

Electrostatic filter with Magnetic Adiabatic Collimation

MAC-E-Technique

adiabatic guiding of β -particles along the magnetic field lines

inhomogen. B-Feld:
stray field of 2 super-
conducting magnets

$$B_{\max} = 3 - 6 \text{ T}$$

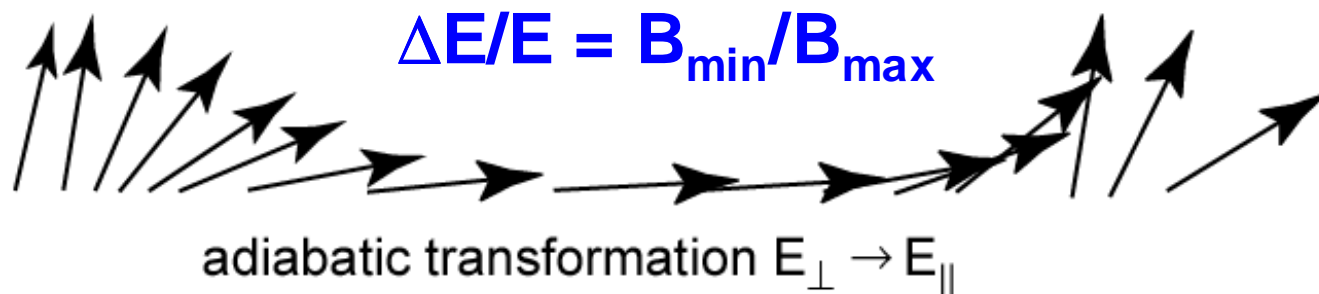
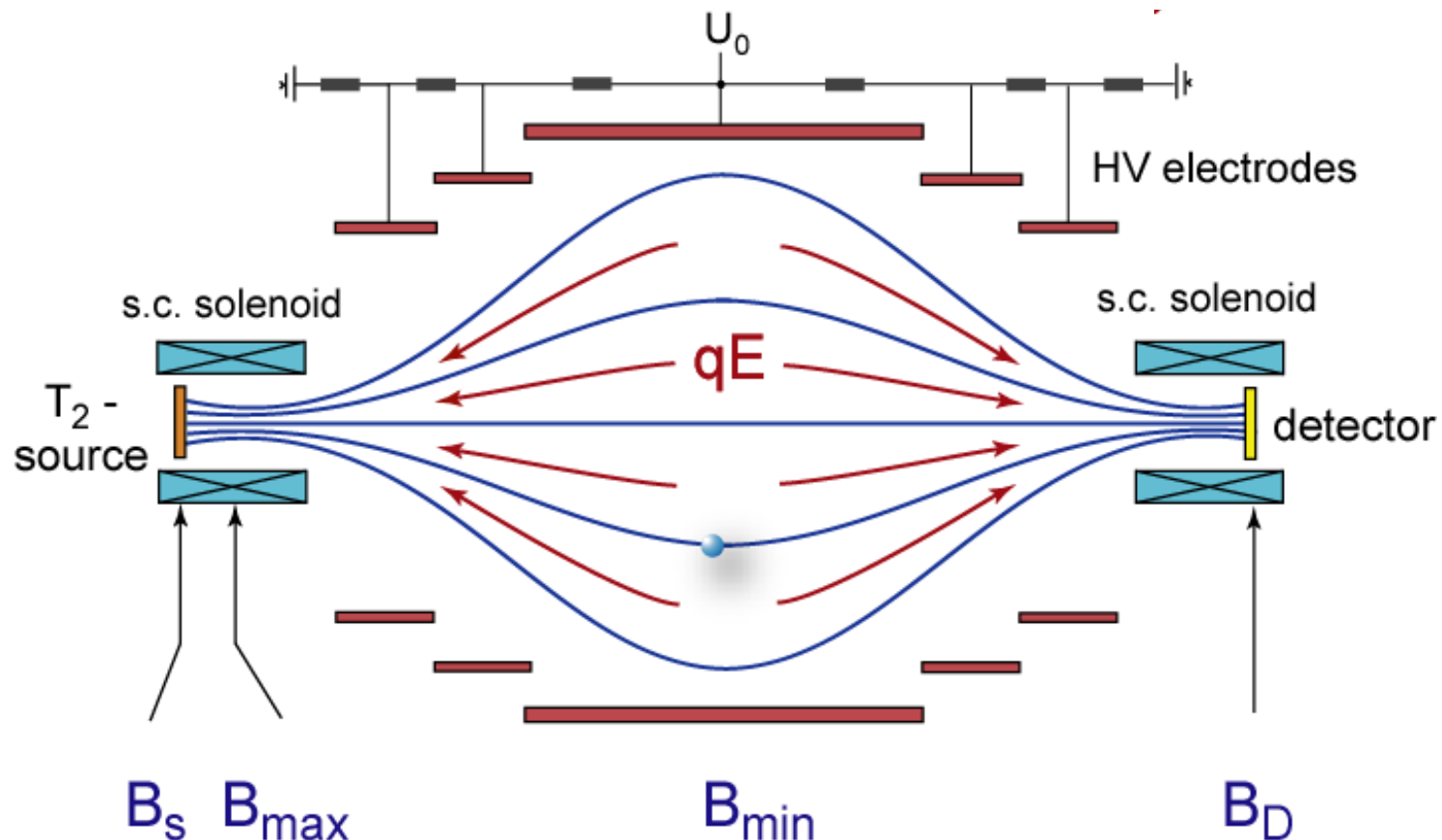
$$B_{\min} < 1 \text{ mT}$$

very large solid angle !

$$\Delta\Omega \sim 2\pi$$

$$\vec{F} = (\vec{\mu} \cdot \nabla) \vec{B} + q \vec{E}$$

$$\mu = E_{\perp} / B = \text{const}$$

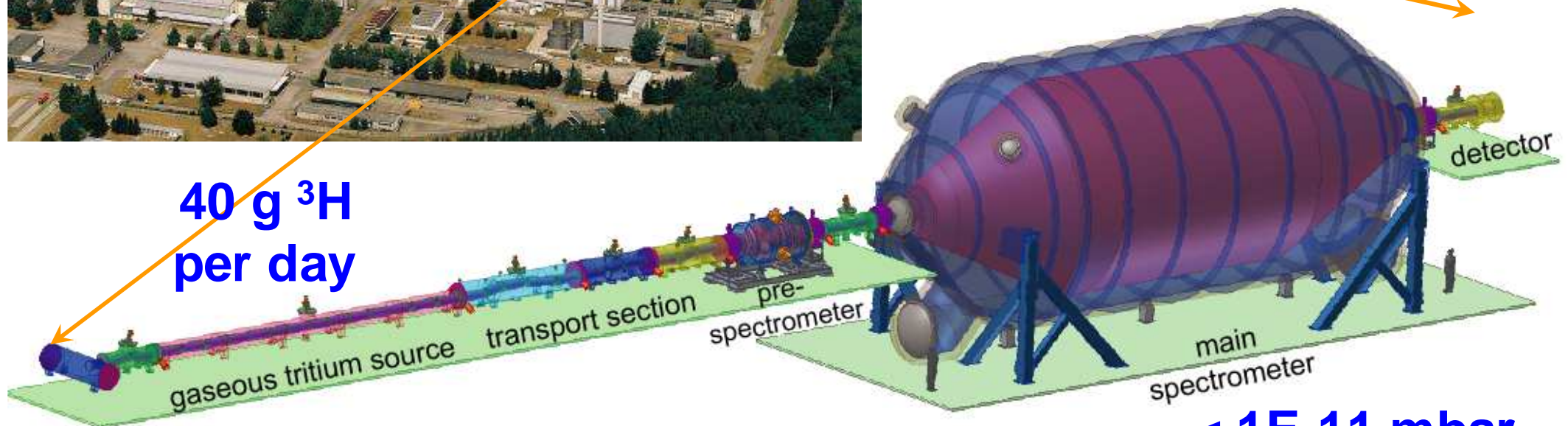


KATRIN experiment



Karlsruhe Tritium Neutrino Experiment

at **Forschungszentrum Karlsruhe**
unique facility for closed T₂ cycle:
Tritium Laboratory Karlsruhe



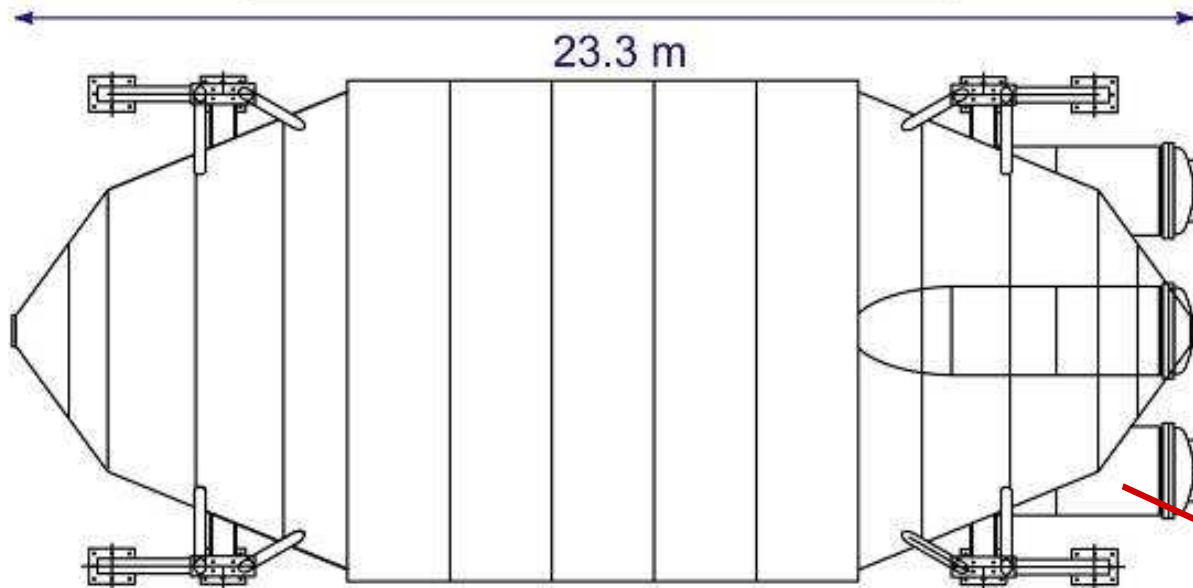
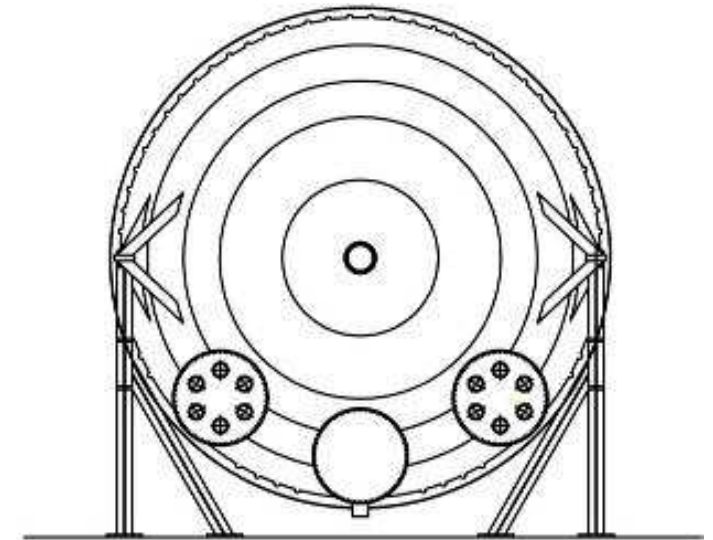
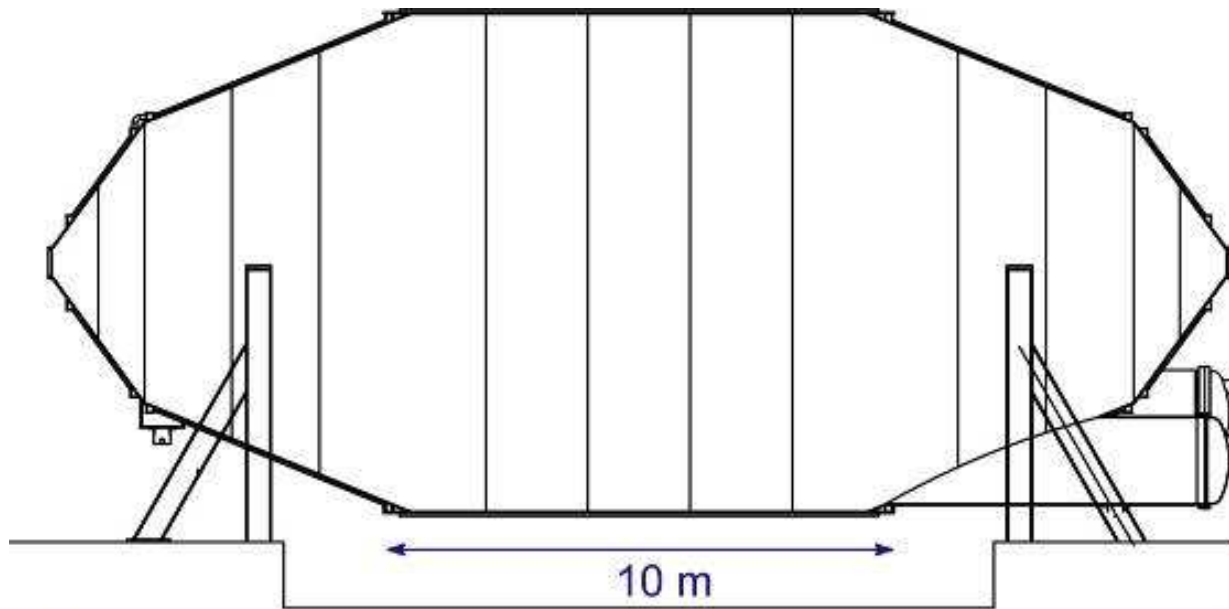
**40 g ³H
per day**

~ 75 m linear setup with 40 s.c. solenoids

< 1E-11 mbar

< 1E-20 mbar ³H

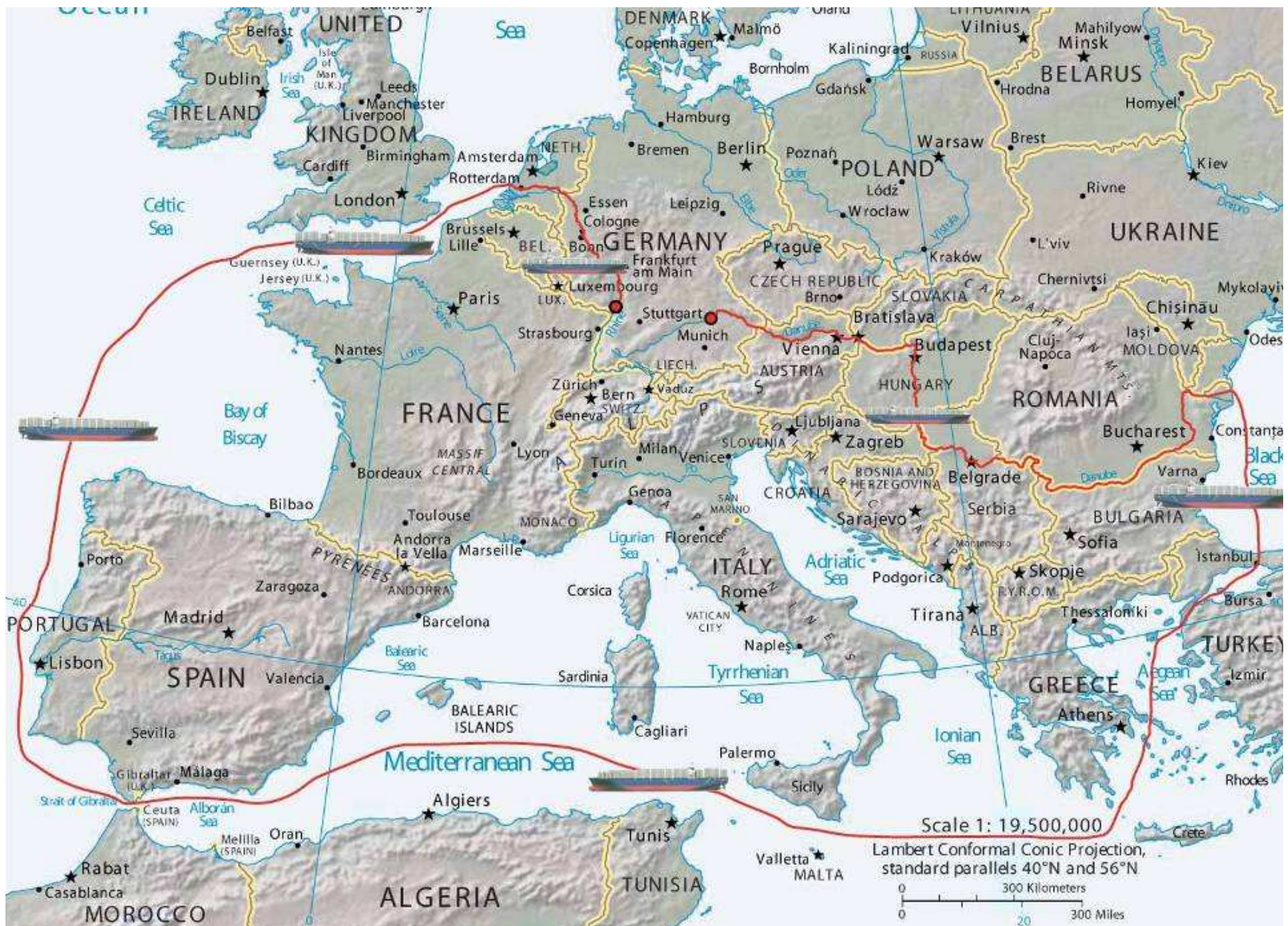
main spectrometer – design



design parameters:

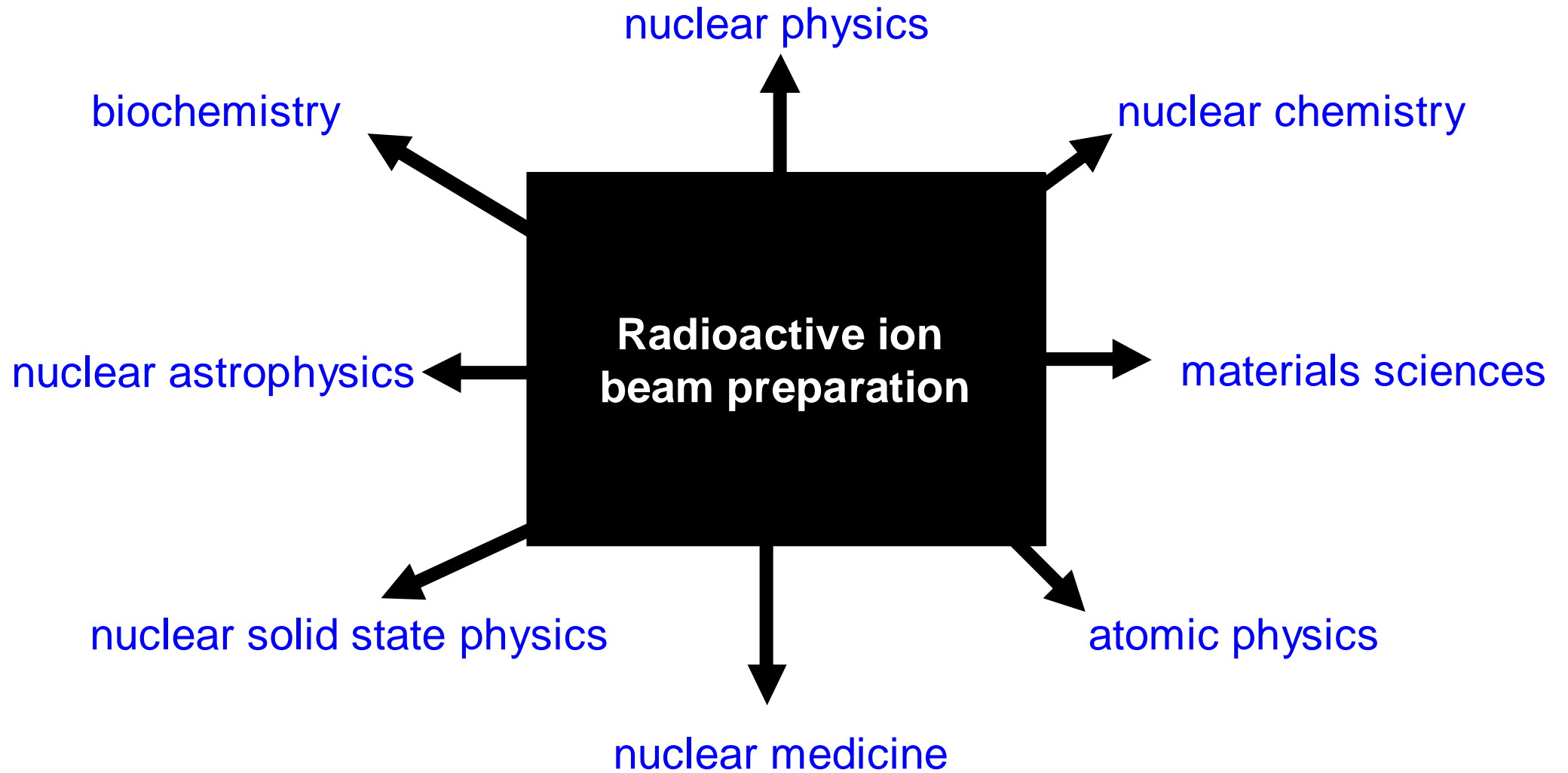
volume:	1258 m ³
surface:	605 m ²
thickness:	32 mm
material weight:	1.4429
	192 t

pumping port
for getters





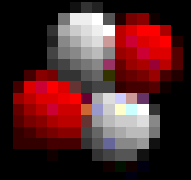
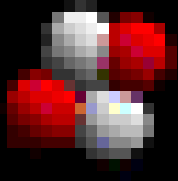
Applications



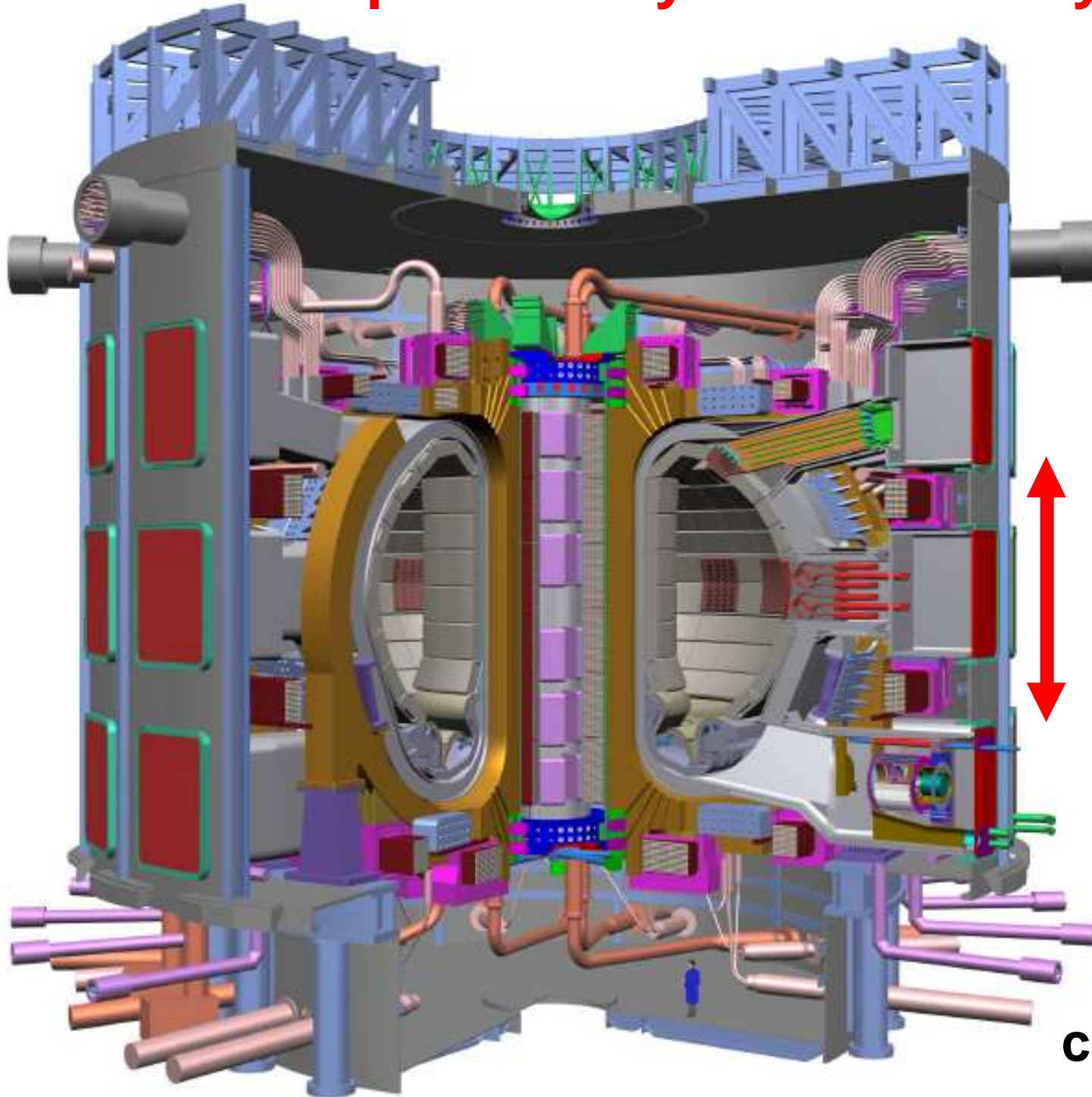
Production of ^{12}C in stars

		C 8 2E-21 s	C 9 127 ms	C 10 19.3 s	C 11 20 m	C 12	C 13	C 14 5.7 ka
		B 7 4E-24 s	B 8 770 ms	B 9 8E-19 s	B 10	B 11	B 12 20 ms	B 13 17 ms
		Be 6 5E-21 s	Be 7 53.3 d	Be 8 7E-17 s	Be 9	Be 10 1.5 Ma	Be 11 13.8 s	Be 12 21 ms
		Li 5 4E-22 s	Li 6	Li 7	Li 8 840 ms	Li 9 178 ms	Li 10 2E-21 s	Li 11 8.5 ms
	He 3	He 4	He 5 7E-22 s	He 6 807 ms	He 7 3E-21 s	He 8 119 ms	He 9 7E-21 s	He 10 3E-21 s
H 1	H 2	H 3 12.3 a						

The triple-alpha process: rate



Setup for study of three-body-fusion?



Exercise:

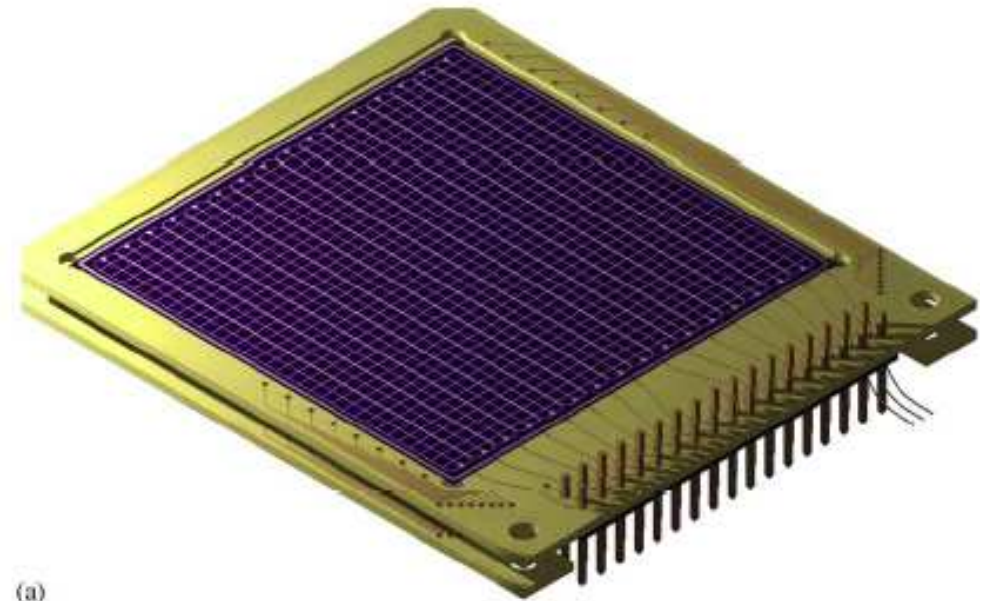
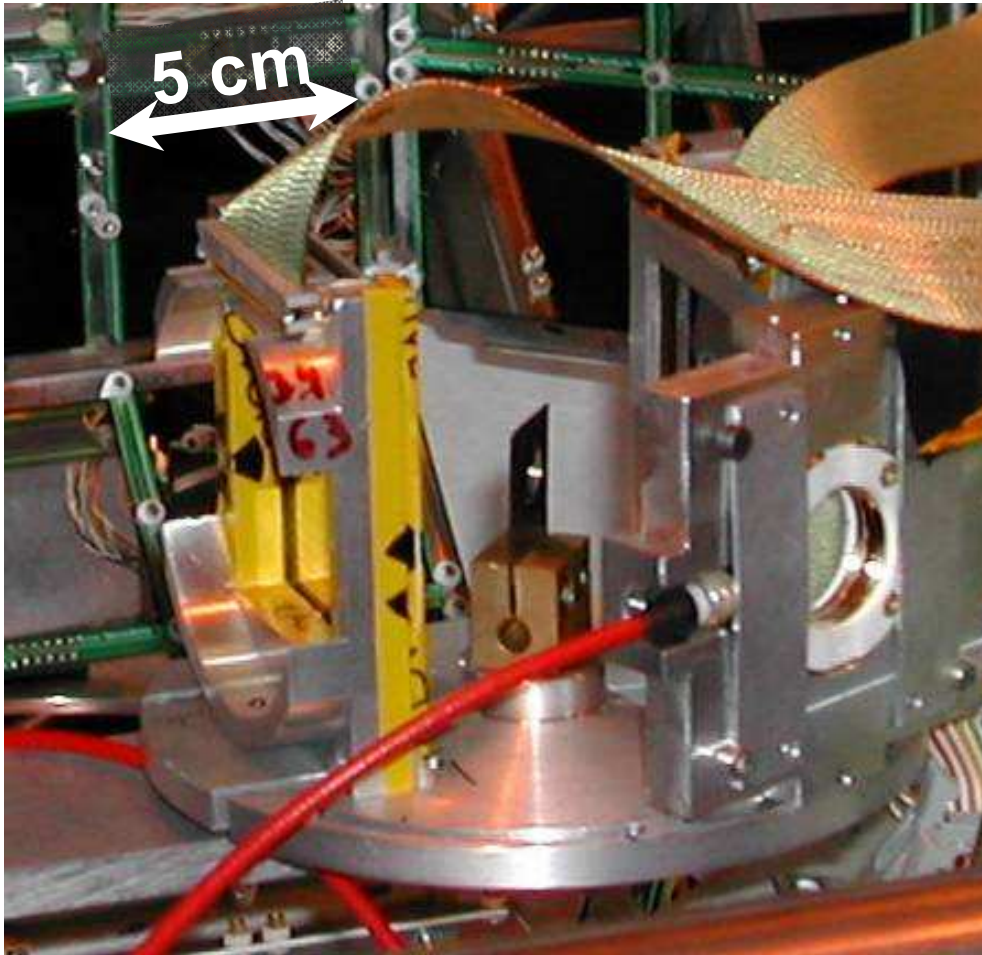
scale from two-body fusion $d+t$ to three-body fusion ${}^4\text{He}+{}^4\text{He}+{}^4\text{He}!$

14 m

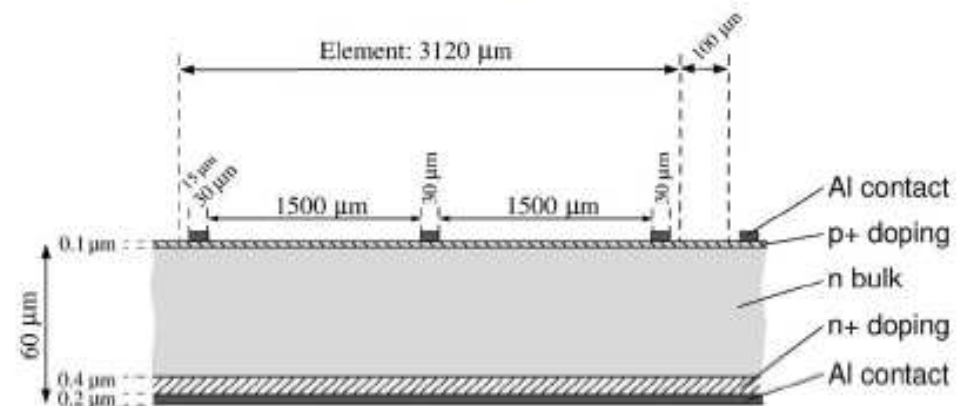
cost ca. 10^{10} EUR

Setup for study of triple alpha reaction!

New detector design



(a)



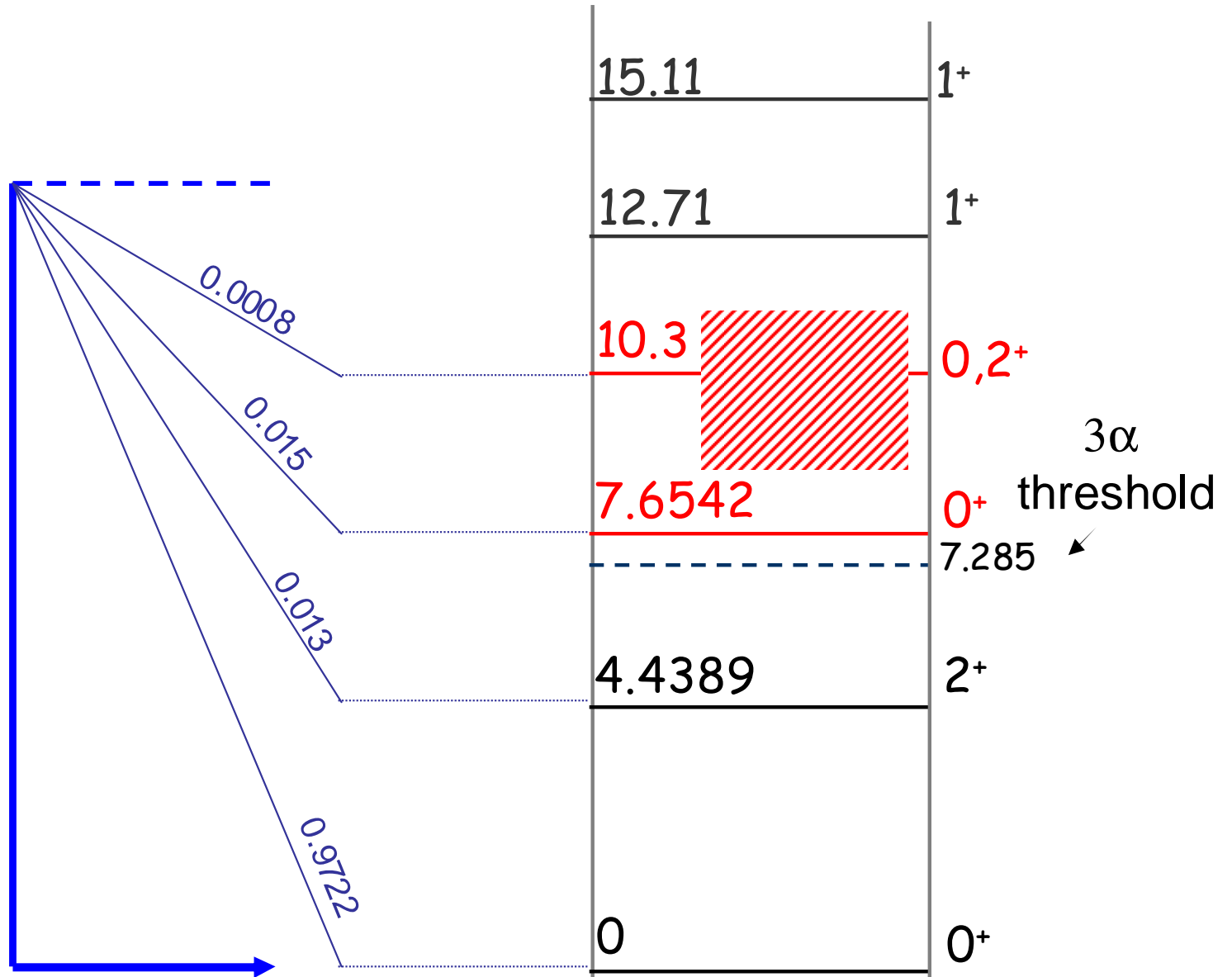
(b)

Reduced deadlayer

Inverse reaction: $^{12}\text{B}(\beta, 3\alpha)$ decay

^{12}B 1^+

B 12
20.2 ms



$^{12}\text{C}^*$ from the beta-decays of ^{12}N and ^{12}B

β -delayed α -spectrum measured:

• 1950 by Alvarez



^{12}N

• 1957 by Fowler *et al.*



^{12}B

• 1963 by Wilkinson *et al.*



$^{12}\text{B}/^{12}\text{N}$

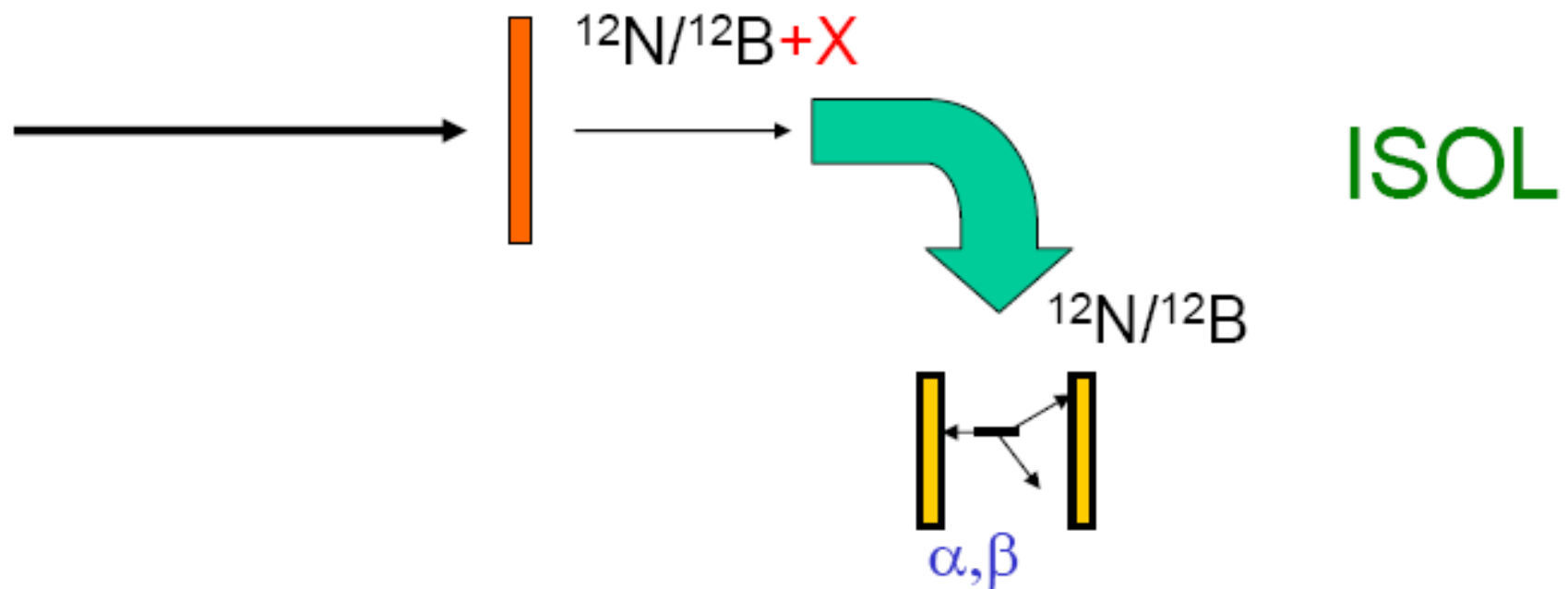
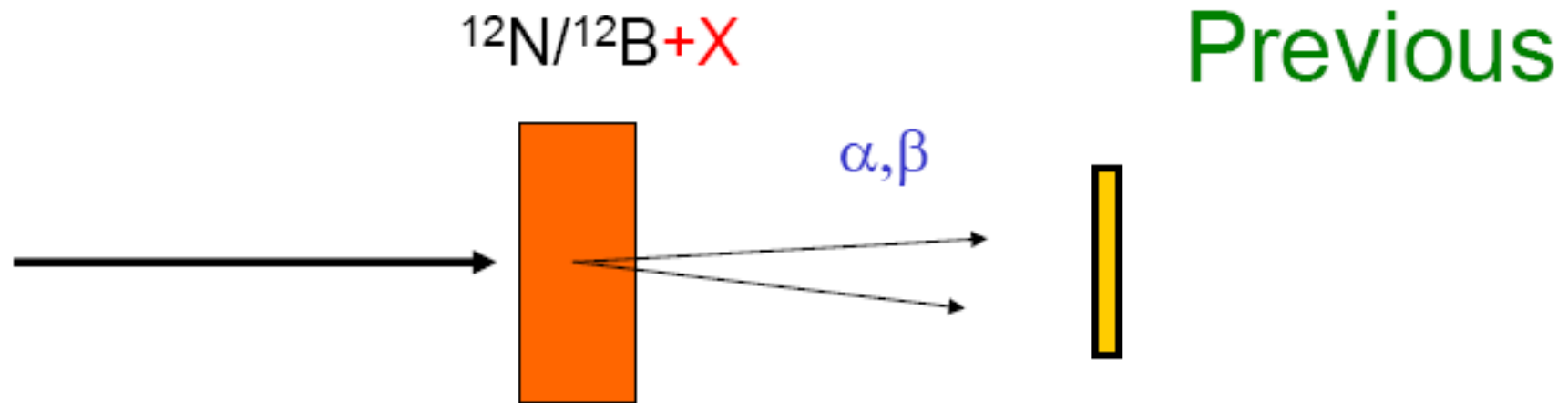
• 1966 by Schwalm and Povh

$^{12}\text{B}/^{12}\text{N}$

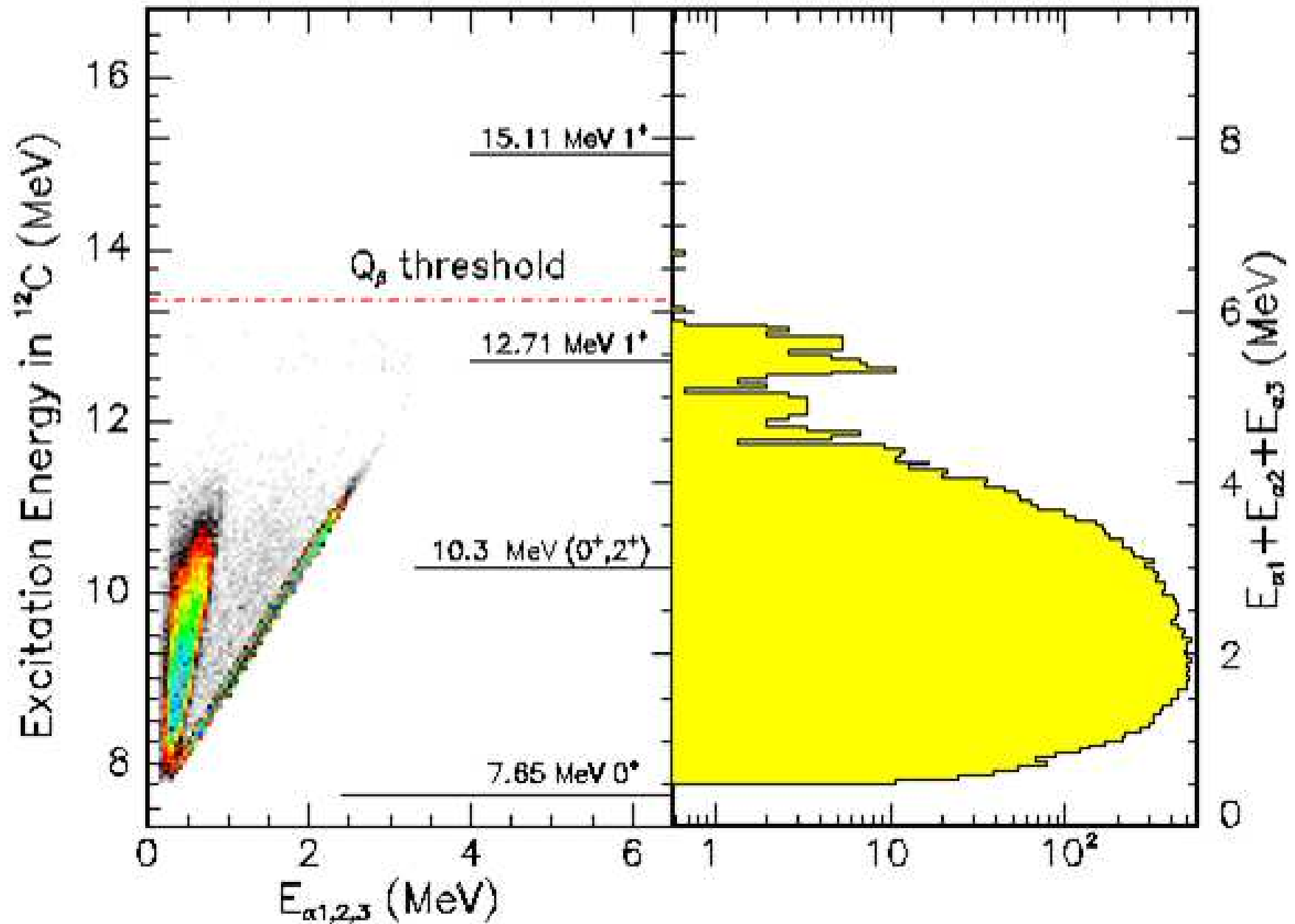
• 1978 by Schwalm and Gergely
(unpublished)

^{12}N

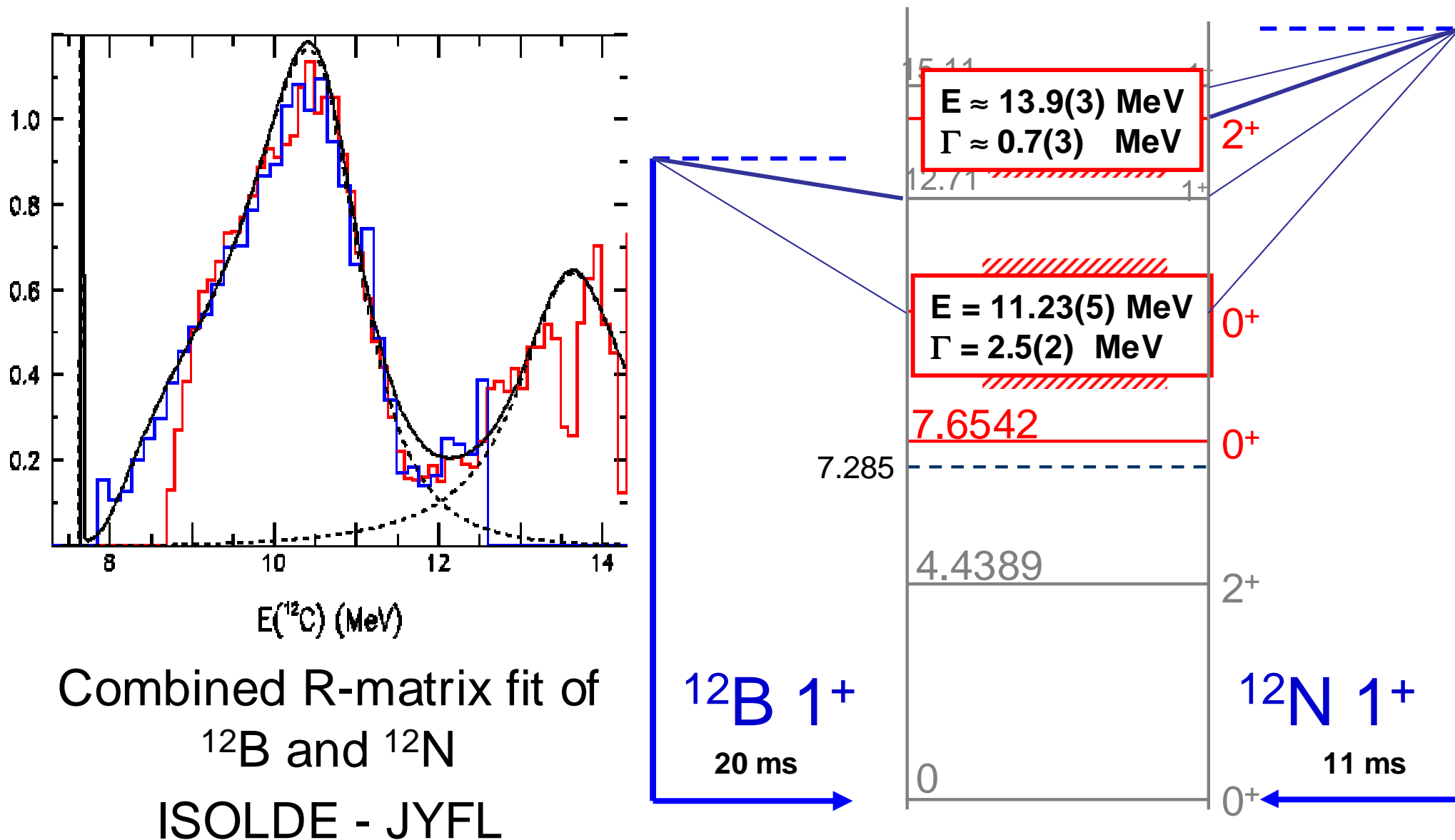
How to measure beta-delayed particle emission?



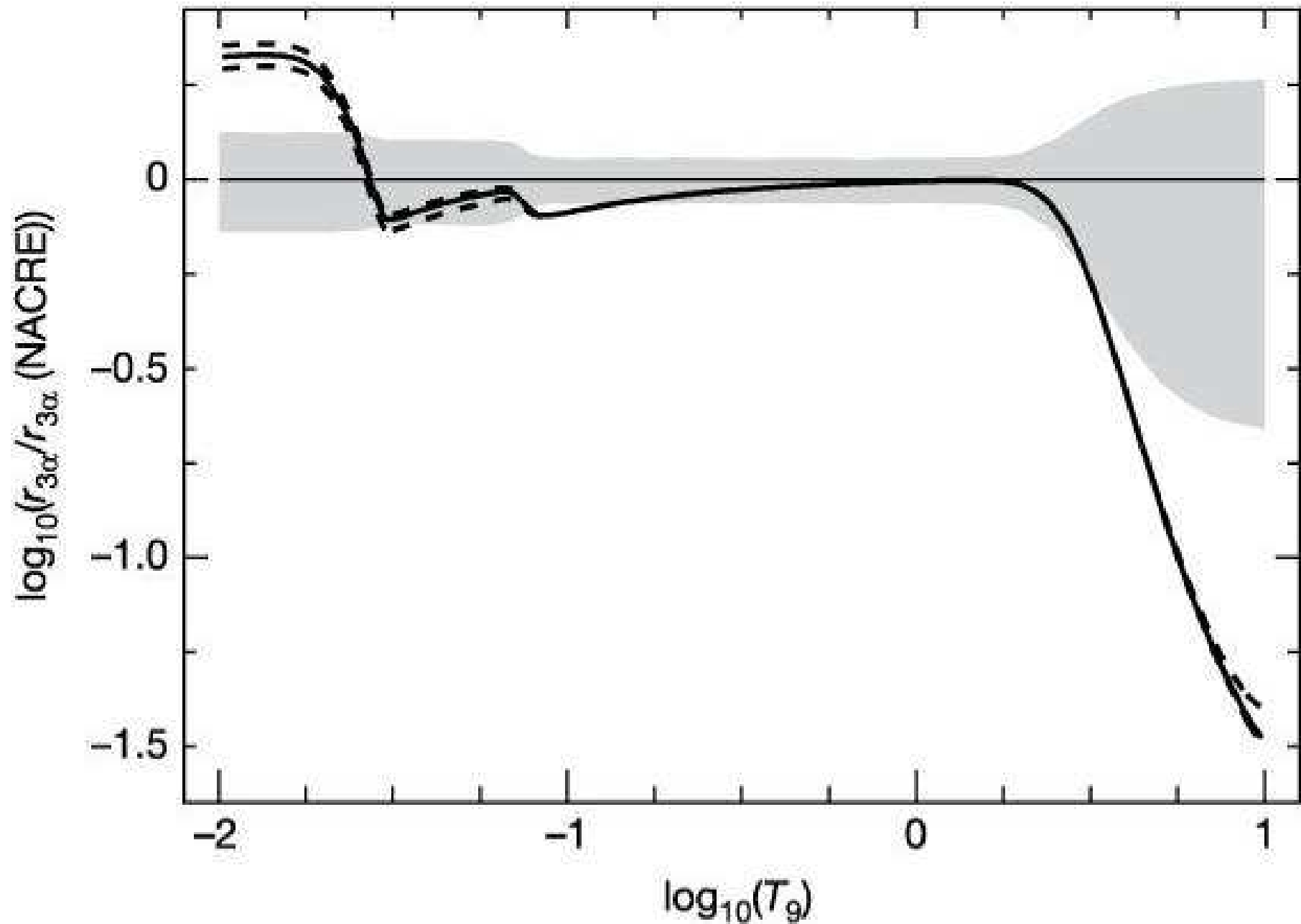
$^{12}\text{Be}(\beta^-)^{12}\text{B}$ beta decay to $^{12}\text{C}^* \rightarrow 2\alpha$ detected



The triple-alpha process: ^{12}B and ^{12}N decays

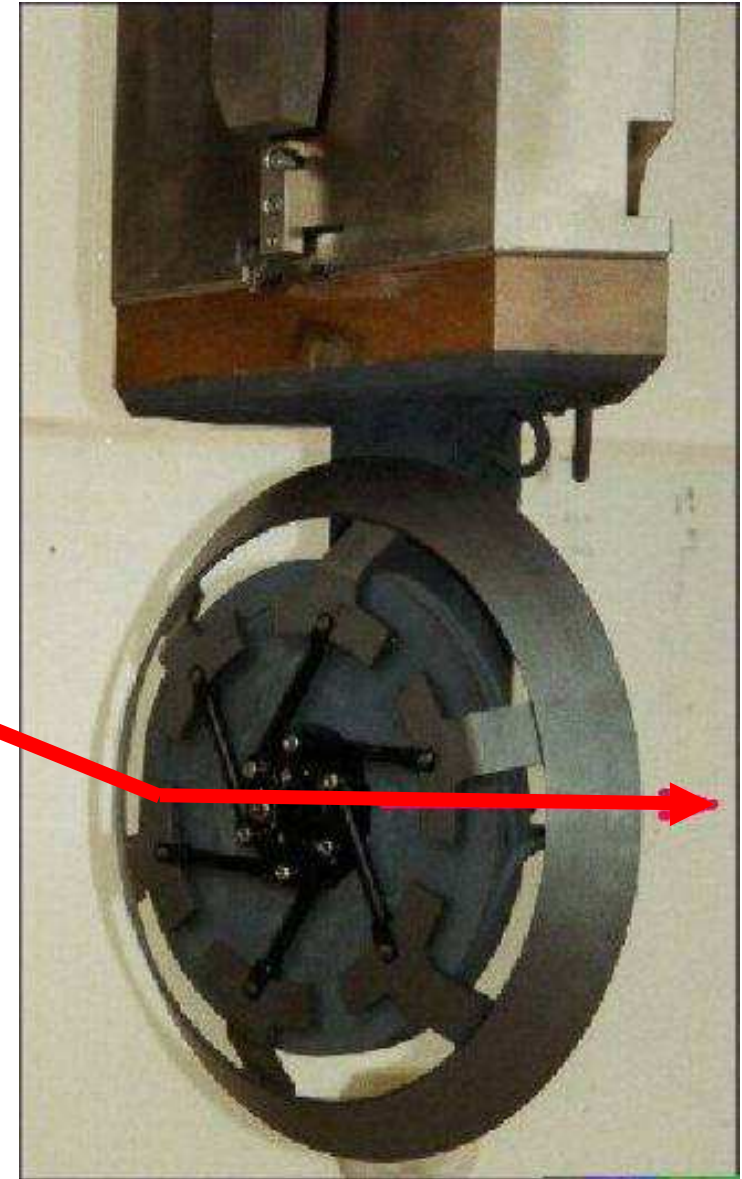
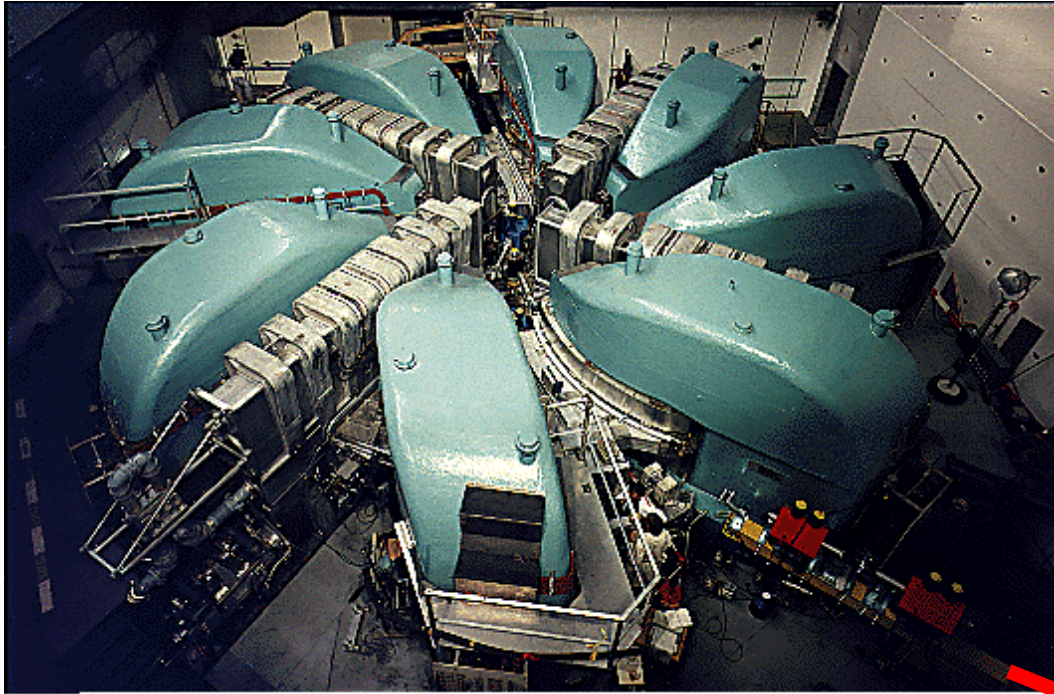


New rates for the triple-alpha process



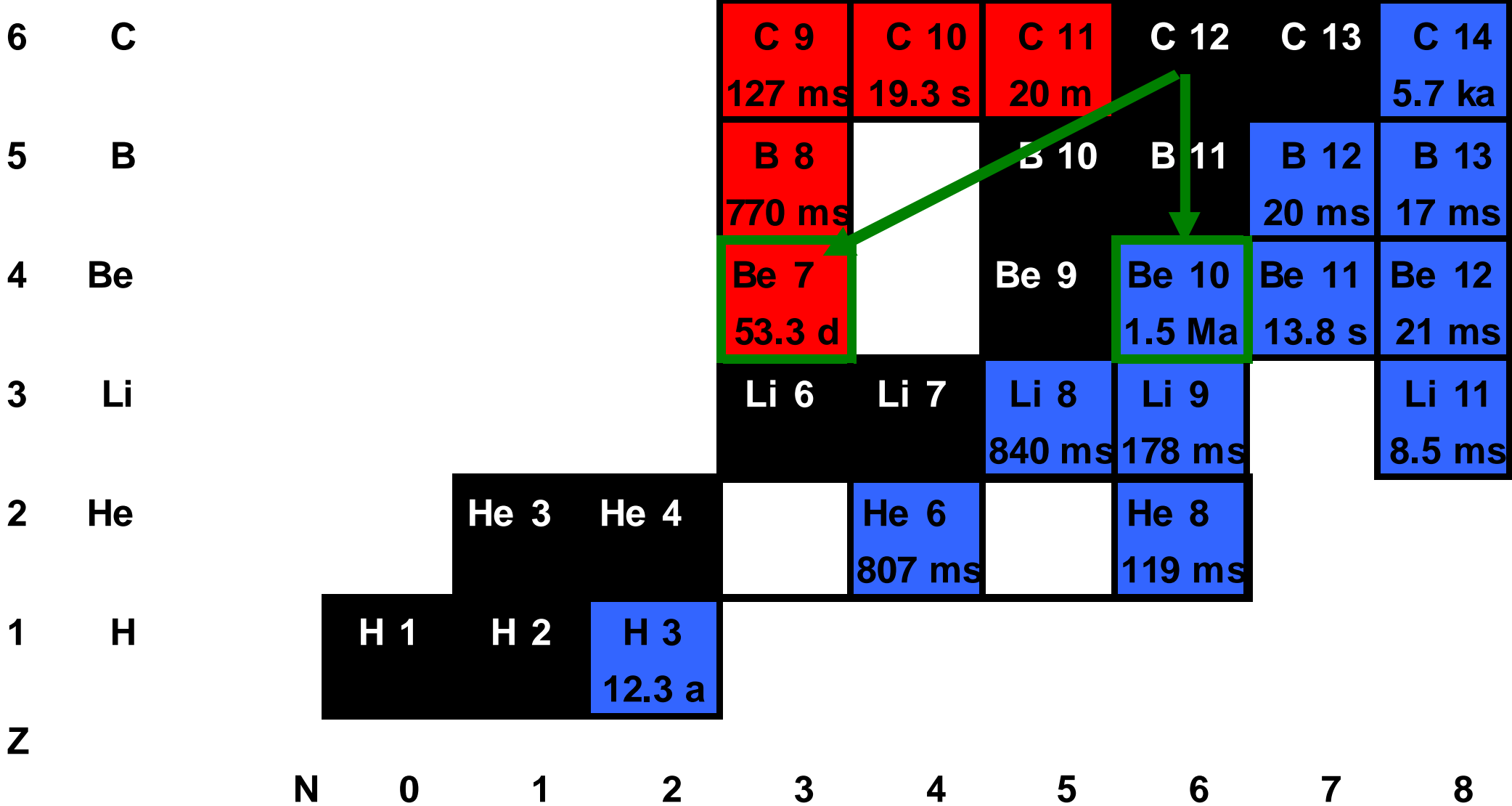
H.O.U. Fynbo et al., Nature 433 (2005) 136.

$7,10\text{Be}$ from Paul Scherrer Institute

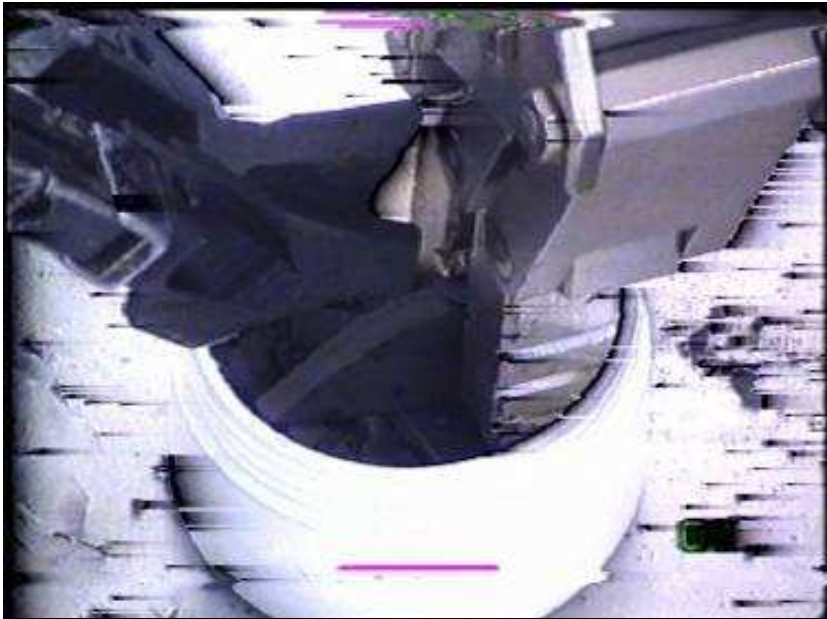


PSI: 2 mA 590 MeV protons onto graphite target for pion production

Spallation products



Procedure for off-line produced ^7Be beams



1. Break graphite into pieces
2. Fill ISOLDE target container
3. Put into Pb-shielded container
4. Transport to ISOLDE
5. Heat target container to $1700\text{ }^{\circ}\text{C}$
6. Ionize Be with RILIS

RIB with “macroscopic” intensity!

HRS.MAG90 Mass Control

Mass Control Program

High Voltage: 60.0472E+03

Calibration Mass: 7.01693

Set Mass: 7Be

Magnetic Field: 0.0938071

Mass Factor: 7.974788E+02

Set Field: 0.0938071

Status: PS On

Mass: 7.01693(0.00000)

HRS.MAG60 Mass Control

Mass Control Program

High Voltage: 60.0472E+03

Calibration Mass: 7.01693

Set Mass: 7Be

Magnetic Field: 0.0935791

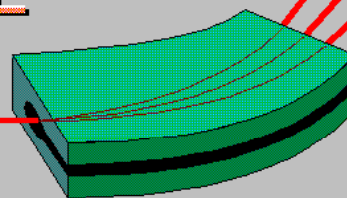
Mass Factor: 8.013679E+02

Set Field: 0.0935791

Status: PS On

Mass: 7.01692(0.00000)

ISOLDE
CERN



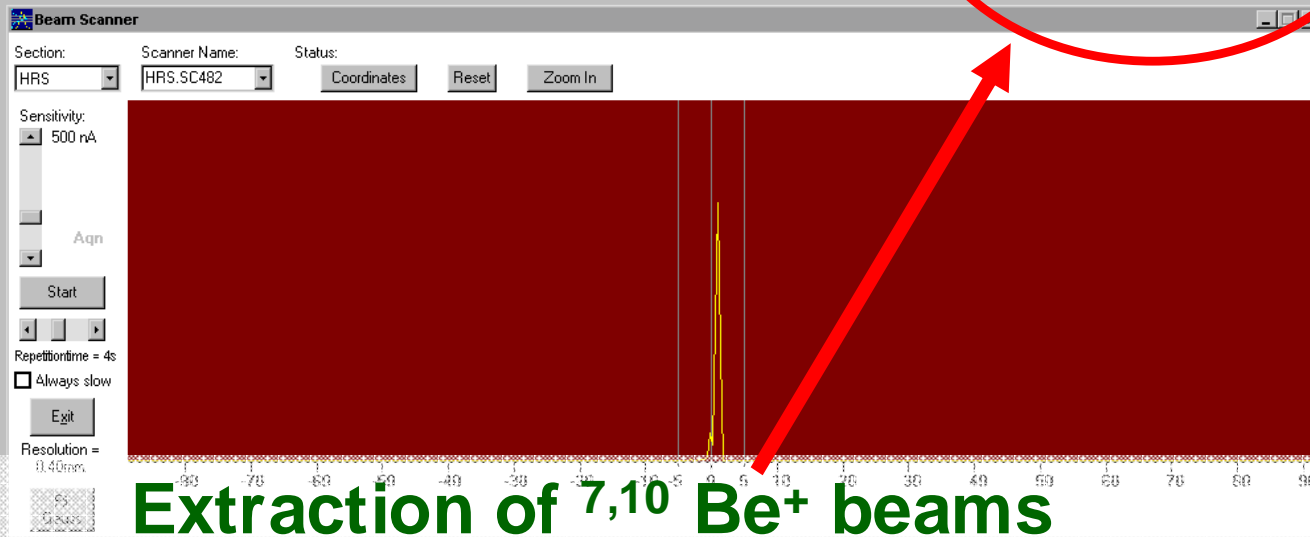
Isotope Separator On Line

HRS.FC490

Position: In, Out, Exit

Measure: Read electrometer, Refresh

Measure: 3.3534e-7 A

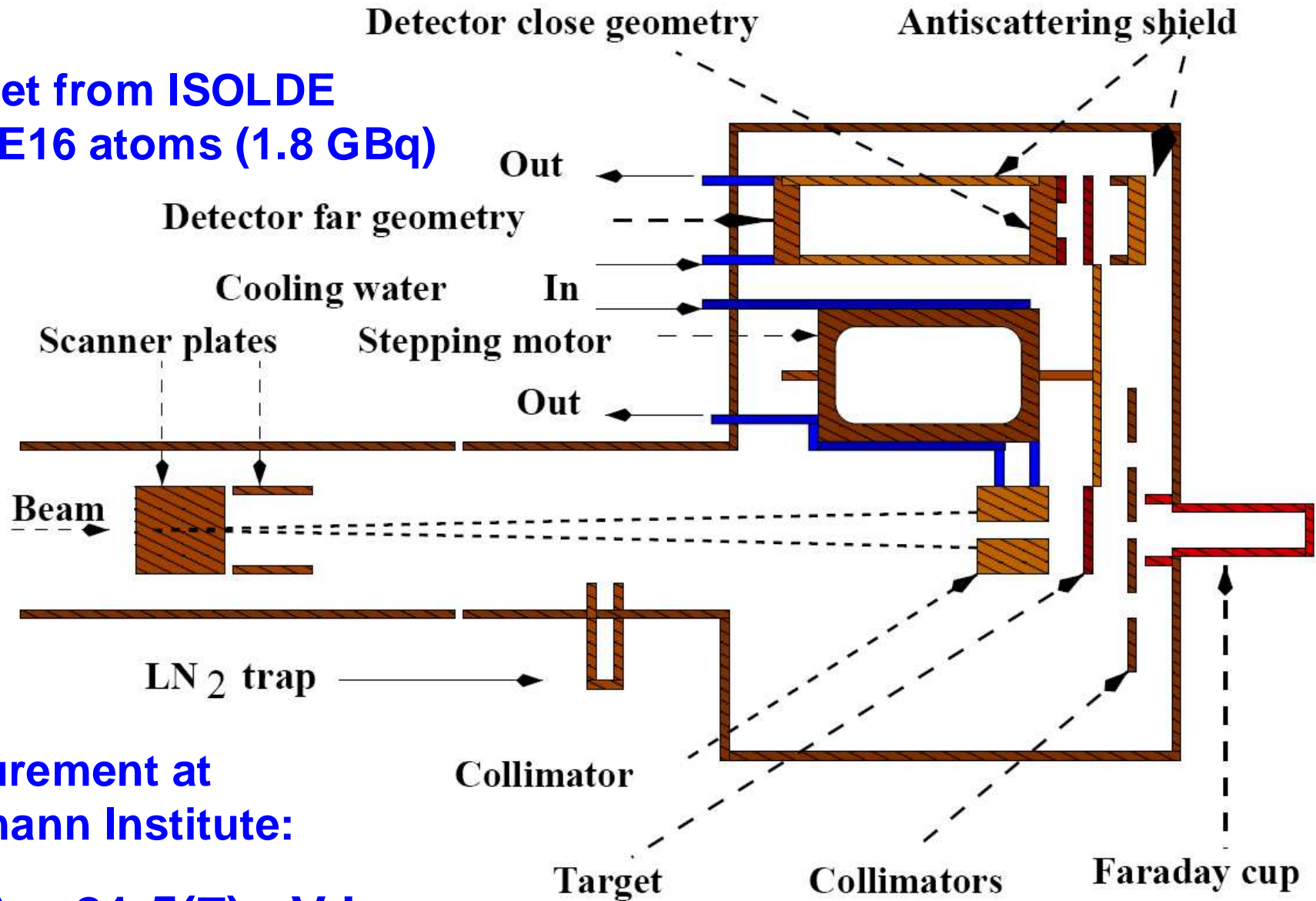


with 300 pA ($2E12$ ions per s) for many hours!

U.K. et al., Nucl. Instr. Meth. B204 (2003) 343.

Measurement of ${}^7\text{Be}(p,\gamma)$ with ion-implanted target

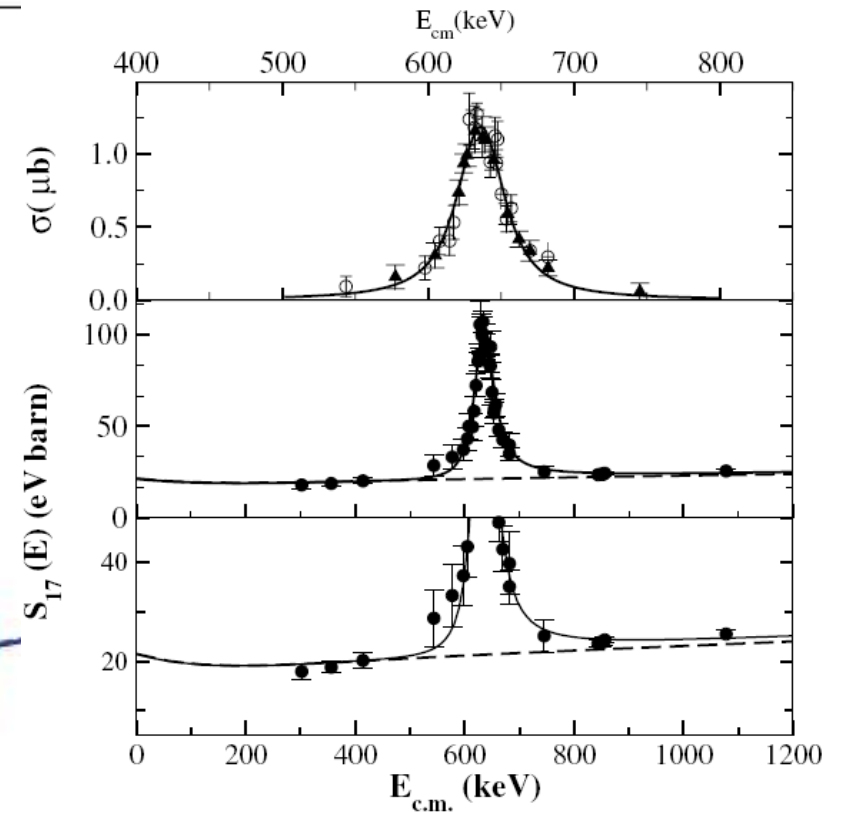
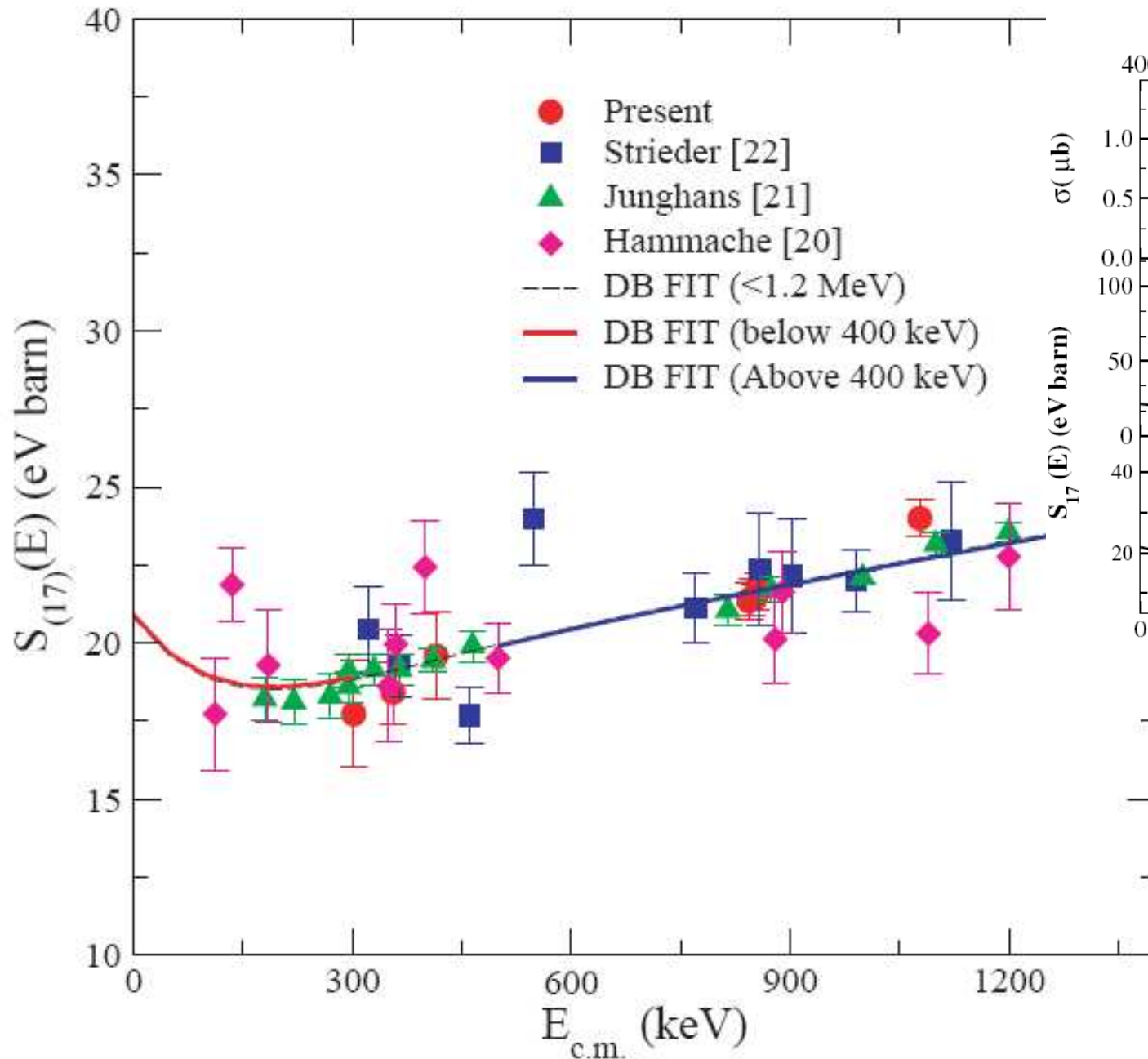
${}^7\text{Be}$ target from ISOLDE
with $1.2\text{E}16$ atoms (1.8 GBq)



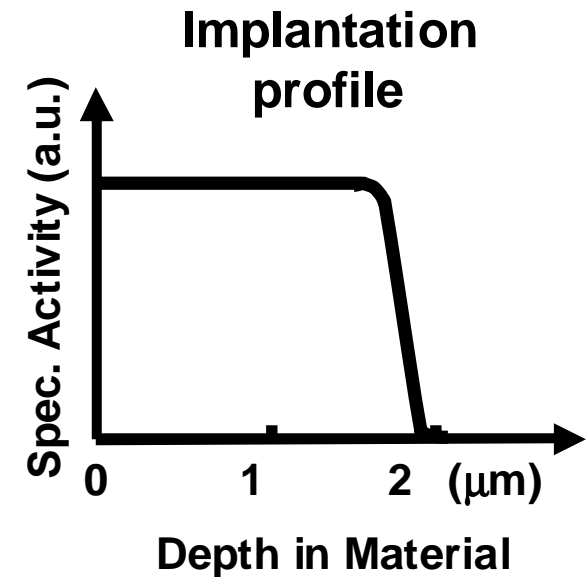
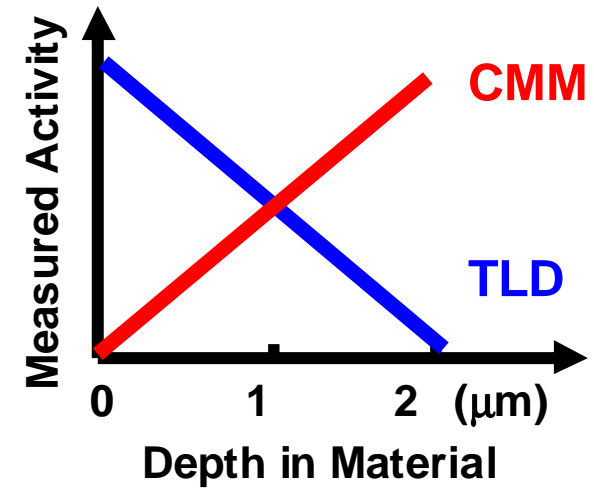
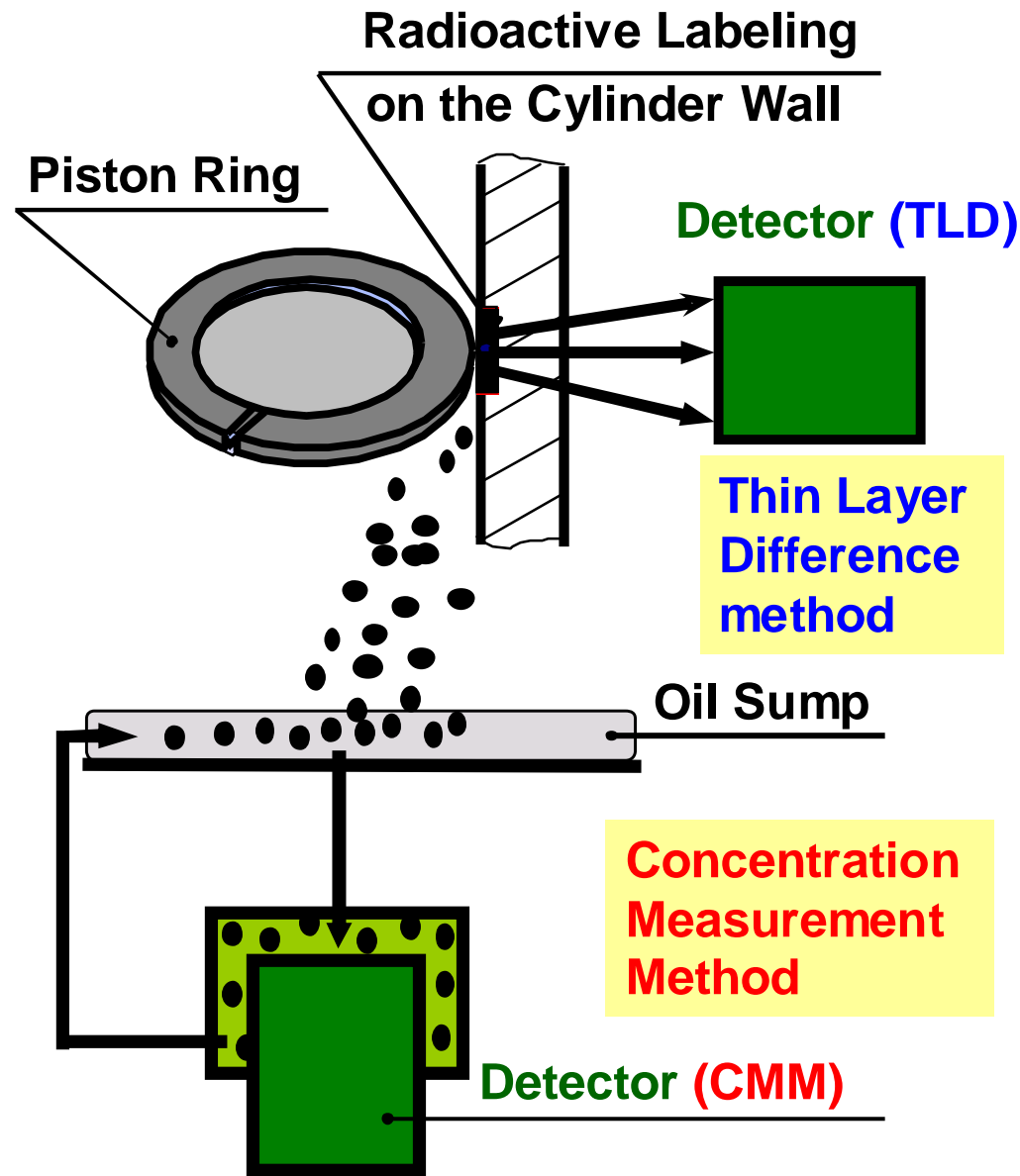
Measurement at
Weizmann Institute:

$$S_{17}(0) = 21.5(7) \text{ eV b}$$

Measurement of ${}^7\text{Be}(p,\gamma)$ with ion-implanted target



High sensitivity wear measurements



M. Hoffmann et al., Nucl. Instr. Meth. B183 (2001) 419.

P. Fehsenfeld et al., Nucl. Phys. A701 (2002) 235c.

⁷Be for wear analysis

Material	Density g/cm ³	Wear rate μm/10 ⁶ cyc.	Implant. depth in μm at beam energy of:				
			60 keV	260 keV	1.2 MeV	6 MeV	15 MeV
UHMWPE	0.97	50	0.36	1.1	2.9	13	43
Ti	4.52		0.17	0.56	1.5	6.1	18
CoCrMo	8.28		0.11	0.39	1.1	4.1	12
Alumina	3.1	0.15	0.20	0.59	1.6	6.7	21
Zirconia	5.5		0.15	0.48	1.3	5.3	16



In-vivo use:

≈ 10⁶ cycles/year

Simulator runs:

(2-10) • 10⁶ cycles

Required dose:

some p nA per cm² (e.g. ball of 22-28 mm diameter)

Cancer

About 1 000 000 new cancer cases per year in EU (15)
58 % local disease, 42 % generalized

45 % cured (5 year survival)

22 % surgery alone

12 % radiation therapy

6 % combination surgery + radiation

5 % chemo-therapy

just beginning of systemic radionuclide therapy

HOW: expose cancer cells or cancer tissue to sufficient radiation doses?

Radioisotopes in therapy = surgery with radiation

	Gamma Knife	β -Knife	α -Knife	Auger Knife
Isotope	^{60}Co $E_\gamma > 1 \text{ MeV}$	^{131}I , ^{90}Y , ^{153}Sm , ^{166}Ho , Others $E_\beta 1 - 3 \text{ MeV}$	$^{212}, ^{213}\text{Bi}$, ^{211}At , ^{149}Tb , $^{223}, ^{224}\text{Ra}$ $E_\alpha 4-8 \text{ MeV}$	^{125}I ^{165}Er $E_e \text{ few eV}$
Range	Full body penetration	about 1 cm	30 – 80 μm	1 μm
Application	Brain cancer	RIT Radio-immuno therapy	Leukemia, lymphoma	future
	Tissue surgery ex vivo	Tissue surgery in vivo	Cell surgery	Molecular surgery

Cancer therapy with alpha emitters

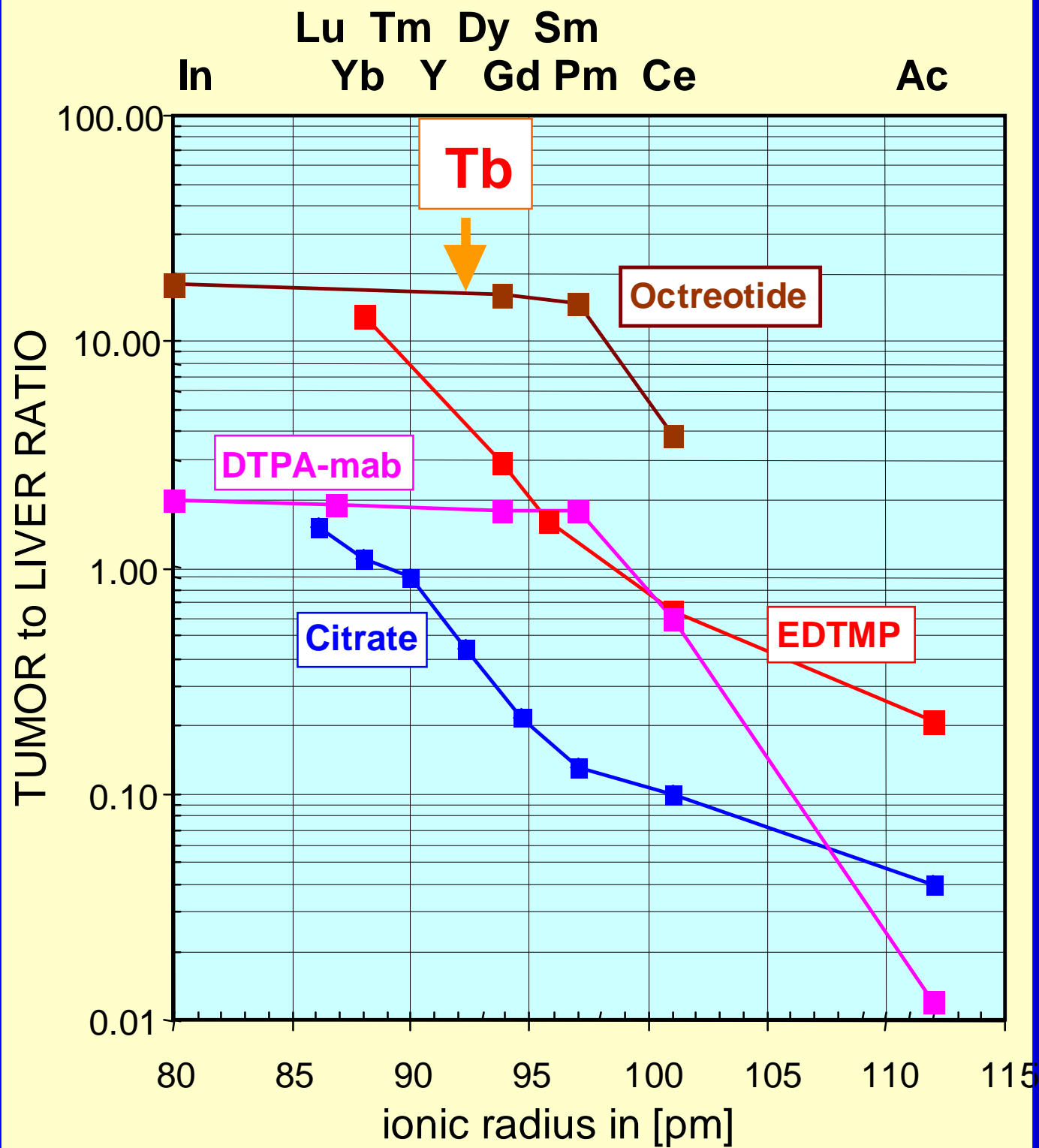
- half-life
- radiotoxicity of daughter isotopes
- biokinetics: in-vivo stability of chelating agent, clearance,...
- affordable
- reliable supply: more than one facility per geographic area

Isotopes for targeted alpha therapy

12 s	Ac 213 0.80 s	Ac 214 8.2 s	Ac 215 0.17 s	Ac 216 0.44 ms	Ac 217 0.74 μ s 69 ns	Ac 218 1.1 μ s	Ac 219 11.8 μ s	Ac 220 26 ms	Ac 221 52 ms	Ac 222 63 s 5.0 s	Ac 223 2.10 m	Ac 224 2.9 h	Ac 225 10.0 d	Ac 226 29 h
11 s	Ra 212 13.0 s	Ra 213 2.1 ms 2.74 m	Ra 214 2.46 s	Ra 215 1.67 ms	Ra 216 2.0 ns 0.18 μ s	Ra 217 1.6 μ s	Ra 218 25.6 μ s	Ra 219 10 ms	Ra 220 23 ms	Ra 221 28 s	Ra 222 38 s	Ra 223 11.43 d	Ra 224 3.66 d	Ra 225 14.8 d
10 m	Fr 211 3.10 m	Fr 212 20.0 m	Fr 213 34.6 s	Fr 214 3.35 ms 5.0 ms	Fr 215 0.09 μ s	Fr 216 0.70 μ s	Fr 217 16 μ s	Fr 218 22 ms 1.0 ms	Fr 219 21 ms	Fr 220 27.4 s	Fr 221 4.9 m	Fr 222 14.2 m	Fr 223 21.8 m	Fr 224 3.3 m
09 m	Rn 210 2.4 h	Rn 211 14.6 h	Rn 212 24 m	Rn 213 19.5 ms	Rn 214 63 ns 0.7 ns 0.27 μ s	Rn 215 2.3 μ s	Rn 216 45 μ s	Rn 217 0.54 ms	Rn 218 35 ms	Rn 219 3.96 s	Rn 220 55.6 s	Rn 221 25 m	Rn 222 3.825 d	Rn 223 23.2 m
08 h	At 209 5.4 h	At 210 8.3 h	At 211 7.22 h	At 212 119 ms 214 ms	At 213 0.11 μ s	At 214 0.76 μ s 0.27 μ s 0.86 μ s	At 215 0.1 ms	At 216 ?	At 217 32.3 ms	At 218 ~2 s	At 219 0.9 m	At 220 3.71 m	At 221 2.3 m	At 222 54 s
07 h	Po 208 2.898 a	Po 209 102 a	Po 210 138.38 d	Po 211 26.2 s 0.616 s	Po 212 45.1 s 17.1 ms 63 μ s	Po 213 4.2 μ s	Po 214 164 μ s	Po 215 1.78 ms	Po 216 0.15 s	Po 217 1.53 s	Po 218 3.05 m	Po 219 >300 ns	Po 220 >300 ns	
06 d	Bi 207 31.55 a	Bi 208 3.68 · 10 ⁸ a	Bi 209 100	Bi 210 5.013 d	Bi 211 2.17 m	Bi 212 25 m 60.60 m	Bi 213 45.59 m	Bi 214 19.9 m	Bi 215 36.9 s 7.7 m	Bi 216 3.6 m 2.17 m	Bi 217 98.5 s	Bi 218 33 s		136
05 a	Pb 206 24.1	Pb 207 22.1	Pb 208 52.4	Pb 209 3.253 h	Pb 210 22.3 a	Pb 211 36.1 m	Pb 212 10.64 h	Pb 213 10.2 m	Pb 214 26.8 m					134
04 a	Tl 205 70.48	Tl 206 3.7 m 4.23 m	Tl 207 1.33 s 4.77 m	Tl 208 3.053 m	Tl 209 2.16 m	Tl 210 1.30 m	Tl 211 >300 ns	Tl 212 >300 ns						132

^{149}Tb for targeted alpha therapy

Er 148 4.6 d α 1312-244; 143-210	Er 149 88a -4a	Er 150 16.5 s	Er 151 661a 288a	Er 152 10.3 a 4.80 1000 179	Er 153 37.1 a 4.077 261-286 146-176	Er 154 3.73 m 4.17 27.4	Er 155 5.3 m 4.492 140-242 204	Er 156 18.6 m 4.18 26.36	Er 157 18.65 m 4.1723 106-187 171	Er 158 2.25 h 4.184 776-981 107	Er 159 36 m 4.171 1404-1600 217	Er 160 28.6 h 4.17 171	Er 161 3.24 d 4.167 1027 217	Er 162 0.139 4.16 1411-1621	Er 163 75 m 4.164 171	Er 164 1.601 4.16 1411-1621
Ho 147 5.8 s 4.159-204 427-1284 171	Ho 148 33a 32a	Ho 149 36a 31a	Ho 150 84a 79a	Ho 151 42a 32a	Ho 152 89a 34m	Ho 153 99a 24m	Ho 154 32a 12m	Ho 155 48 m 4.14 700-106	Ho 156 73m 63a 39a	Ho 157 12.8 m 4.131-13 1880-242 182-217	Ho 158 37a 27a 11a	Ho 159 82a 33m	Ho 160 3a 15a 29m	Ho 161 82a 33a	Ho 162 89a 18m	Ho 163 1.1a 476a
Dy 146 29 s 4.121 280-241 381-2107	Dy 147 89a -89a	Dy 148 2.1 m 4.117-15 1500-1517	Dy 149 41a 43m	Dy 150 7.2 m 4.112 4.22 287	Dy 151 17 m 4.107 326-42 546-176	Dy 152 2.4 h 4.103 1327	Dy 153 6.29 h 4.1 181-114 181-264	Dy 154 3.0 · 10 ⁶ a 4.097	Dy 155 10.0 h 4.094-11 237	Dy 156 0.056 4.09 76a-1000	Dy 157 6.1 h 4.086	Dy 158 0.085 4.082 96a-1000	Dy 159 144.4 d 4.077 100	Dy 160 2.329 4.07 10a-1000	Dy 161 18.889 4.066 10a-1000	Dy 162 25.475 4.060
Tb 145 39a 37a	Tb 146 33a 34a	Tb 147 139a 139a	Tb 148 22m 28a	Tb 149 42m 417	Tb 150 58m 547a	Tb 151 39a 189a	Tb 152 42m 173a	Tb 153 2.39 d 4.07 174-176 176-200	Tb 154 89a 189a 219	Tb 155 5.37 d 4.07 181-200	Tb 156 2451 21a 24a	Tb 157 39 a 4.061	Tb 158 189a 189a	Tb 159 100 4.052	Tb 160 72.3 d 4.046-12 471-200 165 470	Tb 161 6.99 d 4.038-26 125-41-76 37
Gd 144 4.5 m 4.031 332-245 600-247	Gd 145 45a 33a	Gd 146 48.3 d 4.027 106-116 116	Gd 147 59.1 h 4.021 226-296 323	Gd 148 74.6 a 4.016 170-290 347	Gd 149 9.26 d 4.012 170-290 347	Gd 150 1.8 · 10 ⁶ a 4.008	Gd 151 120 d 4.004 134-243 171	Gd 152 0.20 4.001 216-300 347	Gd 153 239-47 d 4.000 151-170 171	Gd 154 2.18 4.000	Gd 155 14.80 4.00000 100-1000	Gd 156 20.47 4.00000 100-1000	Gd 157 15.85 4.00000 100-1000	Gd 158 24.84 4.00000 100-1000	Gd 159 18.48 h 4.00000 100-1000	Gd 160 21.83 4.00000 100-1000
Eu 143 2.6 m 4.011-1000 1012-106 102-10	Eu 144 13.2 s 4.007 1000	Eu 145 5.93 d 4.003 204-1000 324	Eu 146 4.51 d 4.001 747-833 324	Eu 147 24.6 d 4.000 100-100 324	Eu 148 55.6 d 4.000 280-330 324	Eu 149 93.1 d 4.000 100-100 324	Eu 150 47.81 4.000 100-100 324	Eu 151 47.81 4.000 100-100 324	Eu 152 47.81 4.000 100-100 324	Eu 153 52.19 4.000 100-100 324	Eu 154 52.19 4.000 100-100 324	Eu 155 4.761 a 4.000 100-100 324	Eu 156 15.2 d 4.000 100-100 324	Eu 157 15.16 h 4.000 100-100 324	Eu 158 48 m 4.000 100-100 324	Eu 159 181 m 4.000 100-100 324
Sm 142 72.4 m 4.000 1000	Sm 143 8a 38m	Sm 144 3.07 4.000	Sm 145 340 d 4.000 100-100 100	Sm 146 1.03 · 10 ⁶ a 4.000	Sm 147 14.99 4.000 100-100 100	Sm 148 11.24 4.000 100-100 100	Sm 149 13.82 4.000 100-100 100	Sm 150 7.39 4.000 100-100 100	Sm 151 93 a 4.000 100-100 100	Sm 152 36.75 4.000 100-100 100	Sm 153 46.37 h 4.000 100-100 100	Sm 154 22.75 4.000 100-100 100	Sm 155 22.4 m 4.000 100-100 100	Sm 156 9.4 h 4.000 100-100 100	Sm 157 8.11 m 4.000 100-100 100	Sm 158 5.51 m 4.000 100-100 100
Pm 141 23.9 m 4.000 100-100 100	Pm 142 40.5 s 4.000 100	Pm 143 265 d 4.000 100	Pm 144 1.0 a 4.000 100-100 100	Pm 145 17.7 a 4.000 100-100 100	Pm 146 5.93 a 4.000 100-100 100	Pm 147 2.62 a 4.000 100-100 100	Pm 148 53.1 h 4.000 100-100 100	Pm 149 53.1 h 4.000 100-100 100	Pm 150 2.7 h 4.000 100-100 100	Pm 151 39.4 h 4.000 100-100 100	Pm 152 39.4 h 4.000 100-100 100	Pm 153 5.3 m 4.000 100-100 100	Pm 154 5.3 m 4.000 100-100 100	Pm 155 41.5 s 4.000 100-100 100	Pm 156 26.7 s 4.000 100-100 100	Pm 157 10.6 s 4.000 100-100 100
Nd 140 3.37 d 4.000 100	Nd 141 42a 33a	Nd 142 272 4.000	Nd 143 12.2 4.000 100-100 100	Nd 144 23.8 4.000 100-100 100	Nd 145 8.3 4.000 100-100 100	Nd 146 17.2 4.000 100-100 100	Nd 147 10.99 d 4.000 100-100 100	Nd 148 5.7 4.000 100-100 100	Nd 149 1.73 h 4.000 100-100 100	Nd 150 5.6 4.000 100-100 100	Nd 151 12.4 m 4.000 100-100 100	Nd 152 11.4 m 4.000 100-100 100	Nd 153 38.9 s 4.000 100-100 100	Nd 154 25.0 s 4.000 100-100 100	Nd 155 8.9 s 4.000 100-100 100	Nd 156 5.6 s 4.000 100-100 100
Pr 139 4.5 h 4.000 100	Pr 140 3.4 m 4.000 100	Pr 141 100 4.000	Pr 142 14.3 m 19.3a	Pr 143 13.07 d 4.000 100-100 100	Pr 144 72m 113a	Pr 145 5.98 h 4.000 100-100 100	Pr 146 24.0 m 4.000 100-100 100	Pr 147 13.6 m 4.000 100-100 100	Pr 148 287 217m	Pr 149 2.25 m 4.000 100-100 100	Pr 150 -3a 4.000 100-100 100	Pr 151 18.9 s 4.000 100-100 100	Pr 152 3.8 s 4.000 100-100 100	Pr 153 4.3 s 4.000 100-100 100	Pr 154 2.3 s 4.000 100-100 100	Pr 155 >300 ns 4.000
Ce 138 0.251 4.000 100	Ce 139 38.1a 37.8a	Ce 140 88-450 4.000	Ce 141 32.68 d 4.000 100-100 100	Ce 142 11.114 4.000 100-100 100	Ce 143 33.0 h 4.000 100-100 100	Ce 144 294.6 d 4.000 100-100 100	Ce 145 2.95 m 4.000 100-100 100	Ce 146 13.5 m 4.000 100-100 100	Ce 147 57 s 4.000 100-100 100	Ce 148 48 s 4.000 100-100 100	Ce 149 5 s 4.000 100-100 100	Ce 150 4.1 s 4.000 100-100 100	Ce 151 1.0 s 4.000 100-100 100	Ce 152 1.4 s 4.000 100-100 100	Ce 153 >300 ns 4.000	Ce 154 >300 ns 4.000



Comparison

of the
bio-distribution
of different
tumor seeking tracers
labeled with
radio-lanthanides,
 ^{225}Ac and ^{111}In

free chelates:

Citrate
EDTMP

specific tracers:

Octreotide
and
Mab

Linker:

Aminobenzyl-DTPA

G.J.Beyer, Hyperfine Interactions 129 (2000) 529.

Principle of Radio Immuno Therapy

DAUDI cells

Cell membrane

Proteins in healthy cells

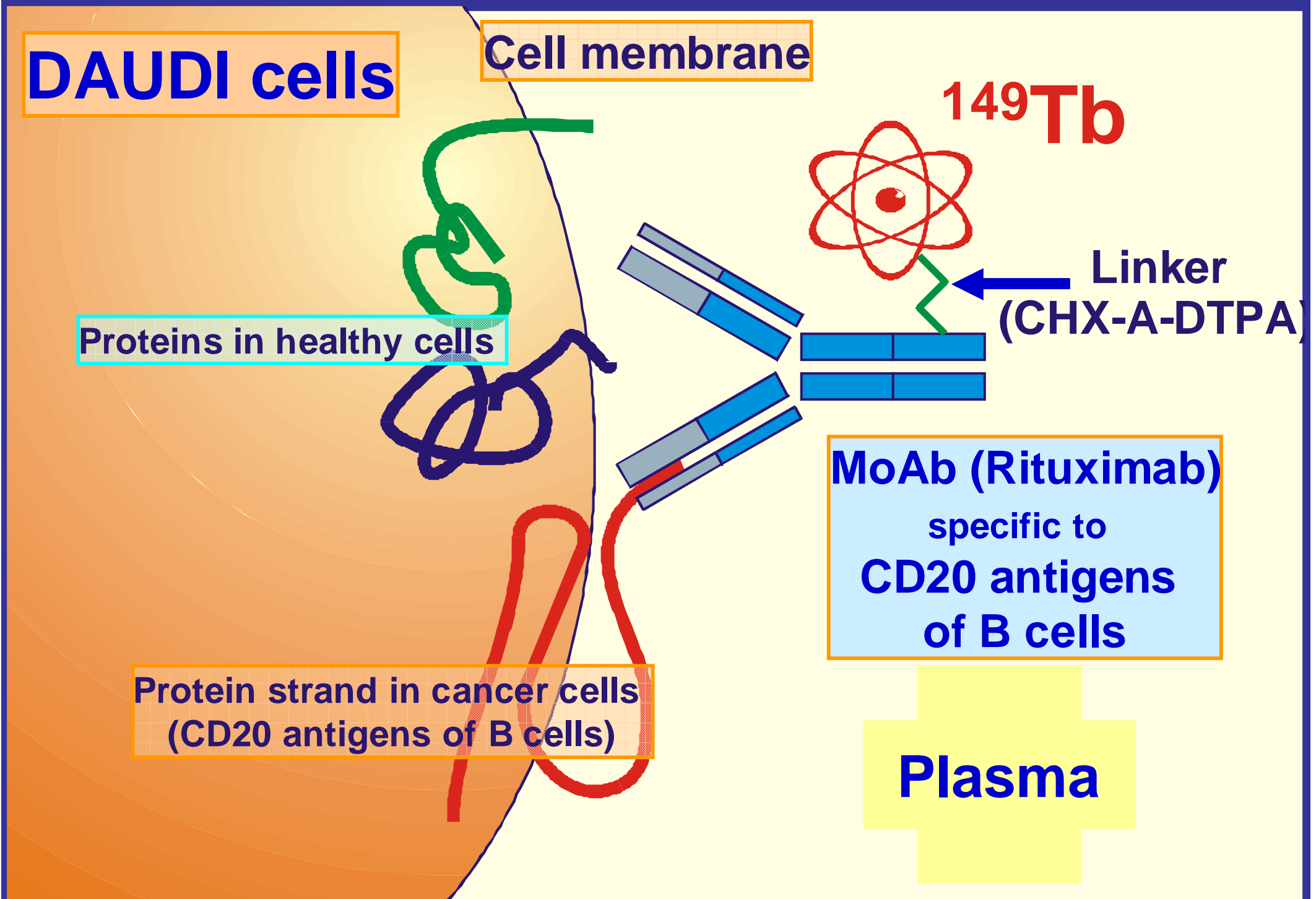
**Protein strand in cancer cells
(CD20 antigens of B cells)**

^{149}Tb

**Linker
(CHX-A-DTPA)**

**MoAb (Rituximab)
specific to
CD20 antigens
of B cells**

Plasma



$1 \cdot 10^5$ limfoma cells injected to all mice (Daudi cells of Burkitt limfoma)



NO treatment



1

2 days later the mice have been divided into 4 groups:

5 μ g MoAb (Rituximab, specific to CD20 antigens of B cells)



300 μ g MoAb



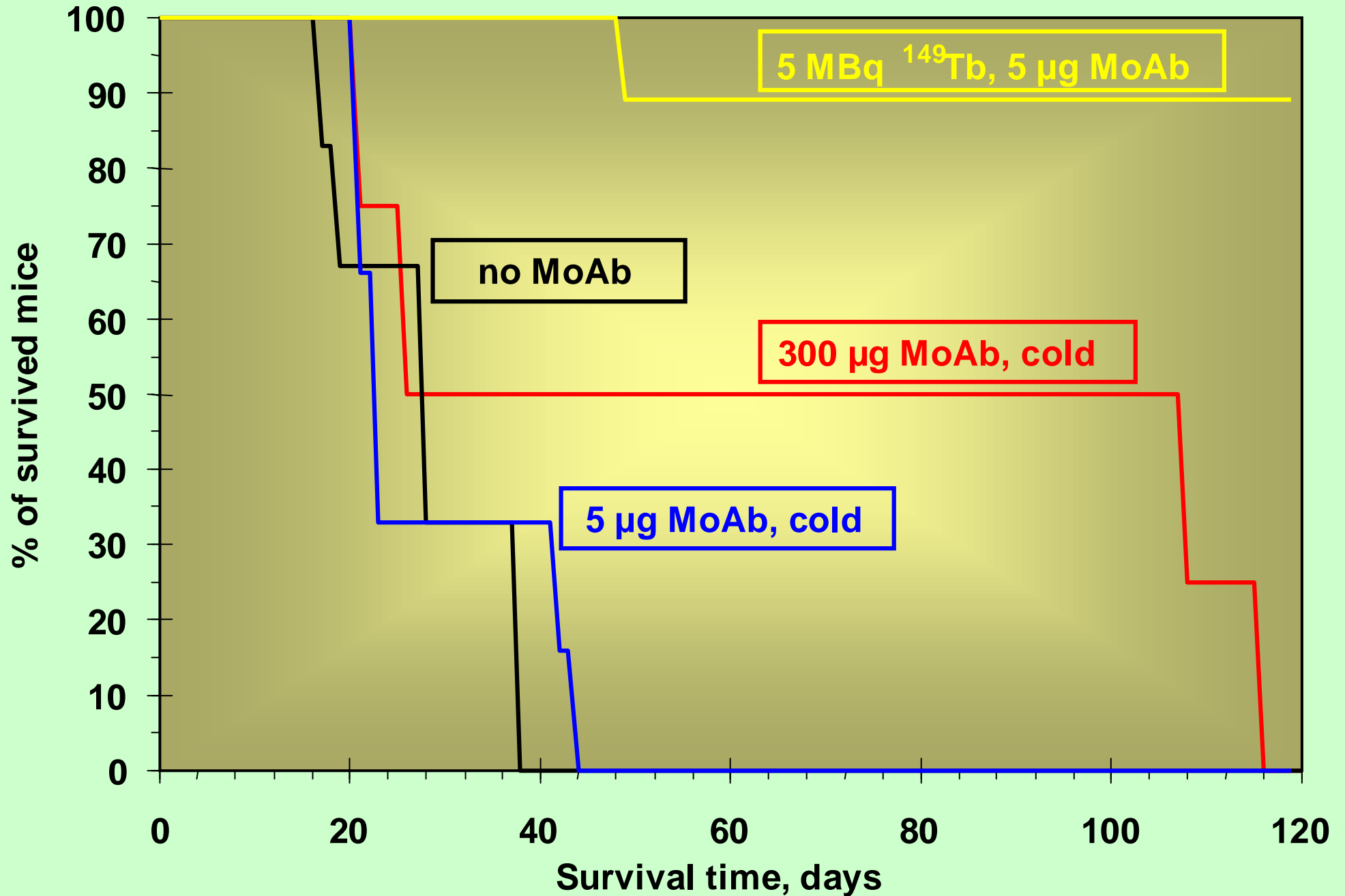
5 MBq ^{149}Tb -MoAb (5 μ g MoAb)



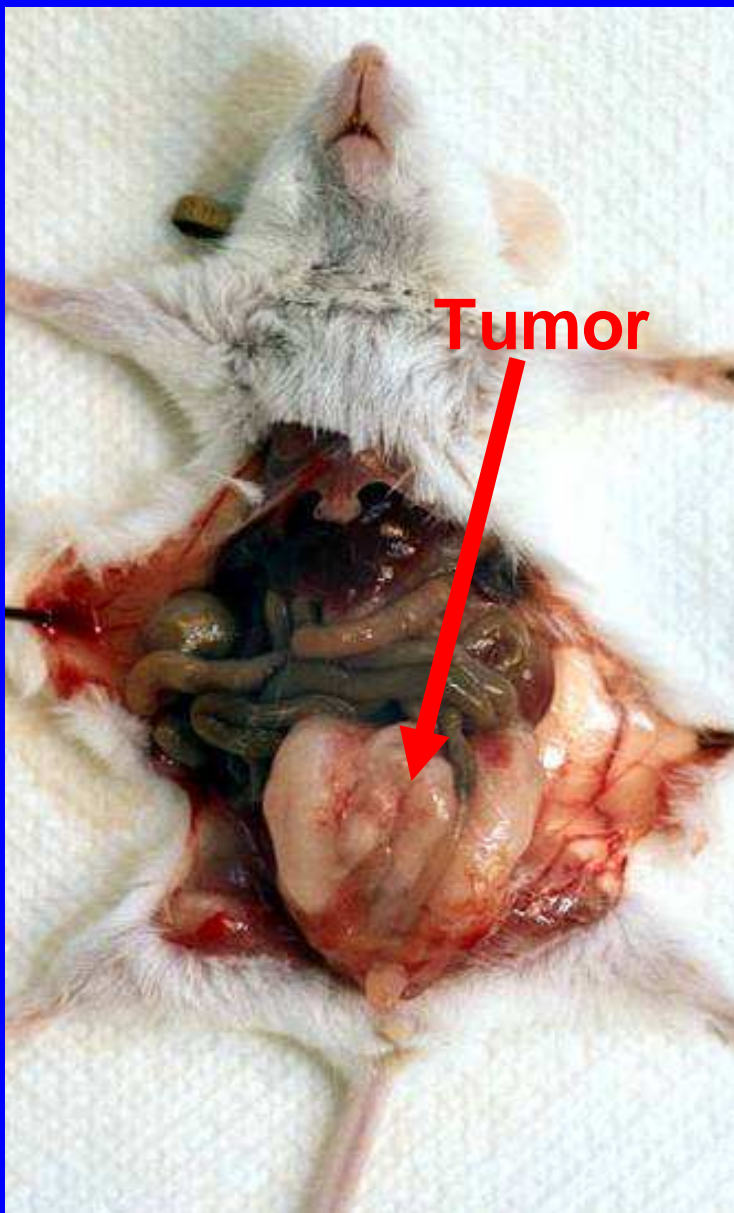
First in vivo experiment to demonstrate the efficiency of alpha targeted therapy using ^{149}Tb produced at ISOLDE, Summer 2001

SCID mice (Severe Combined Immunodeficient)

Survival of SCID mice



103 d p.i.



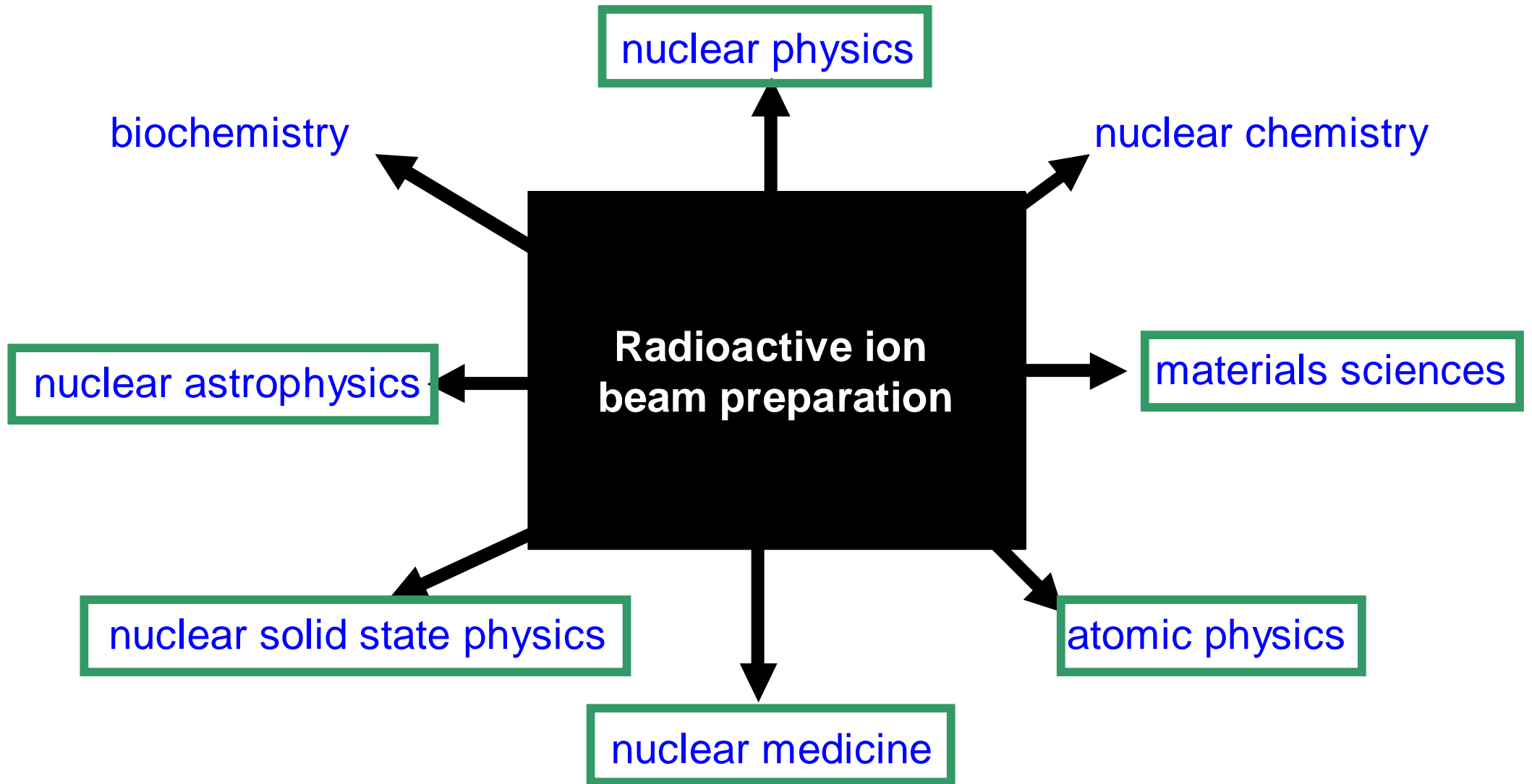
300 μ g MoAb cold

108 d p.i.



5 MBq ^{149}Tb -MoAb (5 μ g)

Applications



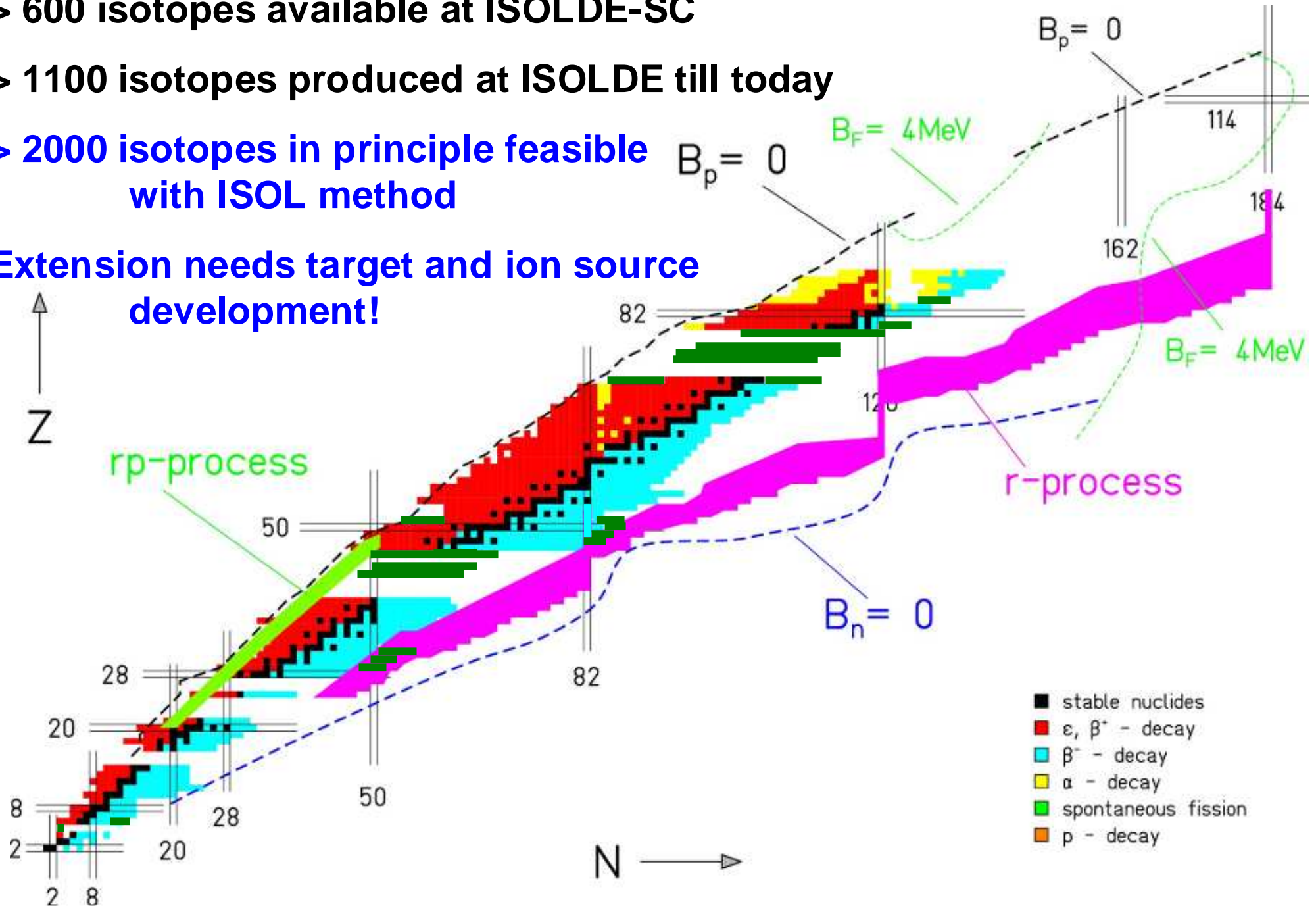
CERN provides physics results even when the LHC does not run!

ISOL beams are not a subfield of nuclear physics but have incredibly many applications across many disciplines

Nuclear chart at ISOLDE

- > 600 isotopes available at ISOLDE-SC
- > 1100 isotopes produced at ISOLDE till today
- > 2000 isotopes in principle feasible with ISOL method

Extension needs target and ion source development!



Existing thick-target ISOL beams

Isotopes with $T_{1/2} < 0.1$ s separated																					
Isotopes with $T_{1/2} < 10$ s separated																					
Isotopes with $T_{1/2} > 10$ s separated																					
1																	2				
H																	He				
3	4															5	6	7	8	9	10
Li	Be															B	C	N	O	F	Ne
11	12															13	14	15	16	17	18
Na	Mg															Al	Si	P	S	Cl	Ar
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36				
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr				
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54				
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe				
55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86				
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn				
87	88	89	104	105	106	107	108	109	110	111	112	113	114	115	116		118				
Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg											

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Identification \neq Separation

- **Identification:**

The beam composition is determined but not changed.

e.g. time-of-flight measurement, ΔE measurement,...

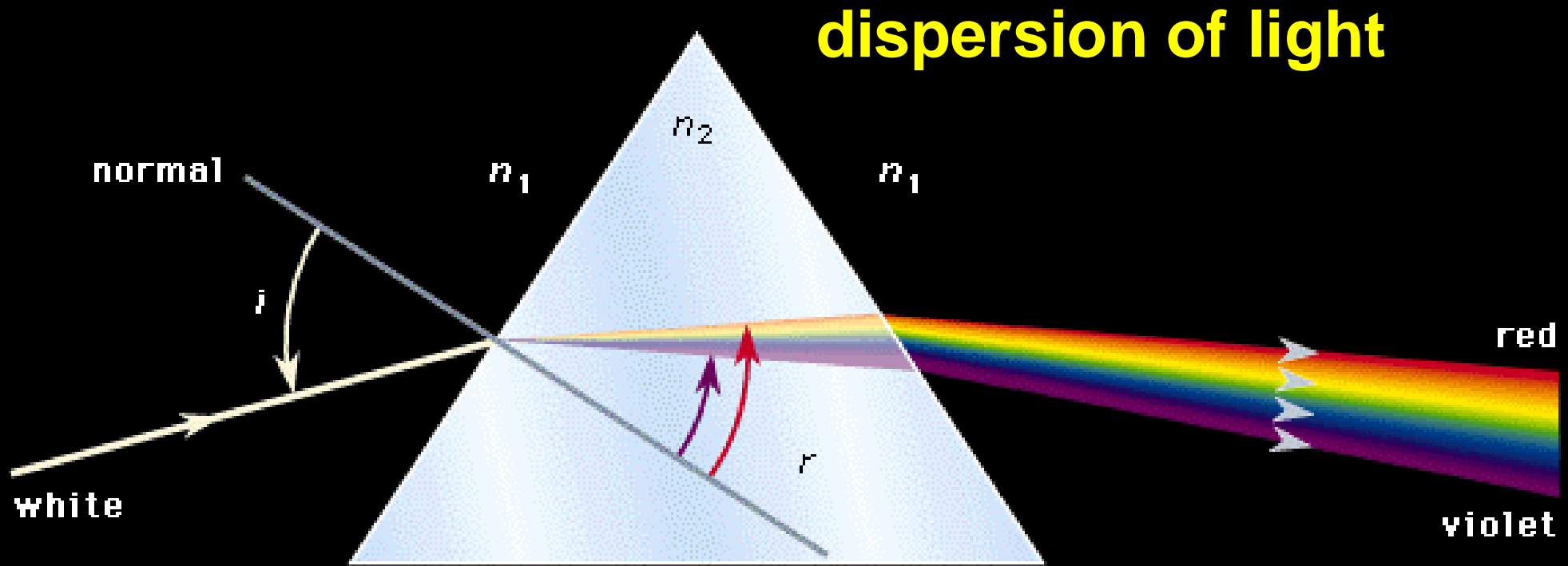
- **Separation:**

Beam contaminations are removed.

e.g. mass separation, chemical separation,...

- **Unique isotope selection** requires the combination of **at least two** different identification/separation methods.
- A **higher-fold** combination gives improved suppression factors.

Prism



The angles i and r that the rays make with the normal are the angles of incidence and refraction. Because n_2 depends upon wavelength, the incident white ray separates into its constituent colours upon refraction, with deviation of the red ray the least and the violet ray the most.

Dispersive ion optical elements

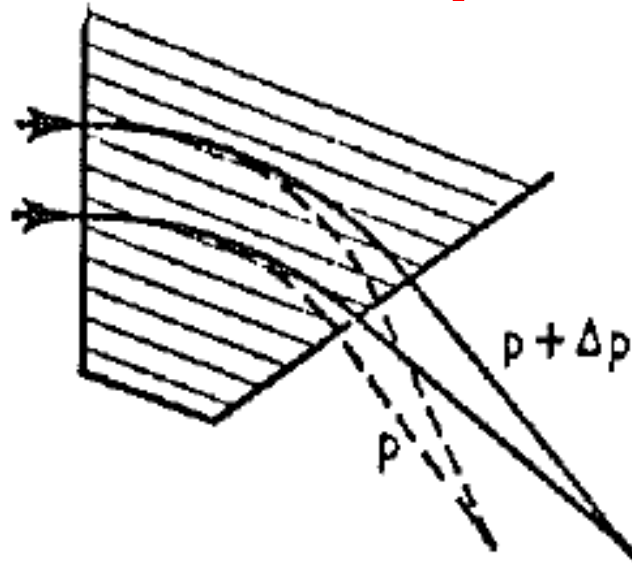


FIG. 5.15 A system with momentum dependent deflection of the central ray, showing lateral displacement due to momentum spread.

- magnets are momentum dispersive
- electrostatic deflectors are energy dispersive
- Wien filters are velocity dispersive

Focusing by tilted entrance/exit of magnetic field

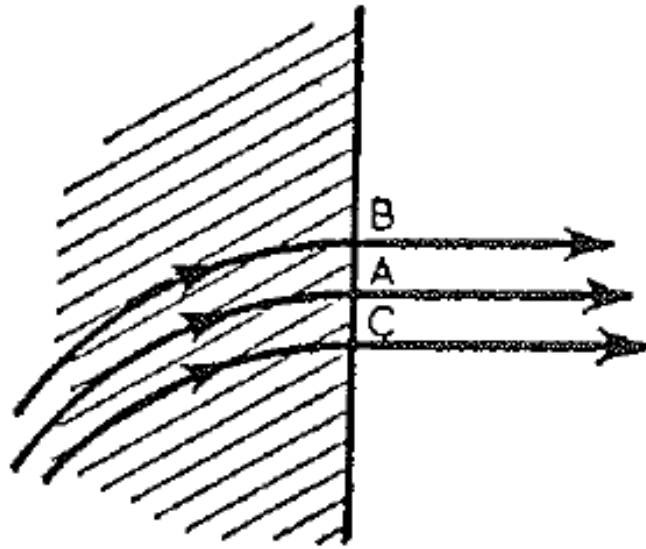


FIG. 5.3 Particles leaving a magnetic field normal to the edge.

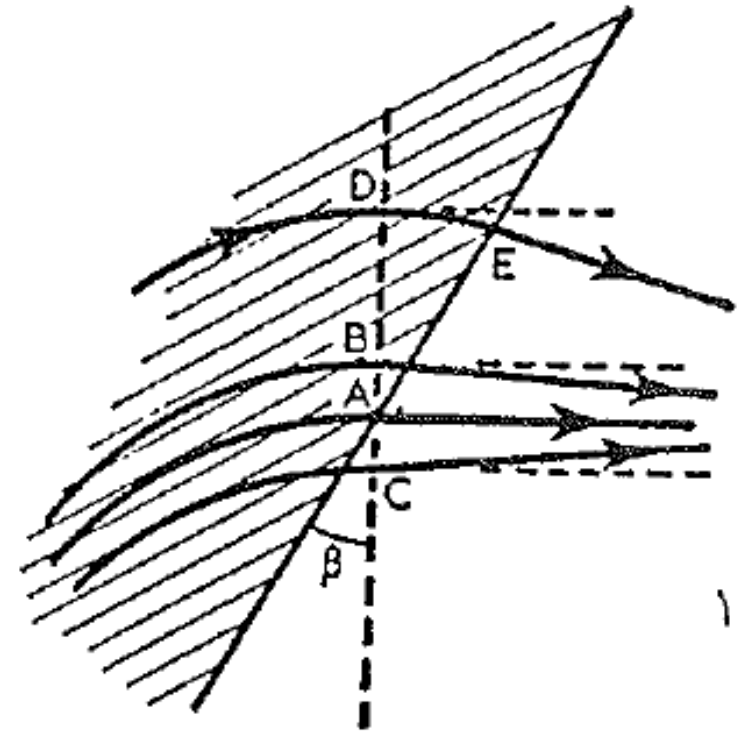


FIG. 5.4 Particles leaving a magnetic field at an angle to the edge. Dotted lines are for normal exit (cf. Fig. 5.3).

horizontal focusing effect

Focusing by tilted entrance/exit of magnetic field

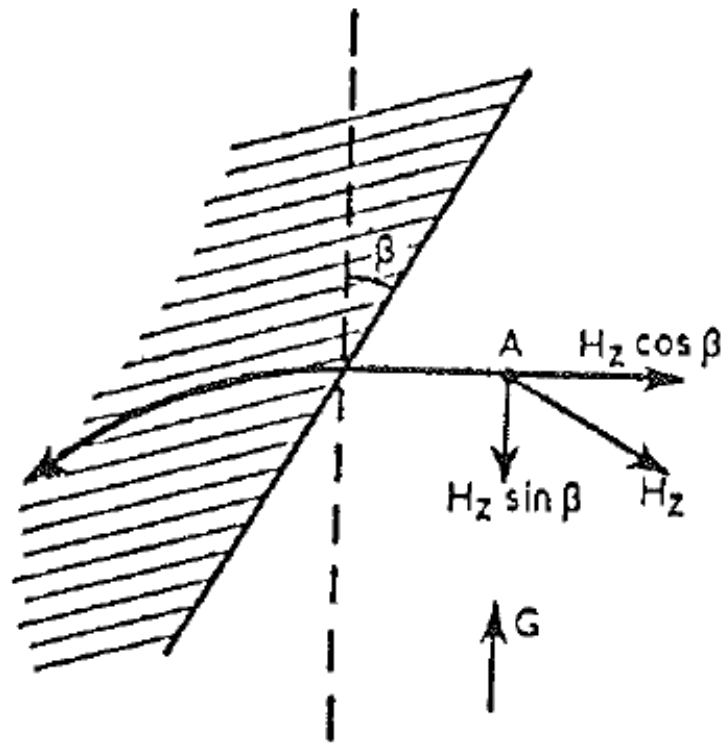


FIG. 5.5 Plan view of a positively charged particle entering a magnetic field directed into the paper. The trajectory makes an angle β with the normal. For view in the direction of arrow G see Fig. 5.6.

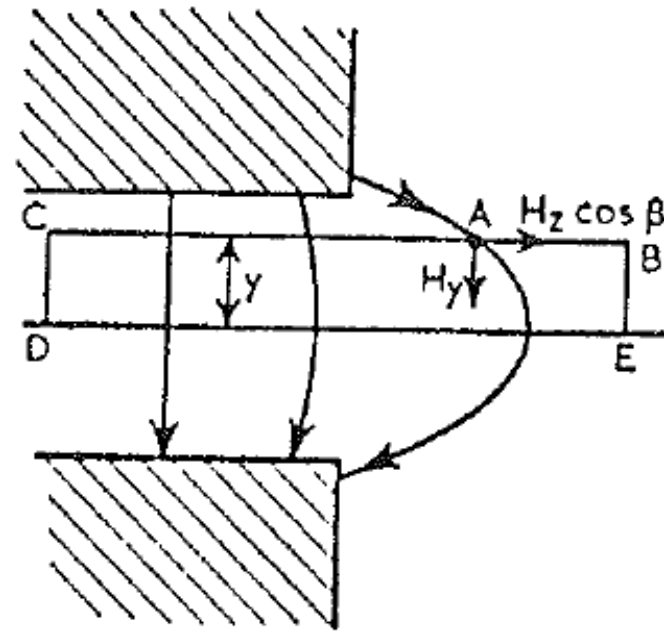


FIG. 5.6 View of Fig. 5.5 in the direction of arrow G . DE is the median plane on which $H_z = 0$.

vertical defocusing effect

Quadrupoles

Electrostatic:

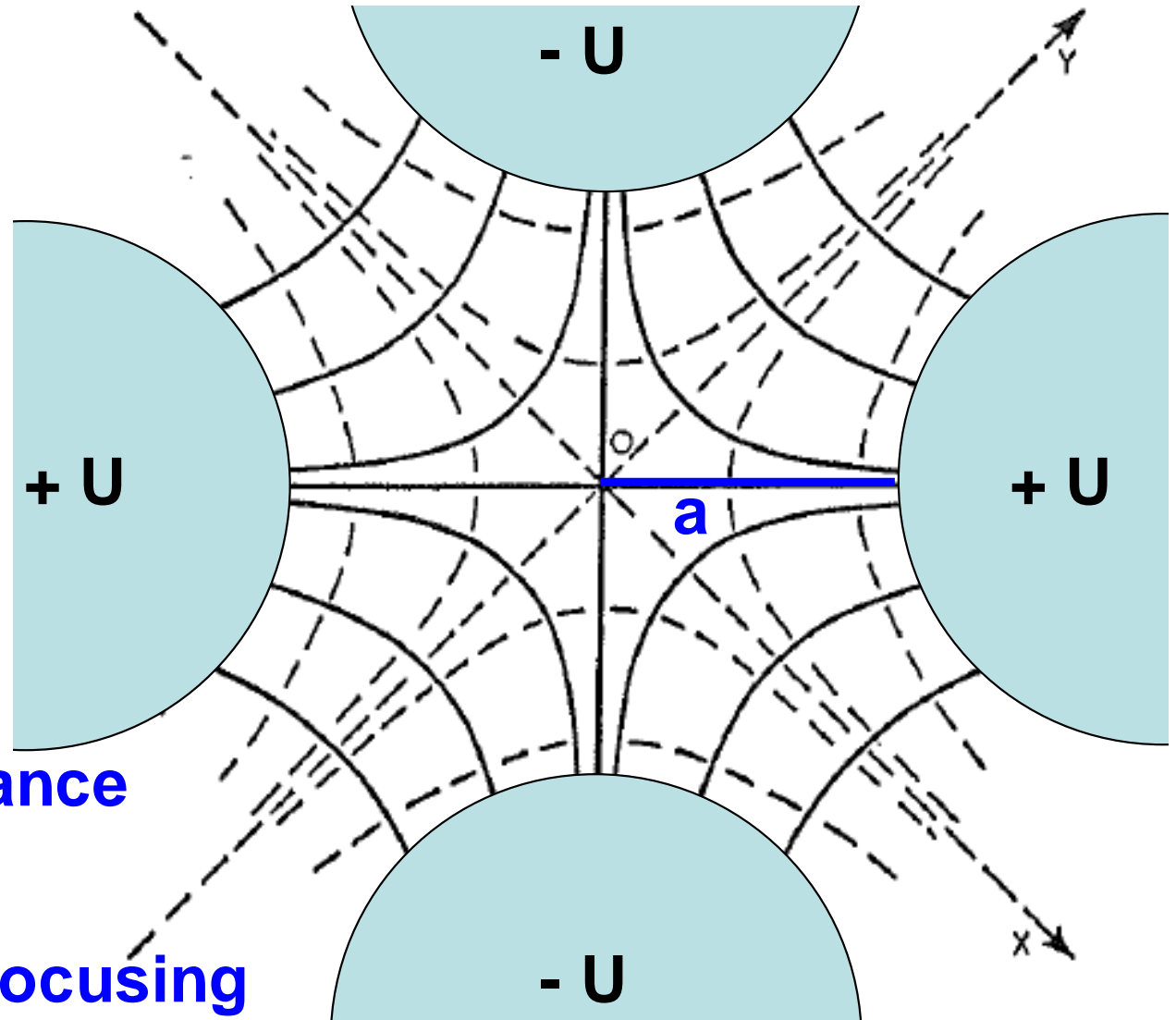
$$V = U (x^2 - y^2)/a^2$$

$$\mathcal{E} = - \text{grad } V$$

$$E_x = -dV/dx = -2U/a^2 x$$

$$E_y = -dV/dy = 2U/a^2 y$$

1. Force increases proportionally to distance from origin.
 2. Focusing in x and defocusing in y (or vice versa).
- ⇒ requires quadrupole doublet or triplet to focus in x and y



Multipole correction elements

Correction of higher-order effects (aberrations) by hexapole, octupole, etc. fields. Often limited by beam diagnostics!

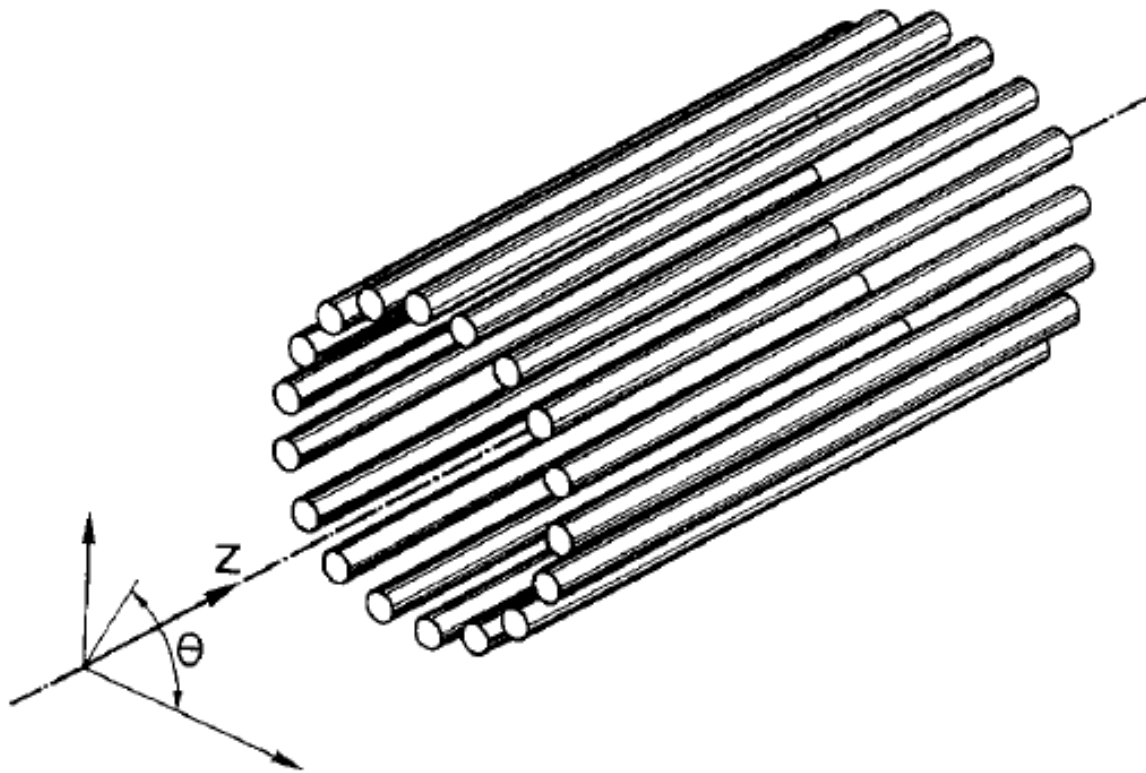


Fig. 1. Squirrel-cage-like electrode arrangement of an electrostatic $2(n+1)$ pole consisting of 18 wires, i.e. a squirrel-cage with $n=8$. In this multipole the potential of each wire is controlled by a separate power supply.

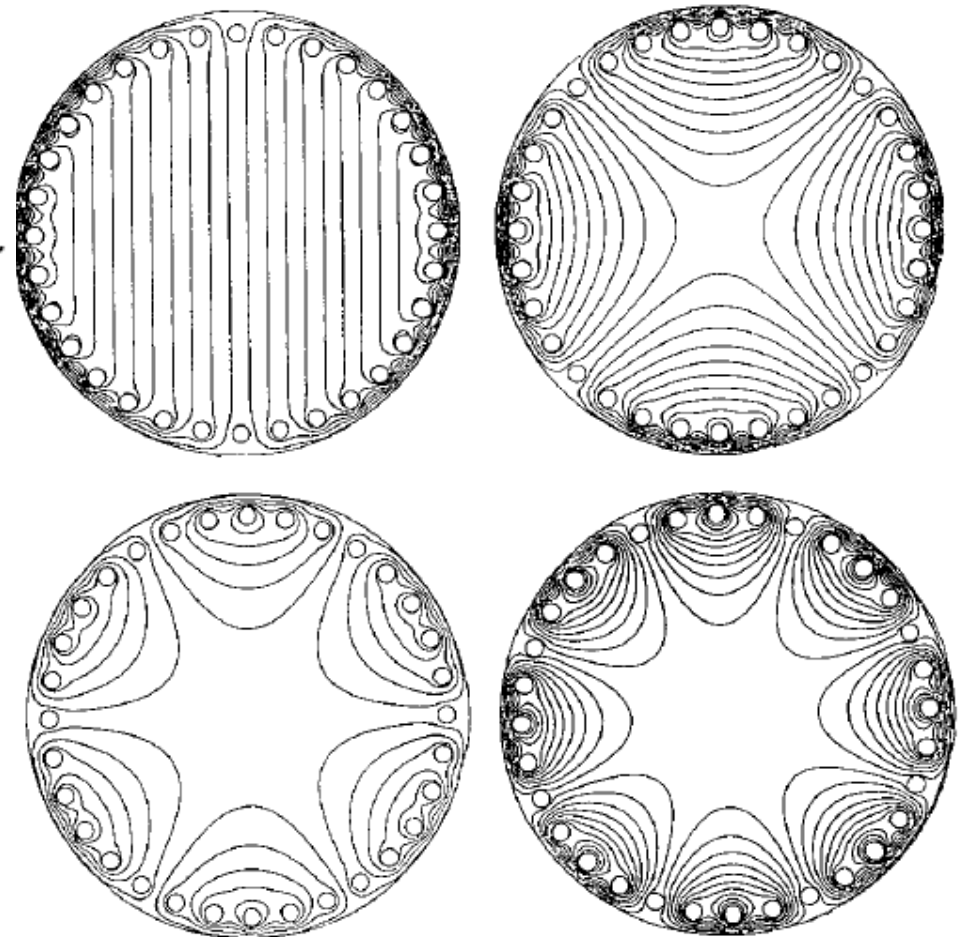


Fig. 3. Calculated potential distributions in a 32-wire squirrel-cage multipole for the cases of dipole ($V_1 \neq 0$), quadrupole ($V_2 \neq 0$), hexapole ($V_3 \neq 0$), octupole ($V_4 \neq 0$) excitations [see eq. (1)] with vanishing $\psi_1, \psi_2, \psi_3, \psi_4, \dots$.

Ion-optical calculations

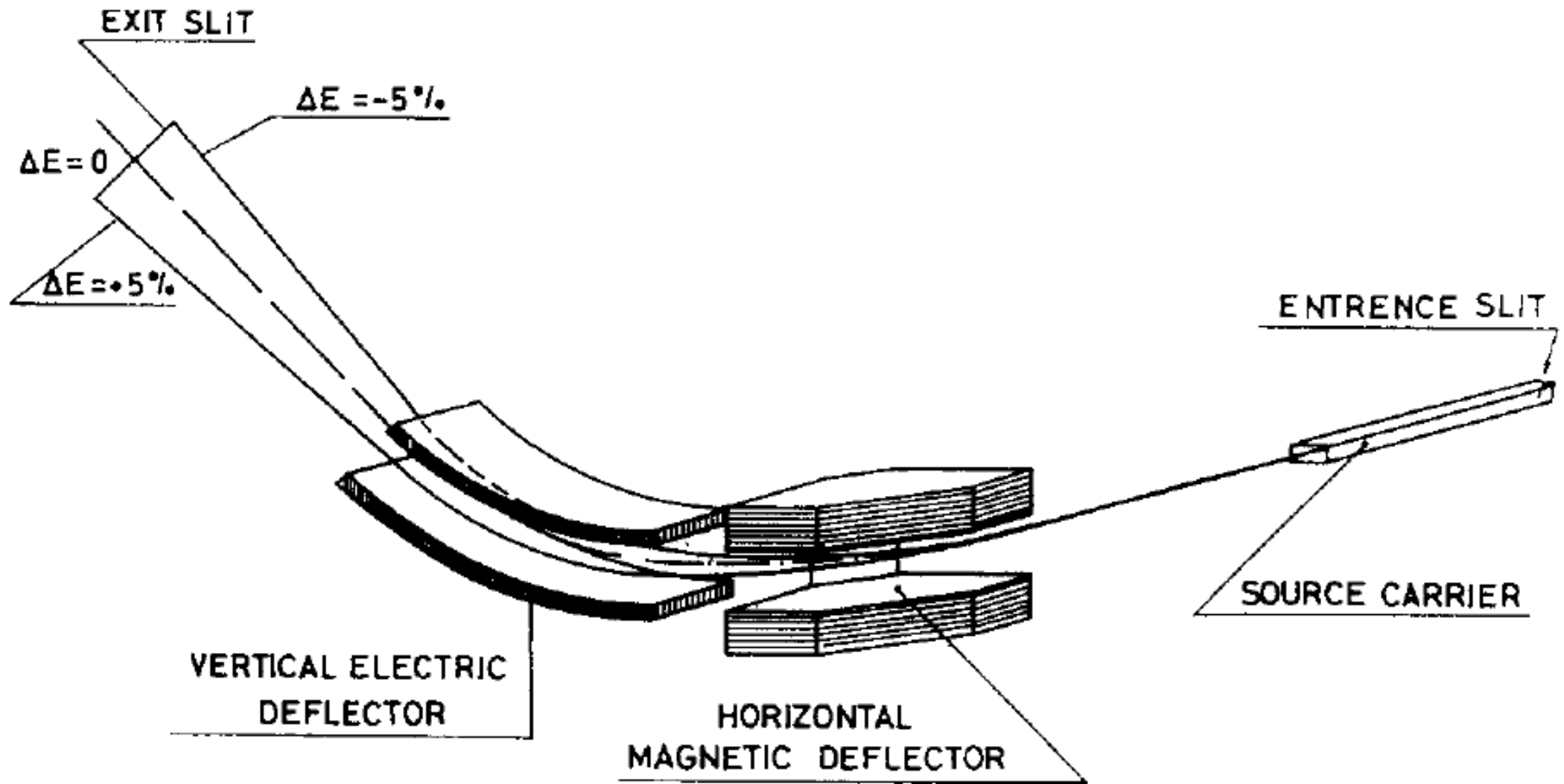
1. Matrix calculation:

TRANSPORT, COSY-INFINITY, GIOS, GICO, LISE++,...

2. MC simulations/ray tracing:

SIMION, ZGOUBY, RAYTRACE, LISE++, MOCADI,...

Focal plane of LOHENGRIN

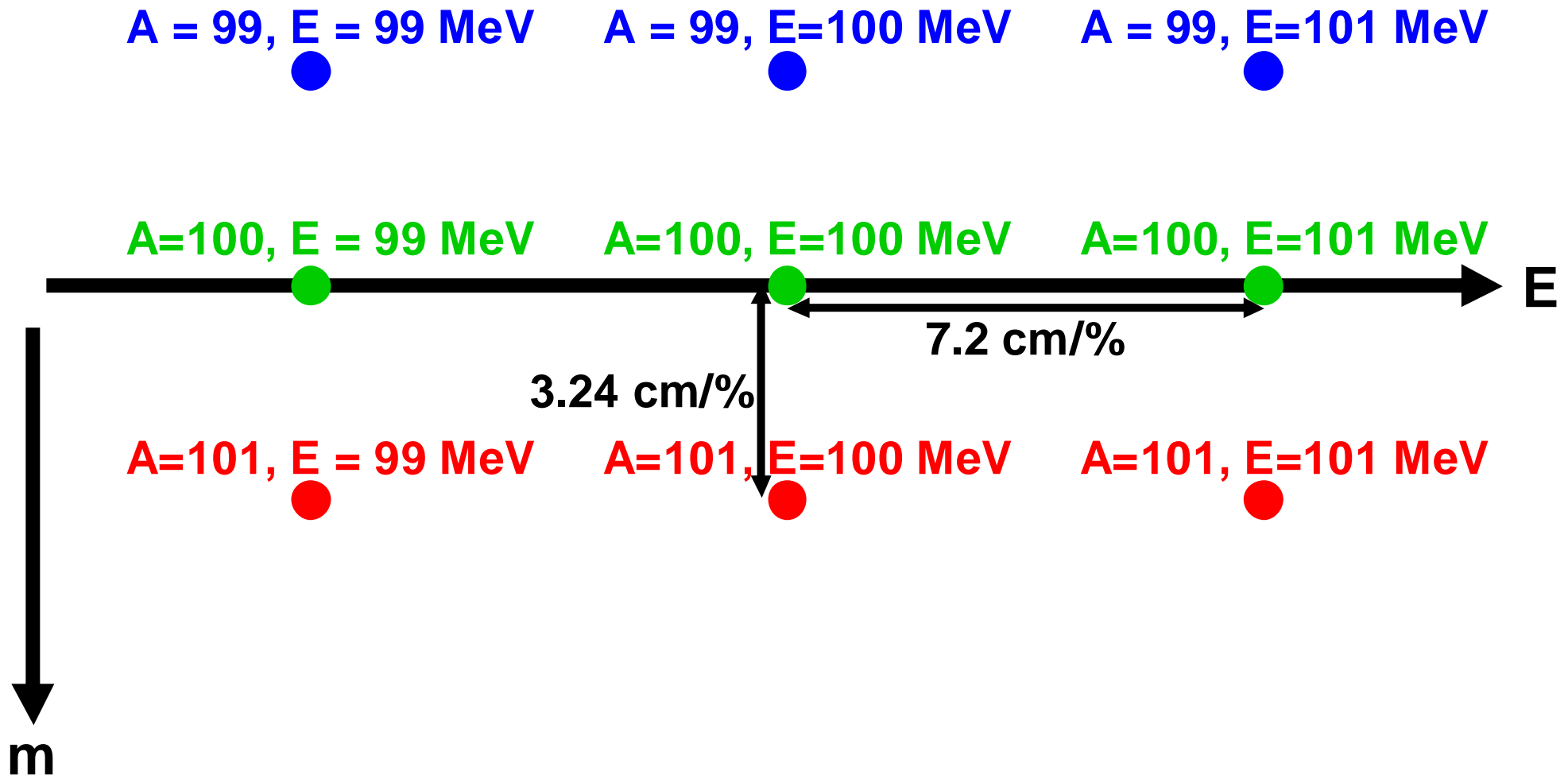


P. Armbruster et al., Nucl. Instr. Meth. 139 (1976) 213.

LOHENGRIN focal plane

Energy dispersion: $\Delta x = D_E \Delta E/E$; $D_E = 7.2 \text{ cm}/\% = 7.2 \text{ m}$

Mass dispersion: $\Delta y = D_m \Delta m/m$; $D_m = 3.24 \text{ cm}/\% = 3.24 \text{ m}$



LOHENGRIN energy resolution

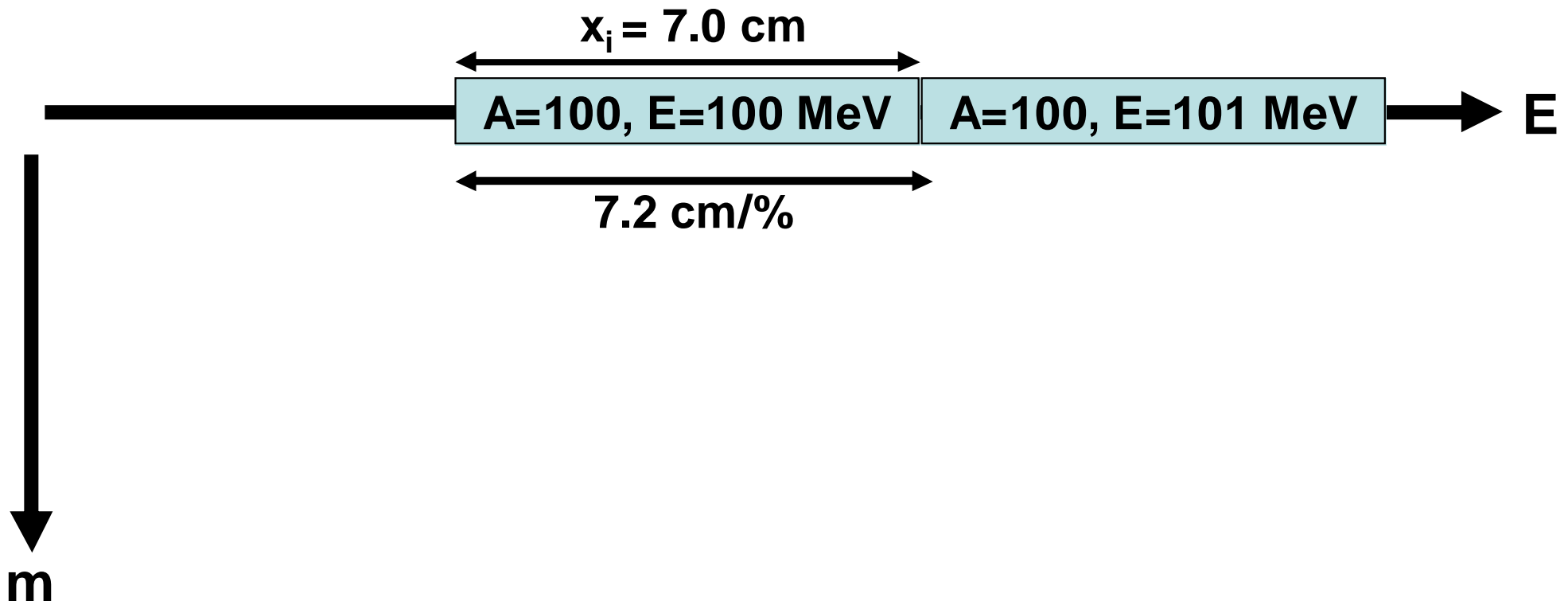
Magnification in x:

$$x_i = M_x x_o$$

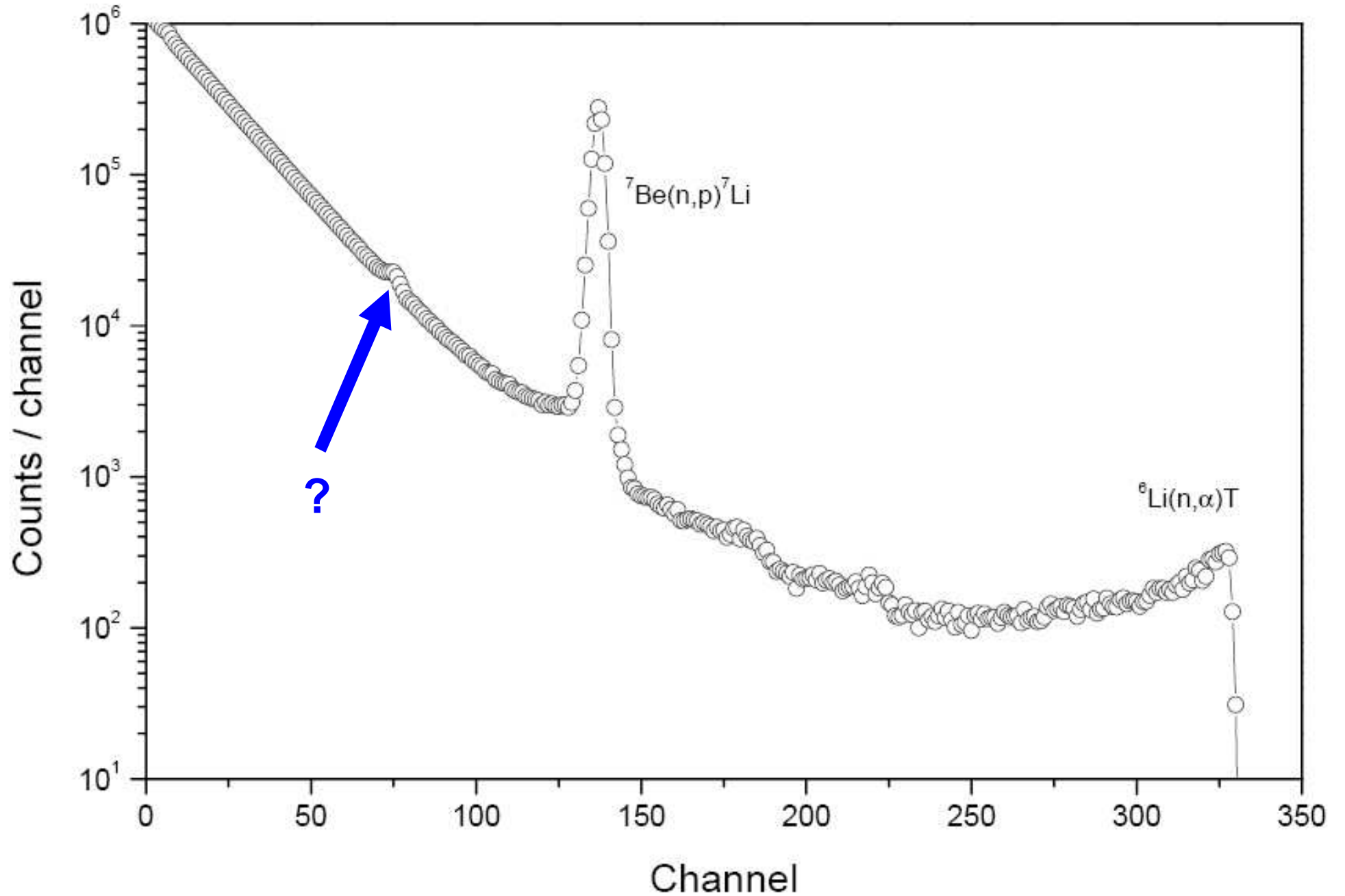
Energy resolution:

$$R_E = x_i / D_E = M_x \cdot x_o / D_E =$$
$$= 1.0 \cdot 7.0 \text{ cm} / 7.2 \text{ m} \approx 1/100$$

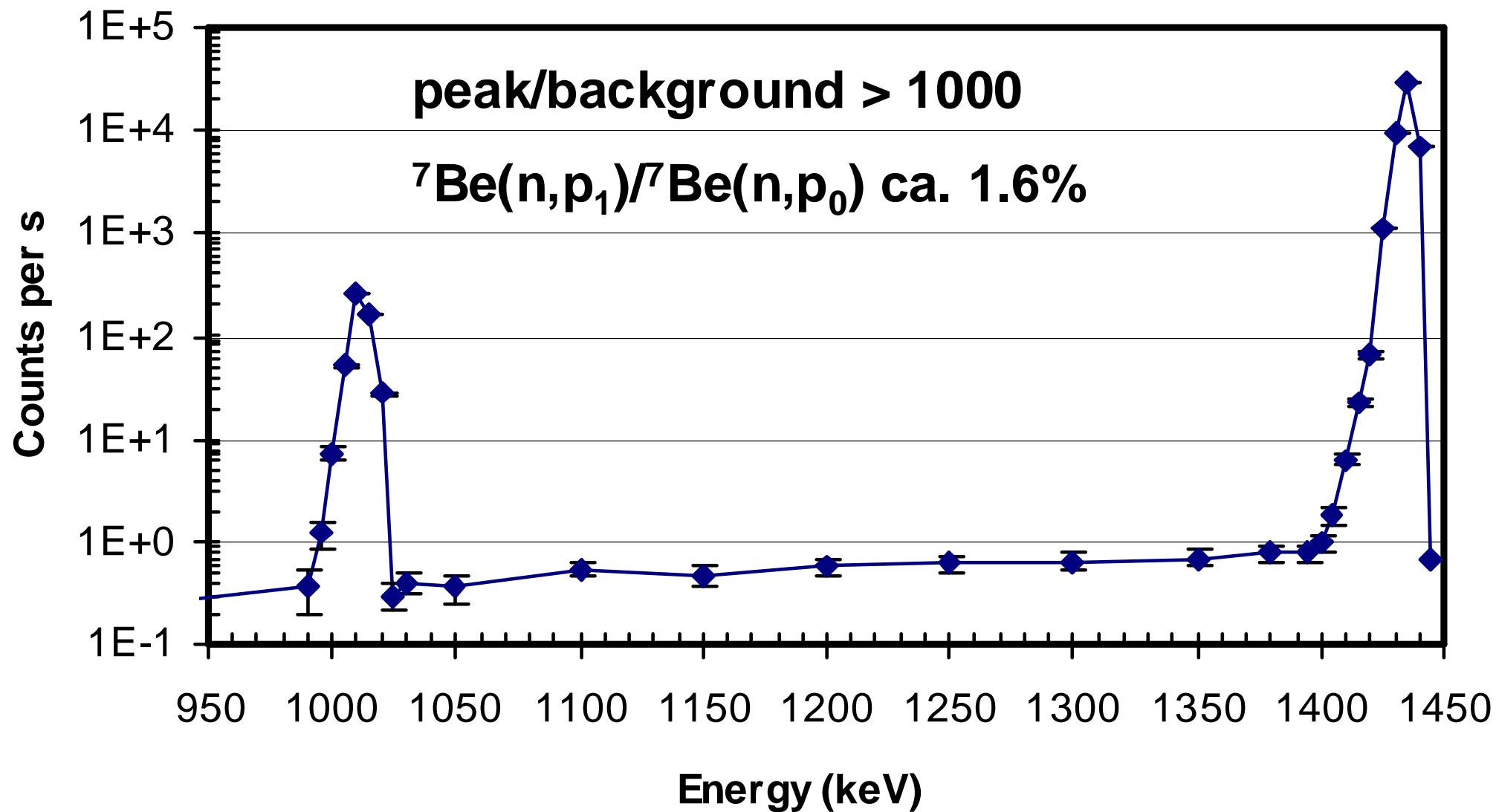
depends on target length!



${}^7\text{Be}(n,p)$ spectrum measured at neutron beam



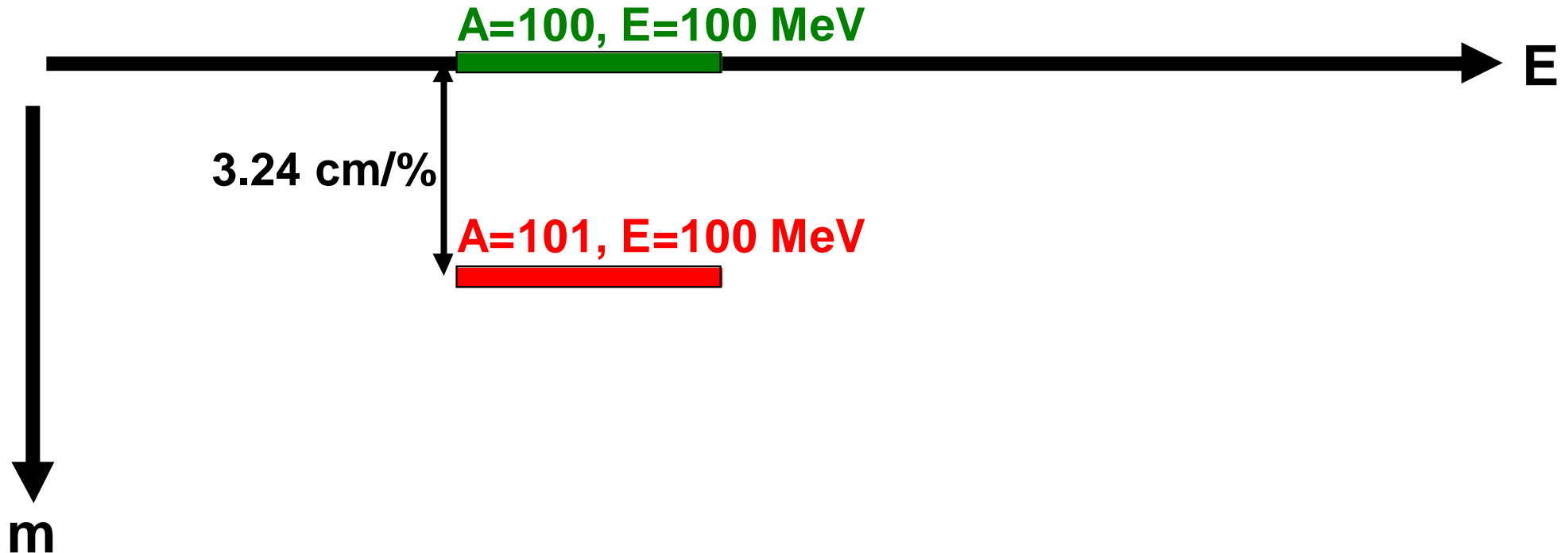
${}^7\text{Be}(n,p)$ measured at LOHENGRIN



LOHENGRIN mass resolution

Magnification in y : $y_i = M_y y_o$

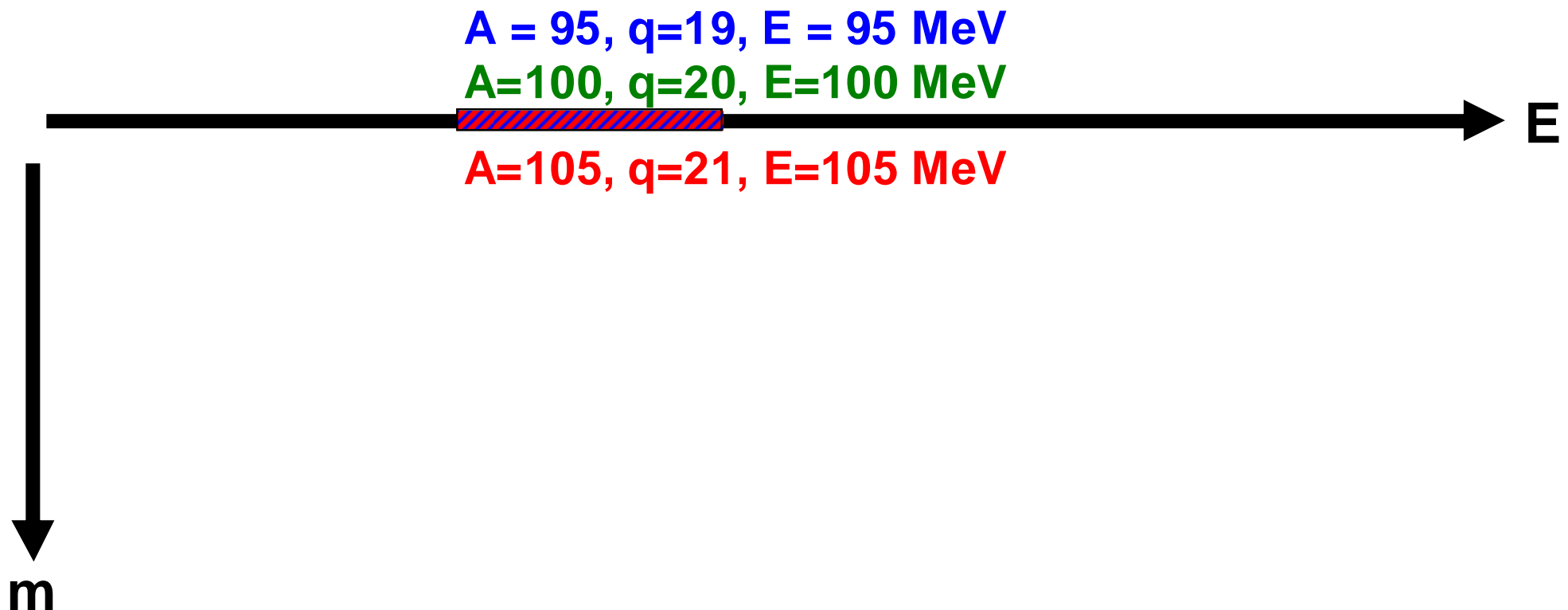
Mass resolution: $R_m = y_i / D_m = M_y \cdot y_o / D_m =$
 $= 1.0 \cdot 0.3 \text{ cm} / 3.24 \text{ m} \approx 1/1000$
depends on target width!



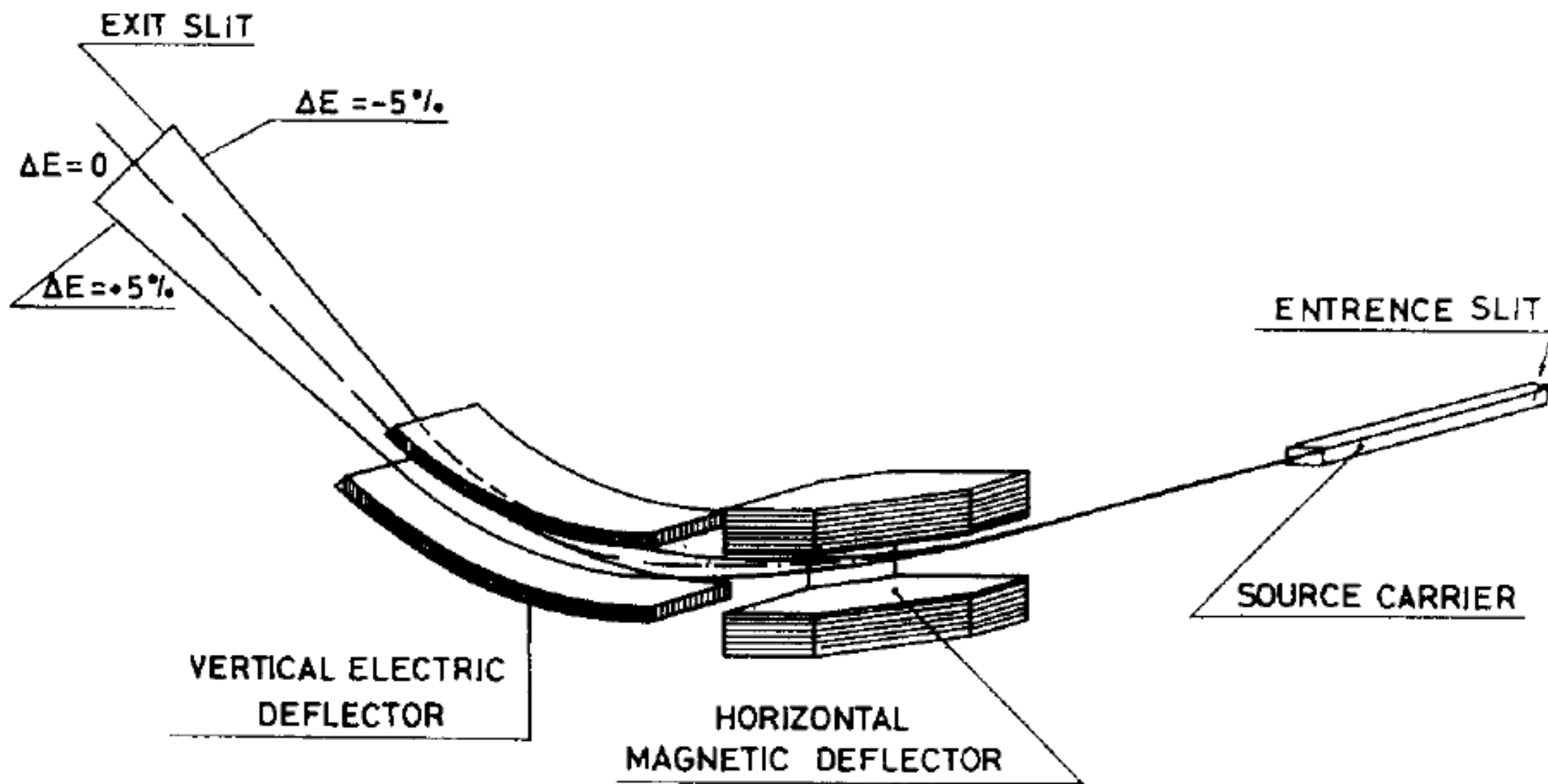
A/q separator

A “mass” separator is in reality an A/q separator and will mix masses with the same A/q and same E/q.

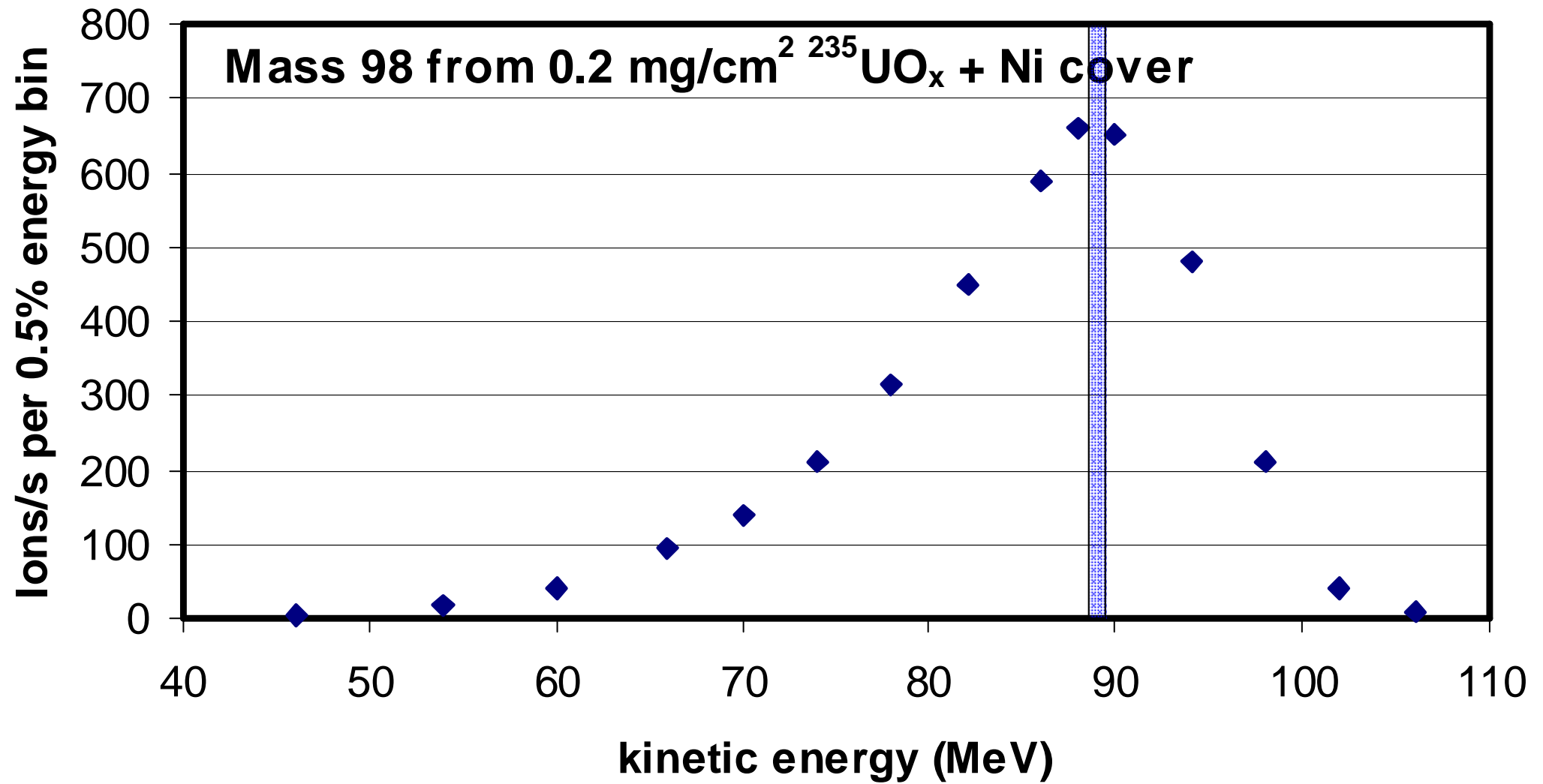
Avoid the use of A/q with (near-)integer ratios!



Focal plane of LOHENGRIN



Measured kinetic energy distribution



kinetic energy acceptance: $\Delta E/E = 1\%$

Reverse Energy Dispersion magnet

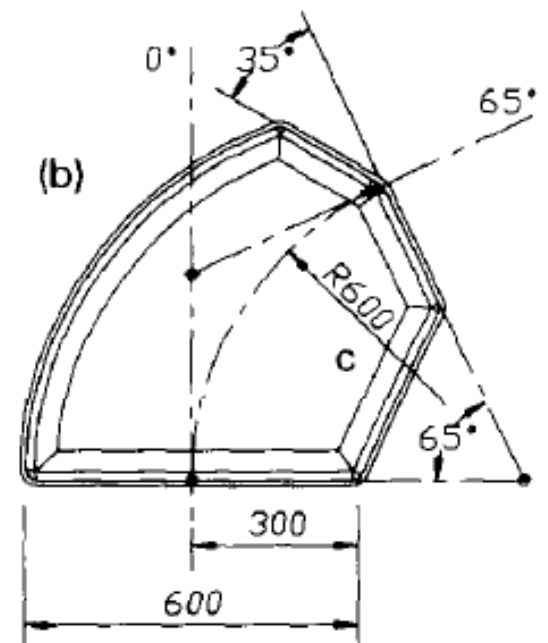
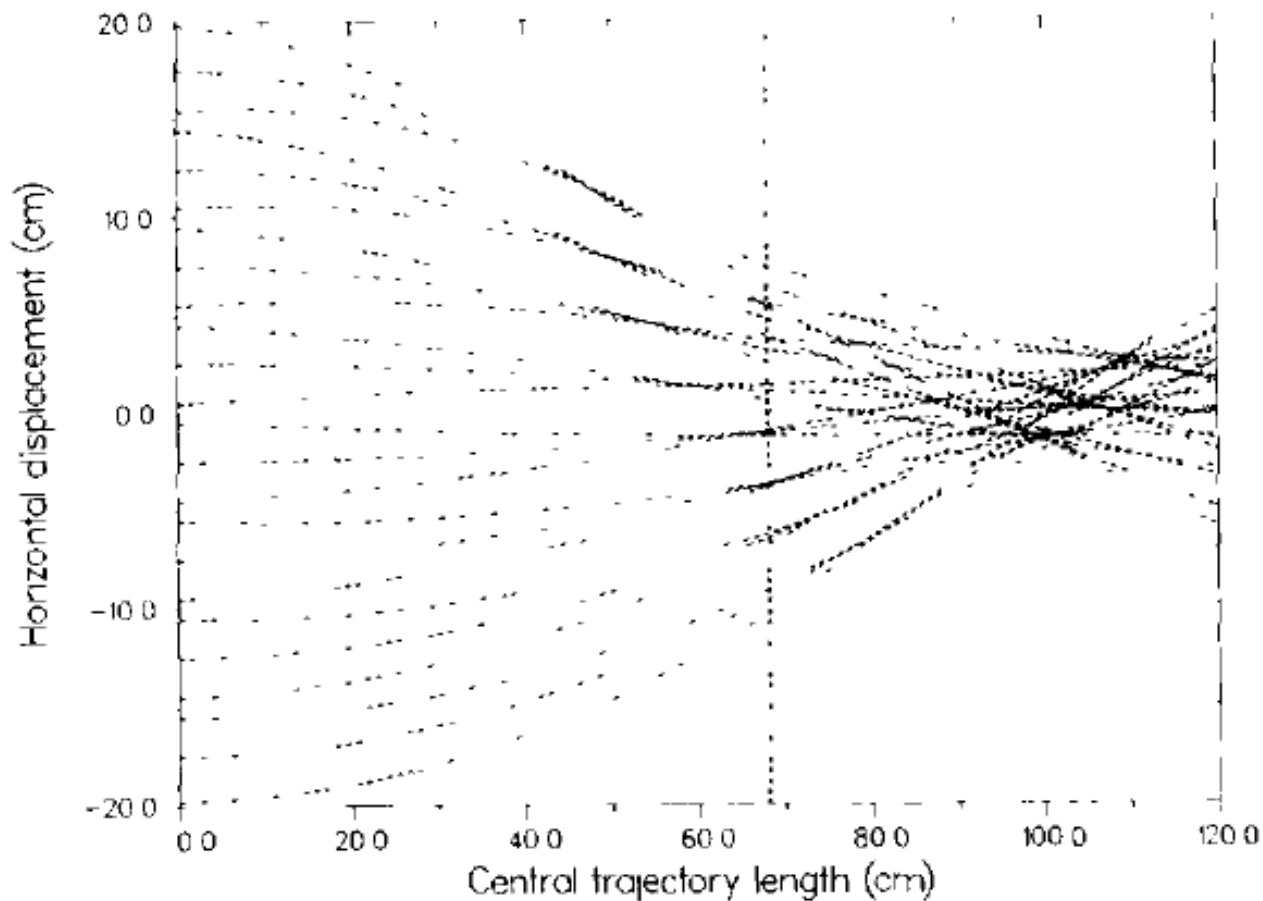
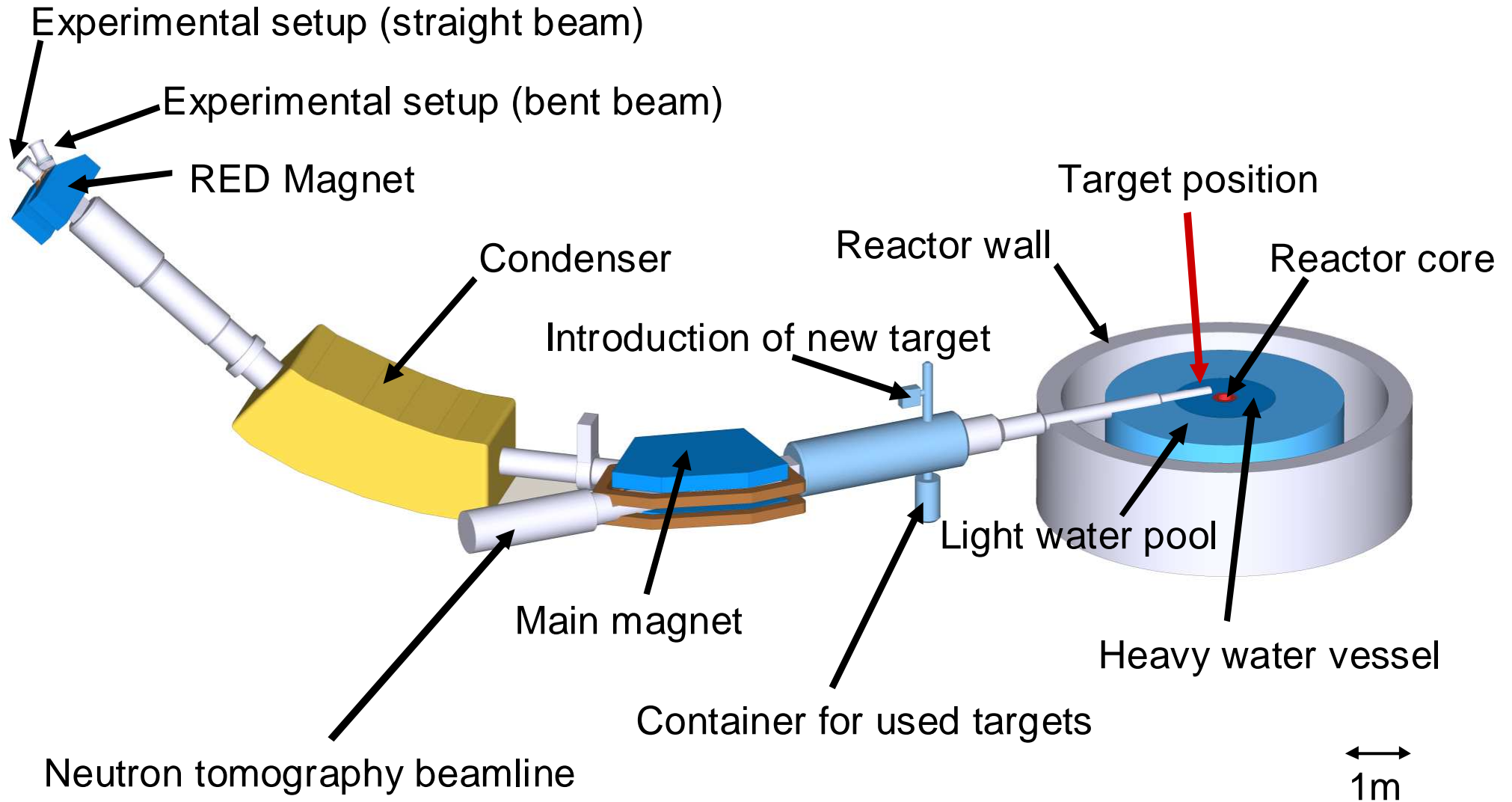
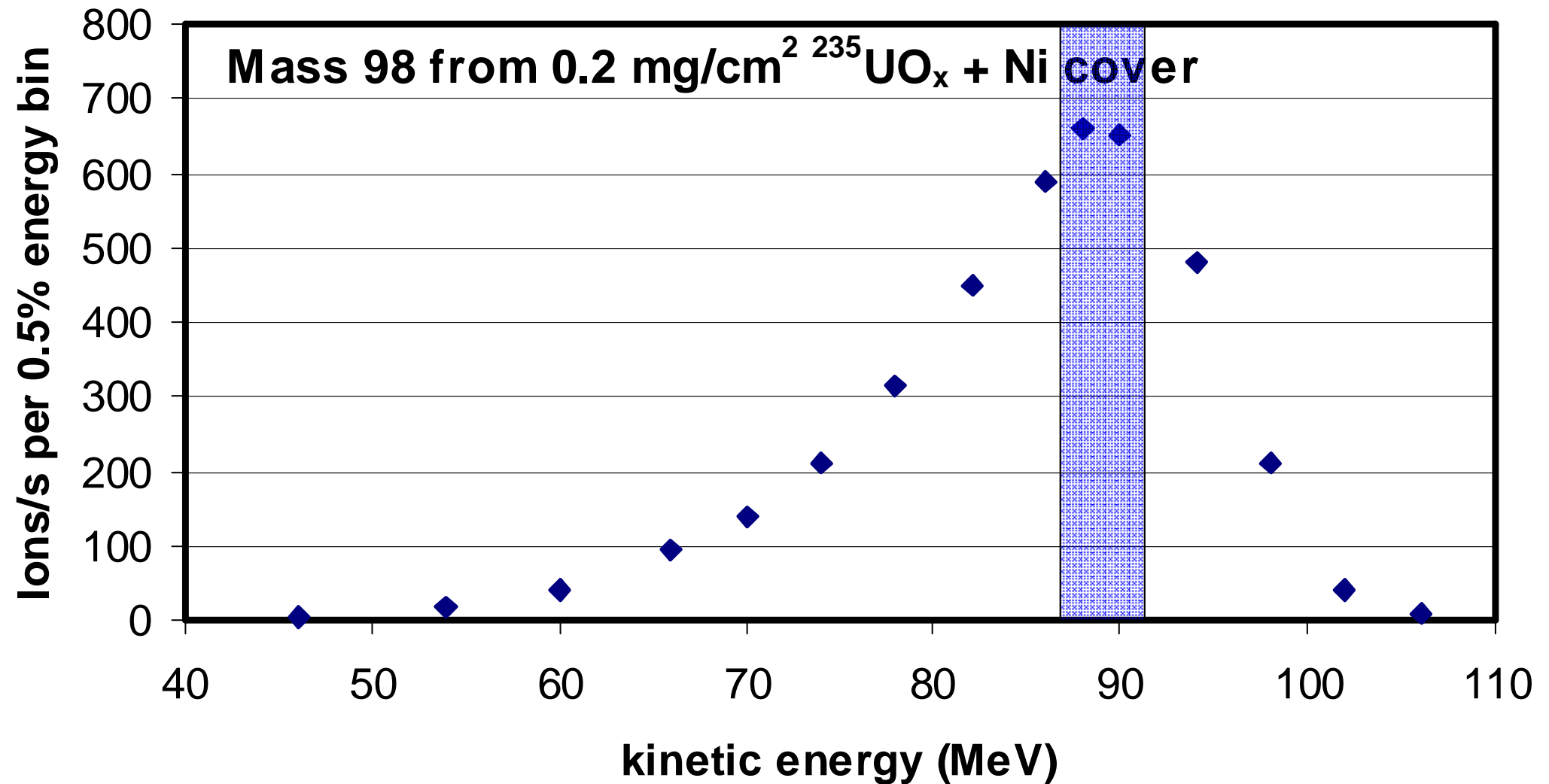


Fig. 5. Horizontal displacement with respect to the central trajectory of a beam arising from a $5 \times 70 \text{ mm}^2$ target vs the central trajectory length. The vertical dashed and dotted lines show respectively the extent of the pole pieces and the focal position.

LOHENGGRIN Setup



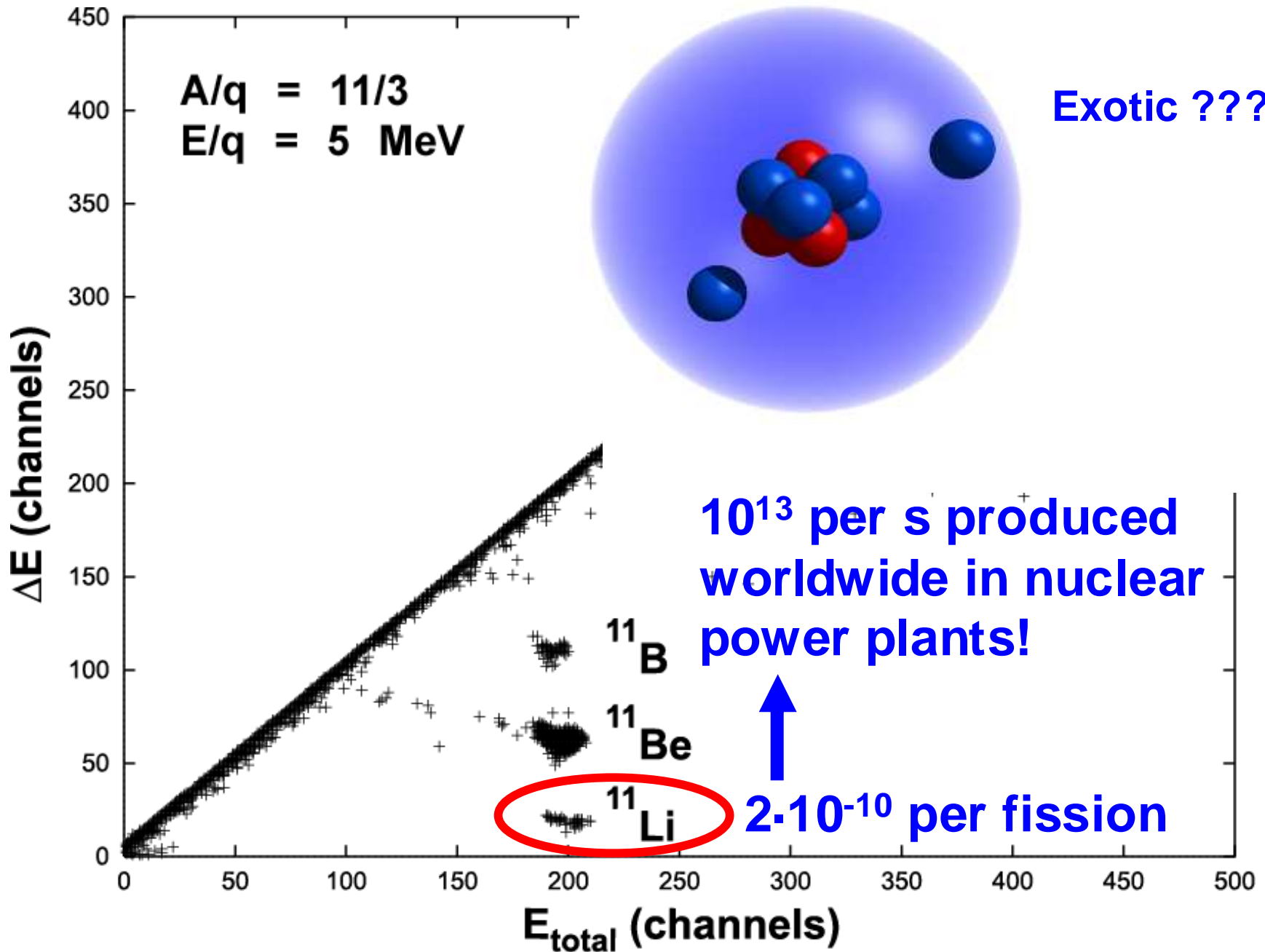
Measured kinetic energy distribution



kinetic energy acceptance: $\Delta E/E = 5\%$

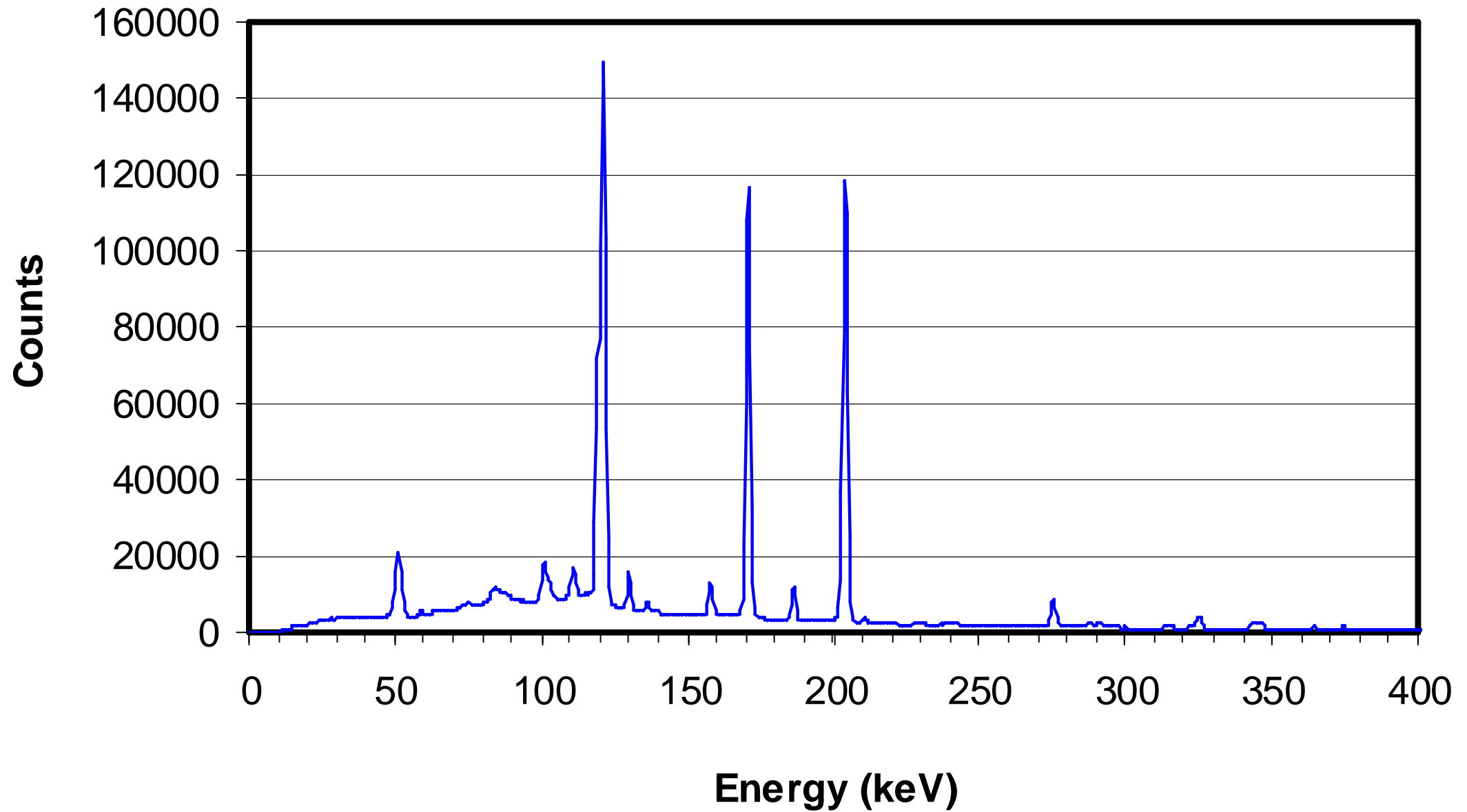
\Rightarrow 10-60% transmission (low for thick spectroscopy targets)

Detection of rare ternary particles

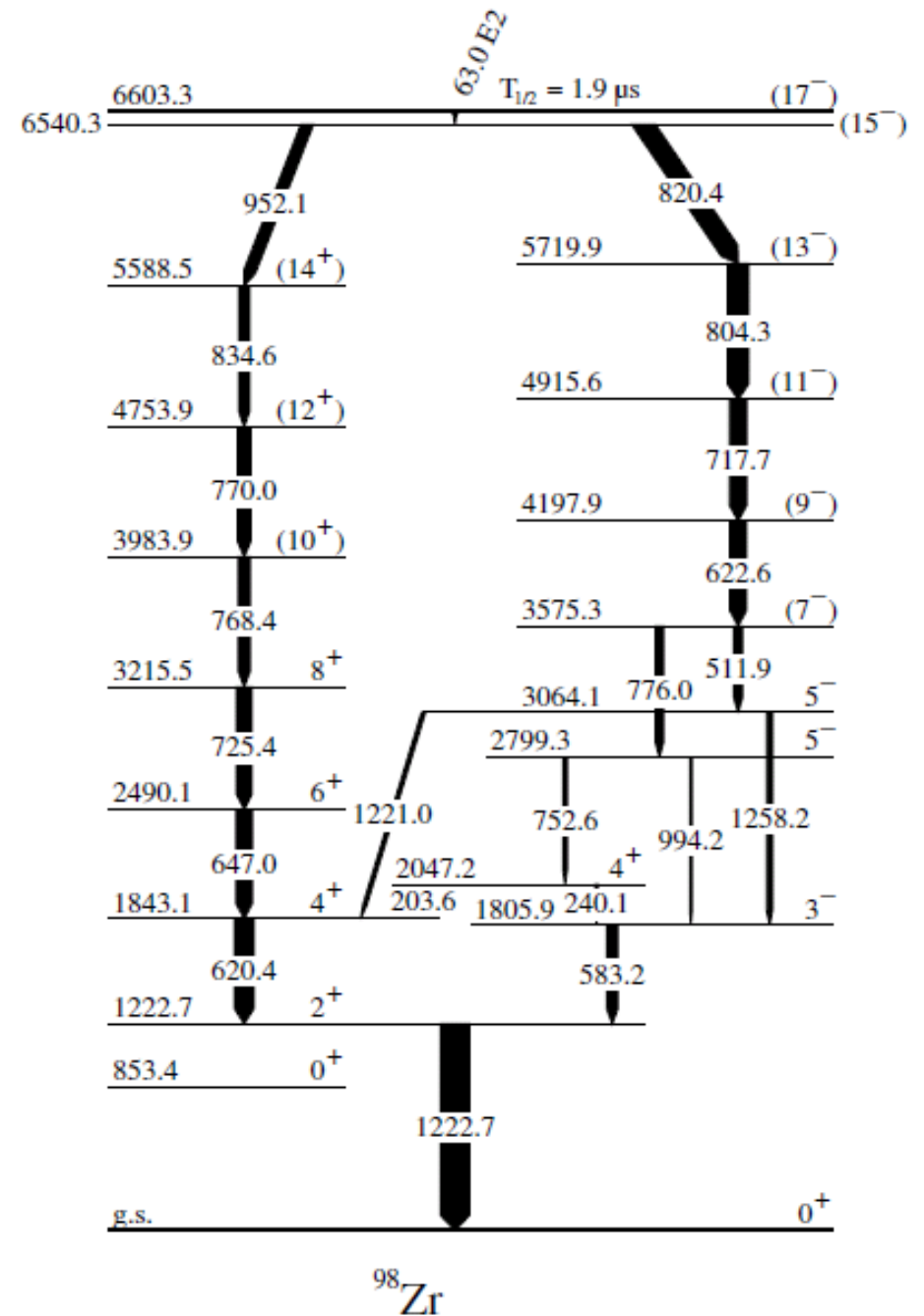
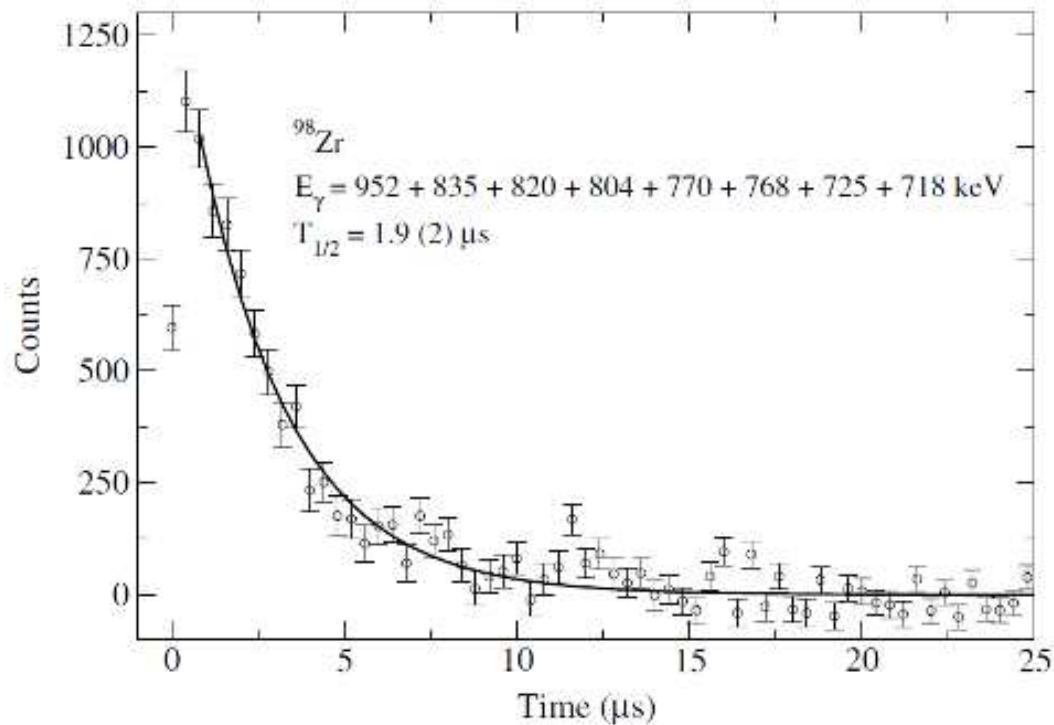
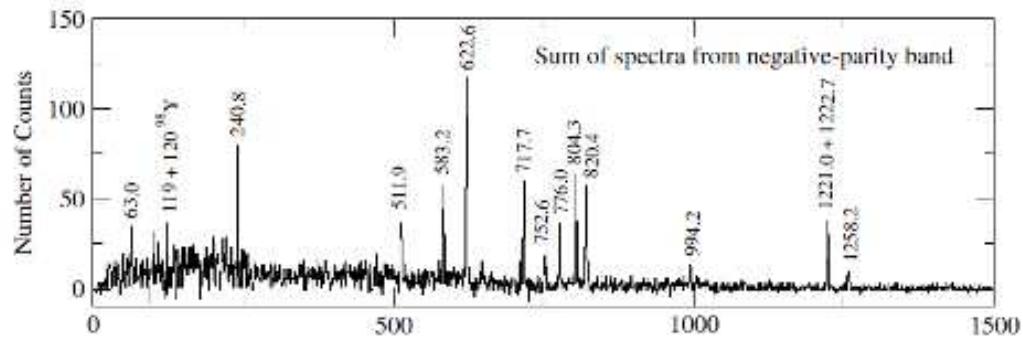
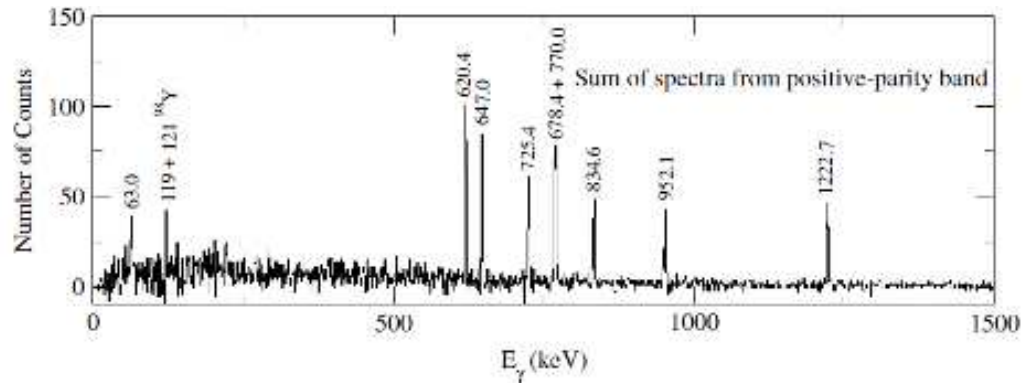




Gamma decay of 7.6 μs ^{98}Y isomer



17- isomer at 6.6 MeV in ^{98}Zr



G. Simpson et al.,
Phys. Rev. C 74 (2006) 064308.

Existing thick-target ISOL beams

Isotopes with $T_{1/2} < 0.1$ s separated																					
Isotopes with $T_{1/2} < 10$ s separated																					
Isotopes with $T_{1/2} > 10$ s separated																					
H 1																	He 2				
Li 3	Be 4															B 5	C 6	N 7	O 8	F 9	Ne 10
Na 11	Mg 12															Al 13	Si 14	P 15	S 16	Cl 17	Ar 18
K 19	Ca 20	Sc 21	Ti 22	V 23	Cr 24	Mn 25	Fe 26	Co 27	Ni 28	Cu 29	Zn 30	Ga 31	Ge 32	As 33	Se 34	Br 35	Kr 36				
Rb 37	Sr 38	Y 39	Zr 40	Nb 41	Mo 42	Tc 43	Ru 44	Rh 45	Pd 46	Ag 47	Cd 48	In 49	Sn 50	Sb 51	Te 52	I 53	Xe 54				
Cs 55	Ba 56	La 57	Hf 72	Ta 73	W 74	Re 75	Os 76	Ir 77	Pt 78	Au 79	Hg 80	Tl 81	Pb 82	Bi 83	Po 84	At 85	Rn 86				
Fr 87	Ra 88	Ac 89	Rf 104	Db 105	Sg 106	Bh 107	Hs 108	Mt 109	Ds 110	Rg 111	112	113	114	115	116		118				

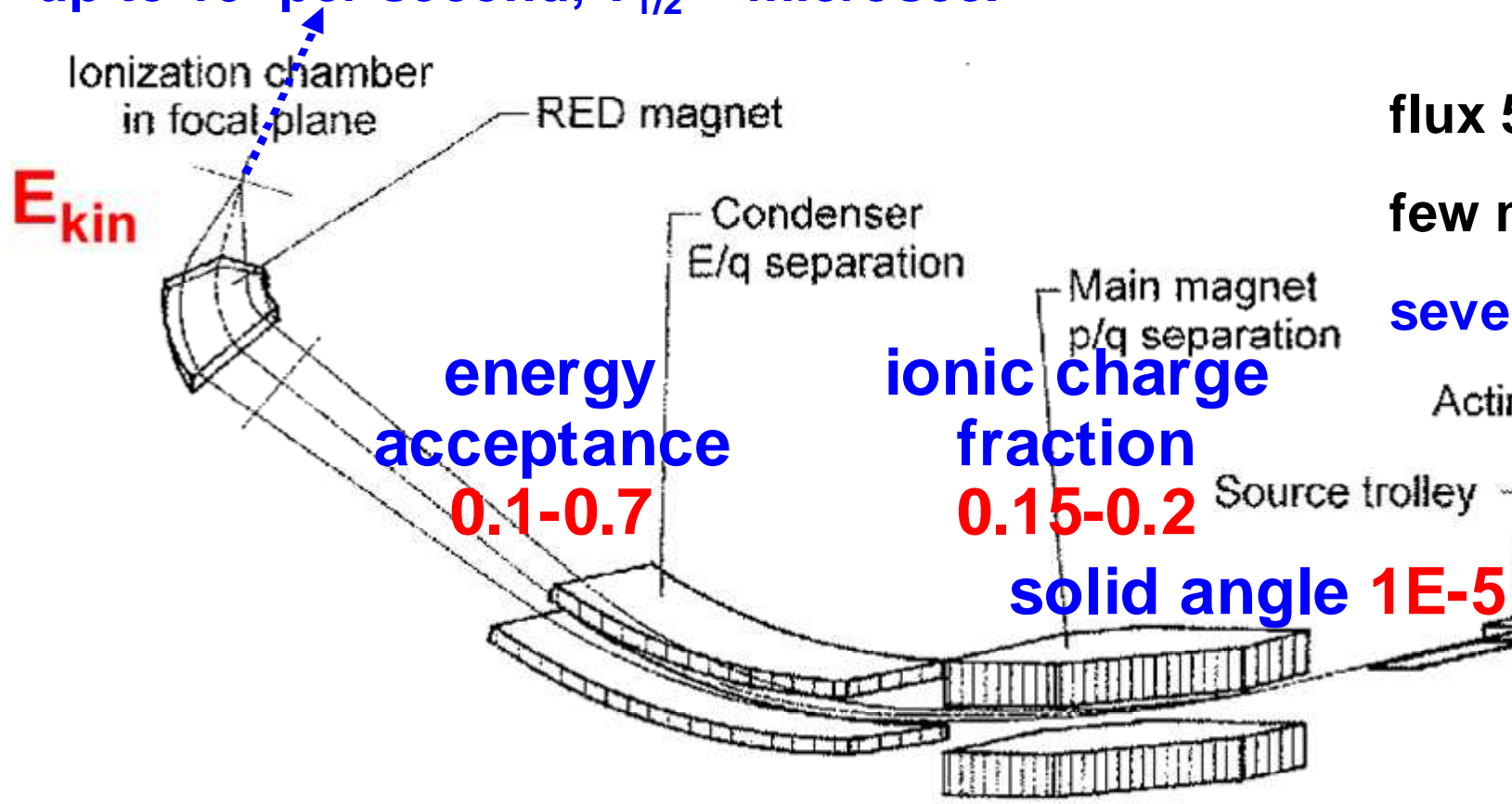
58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

The LOHENGRIN fission fragment separator

mass-separated fission fragments,
up to 10^5 per second, $T_{1/2} \geq$ microsec.

$$\Delta A/A = 3E-4 - 3E-3$$

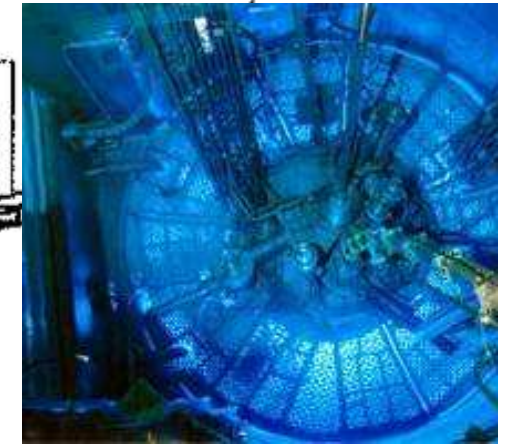
$$\Delta E/E = 1E-3 - 1E-2$$



flux $5.5 \cdot 10^{14}$ n./cm²/s

few mg fission target

several 10^{12} fissions/s



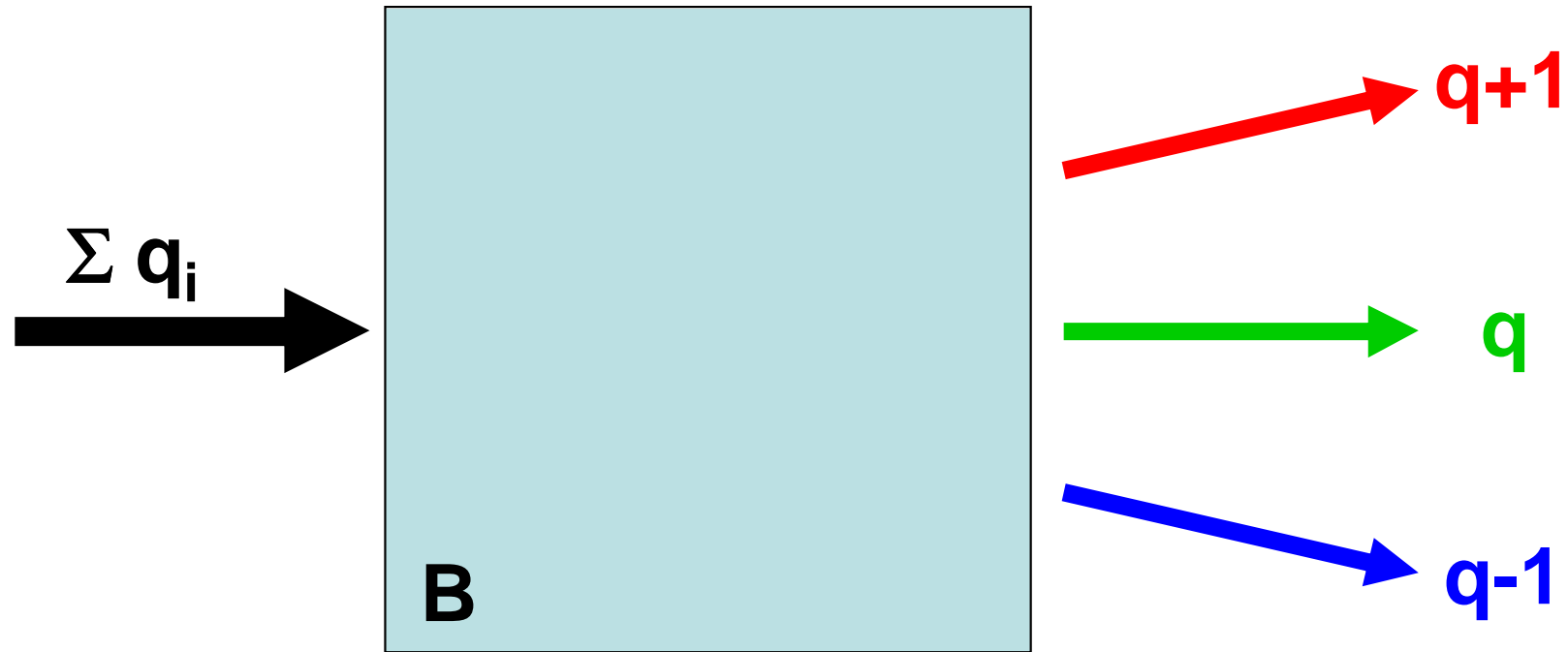
$$m v^2 / r_{el} = q E$$

$$E_{kin} / q = E / 2 r_{el}$$

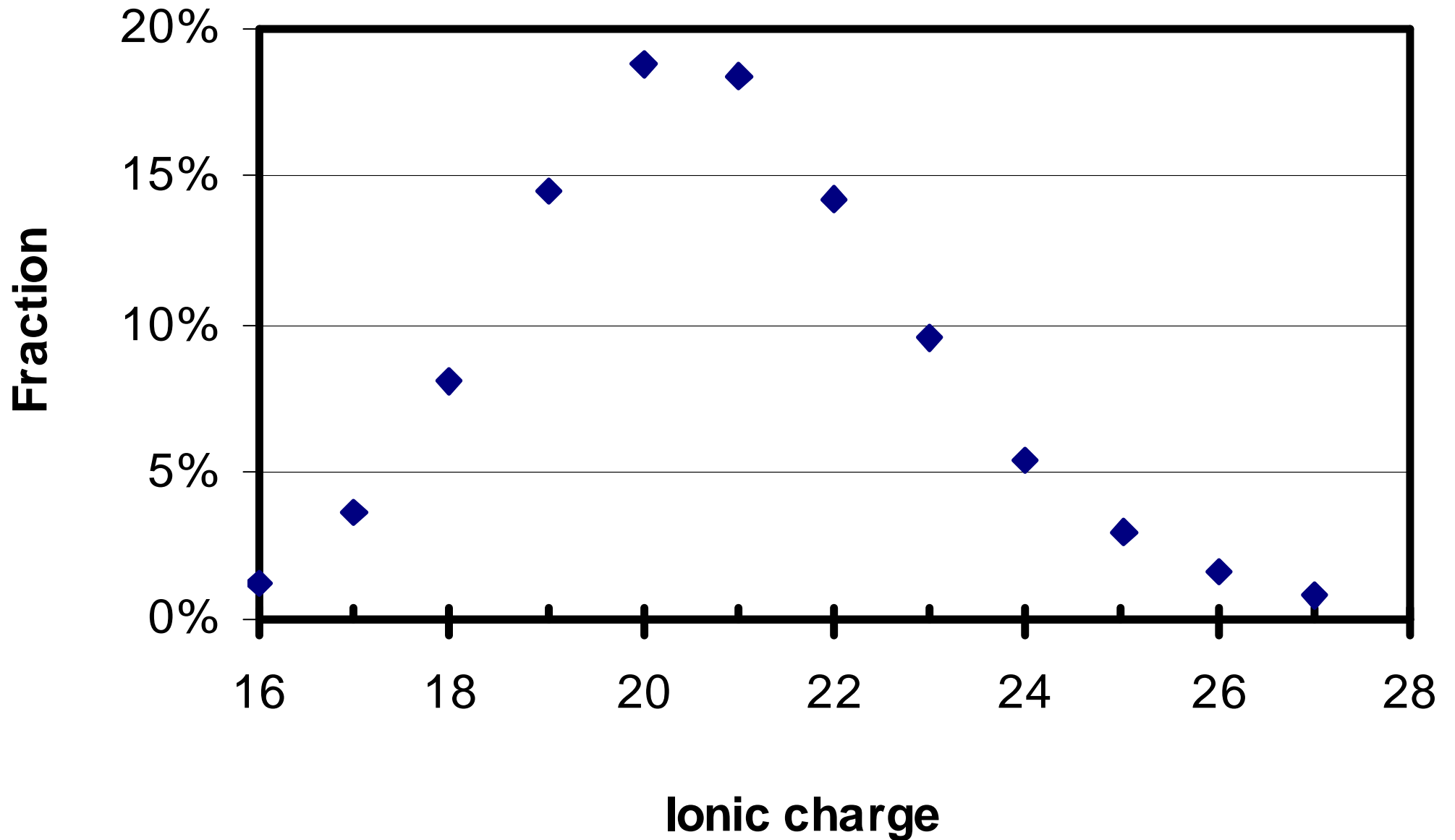
$$m v^2 / r_{magn} = q v B$$

$$m v / q = B r_{magn}$$

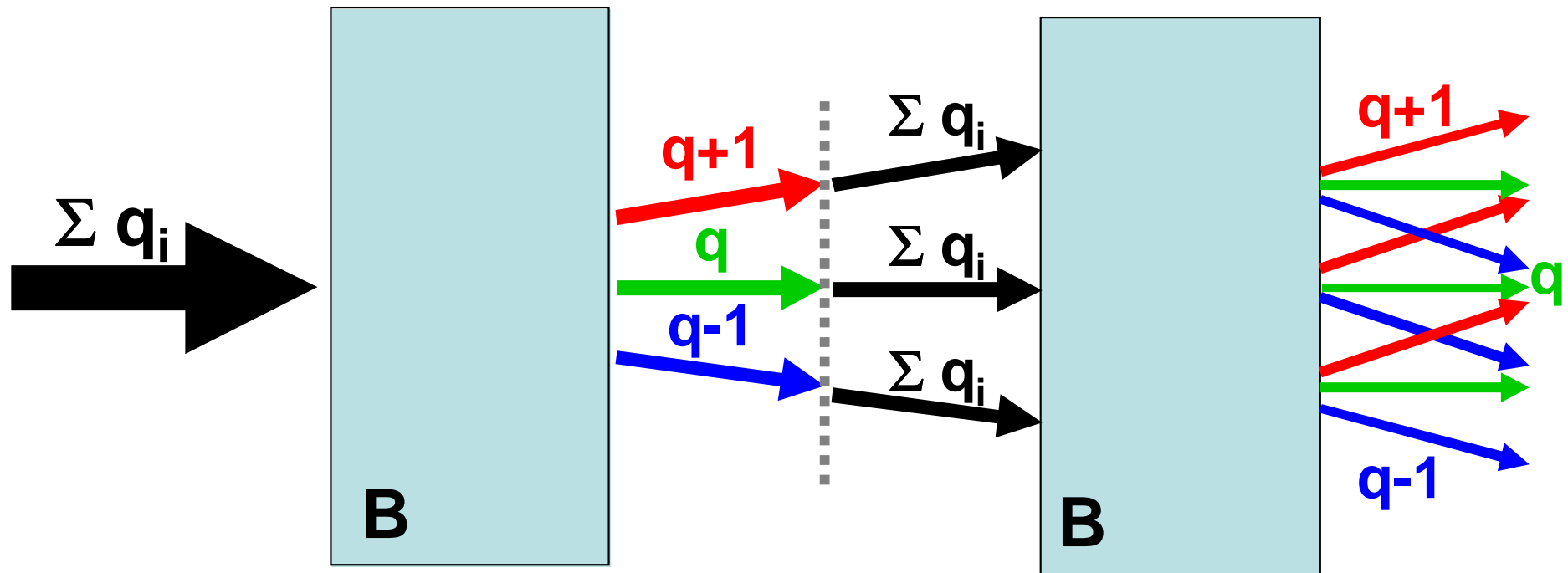
Ionic charge separation



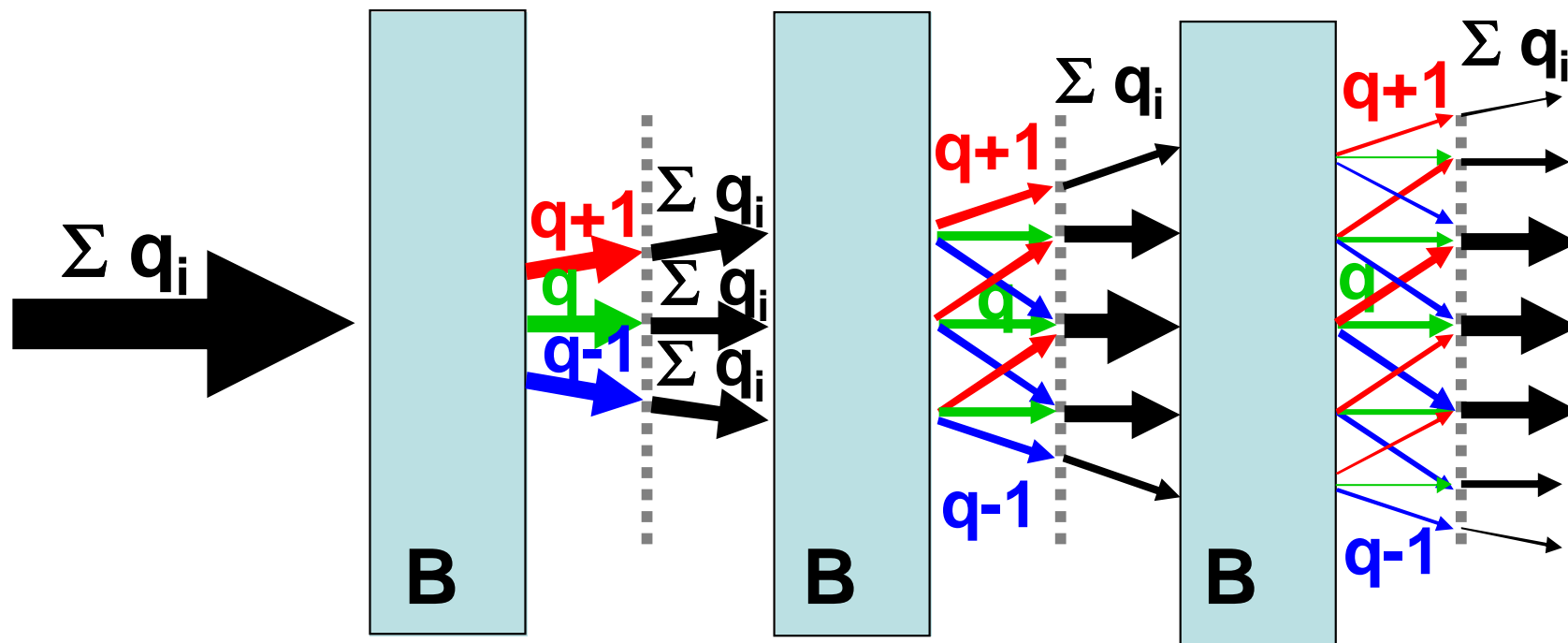
Ionic charge state distribution



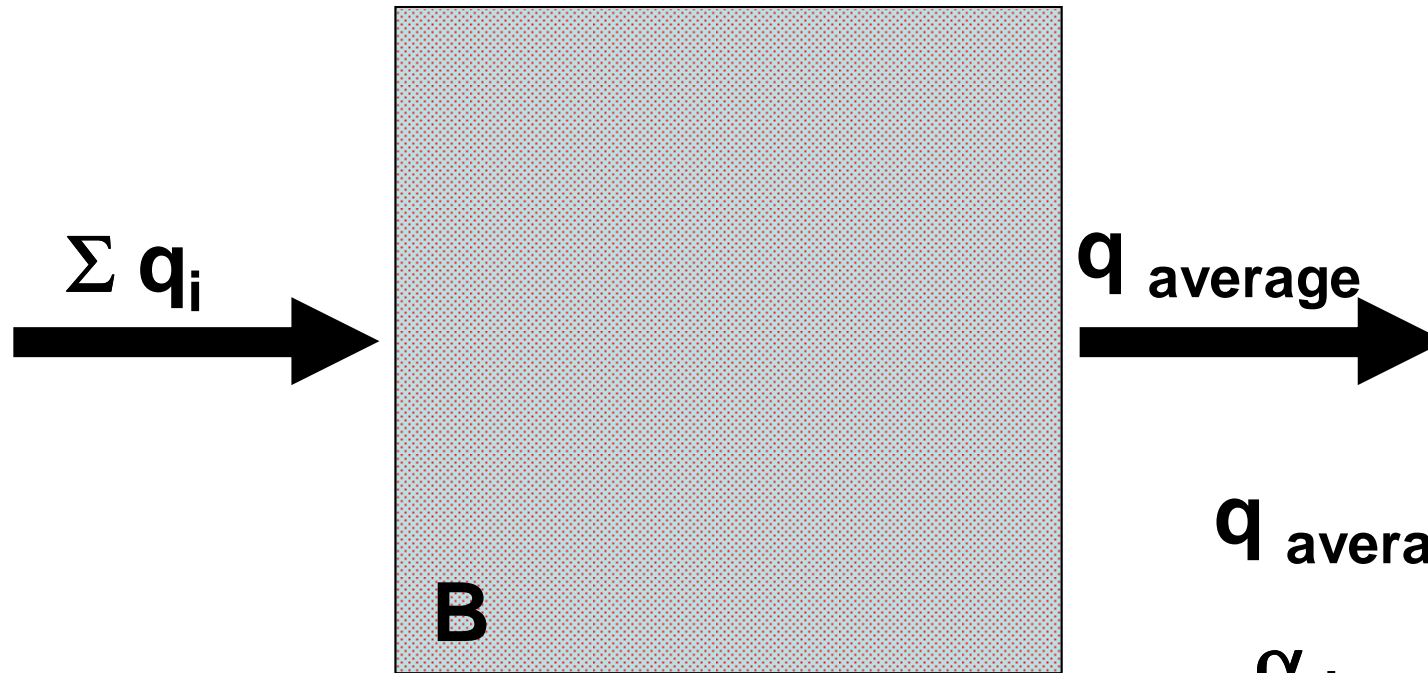
Ionic charge separation



Ionic charge separation



Separation with gas-filled magnet



He or N₂ at few mbar

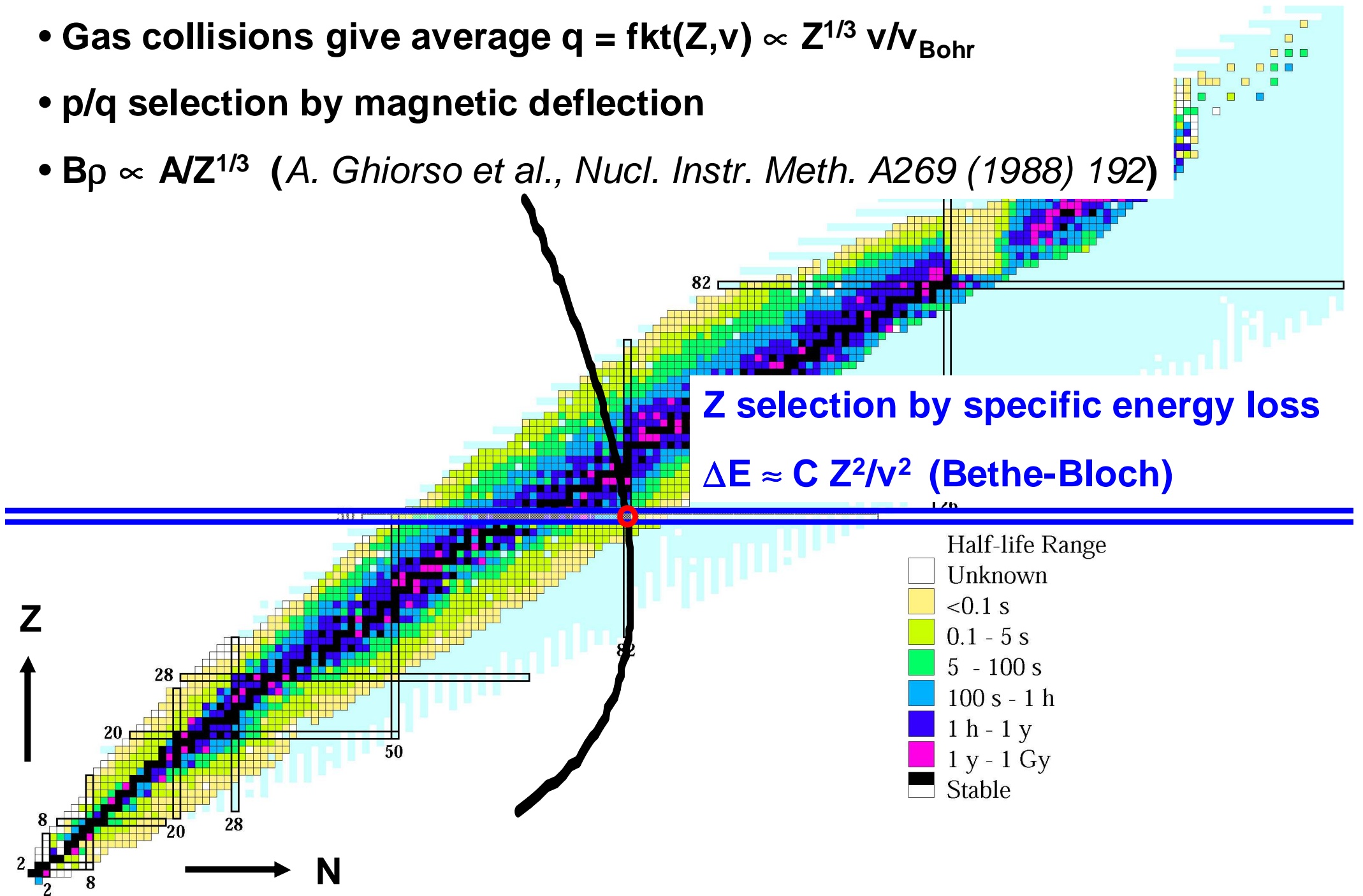
$$q_{\text{average}} = v/v_{\text{Bohr}} Z^{\alpha}$$

$$\alpha_{\text{theo}} = 1/3 \text{ (Bohr)}$$

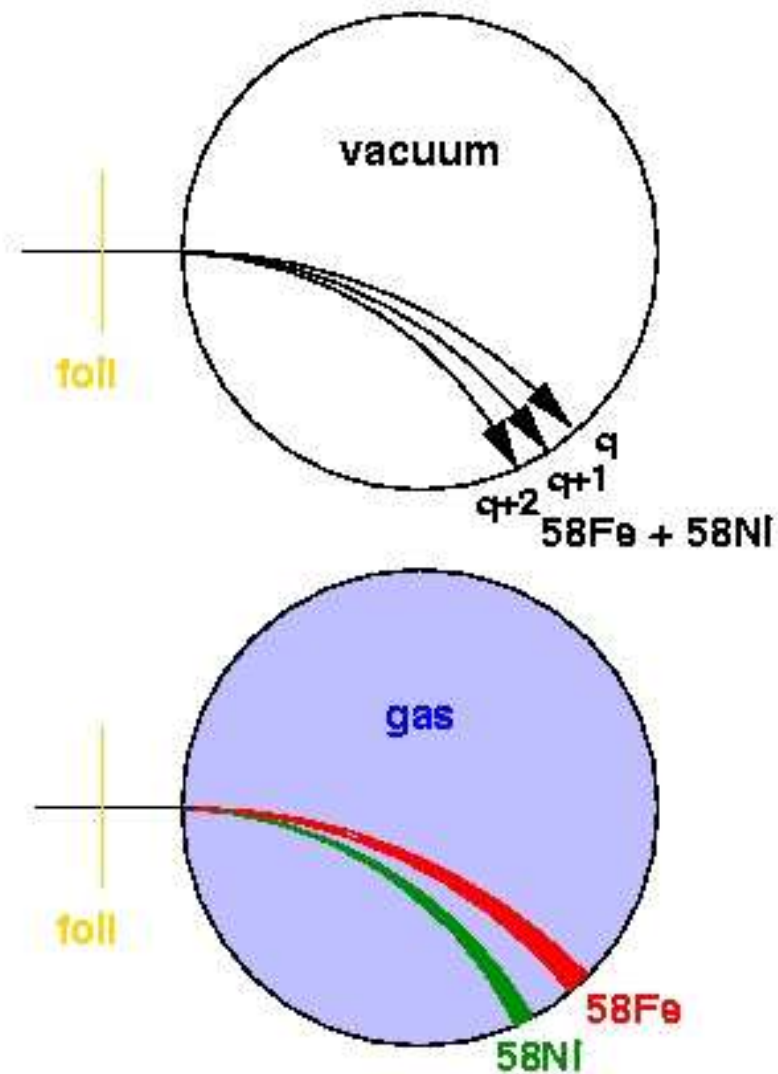
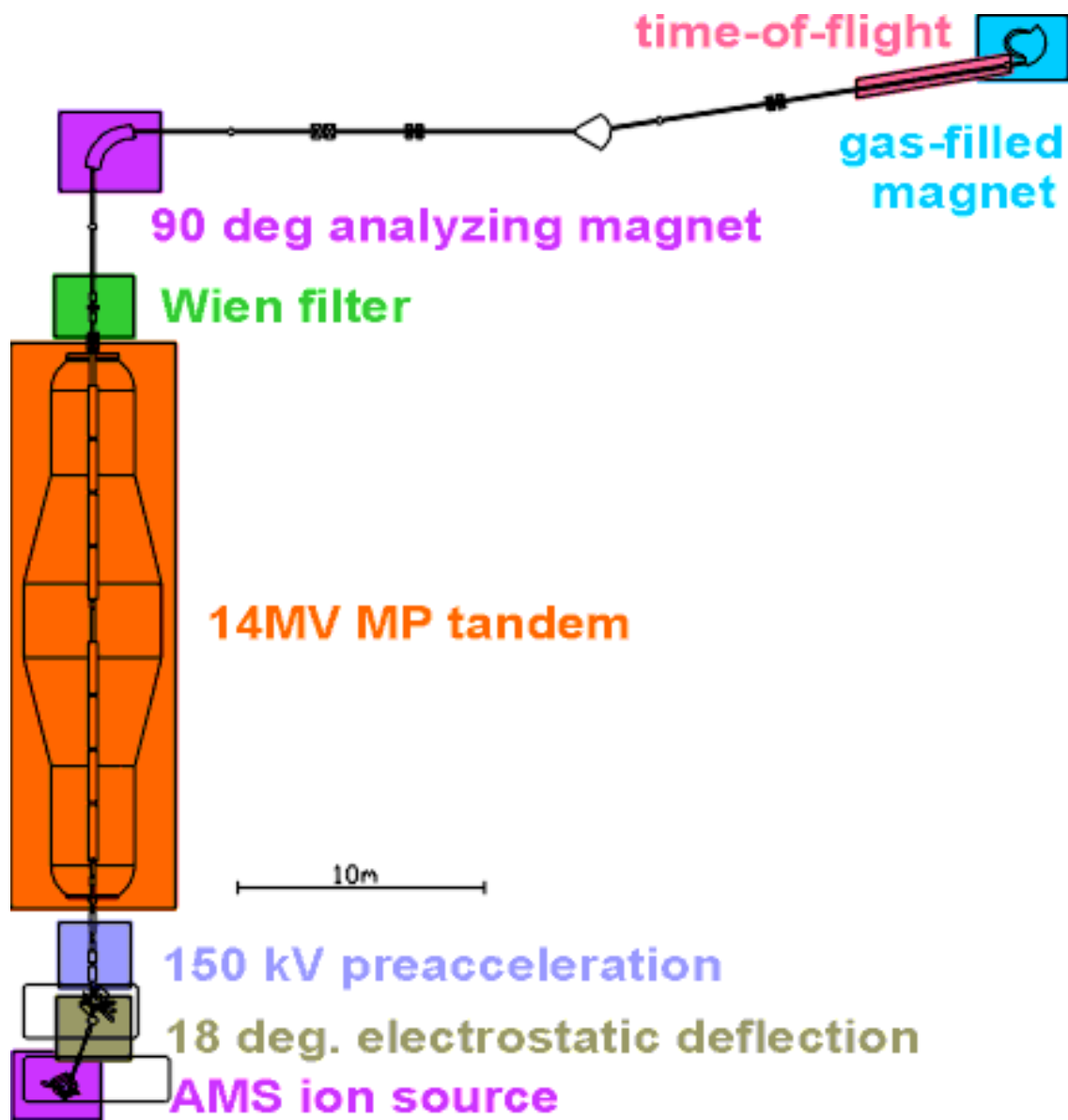
$$\alpha_{\text{exp}} = 0.33 - 0.7$$

Isotope selection with gas-filled separators

- Gas collisions give average $q = fkt(Z,v) \propto Z^{1/3} v/v_{\text{Bohr}}$
- p/q selection by magnetic deflection
- $B\rho \propto AZ^{1/3}$ (A. Ghiorso et al., Nucl. Instr. Meth. A269 (1988) 192)



Multistep-Separation in Accelerator Mass Spectrometry



10^{-15} sensitivity!

From ISOL beams to RIBs with higher energies

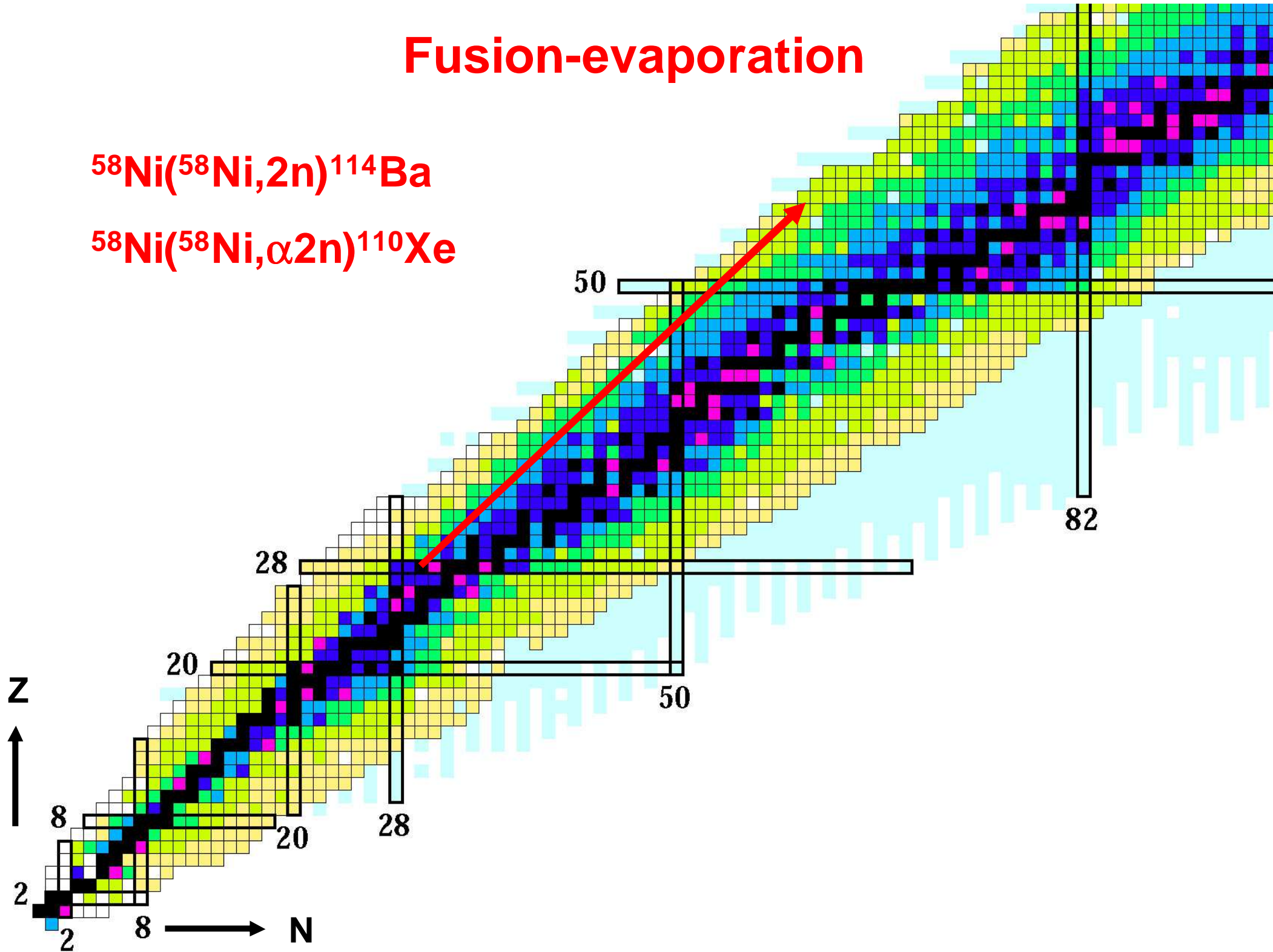
ISOL beams

- have well-defined energy ($\Delta E/E \approx 1 \text{ eV}/60 \text{ keV}$)
- have usually small emittance (e.g. $10 \pi \text{ mm mrad}$), i.e. limited opening angle
- have often well-defined ionic charge $q=1$
- Z selection is performed before the mass separator

Recoils or fragments of nuclear reactions:

- have large energy spread
- large angular spread
- different ionic charge states
- depending on nuclear reaction different Z

Fusion-evaporation

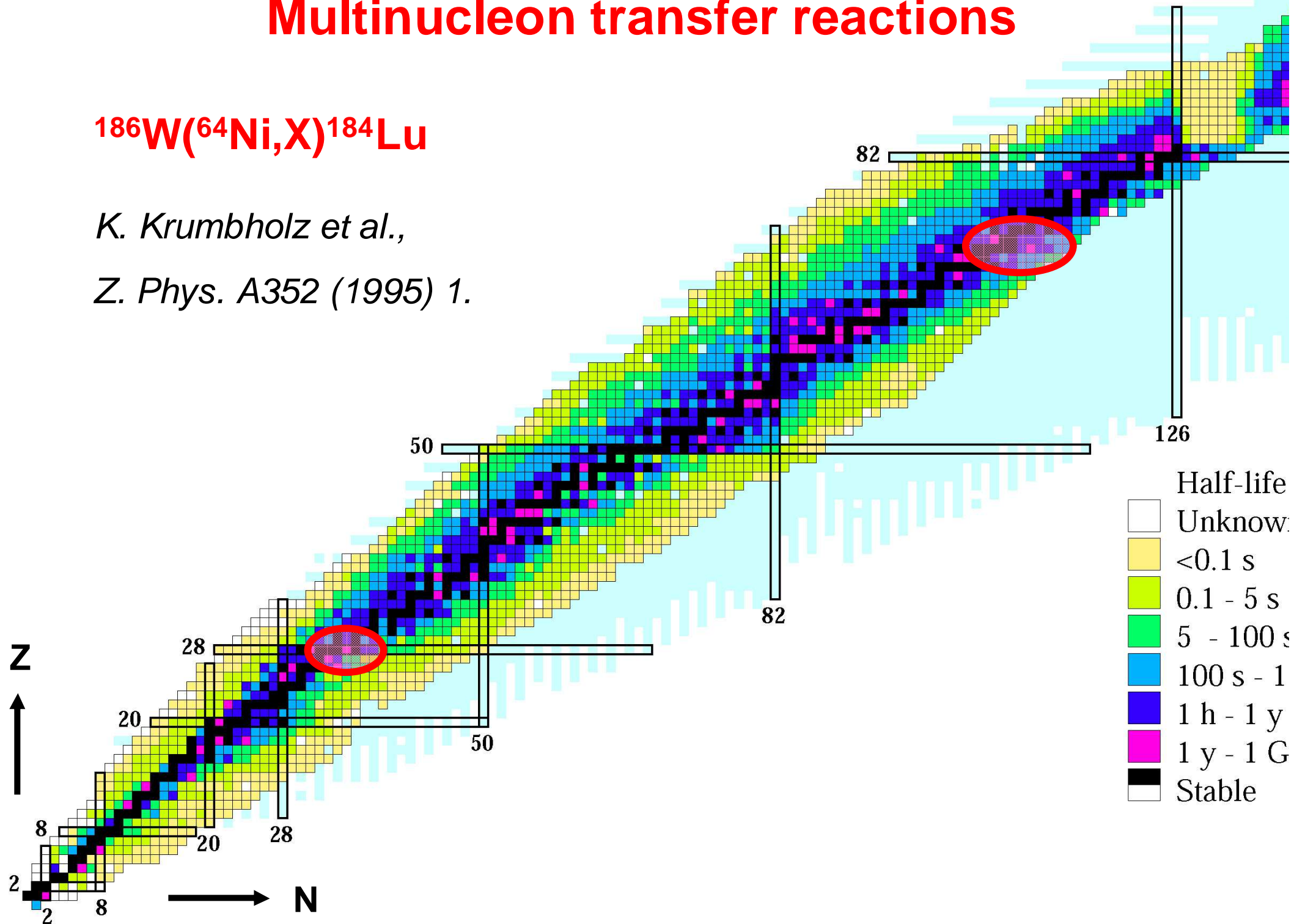


Multinucleon transfer reactions



K. Krumbholz et al.,

Z. Phys. A352 (1995) 1.



Requirements for in-flight separators

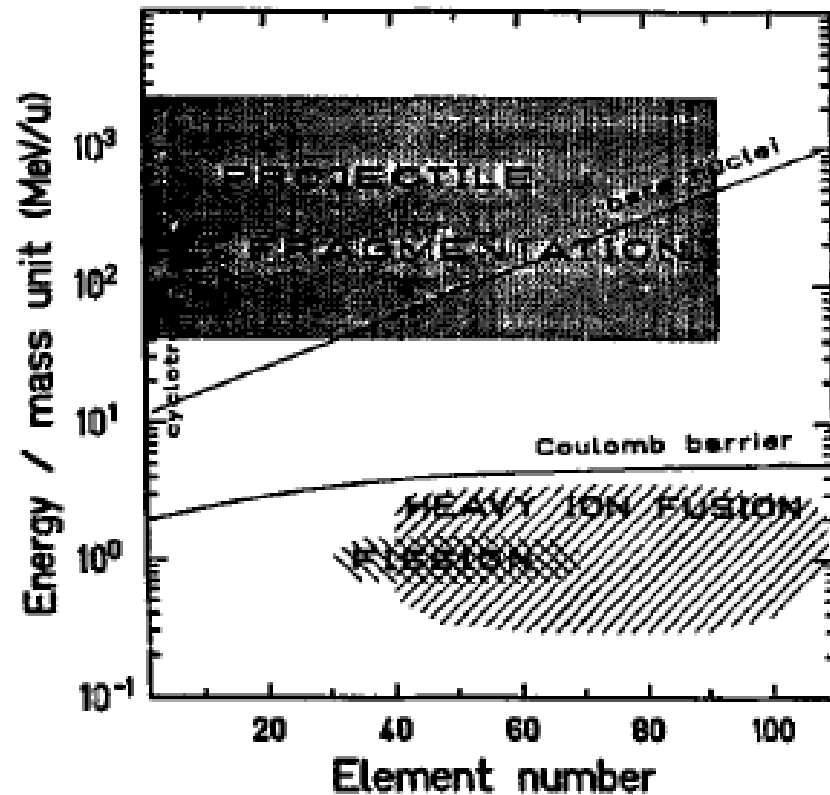


Fig. 2. Domains of kinetic energies of the reaction products from various nuclear reactions.

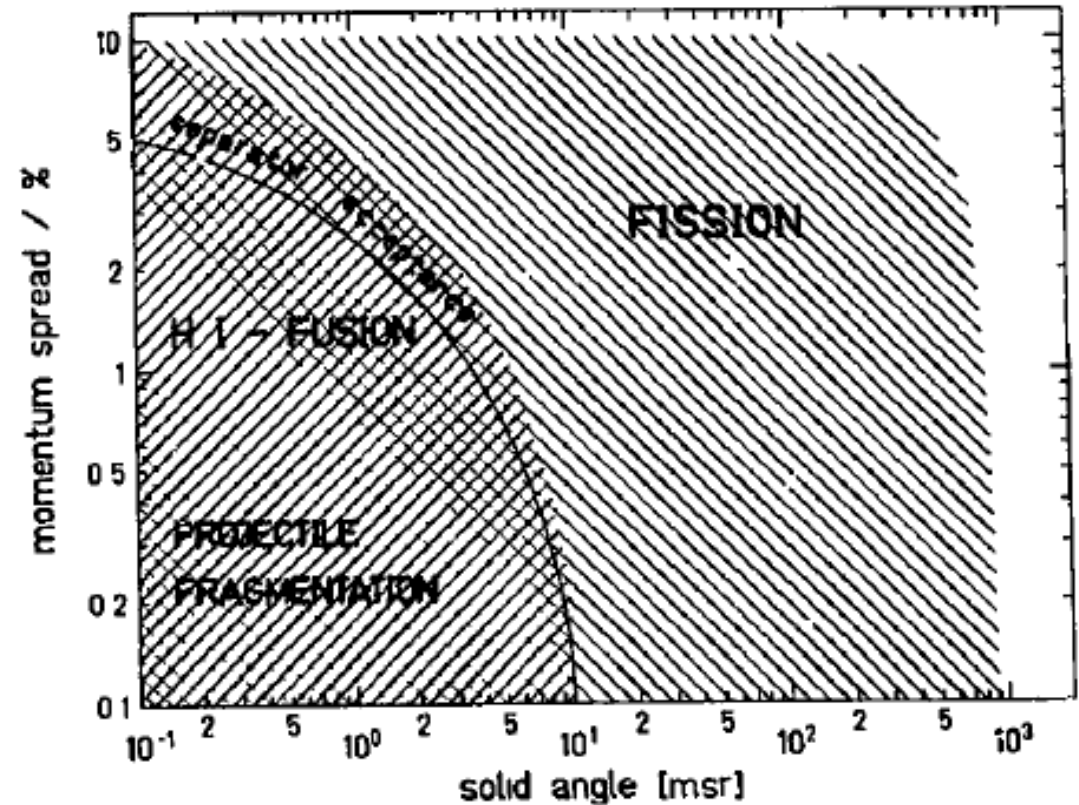
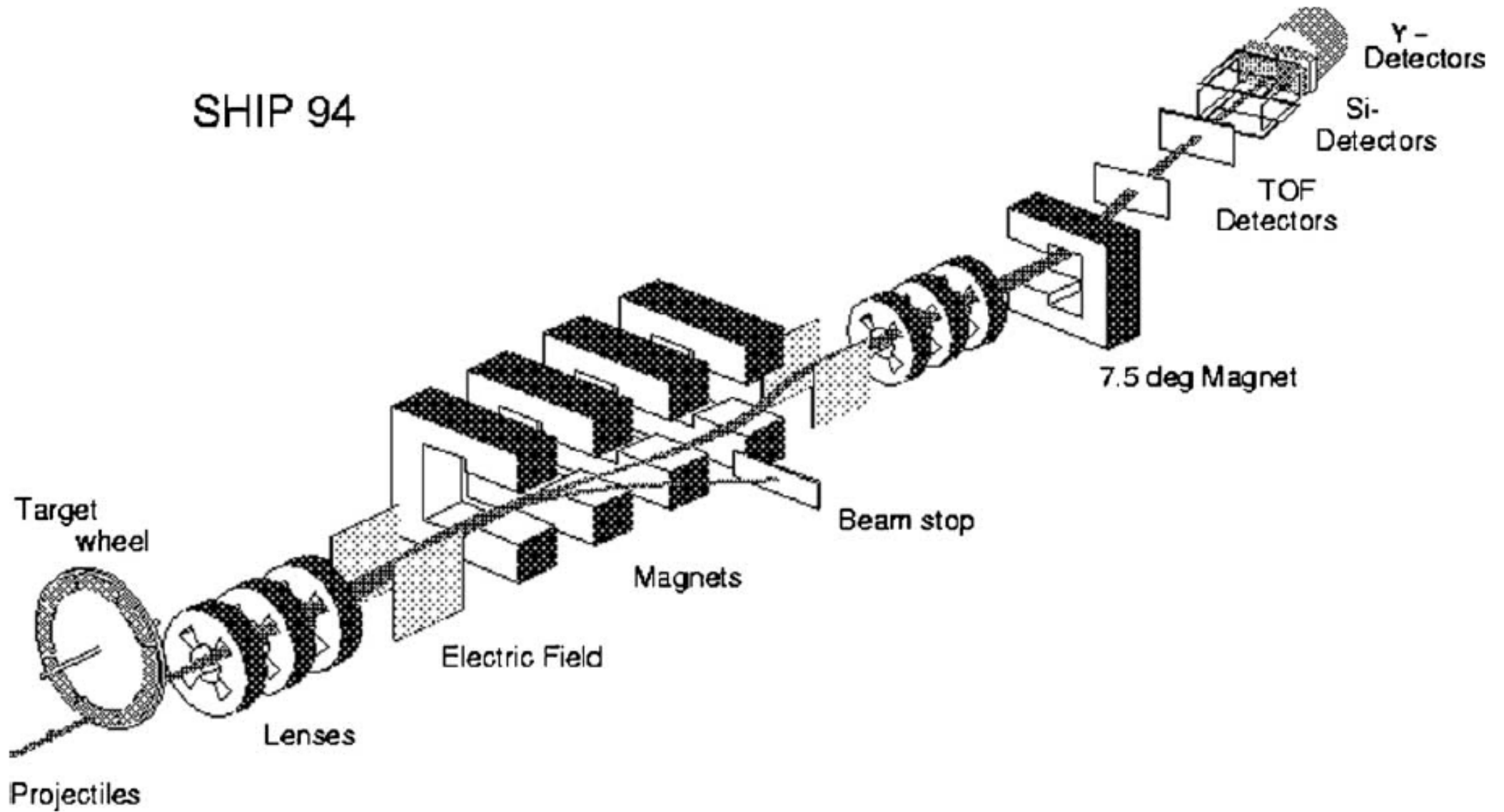


Fig. 3. Solid angle and momentum spread of the reaction products compared to the acceptances of spectrometers.

Recoil separators

- **separate the products of a nuclear reaction (recoils) from the projectile beam**
- **early dumping of unwanted beam**
- **optionally also A/q separation of reaction products**
- **usually kinetic energies up to 10 MeV/nucleon**
- **mass dispersion achieved by combination of magnetic dipoles, electric dipoles or Wien filter**
- **usually additional quadrupoles for focusing**

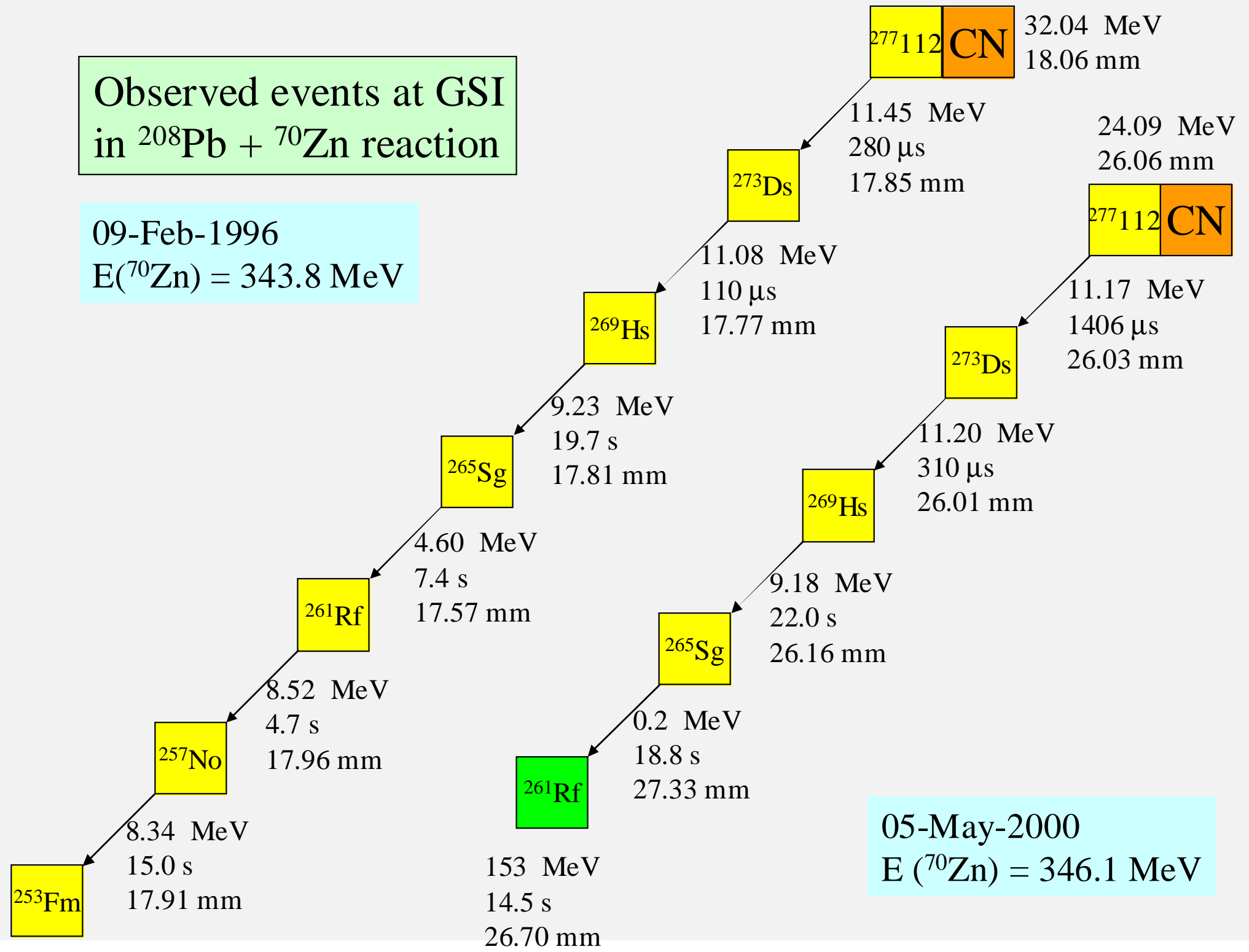
SHIP at GSI Darmstadt



G. Münzenberg et al., Nucl. Instr. Meth. 141 (1979) 65.

**Observed events at GSI
in $^{208}\text{Pb} + ^{70}\text{Zn}$ reaction**

09-Feb-1996
 $E(^{70}\text{Zn}) = 343.8 \text{ MeV}$



05-May-2000
 $E(^{70}\text{Zn}) = 346.1 \text{ MeV}$

VASSILISSA

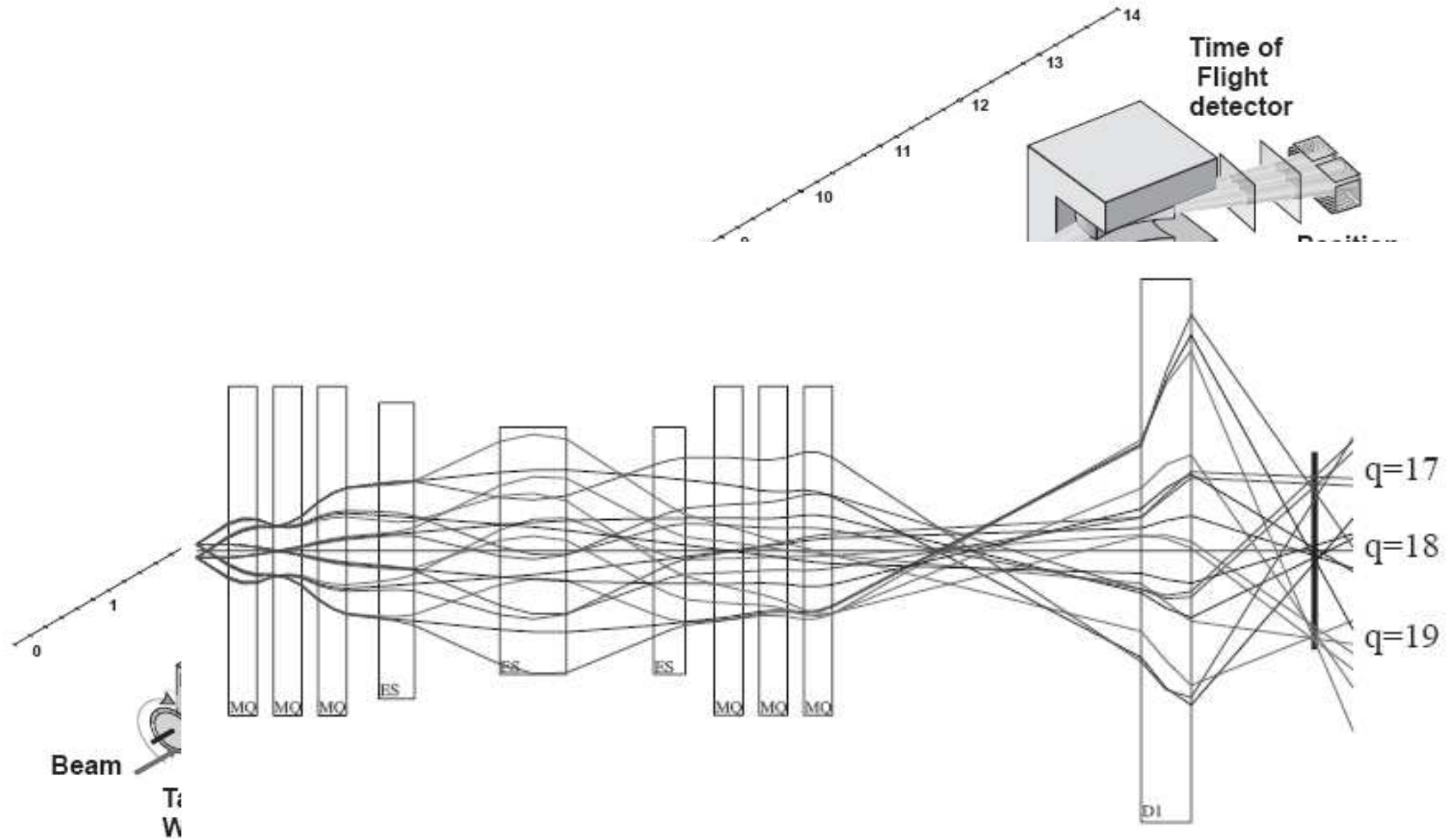
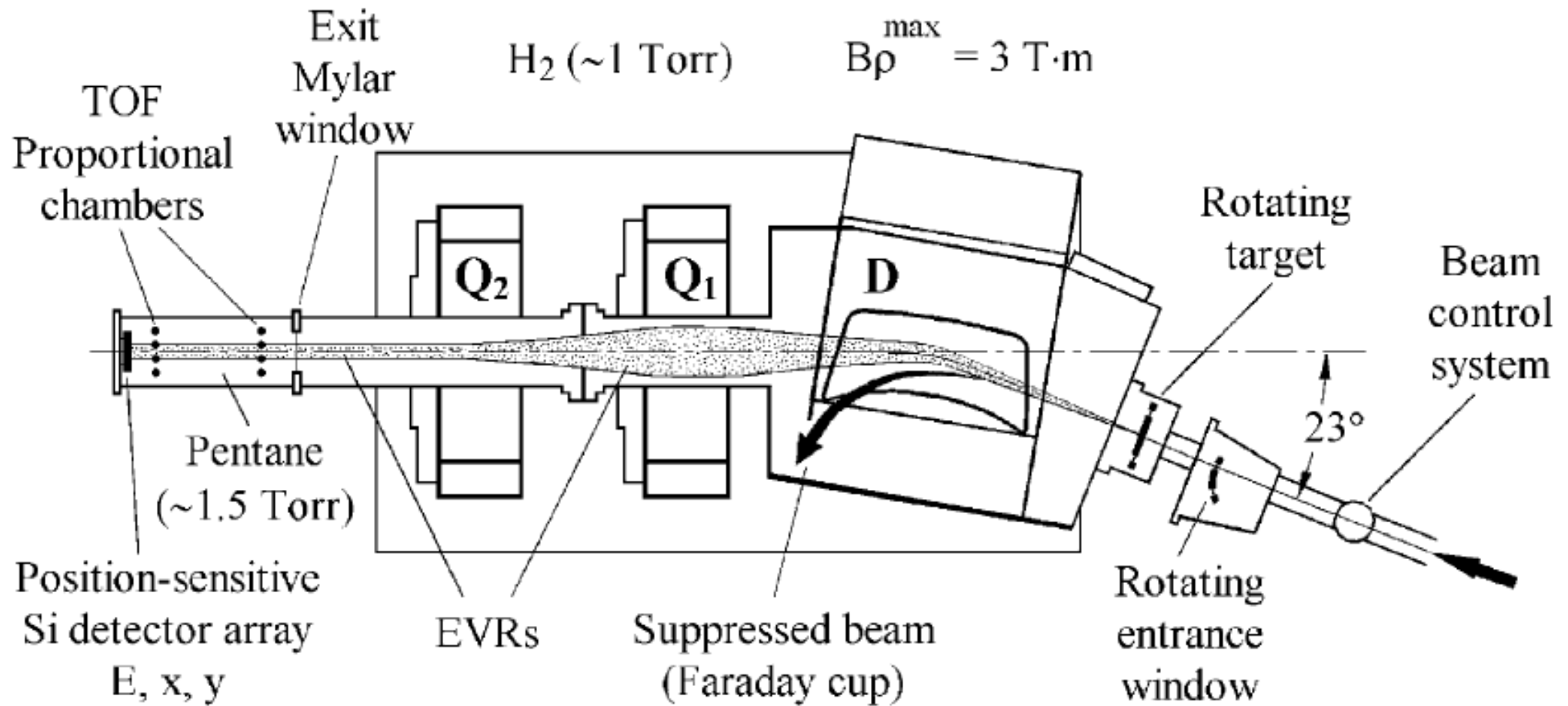


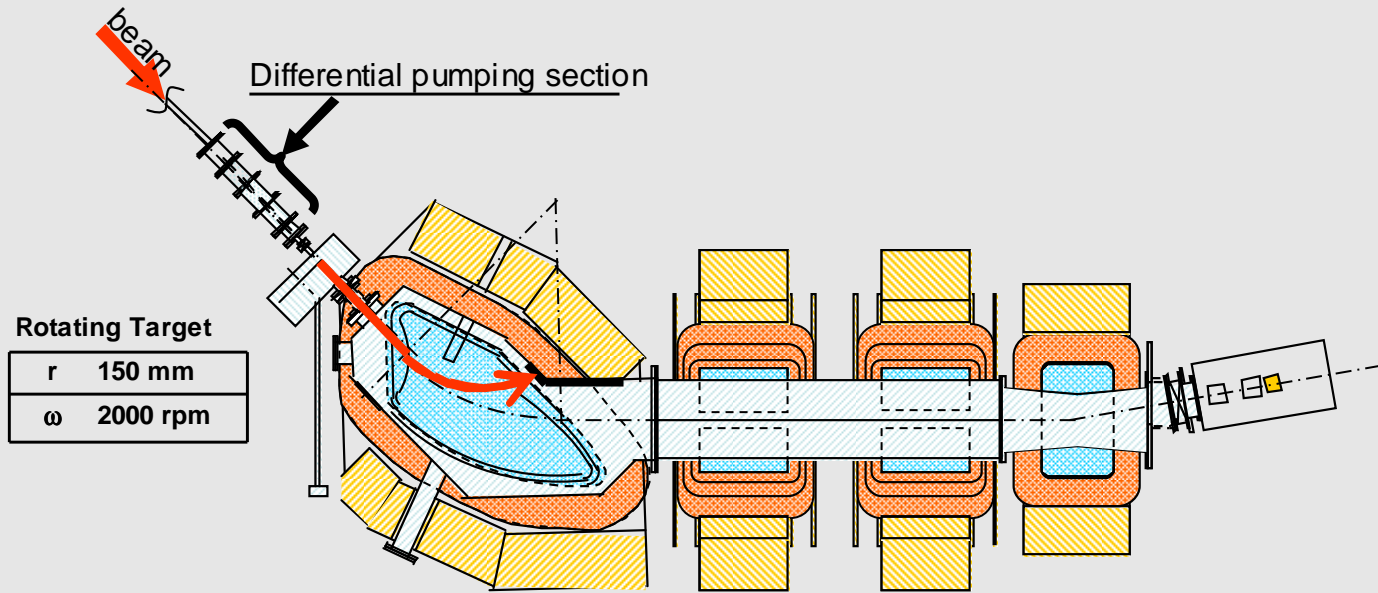
Fig. 5. Ion-optical elements of the VASSILISSA and computer-simulated trajectories of ^{198}Po ions.

DGFRS: Dubna Gas-Filled Recoil Separator



K. Subotic et al., Nucl. Instr. Meth. A481 (2002) 71.

RIKEN Gas-filled Recoil Separator GARIS



D1

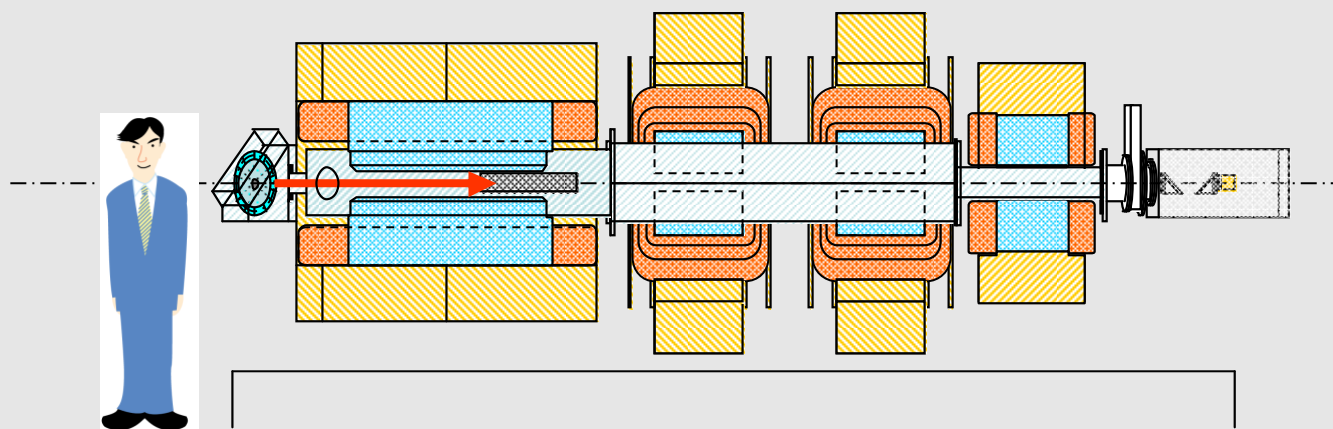
Bending angle	45 degree
Pole gap	150 mm
Radius of central ray	1200 mm
Maximum field	1.54 T

Q1, Q2

Pole length	500 mm
Bore radius	150 mm
Maximum field gradient	5.2 T/m

D2

Bending angle	10 degree
Pole gap	160 mm
Pole length	400 mm
Maximum Field	1.04 T



Magnification	X	-0.76
	Y	-1.99
Dispersion		0.97 cm/%
Total length		5760 mm
Acceptance	$\Delta\theta$	± 68 mrad
	$\Delta\phi$	± 57 mrad
	$\Delta\Omega$	12.2 msr

GANIL/SPIRAL1/SPIRAL2 facility

GANIL/SPIRAL 1 today

SP2 Beam time: 44 weeks/y
GANIL Beam time: 35 weeks/y
ISOL RIB Beams: 28-33 weeks/y
GANIL+SP 2 Users: 700-800/y

DESIR Facility
low energy RIB

CIME cyclotron RIB at 1-20 AMeV
(up to 9 AMeV for fiss. fragments)

HRS+RFQ Cooler

S3 separator-spectrometer

RIB Production Cave
Up to 10^{14} fiss./sec.

LINAC: 33MeV p, 40 MeV d, 14.5 A MeV HI

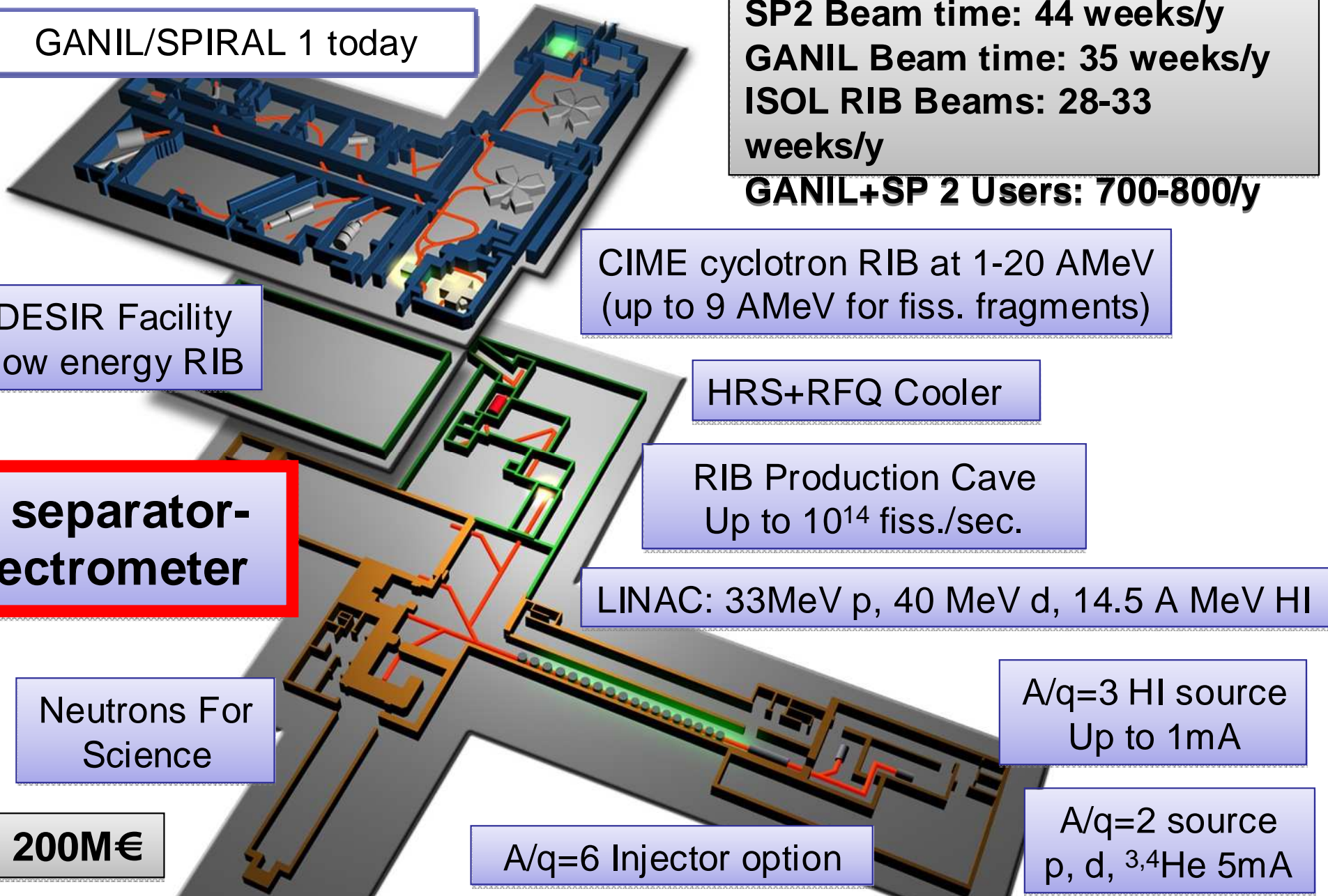
Neutrons For Science

A/q=3 HI source
Up to 1mA

Cost: 200M€

A/q=6 Injector option

A/q=2 source
p, d, ^3He , ^4He 5mA



S3 at SPIRAL2, GANIL, Caen

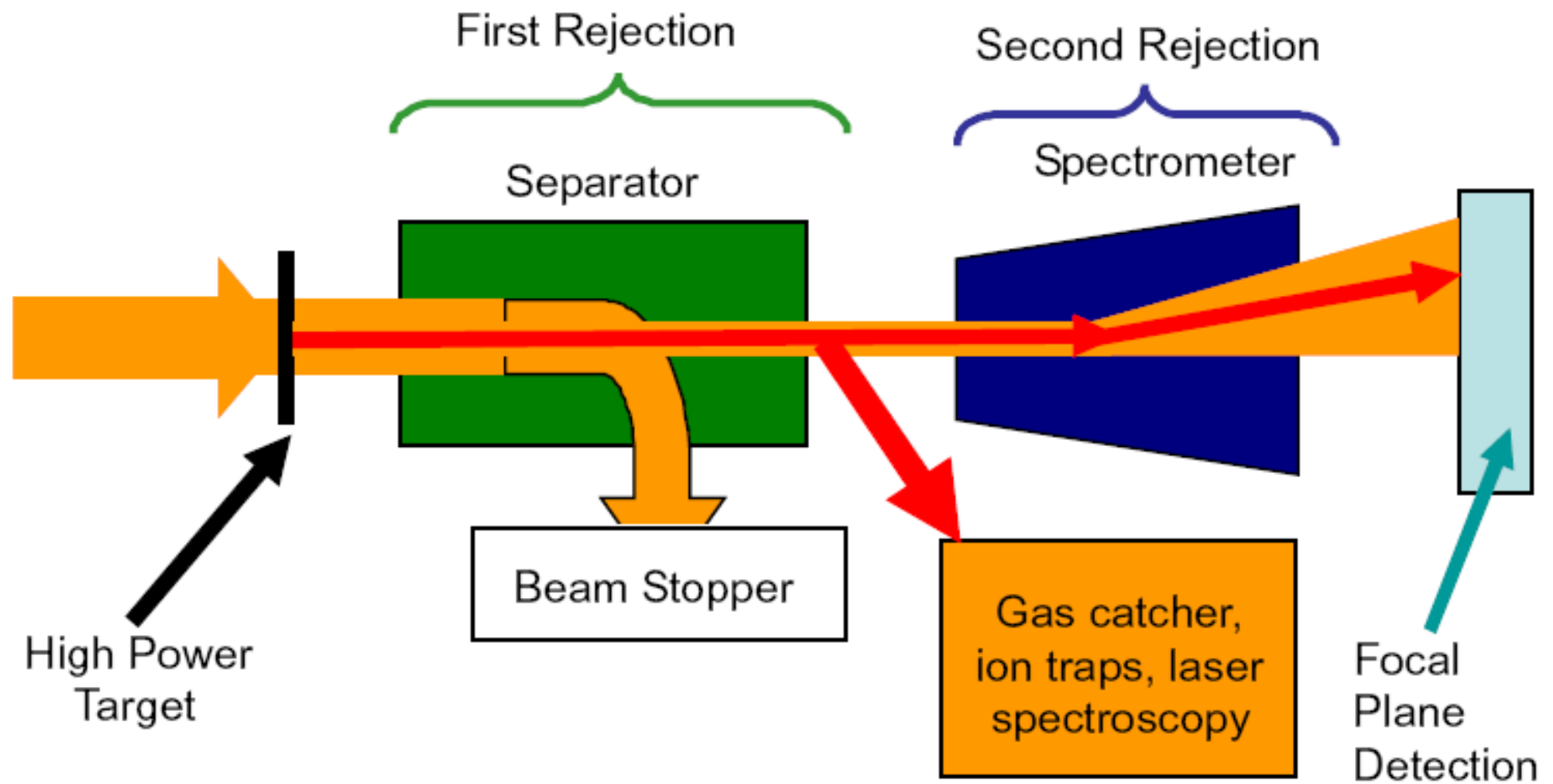
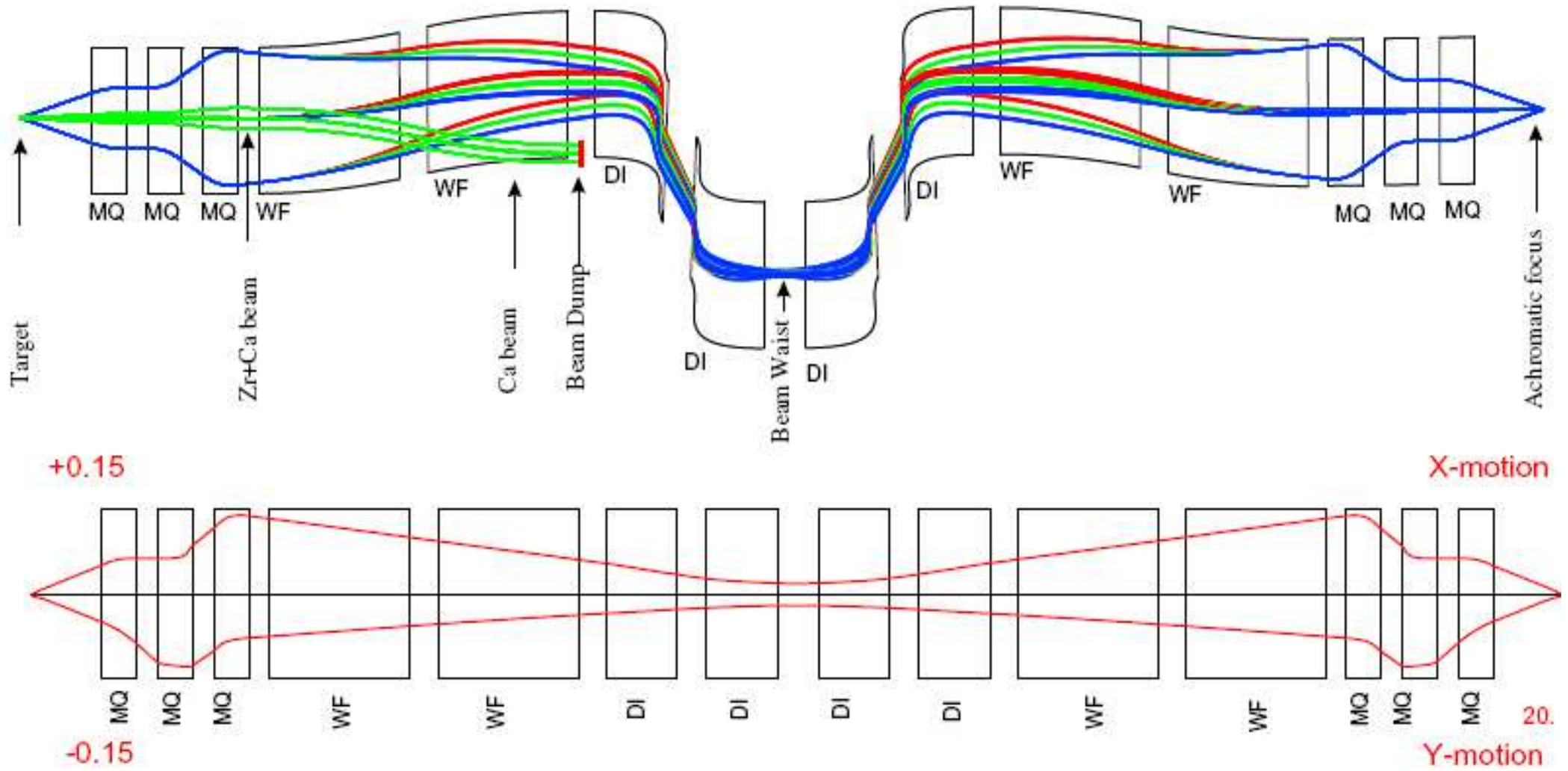


Fig. 1. Schematic idea for S^3 showing the two stage separator.

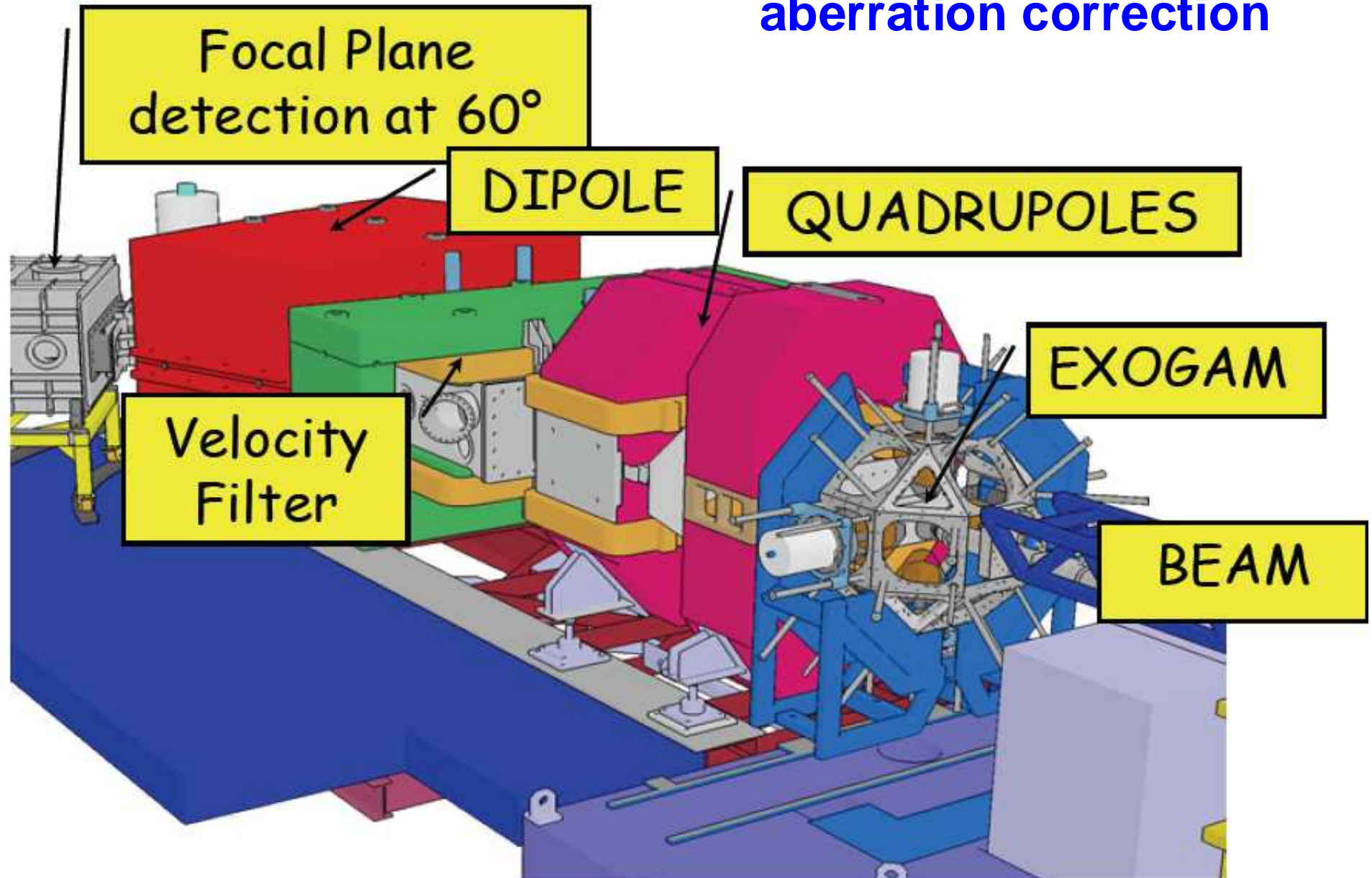
S3 at SPIRAL2, GANIL, Caen



A. Drouart et al., Nucl. Instr. Meth. B266(2008) 4162.

VAMOS at GANIL

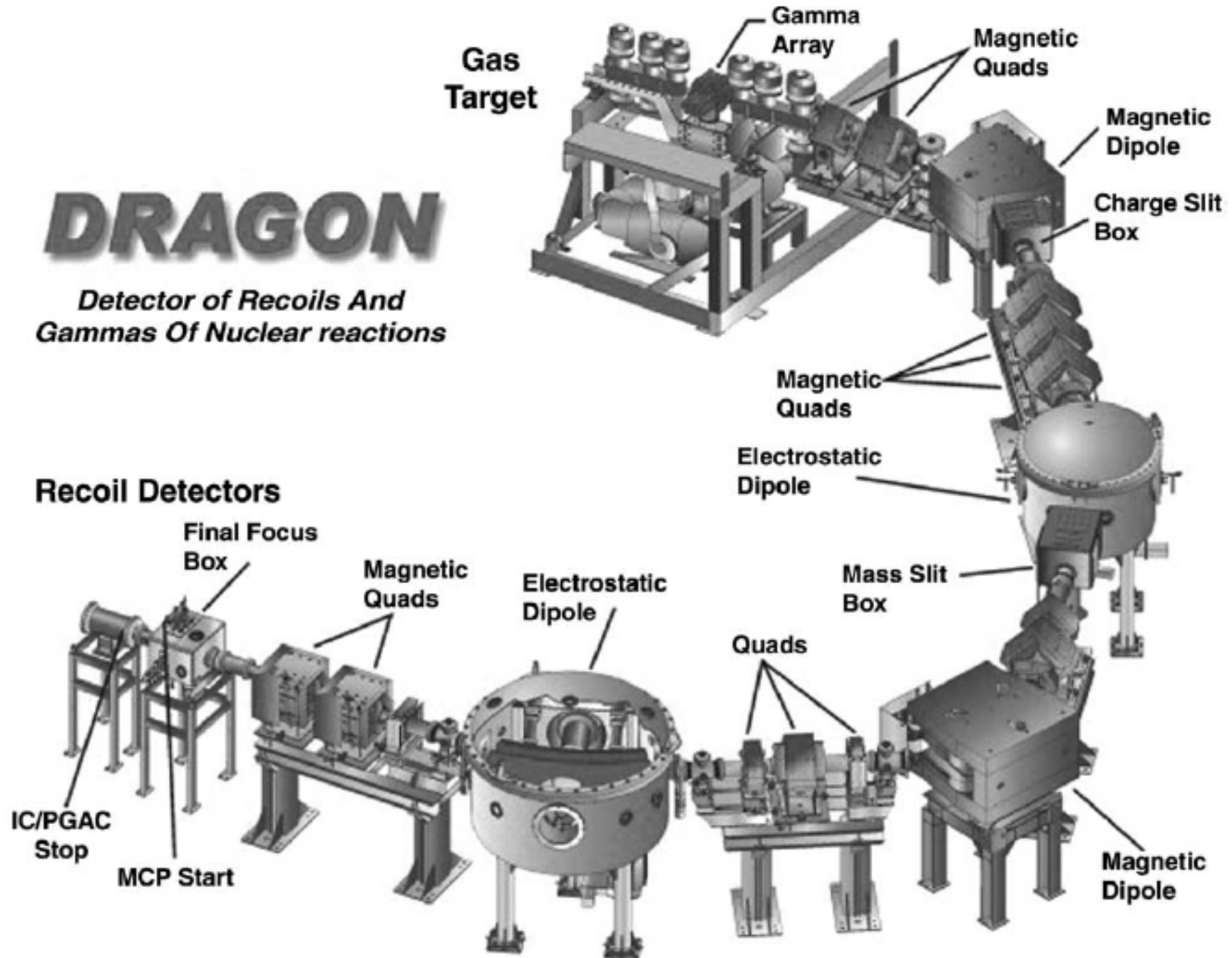
Very wide acceptance \Rightarrow trajectory reconstruction for aberration correction



DRAGON

DRAGON

*Detector of Recoils And
Gammas Of Nuclear reactions*

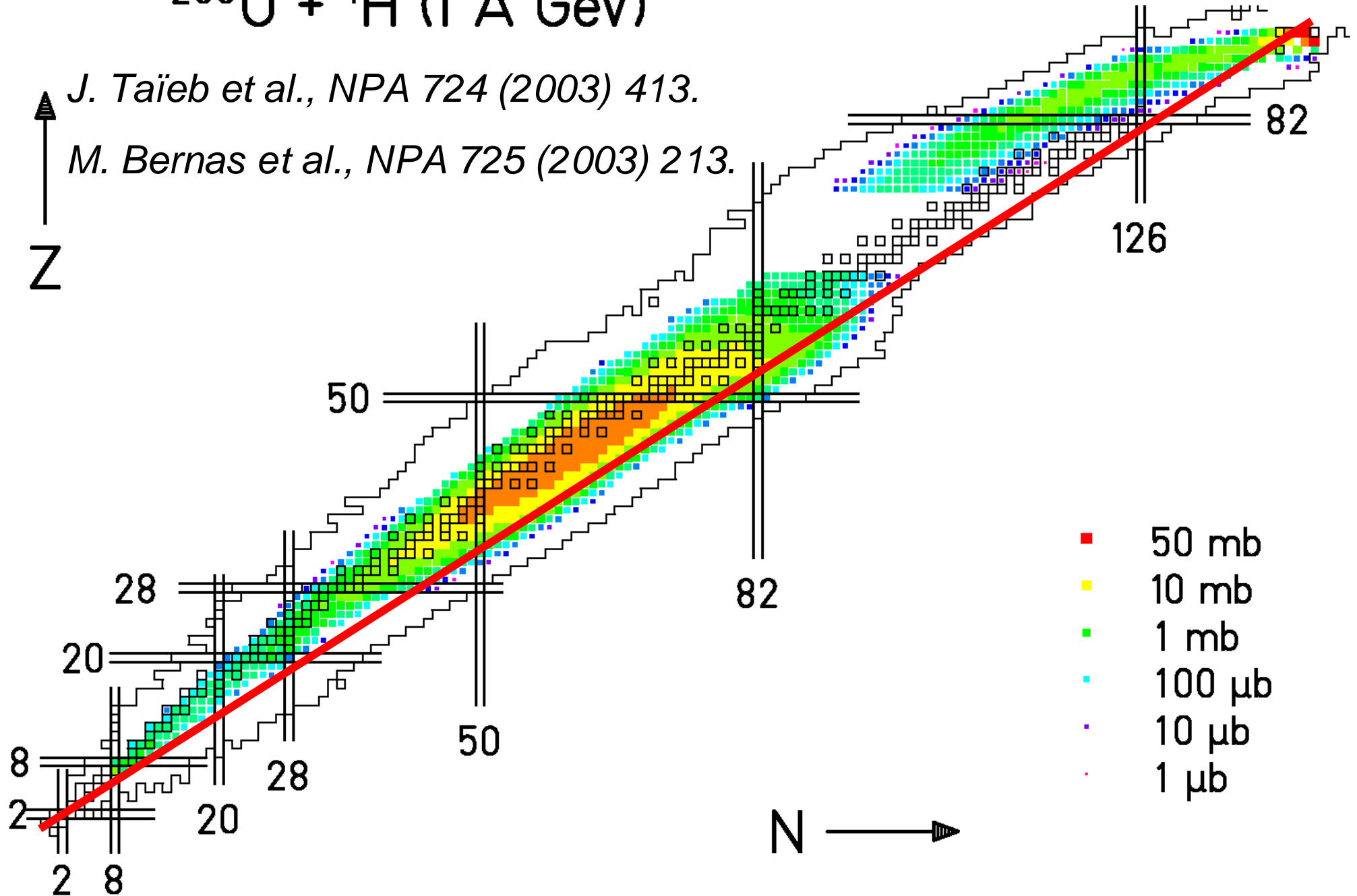


Spallation + Fragmentation + Fission

$^{238}\text{U} + ^1\text{H}$ (1 A GeV)

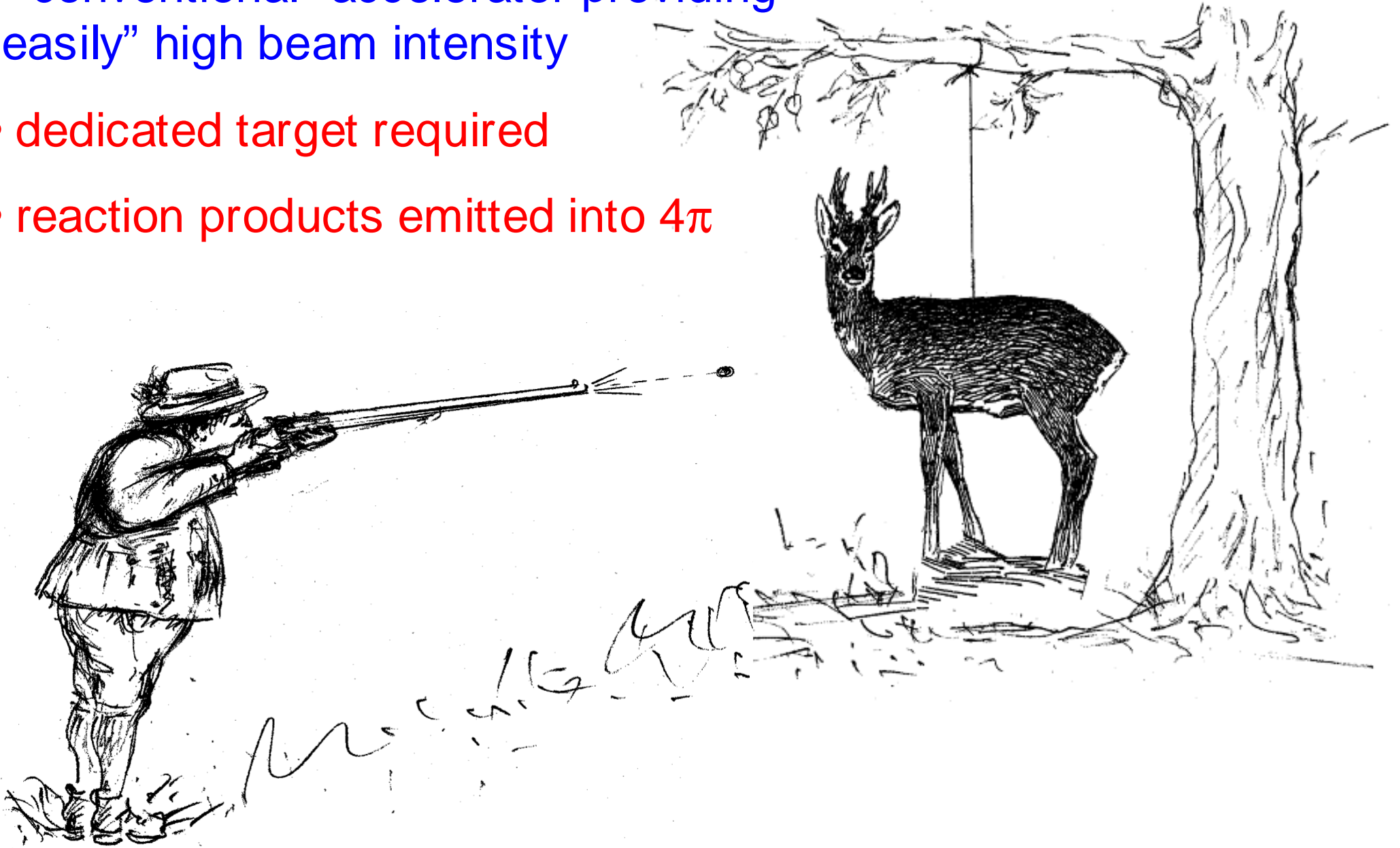
J. Taïeb et al., NPA 724 (2003) 413.

M. Bernas et al., NPA 725 (2003) 213.



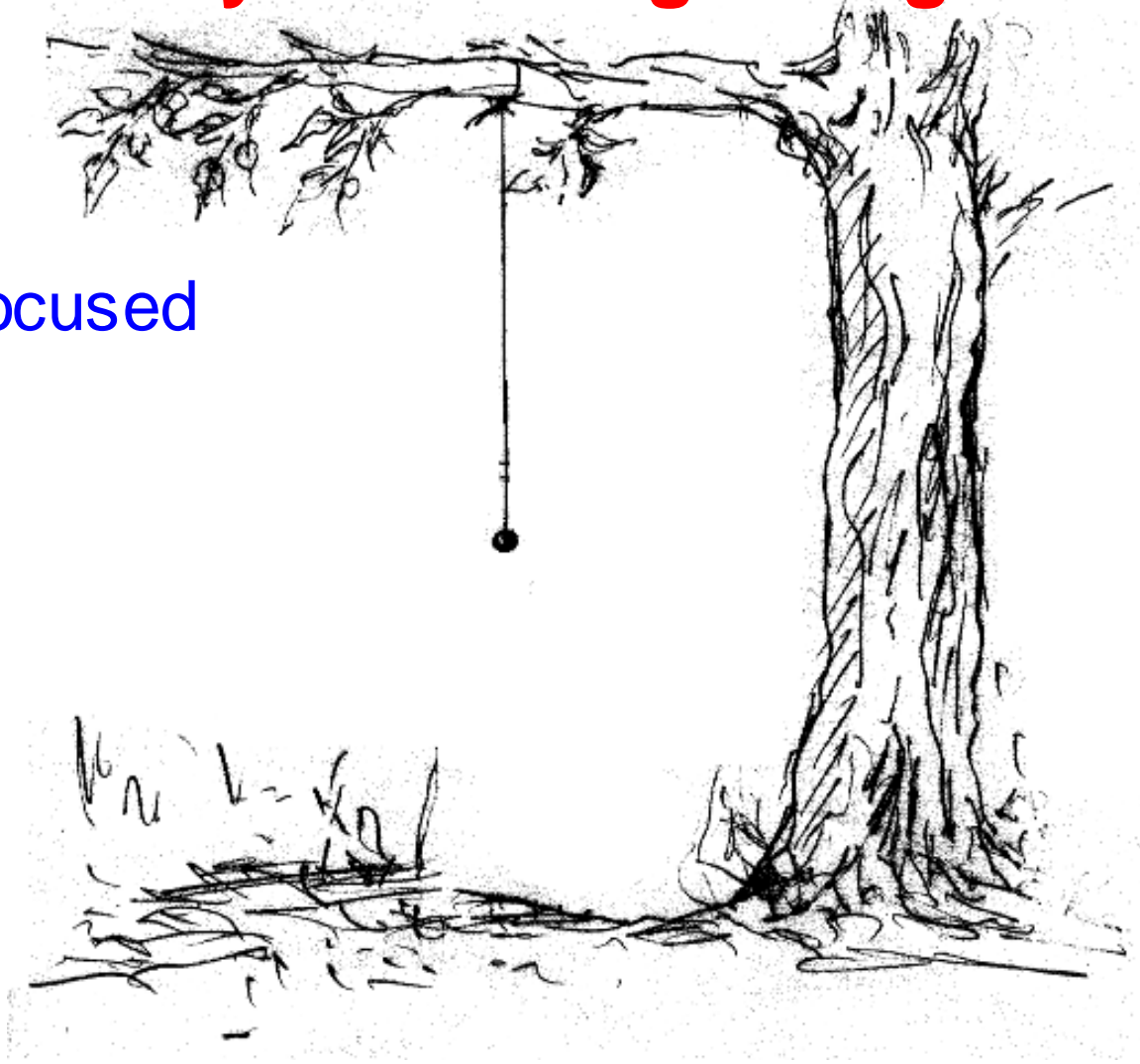
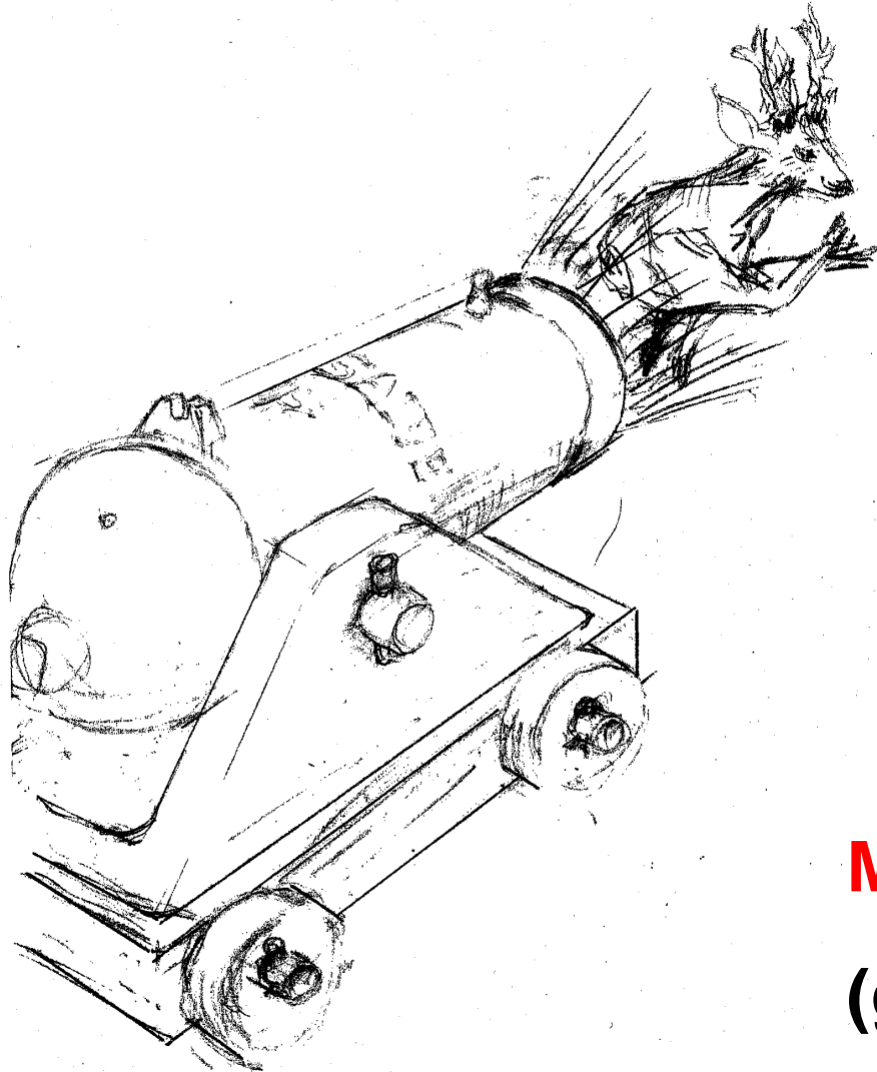
Normal kinematics: n, p or light ions on heavy target

- “conventional” accelerator providing “easily” high beam intensity
- dedicated target required
- reaction products emitted into 4π



Inverse kinematics: heavy ions on light target

- complex and expensive accelerator
- reaction products forward focused



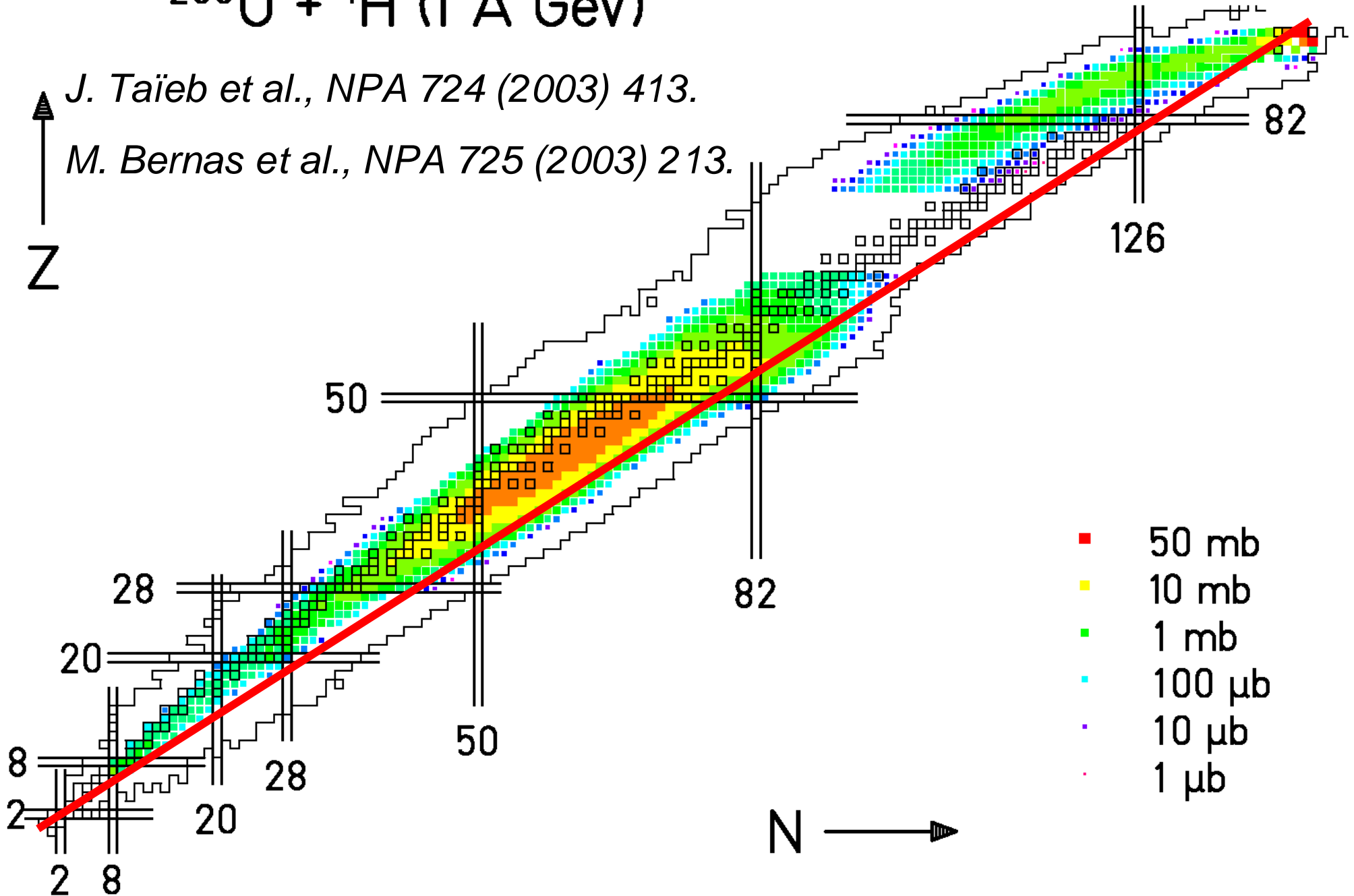
Mind the losses during stopping!
(graphical representation censored)

Spallation + Fragmentation + Fission

$^{238}\text{U} + ^1\text{H}$ (1 A GeV)

J. Taïeb et al., NPA 724 (2003) 413.

M. Bernas et al., NPA 725 (2003) 213.



Momentum-loss achromat (Wedge separation)

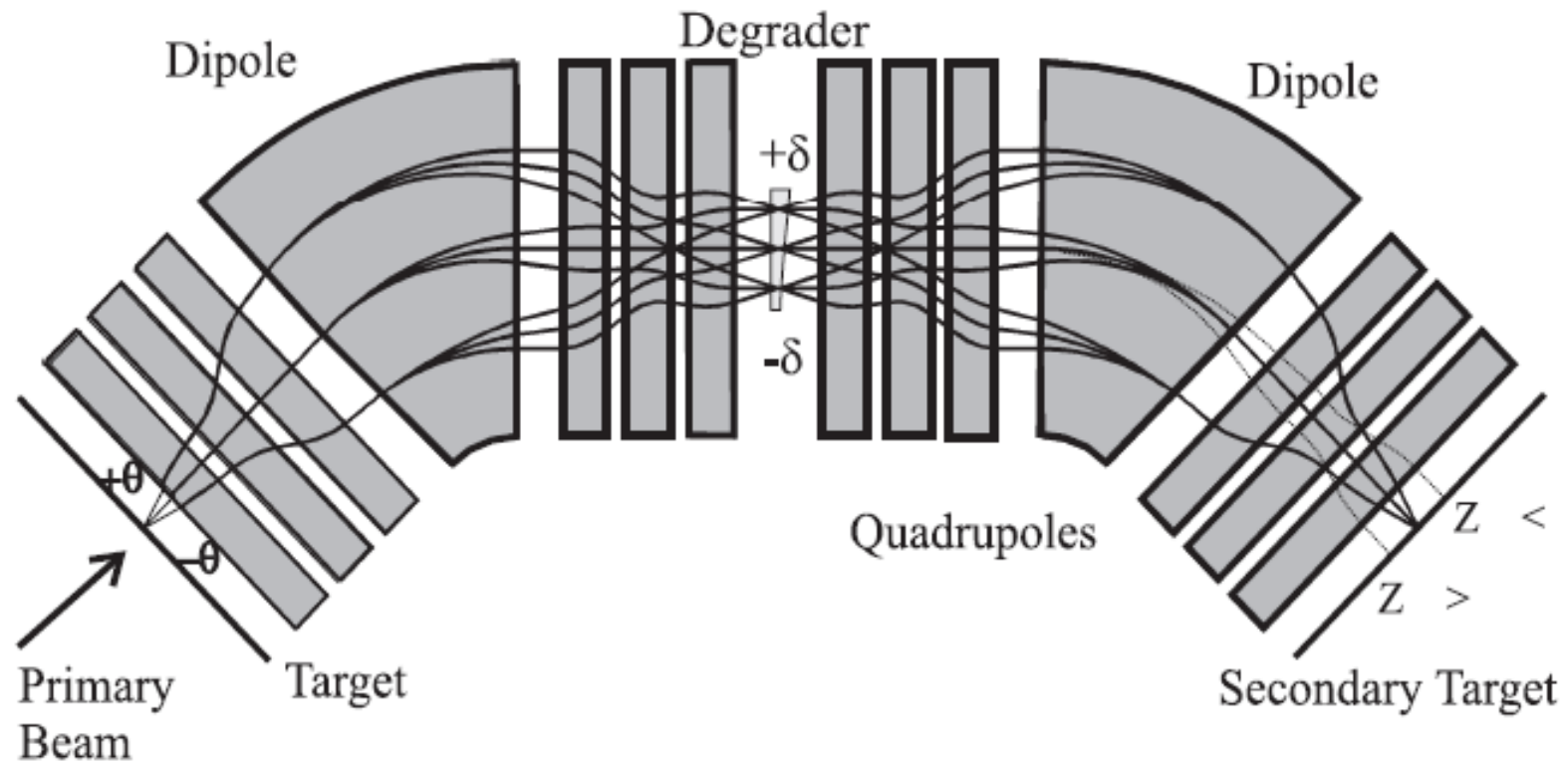
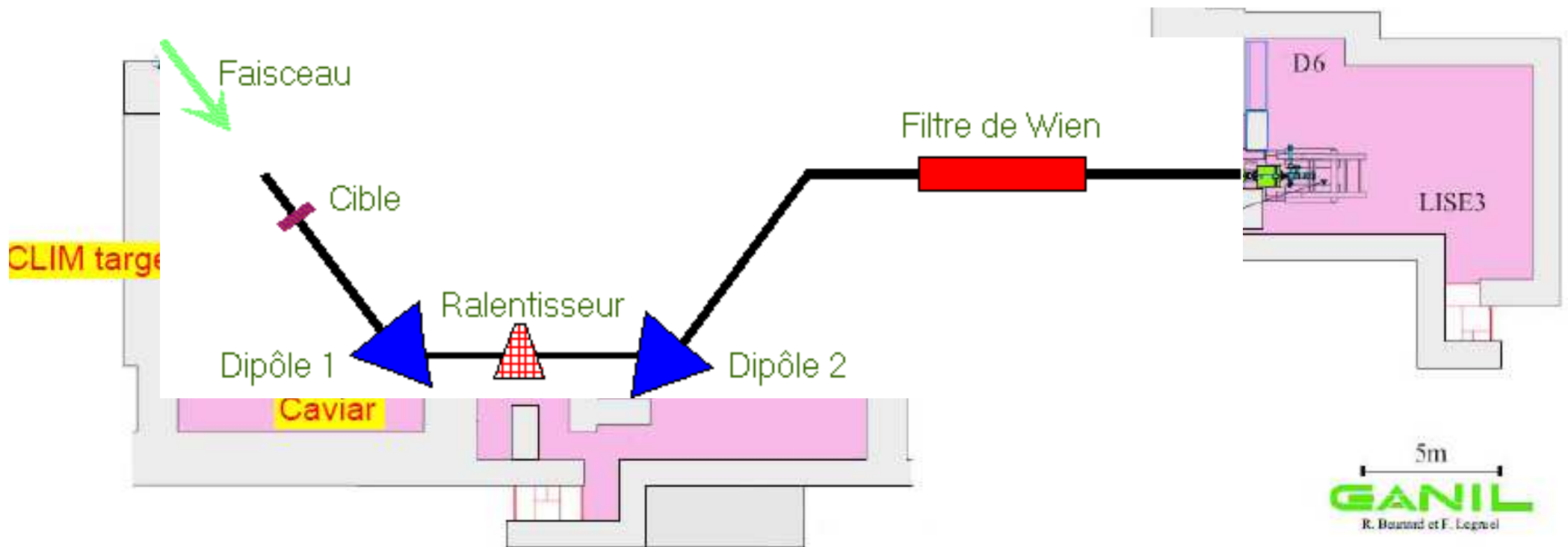


Fig. 4. Schematic representation of the ion-optics used in a momentum-loss achromat to separate projectile fragments.

LISE



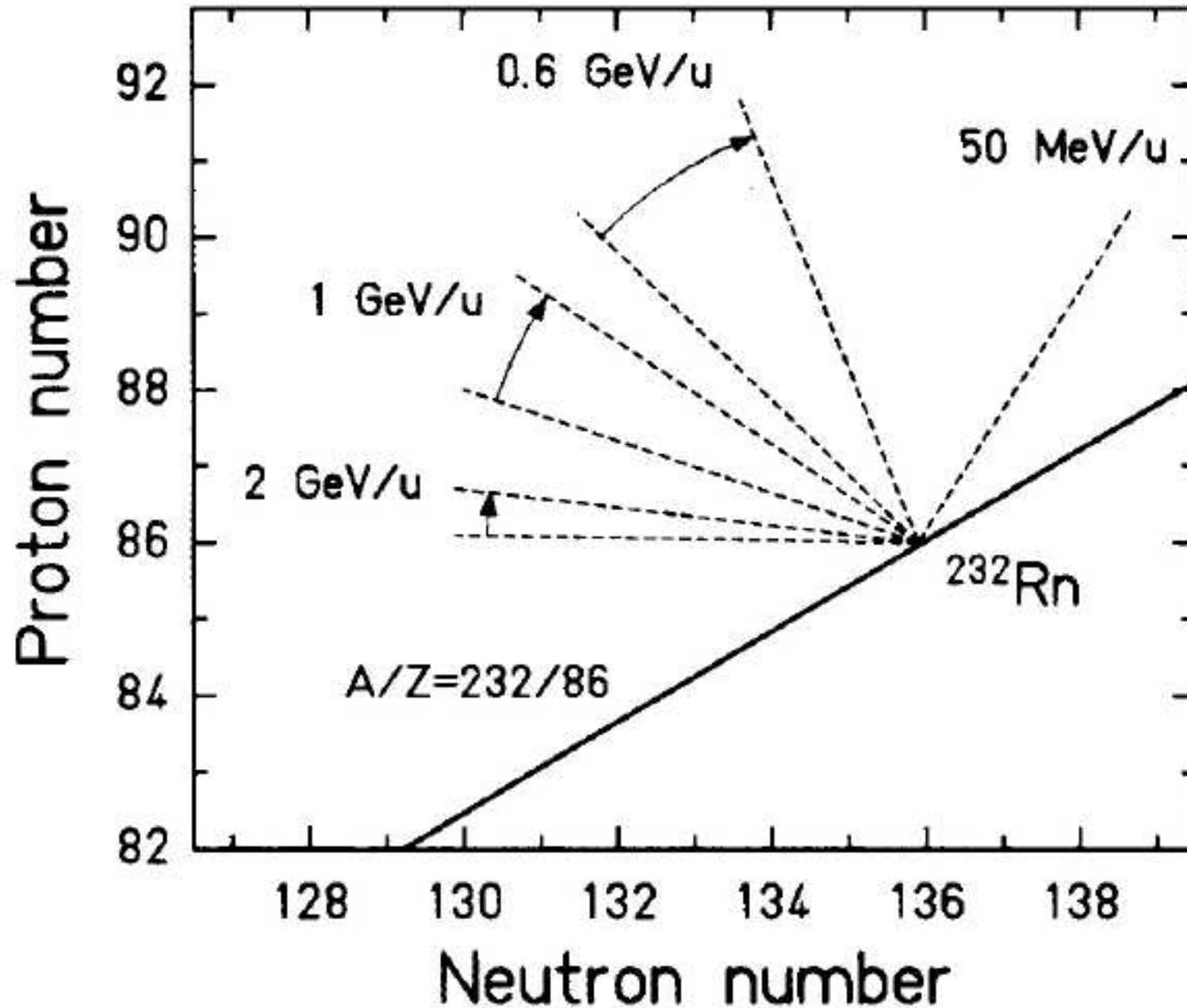
R. Anne et al., Nucl. Instr. Meth. A257 (1987) 215.

R. Anne et al., Nucl. Instr. Meth. B70 (1992) 276.

Dispersive ion optical elements

- magnets are momentum dispersive
- electrostatic deflectors are energy dispersive
- Wien filters are velocity dispersive
- achromatic wedges are dispersive in mZ^2/E or $(Z/v)^2$
- RF kicker are flight time selective

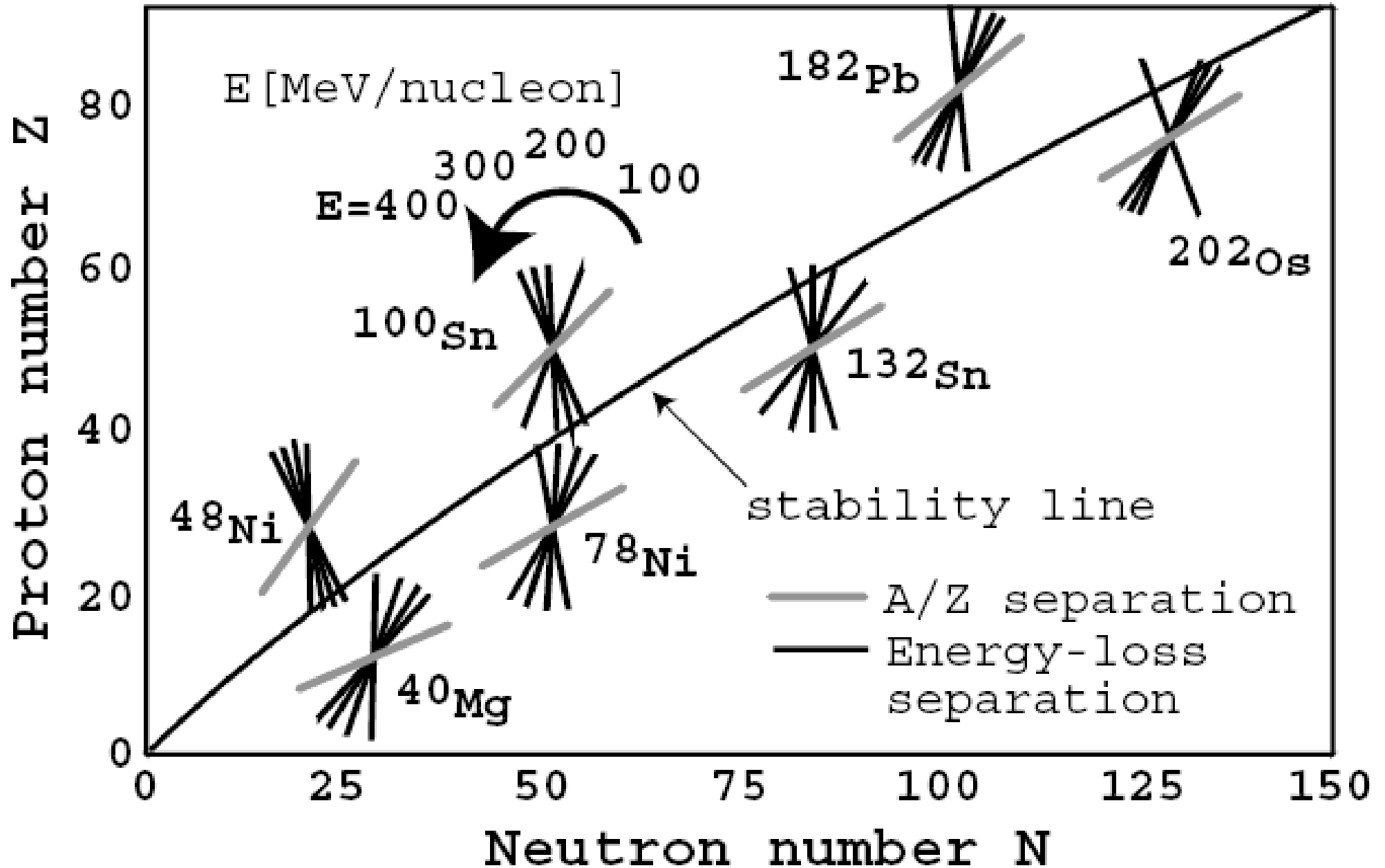
Effect of wedge selection



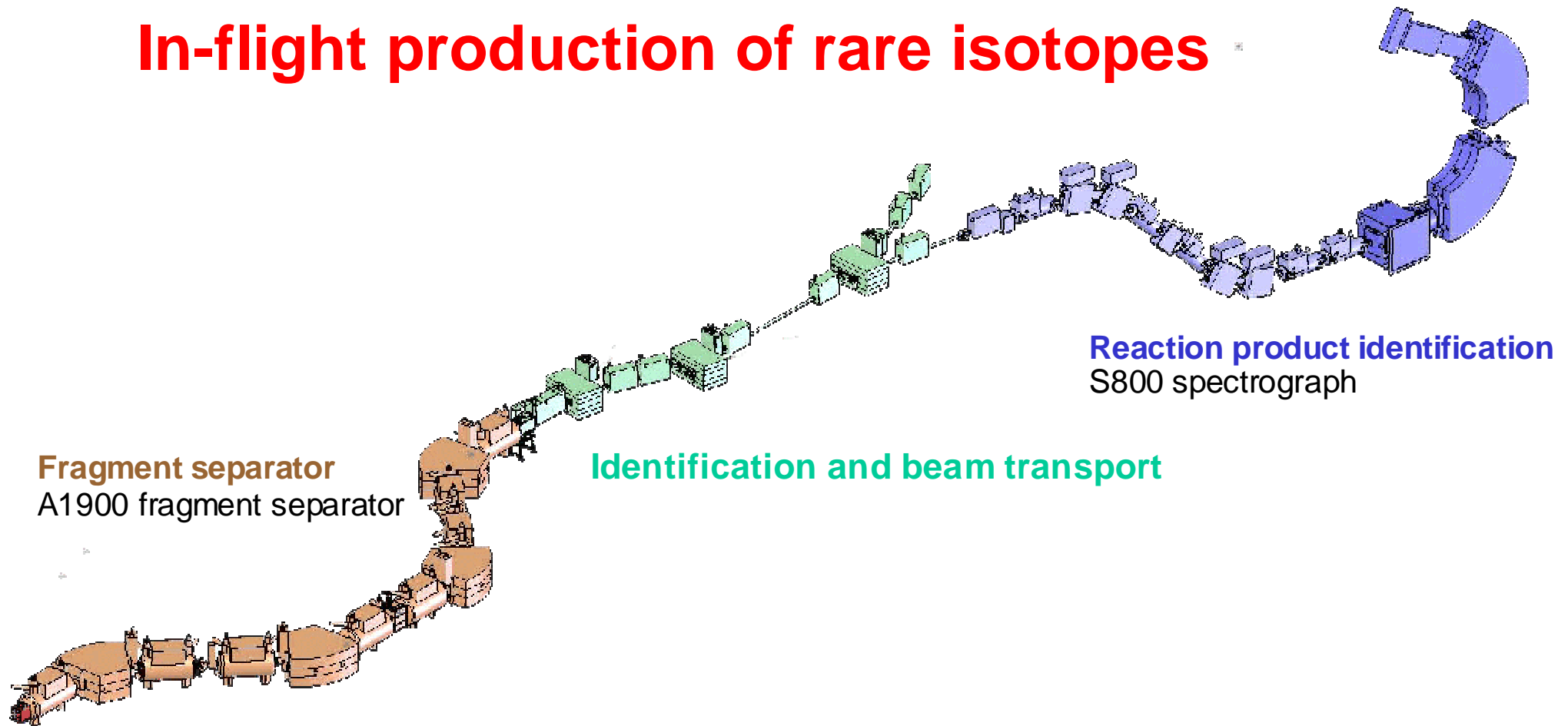
K.H. Schmidt et al., Nucl. Instr. Meth. A260 (1987) 287.

Effect of wedge selection

$d/R=0.5$



In-flight production of rare isotopes



Fragment separator
A1900 fragment separator

Identification and beam transport

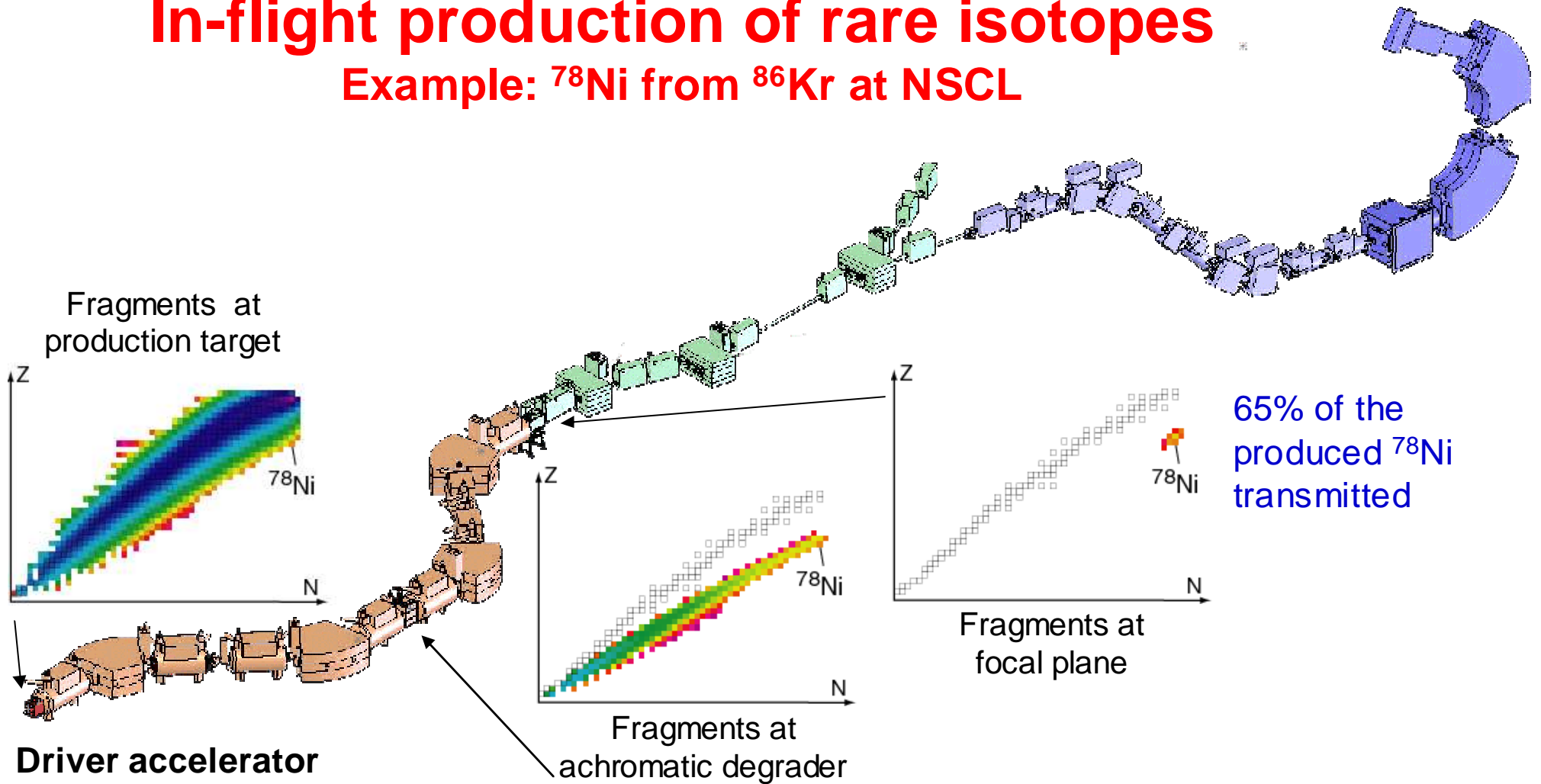
Reaction product identification
S800 spectrograph

Driver accelerator ($v = 0.25-0.6 c$)

- **Fragment separator** (A1900_{NSCL}, FRS_{GSI}, BigRIPS_{RIKEN}, ALPHA spectrometer/LISE_{GANIL})
- **Identification and beam transport**
 - Stopped beam experiments, reaccelerated beam experiments
 - Fast beam experiments
 - Secondary reaction
 - **Reaction product identification** (S800 spectrograph, CATE/Aladin, Silicon telescopes/TOF wall, SPEG)

In-flight production of rare isotopes

Example: ^{78}Ni from ^{86}Kr at NSCL



Driver accelerator

➤ **Fragment separator** (A1900_{NSCL}, FRS_{GSI}, BigRIPS_{RIKEN}, ALPHA/LISE_{GANIL})

➤ **Identification and beam transport**

➤ Stopped beam experiments

➤ Fast beam experiments

➤ Secondary reaction

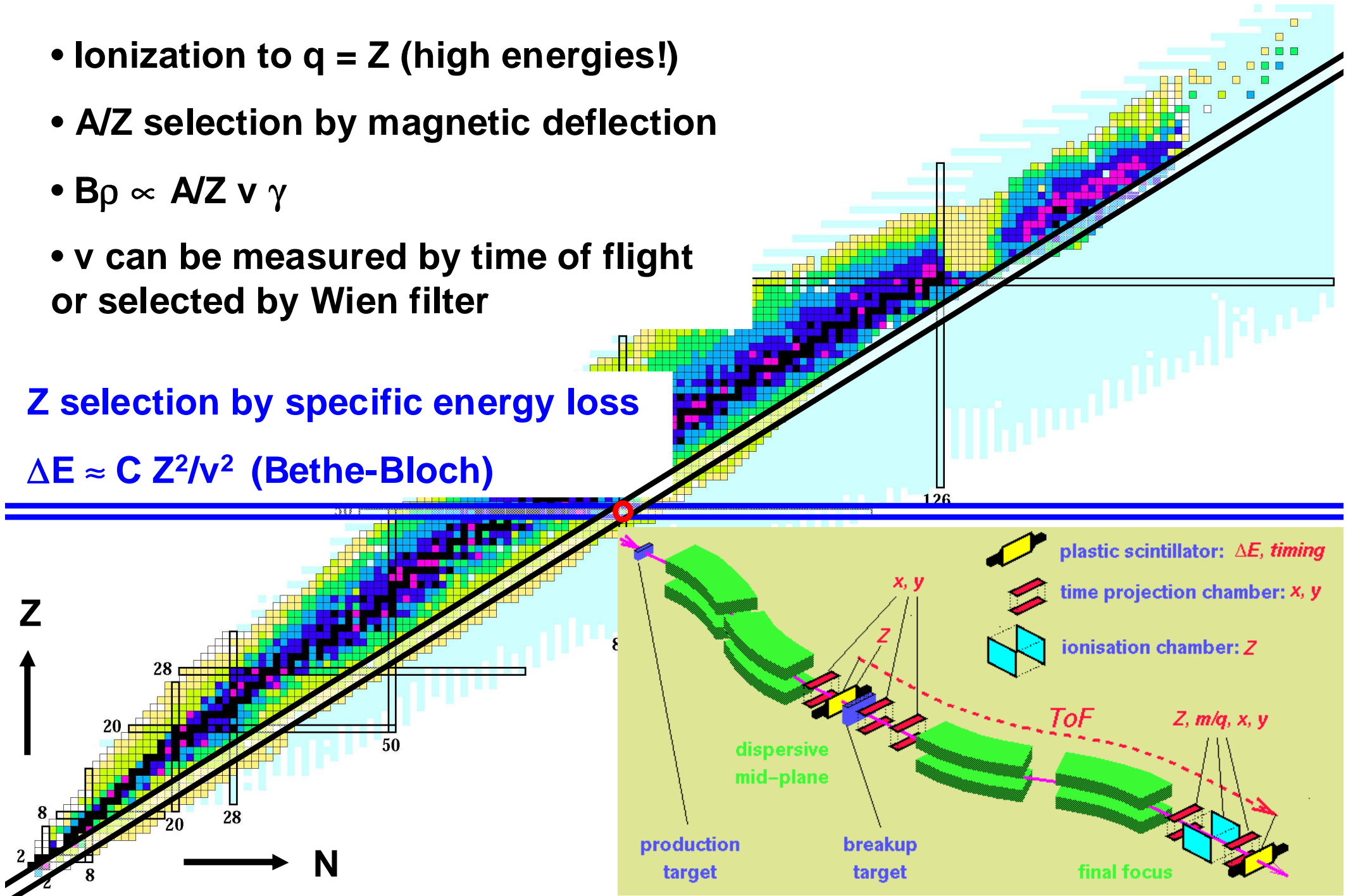
➤ **Reaction product identification** (S800 spectrograph, CATE/Aladin, Silicon telescopes/TOF wall, SPEG)

Isotope selection at (high E) in-flight separators

- Ionization to $q = Z$ (high energies!)
- A/Z selection by magnetic deflection
- $B\rho \propto A/Z v \gamma$
- v can be measured by time of flight or selected by Wien filter

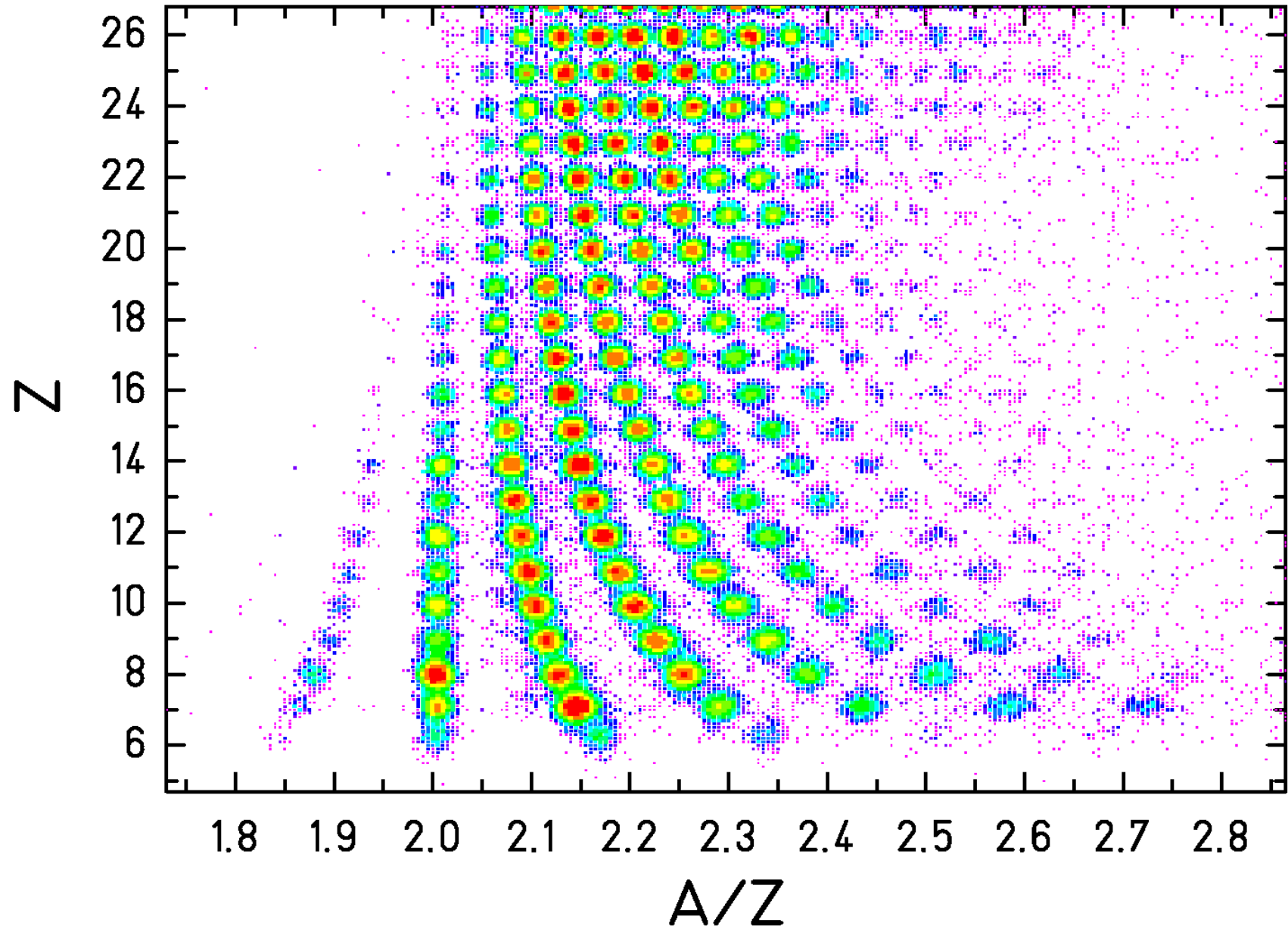
Z selection by specific energy loss

$$\Delta E \approx C Z^2/v^2 \text{ (Bethe-Bloch)}$$



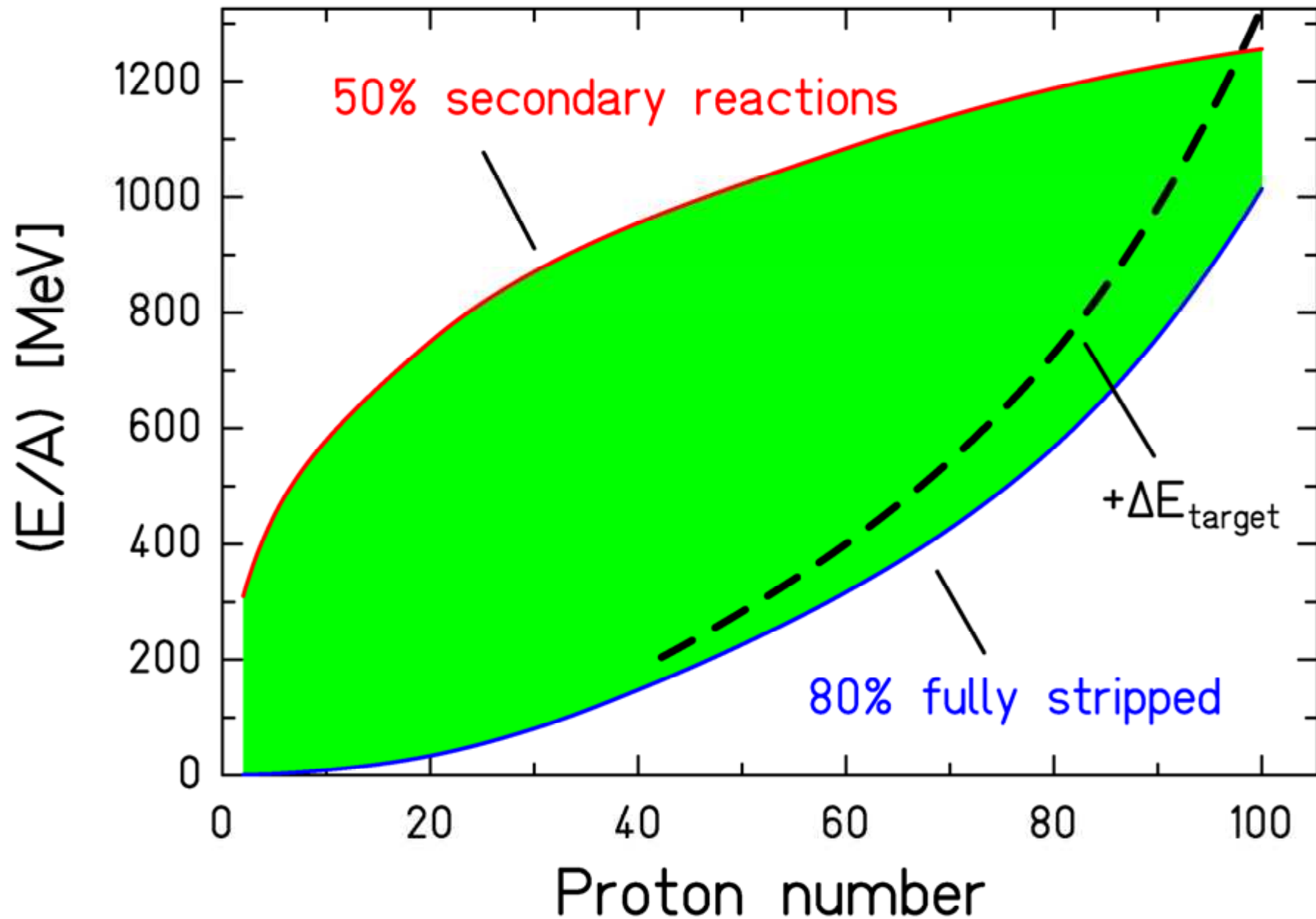
Perfect isotope identification at high energy

1 A GeV ^{238}U on titanium



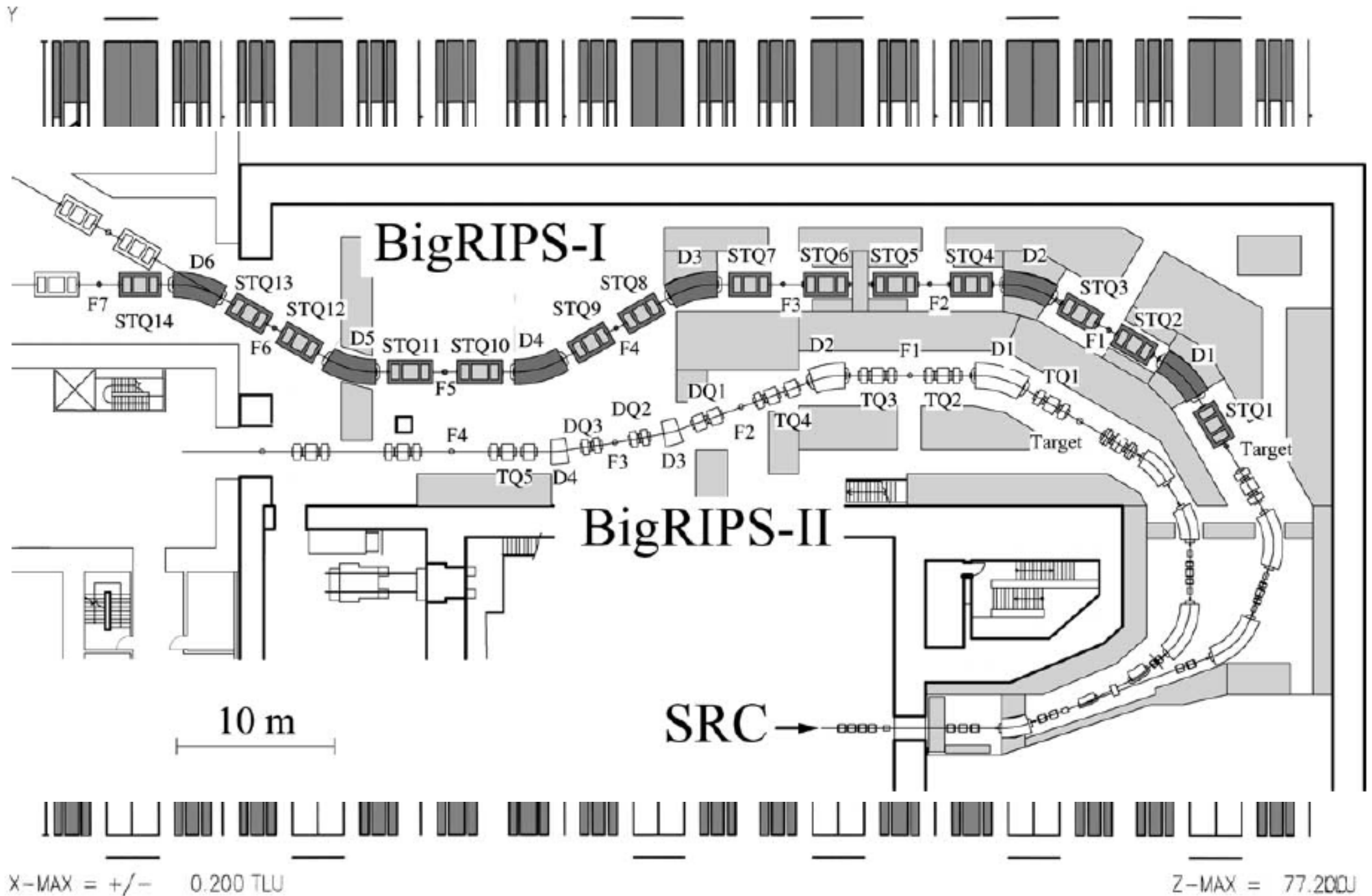
M.V. Ricciardi et al., Nucl. Phys. A733 (2004) 299.

Optimum energy for FRS-like momentum achromat



K.H. Schmidt, Euroschool Leuven 2000.

BigRIPS at RIKEN, Japan



Super-FRS at FAIR, Darmstadt

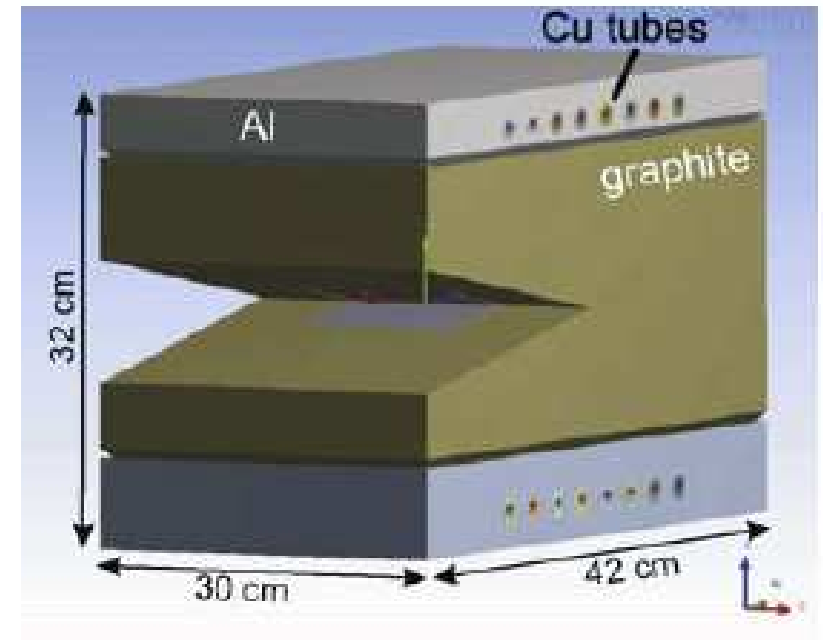
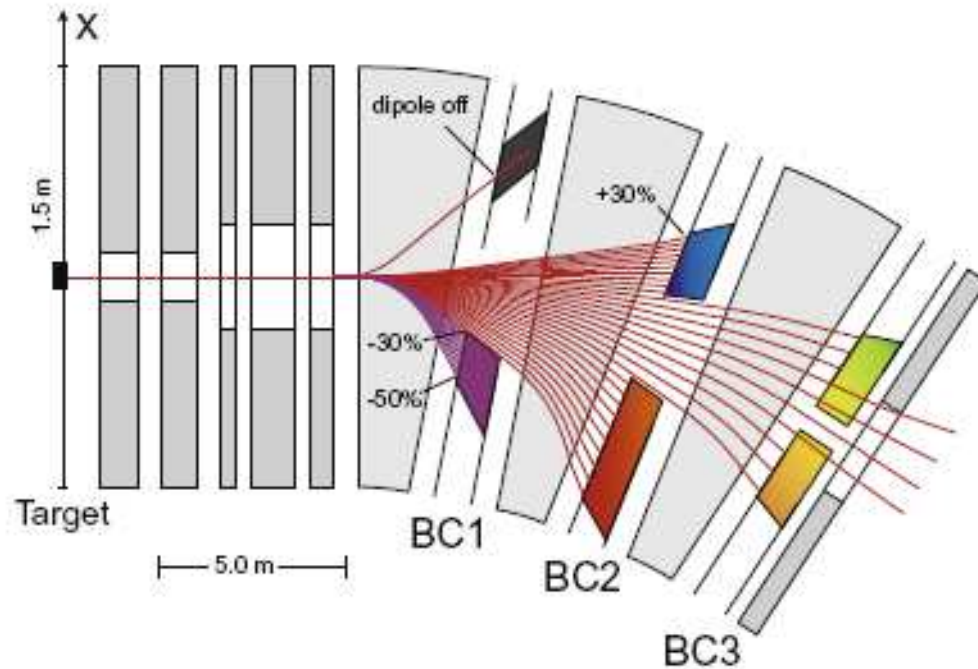
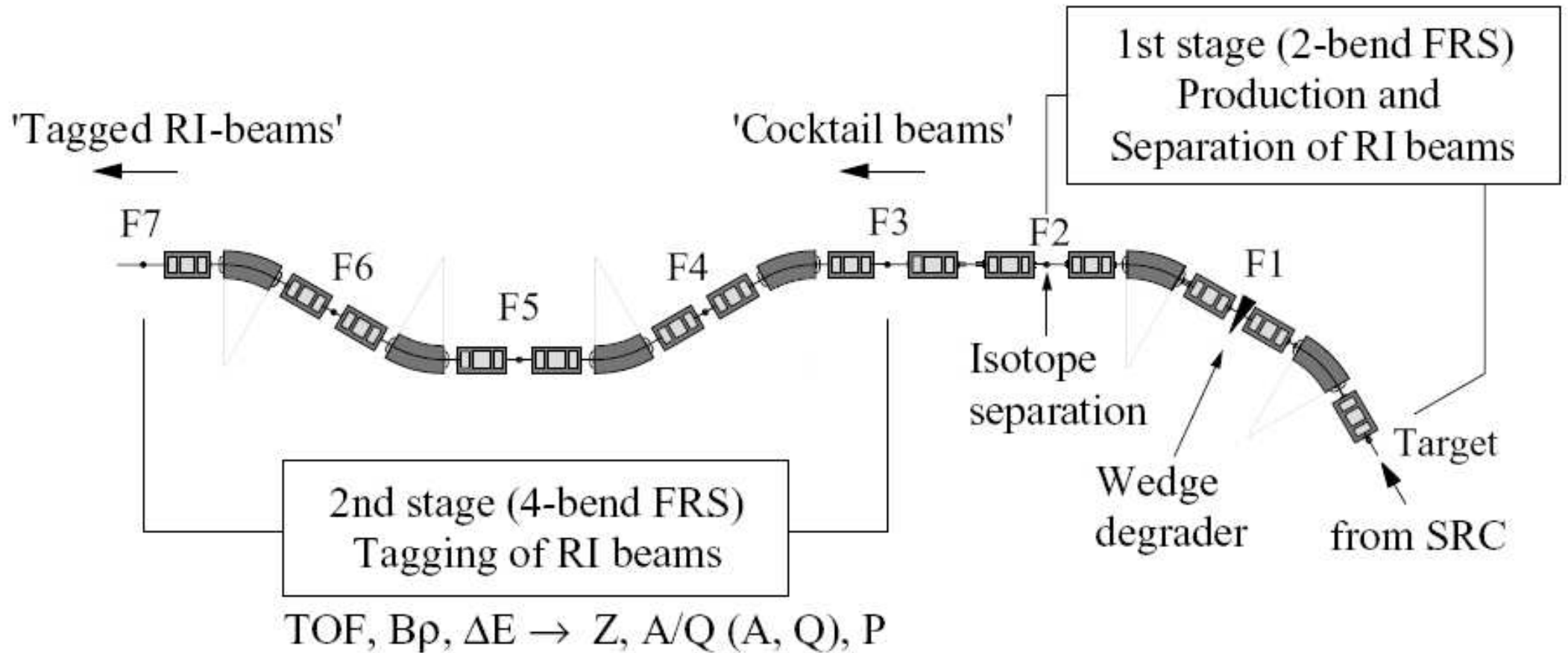


Fig. 4. Beam catcher locations in the first dipole stage of the pre-separator. Depending on the fragment setting the primary beam will be dumped at the position given by the relative difference in magnetic rigidity. Plotted are trajectories of primary beams with different $\delta_{B\rho}$ values in steps of 1%.

Fig. 5. Layout of the front part of the beam catcher. The V-shaped graphite block will absorb the beam energy of up to 50 kW and is actively cooled.

BigRIPS at RIKEN, Japan

BigRIPS : Tandem (Two-stage) Separator



Q3D Spectrometer

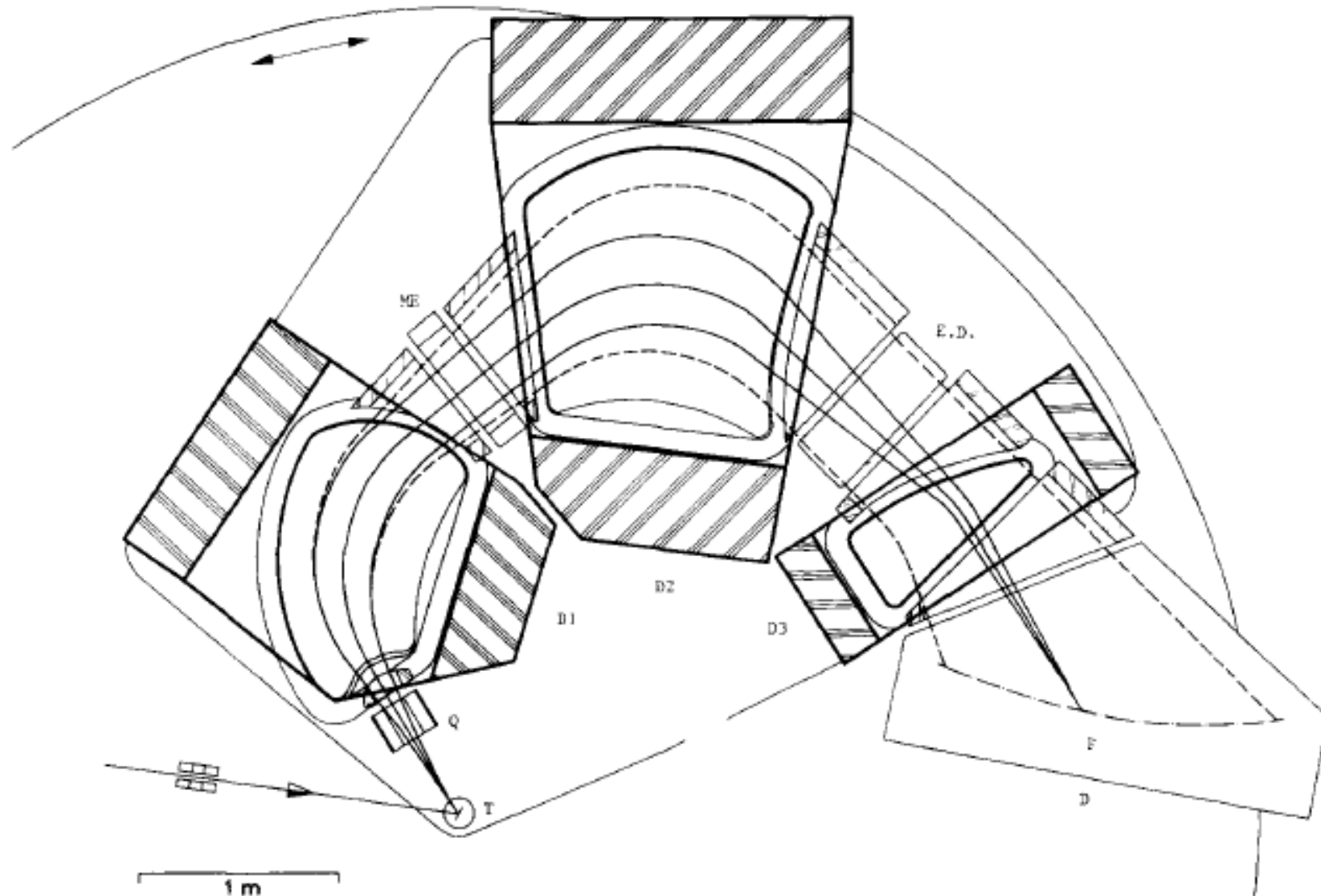


Fig. 1. Ion optical layout of the QDDD spectrograph. T – target chamber; ME – multipole element; D1, D2, D3 – dipole magnets; E.D. – electrostatic deflector; F – focal surface; D – detector chamber.

M. Löffler et al., Nucl. Instr. Meth. 111 (1973) 1.

Example spectrum $^{180}\text{Hf}(d,p)$

V. Bondarenko et al. / Nuclear Physics A 709 (2002) 3–59

19

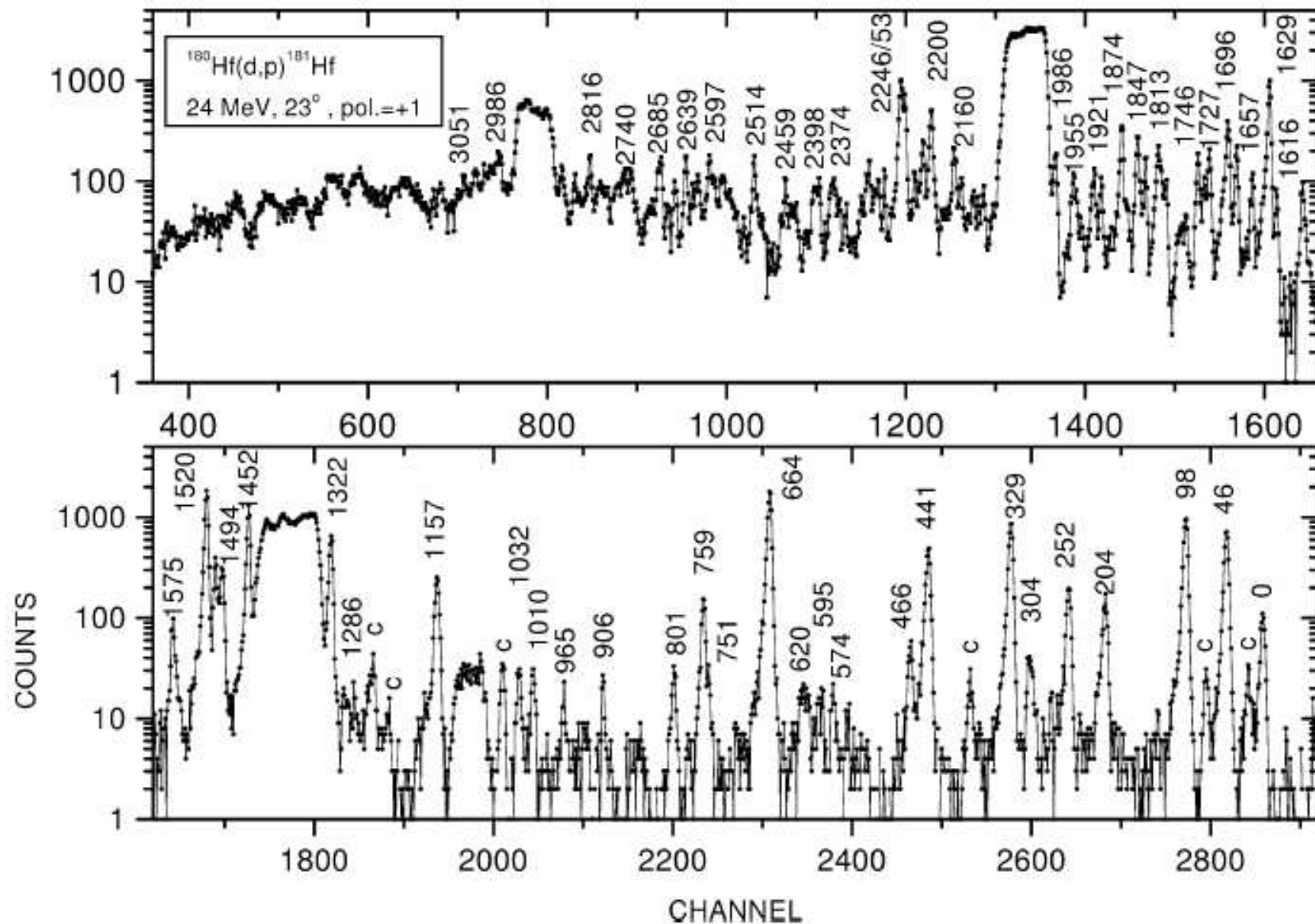


Fig. 3. An example of proton spectra from the reaction $^{180}\text{Hf}(d,p)^{181}\text{Hf}$. The peaks are labelled by the excitation energy in keV. The proton groups labeled with 'c' belong to contaminant isotopes.

The SPEG spectrometer at GANIL

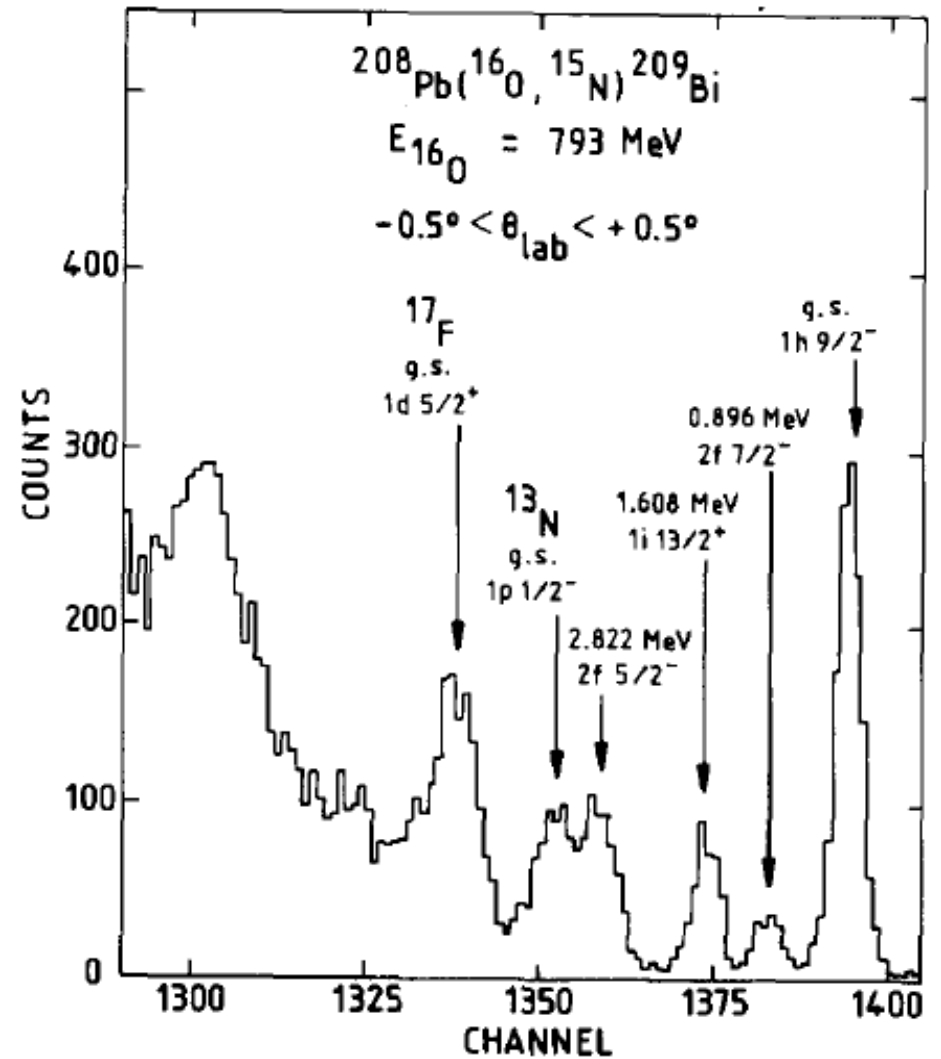
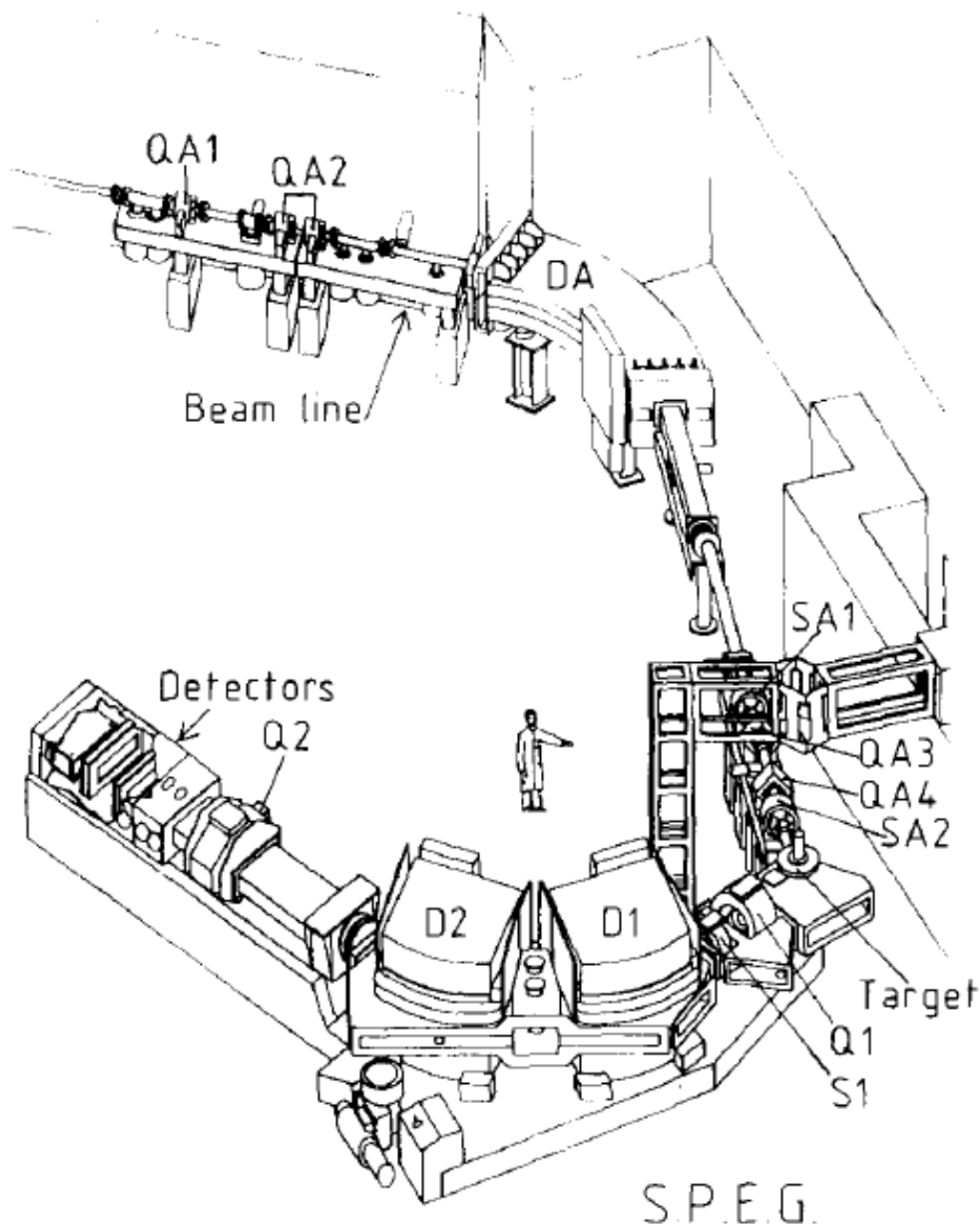


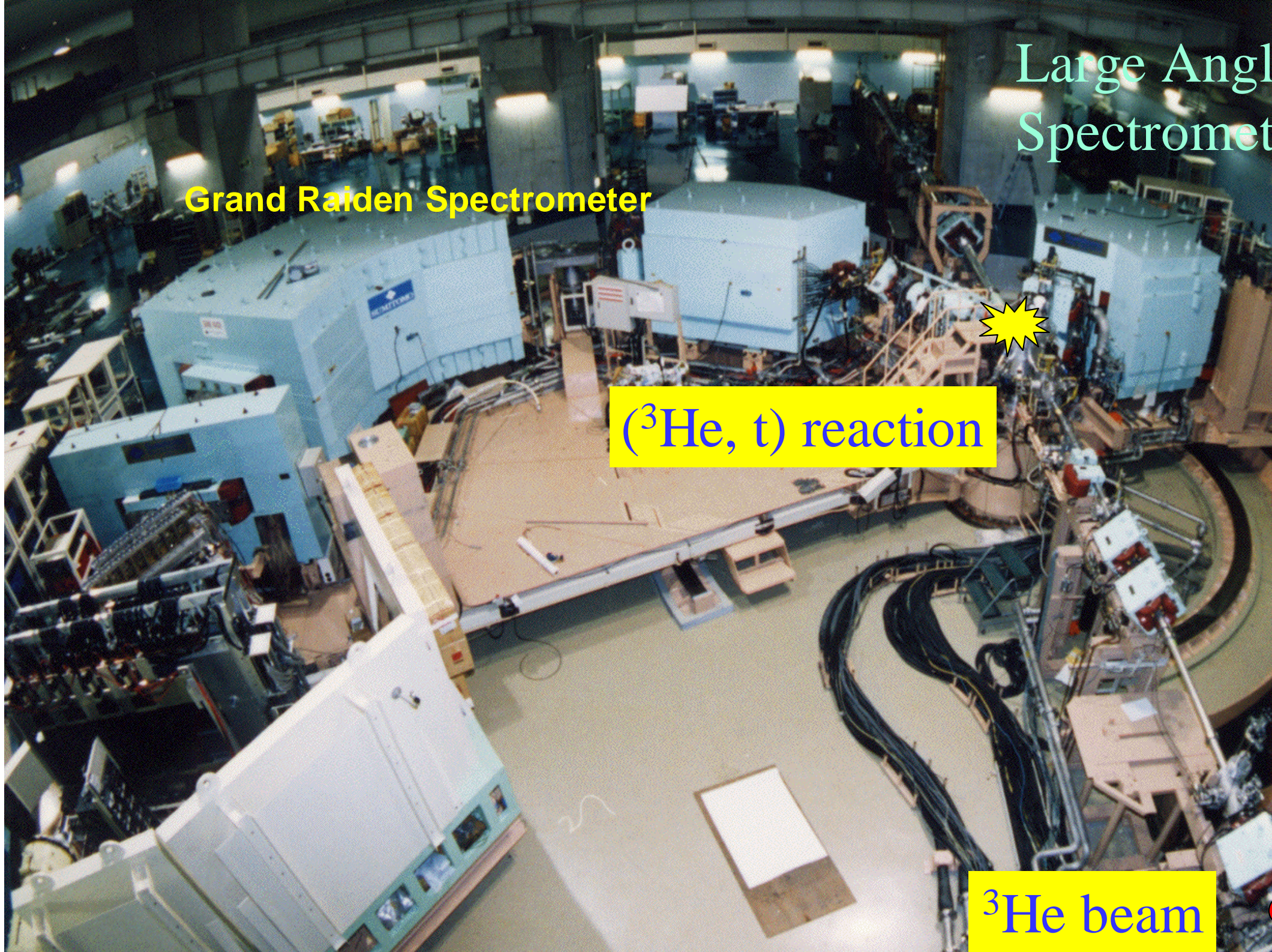
Fig. 20. Zero degree spectrum measured for the $^{208}\text{Pb}(^{16}\text{O}, ^{15}\text{N})^{209}\text{Bi}$ reaction.

Large Angle
Spectrometer

Grand Raiden Spectrometer

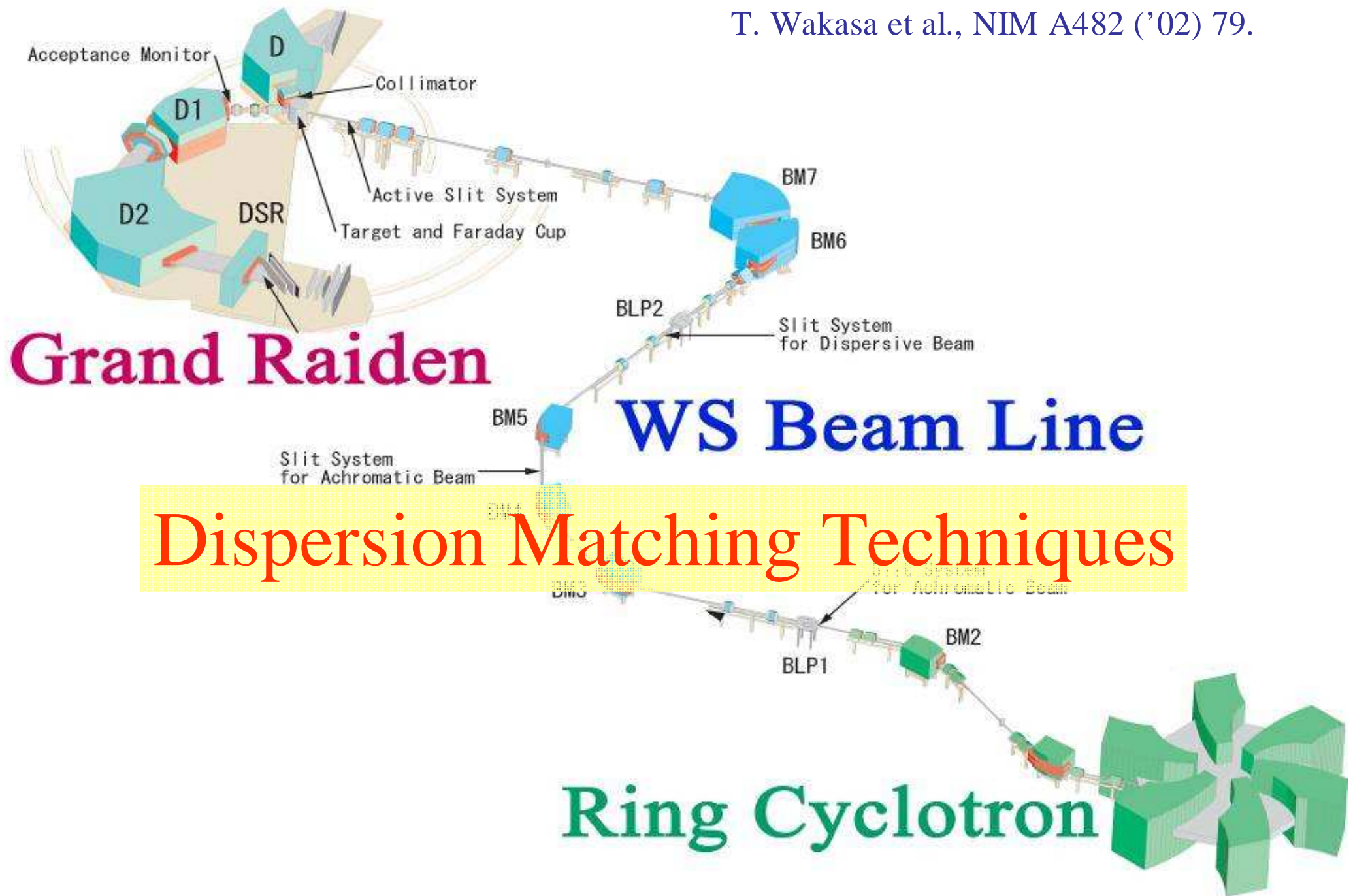
$(^3\text{He}, t)$ reaction

^3He beam

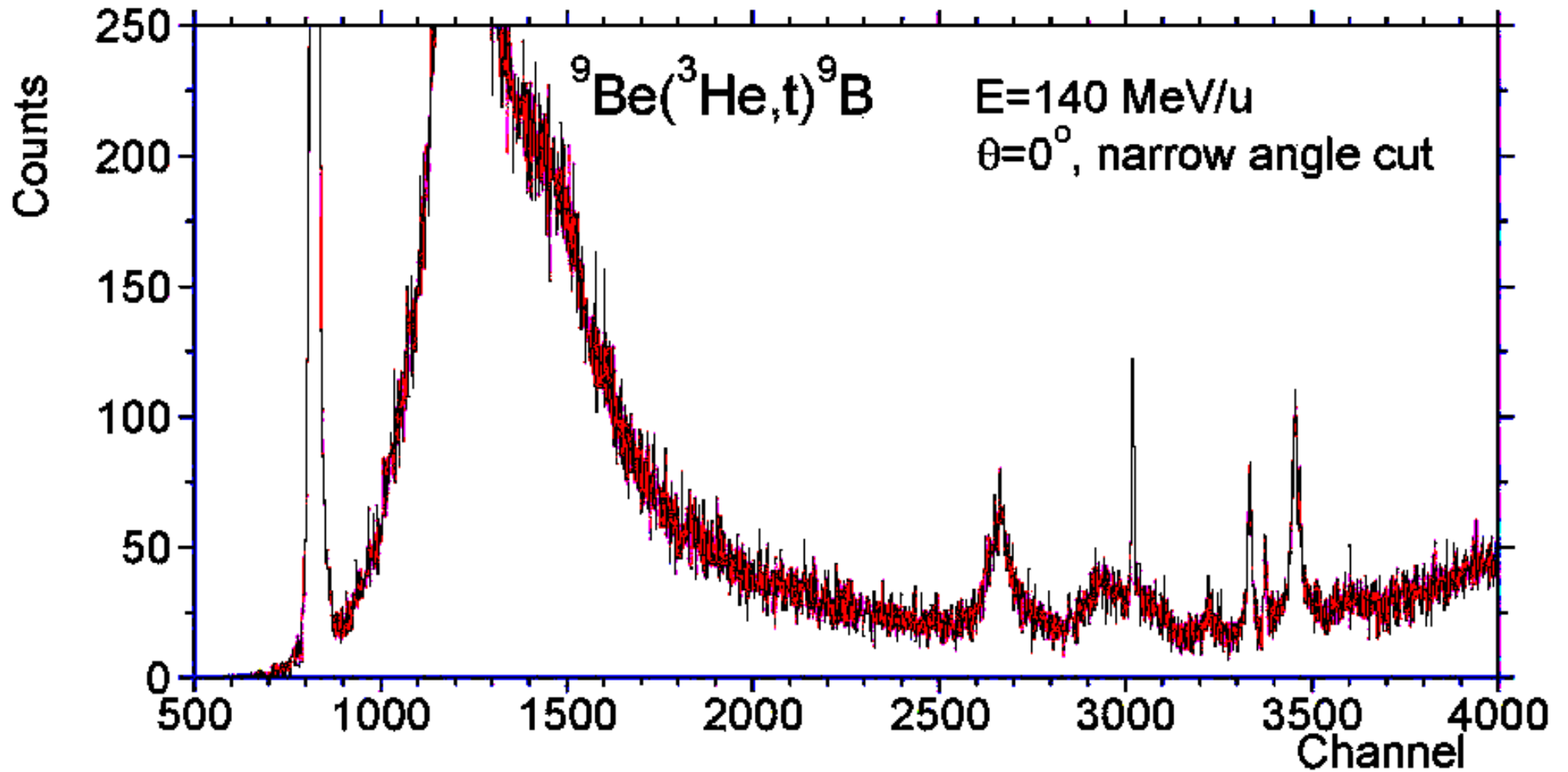


Beam line WS-course at RCNP

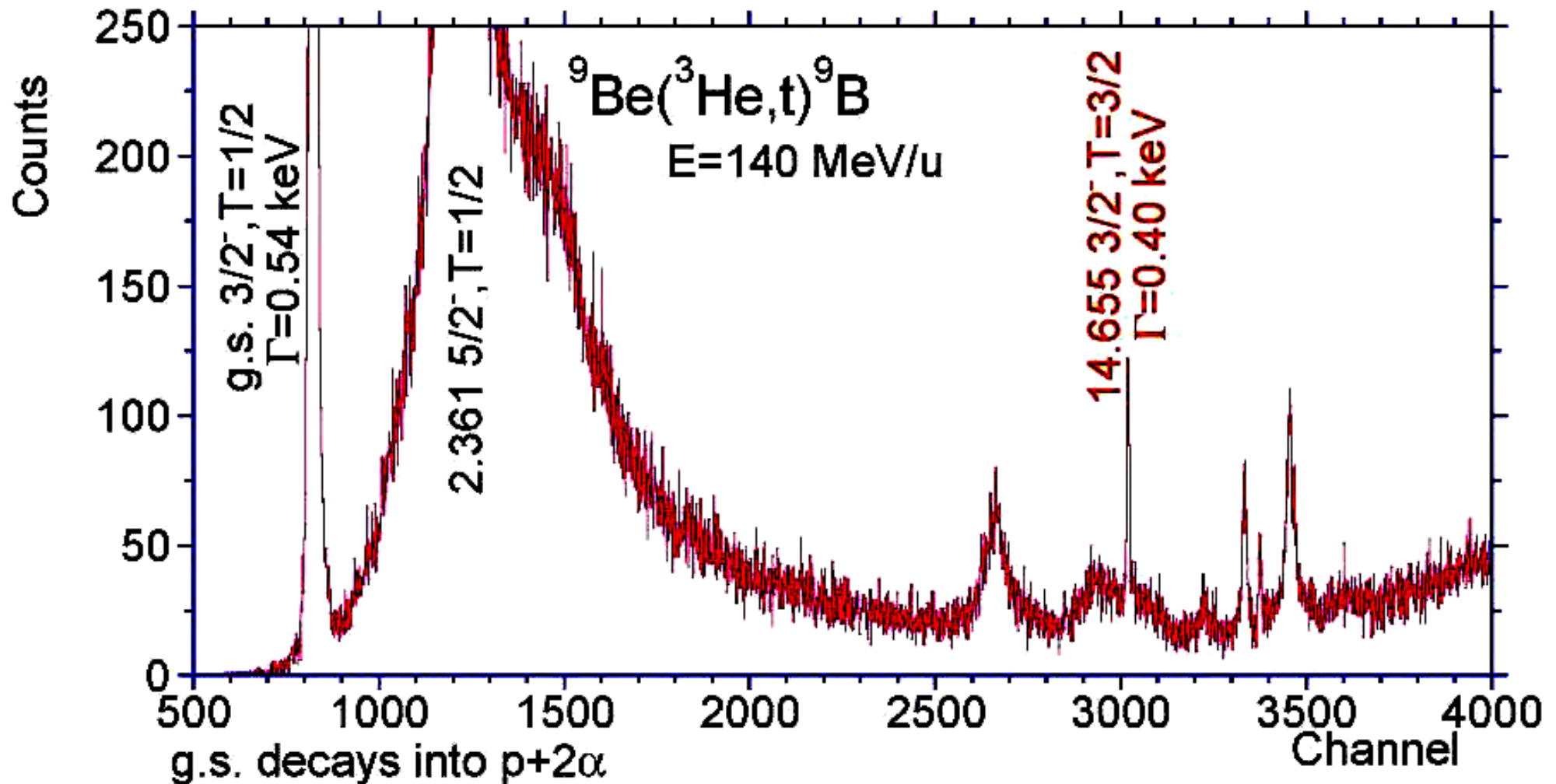
T. Wakasa et al., NIM A482 ('02) 79.



${}^9\text{Be}({}^3\text{He},t){}^9\text{B}$ spectrum (at various scales)



${}^9\text{Be}({}^3\text{He},t){}^9\text{B}$ spectrum (II)



**Isospin selection rule prohibits
proton decay of $T=3/2$ state!**

C. Scholl, Köln

References

- **Inorganic Mass Spectrometry: Principles and Applications, Sabine Becker, Wiley, 2007.**
- **Optics of Charged Particles, Hermann Wollnik, Academic Press 1987.**
- **Mass spectroscopy, H.E. Duckworth et al., Cambridge Univ. Press, 1986.**
- **The transport of charged particle beams, A.P. Banford, E. & F.N. Spon, 1966.**
- **Proceedings of the EMIS (Electromagnetic Isotope Separation) Conferences:
Nucl. Instr. Meth. B266, NIM B204, NIM B126, NIM B70, ...**